DISCOVERY OF DIRECT ORGANOCATALYTIC PUSH-PULL DIENAMINES: SCOPE AND SYNTHETIC APPLICATIONS

A Thesis Submitted for the Degree of

DOCTOR OF PHILOSOPHY

By

RAMAKUMAR KINTHADA



SCHOOL OF CHEMISTRY UNIVERSITY OF HYDERABAD HYDERABAD-500 046, INDIA

September 2010

DEDICATED TO AMMA AND NANNA

DECLARATION

I hereby declare that the entire work embodied in this thesis is the result of investigations carried out by me in the School of Chemistry, University of Hyderabad, Hyderabad, under the guidance of **Dr. Dhevalapally B. Ramachary** and that it has not been submitted elsewhere for any degree or diploma. In keeping with the general practice, due acknowledgements have been made wherever the work described is based on the findings of other investigators.

RAMAKUMAR KINTHADA
(Candidate)

CERTIFICATE

I hereby certify that the entire work embodied in this thesis has been carried out by Mr. Ramakumar Kinthada under my guidance in the School of Chemistry, University of Hyderabad, and that no part of it has been submitted elsewhere for any degree or diploma.

Dr. DHEVALAPALLY B. RAMACHARY
(THESIS SUPERVISOR)

DEAN
SCHOOL OF CHEMISTRY
UNIVERSITY OF HYDERABAD

CONTENTS

DECLA	ARATION	
CERTI	FICATE	
CONTE	ENTS	
ACKNO	OWLEDGEMENTS	
PREFA	CE	
LIST O	F ABBREVIATIONS	
Discove	ery of Direct Organocatalytic Push-Pull Dienamines:	
Scope a	and Synthetic Applications	
1	Abstract	1
2	Introduction	2
3	Sequential one-pot combination of multi-reactions	19
	through multi-catalysis: A general approach	
	to functionalized push-pull olefins and phenols	
4	Organocatalytic cascade reactions based on	58
	push-pull dienamine platform: Synthesis	
	of highly substituted Anilines	
5	Amino acid-catalyzed cascade [3+2]-cycloaddition/	79
	hydrolysis reactions based on the push-pull	
	dienamine platform: Synthesis of NH-1,2,3-Triazoles	
6	Direct organocatalytic asymmetric Michael	99
	reactions based on the push-pull dienamine	
	platform: synthesis of highly substituted	
	chiral Hagemann's esters	
7	References	119
8	Experimental Section	133
A	BOUT THE AUTHOR	xi
AI	DOUL THE AUTHOR	$\lambda \iota$

ACKNOWLEDGEMENTS

I am greatly indebted to **Dr. Dhevalapally B. Ramachary**, my research supervisor, for his inspiring guidance and constant encouragement throughout the course of the present investigations. It has been a great privilege and honour to be associated with him.

I thank present and former Dean of School of Chemistry, Prof. D. Basavaiah and Prof M. Periasamy and all the faculty members of the School of Chemistry for their timely help and cooperation during my M. Sc. & Ph. D. programs.

It is great privilege to express my heartfelt regards to my B. Sc. teachers Dr. Sarakadam Nookaraju and N. V. S. S. S. Sastry for their inspirational teaching, encouragement and motivation towards higher studies and also I am thankful to Padma madam, Jagan sir and also to all my teachers during the entire tenure of my educational carrier.

It gives me great pleasure to thank all my labmates Kishor, Narayana, Rumpa Mondal, Vijay, Sakthidevi, Venkaiah, Siva Prasad, Shashank, Srinivasa Reddy and Madhavachary for their friendly cooperation in the lab. My special thanks to Kishor, Narayana, Vijay and Shashank for their heartful support and helpful discussions.

I would like to express my heartful thanks to my friends Narasimharao and Srinivasarao for their caring and affection during my inter to B. Sc. studies and I extend my utmost gratitude to my friend Dr. Vikram for his invaluable moral support and encouragement.

I thank all my M. Sc. friends Ramu, Shekar, Phani, Dr. Satish kumar, Venu, Nagaraju, Sreedhar Reddy, D. S. Ramakrisha, Vishnu murthy, Sudheer, Basavaiah, Praveen, Dr. Hari, Dr. Venkatramana, Dr. Syamaroy, Innus, Nani, Dr. Baskar Reddy, Venkat Reddy, Dr. Rehman, Koti and Praveen.

I am very thankful to all my friends in School of Chemistry, Dr. Ramesh Reddy, Dr. Y. Srinivasarao, Dr. N. Vijaykumar, Dr. B. N. Murthy, Dr. Jayaprakash, T. K.

Chaitanya, D. K. Srinivas, Ramesh, Viji, Ramaraju, Sateesh, K. Srinu, Ganesh, Ravindrababu, Vanaja, Naveen, Manoj, Mallesh, Sanjeev, Guru, Laxman, Dr. Aravindu, Veeraraghavaiah, Leninbabu, Mallikarjunreddy, Ramasuresh, Anji, Nagarjuna, Ramesh, Gangadar, Arunbabu, Swamy, Dr. Bhuvan, Dr. Narahari, Rambabu, G. Durgaprasad, Bharat, Kishore, Kalyan, Anand, Kishor, Santosh, Ramesh, Dr. Rajesh, Gupta, Hari, Venu, Sheshadri, Suresh, Sekharreddy (junior), Kishore, Anand, Praveen, Haneesh, Karunakar, Srinu, Nandakishore, Dr. Biju, Apparao, Ravikumar, Suresh, Lakshminarayana, Srikant (IICT), Apparao, Ungati Harinarayana, Rajesh, Rajabhashkar, D. Siva, Manojkumar, Rakesh, Praveen, Ajay and Chalapati.

I would like to express my special thanks to Nabakamal nath, Sivaranjan Reddy, Ranjit, Dr. Utpaldas, Dr. Suresh, Dr. Narsi Reddy, Dr. Tejendarthakur, Dr. Jagadeesh, Dr. Raju and Muzahid for their helpful discussions and help throughout the course.

I am so thankful to my childhood friends Simhachalamnaidu, Srinivas Kannamnaidu, Rammohanreddy, Umavasu and Y. V. Ganesh. I am also thankful to my friends particularly Gallaraju (IICT), Ramachandramurthy (IICT), M. Suriapparao (IICT), Baskar(AKP), V. Ramana, Ratnakar, Pandu, V. jagannadham, Ayyappa, O. V. K. Prakash, Nakkasreenu, Brahamaji, Durga Prasad and I. B. Raju.

I am thankful to my family friends especially M. Chittibabu for his lovely affection, A. Satyavati, P. Sinu (Valluru), G. Jayaram, A. V. Ramesh, T. V. Ramana, V. Rajni, K. V. Ramanauncle and K. Suresh and to my beloved sister R. Suneeta.

I express my deepest affection to my mother & father for their support, encouragement and deserves great appreciation. I would like to express my sincere gratitude to my brother veerababu for his constant support, timely help and for always being there for me. The unconditional love of my parents and their blessings made me what I am today and I owe everything to them. Dedicating this thesis to them is a minor recognition for their relentless support and love. I would like to extend my utmost gratitude to my family members Nayanamma, Ammamma, Tatayya, Pedamaya,

Raghavaatta, Raghavamaya, Venkatalakshmi akka, Appalaraju maya, Lakshmi akka, Varalakshmi pinni, Rambabu cinnannagaru, Buchhamma atta, Ramarao uncle, Cinnaatta, Ramakrishna uncle, Rambabumaya, Manjuakka, Venu pinni, Ramadevi, Chinnababu, Asha, Sai, Ramchandu, Ramyata, Veerendra, Triveni, Vasavi and Asish.

I acknowledge the help and support provided by the technical and non-teaching staff of the School of Chemistry. I also thank Dr. P. Raghavaiah, Satyanarayana, Mallaiah Shetty, Bhaskara Rao, Venkataramana, Asiaperwej, Vijaya Laxmi, A. R. Setty, Jayaram, Prasad, Rangaiah and Durgesh for their timely help.

National Single Crystal X-ray Facility, funded by DST (New Delhi) in School of Chemistry is highly acknowledged. I am thankful to IGM library for providing excellent books and journals. Financial assistance by CSIR (New Delhi) is gratefully acknowledged.

Ramakumar Kinthada

PREFACE

Nature is the inspiration to develop a new area of catalysis "Organocatalysis" to synthesize stereochemically complex molecules. In this broad area of organocatalysis mainly enamine catalysis, the catalysis of carbonyl transformations via enamine intermediates by using primary and secondary amines as catalysts is fully appreciated as a powerful synthetic tool to design and implementation of new reactions for construction of highly functionalized molecules with good selectivity. The present thesis entitled "Discovery of Direct Organocatalytic Push-Pull Dienamines: Scope and Synthetic Applications" describes the reactions involving push-pull dienamine intermediates in cascade reactions and synthesis of highly functionalized molecules. In all sections, a brief introduction is provided to keep the present work in proper perspective, the compounds are sequentially numbered (bold), and references are marked sequentially as superscript and listed at the end of the thesis. All the figures included in the thesis were obtained by DIRECT PHOTOCOPY OF THE ORIGINAL SPECTRA, and in some of them uninformative areas have been cut to save the space.

Highly functionalized molecules, which are widely used as intermediates in the pharmaceuticals and natural product synthesis. To construct such functionalized molecules a diversity-oriented general, sustainable and practical process for the sequential cascade synthesis is required. Here we discovered direct sequential one-pot combination of amine- or amino acid-catalyzed cascade Knoevenagel/Michael/aldol condensation/decarboxylation with other reactions like amine- or amino acid-catalyzed cascade Claisen-Schmidt/iso-aromatization, Claisen-Schmidt/isomerization, Claisen-Schmidt/isomerization, Claisen-Schmidt/isomerization/isomerization, Michael addition, Claisen-Schmidt/Michael reactions of alkyl acetoacetates, variety of aldehydes furnished the highly functionalized push-pull olefins and phenols with high yields. Evidence for a new reaction pathway involving formation of novel push-pull dienamines under amine- or amino acid-catalysis is presented along with examples demonstrating the amenability of the process to multi-catalysis cascade (MCC) chemistry.

Highly functionalized diverse o-hydroxydiarylamine compounds, which are having wide applications and considerable importance in a variety of industries. To construct such complex molecules a diversity-oriented synthesis is required. We reported a practical and

novel one-pot organocatalytic selective process for the cascade synthesis of highly substituted o-hydroxydiarylamines and o-pyrrolidin-1-yldiarylamines. Here we achieved using simple starting materials such as alkyl acetoacetates, aldehydes, and nitrosoarenes furnished the highly functionalized anilines with high yields through direct combination of amine-catalyzed cascade Knoevenagel/Michael/aldol condensation/decarboxylation and cascade enamine amination/isoaromatization. Along with we demonstrated the bio-mimetic auto- and self-catalysis in amine-catalyzed cascade reactions.

In continuation to the organocatalytic cascade reactions based on push-pull dienamine platform highly substituted NH-1,2,3-triazole products and unhydrolyzed 1,2,3-triazoles were synthesized from simple starting materials such as Hagemann's esters and ptoluenesulfonyl azide (TsN₃) in the presence of catalytic amount of L-proline. For the first time through cascade [3+2]-cycloaddition/hydrolysis via a practical and environmentally friendly amino acid catalyzed cascade process was developed for the synthesis of highly substituted NH-1,2,3-triazole products and unhydrolyzed 1,2,3-triazoles. The cascade reaction proceeds in good yields with high selectivity using L-proline as the catalyst. Furthermore, we demonstrated the bio-mimetic solvent induced hydrolysis in amino acid-catalyzed cascade reactions.

In a similar manner, a practical and novel organocatalytic chemo- and enantioselective process for the cascade synthesis of highly substituted 2-alkyl-3-(2-nitro-1-aryl-ethyl)-4-oxo-cyclohex-2-enecarboxylic acid alkyl esters is presented using novel chiral push-pull dienamines. Here we described the synthesis of the L-(3,5-Me₂)₂DPP catalyzed asymmetric Michael reactions of Hagemann's esters with nitroolefins at ambient conditions. This novel asymmetric Michael reaction proceeds in good yields with high selectivity. Along with we demonstrated the synthesis of highly substituted cyclohexenones from chiral Michael adducts using simple decarboxylation as a key step.

LIST OF ABBREVIATIONS

Ac acetyl AcOH acetic acid Anal. analysis Aqueous aq. Ār aryl benzyl Bn boiling point Bp br broad Bu butyl

t-Bu or ^tBu tertiary-butyl
Bz benzoyl
Calcd. calculated
cat. catalytic
cm centimeter

DABCO 1,4-diazabicyclo(2.2.2)octane
DBU 1,8-diazabicyclo(5.4.0)undec-7-ene
DCC 1,3-dicyclohexylcarbodiimide

DCE 1,2-dichloroethane
DCM dichloromethane
DCU dicyclohexyl urea
dd doublet of doublet
de diastereomeric excess

DEPT distortionless enhancement by polarization transfer

DMAP dimethylaminopyridine **DME** dimethoxy ethane **DMF** *N*,*N*-dimethylformamide dimethyl sulfoxide **DMSO** DPP diphenyl prolinol diastereomeric ratio dr dt doublet of triplet enantiomeric excess ee

eq. equation equiv. equivalent(s)

Et ethyl

EWG electron withdrawing group

Fig. figure gm gram (s) ĥ hour (s) Hz hertz Hex hexyl i Pr isopropyl IR infrared kilocalories Kcal

LAH lithium aluminum hydride

lit. literature multiplet

M molarity melting point Mp. Me methyl

milligram (s) mg milliliter mĹ mmol millimole

NMM *N*-methylmorpholine nuclear magnetic resonance **NMR**

N-methylpyrrolidine NMP

phenyl Ph

parts per million ppm *p*-toluenesulfonic acid p-TSA

pyridine рy propyl pr q RT quartet

room temperature

singlet S secondary sec triplet

TCA trichloro aceticacid

tert tertiary

TFA trifluoroacetic acid tetrahydrofuran **THF**

thin layer chromatography TLC

trimethylsilyl **TMS** toluenesulphonyl Ts methanesulphonyl Ms

ABOUT THE AUTHOR

The author, **Mr. Ramakumar Kinthada** was born on 16th August 1982 at Chodavaram, Vishakapatnam Dist, Andhra Pradesh. After his initial schooling in Appalaraju puram, Vishakahapatnam (Dist.) he obtained his B. Sc. degree in 2002 from A. M. A. L. College, Anakapalle; and he obtained his M. Sc. degree in 2004 from University of Hyderabad, Hyderabad. He continued as research scholar in the School of Chemistry, University of Hyderabad for the Ph. D. programme from March 2005 onwards. Presently he is working as a research associate in the department.

LIST OF PUBLICATIONS

- D. B. Ramachary, K. Ramakumar and M. Kishor, Direct organocatalytic in situ generation of novel push–pull dienamines: application in tandem Claisen– Schmidt/iso-aromatization reactions, *Tetrahedron Lett.* 2005, 46, 7037–7042.
- 2. D. B. Ramachary, M. Kishor and K. Ramakumar, A novel and green protocol for two-carbon homologation: a direct amino acid/K₂CO₃-catalyzed four-component reaction of aldehydes, active methylenes, Hantzsch esters and alkyl halides, *Tetrahedron Lett.* 2006, 47, 651–656.
- 3. D. B. Ramachary, **K. Ramakumar** and V. V. Narayana, Organocatalytic cascade reactions based on push-pull dienamine platform: synthesis of highly substituted anilines, *J. Org. Chem.* 2007, 72, 1458–1463.
- 4. D. B. Ramachary, V. V. Narayana and K. Ramakumar, Direct ionic liquid promoted organocatalyzed diazo-transfer reactions: diversity-oriented synthesis of diazo-compounds, *Tetrahedron Lett.* 2008, 49, 2704–2709.
- 5. D. B. Ramachary, V. V. Narayana and K. Ramakumar, A new one-pot synthetic approach to the highly functionalized (Z)-2-(buta-1,3-dienyl)phenols and 2-methyl-2H-chromenes: use of amine, Ruthenium and base-catalysis, Eur. J. Org. Chem. 2008, 3907–3911.

- 6. D. B. Ramachary, K. Ramakumar and V. V. Narayana, Amino acid-catalyzed cascade [3+2]-cycloaddition/hydrolysis reactions based on the push–pull dienamine platform: synthesis of highly functionalized NH-1,2,3-triazoles, Chem. Eur. J. 2008, 14, 9143–9147.
- 7. D. B. Ramachary, V. V. Narayana, M. S. Prasad and K. Ramakumar, High-yielding synthesis of Nefopam analogues (functionalized benzoxazocines) by sequential one-pot cascade operations, *Org. Biomol. Chem.* 2009, 7, 3372–3378.
- 8. D. B. Ramachary, K. Ramakumar, A. B. Shashank and V. V. Narayana, Sequential one-pot combination of multi-reactions through multi-catalysis: a general approach to rapid assembly of functionalized push-pull olefins, phenols and 2-methyl-2*H*-chromenes, *J. Comb. Chem.* 2010, *12*, 0000–0000.
- 9. D. B. Ramachary and K. Ramakumar, Direct organocatalytic asymmetric Michael reactions based on the push-pull dienamine platform: synthesis of highly substituted chiral Hagemann's esters (communicated).

POSTERS AND PRESENTATIONS

- 1. Presented a poster entitled "Amino acid-catalyzed cascade [3+2]-cycloaddition/hydrolysis reactions based on the push–pull dienamine platform: synthesis of highly functionalized *NH*-1,2,3-triazoles" in 6th in-house symposium "Chemfest-2009" held at University of Hyderabad, Hyderabad, India on March 7-9, 2009.
- 2. Given a flash oral presentation entitled "Amino acid-catalyzed cascade [3+2]-cycloaddition/hydrolysis reactions based on the push–pull dienamine platform: synthesis of highly functionalized *NH*-1,2,3-triazoles" in 6th in-house symposium "Chemfest-2009" held at University of Hyderabad, Hyderabad, India on March 7-9, 2009.

DISCOVERY OF DIRECT ORGANOCATALYTIC PUSH-PULL DIENAMINES: SCOPE AND SYNTHETIC APPLICATIONS

1. ABSTRACT

A general, sustainable and practical process for the sequential cascade one-pot synthesis of highly substituted push-pull olefins and phenols was reported through multi-catalysis cascade (MCC) reactions. Direct sequential one-pot combination of amine- or amino acid-catalyzed cascade Knoevenagel/Michael/aldol condensation/decarboxylation with other reactions like amine- or amino acid-catalyzed cascade Claisen-Schmidt/iso-aromatization, Claisen-Schmidt/isomerization, Claisen-Schmidt/iso-aromatization/isomerization, Michael addition, Claisen-Schmidt/Michael, reactions of alkyl acetoacetates, variety of aldehydes furnished the highly functionalized push-pull olefins and phenols with high yields. The yields and regioselectivities were good to excellent. Evidence for a new reaction pathway involving formation of novel push-pull dienamines under amine- or amino acid-catalysis is presented along with examples demonstrating the amenability of the process to multi-catalysis cascade (MCC) chemistry.

A practical and novel one-pot organocatalytic selective process for the cascade synthesis of highly substituted *o*-hydroxydiarylamines and *o*-pyrrolidin-1-

yldiarylamines is reported. Direct combination of amine-catalyzed cascade Knoevenagel/Michael/aldol condensation/decarboxylation and cascade enamine amination/isoaromatization of alkyl acetoacetates, aldehydes, and nitrosoarenes furnished the highly functionalized anilines with high yields. Furthermore, we demonstrated the bio-mimetic auto- and self-catalysis in amine-catalyzed cascade reactions.

A practical and environmentally friendly amino acid catalyzed cascade process for the synthesis of highly substituted NH-1,2,3-triazole products and unhydrolyzed 1,2,3-triazoles was achieved for the first time through cascade [3+2]-cycloaddition/hydrolysis from simple starting materials Hagemann's esters and p-toluenesulfonyl azide (TsN₃) in the presence of catalytic amount of proline. The cascade reaction proceeds in good yields with high selectivity using proline as the catalyst. Furthermore, we demonstrated the bio-mimetic solvent induced hydrolysis in amino acid-catalyzed cascade reactions.

A practical and novel organocatalytic chemo- and enantioselective process for the synthesis of highly substituted 2-alkyl-3-(2-nitro-1-aryl-ethyl)-4-oxo-cyclohex-2-enecarboxylic acid alkyl esters is reported. Here we described the synthesis of L-(3,5-Me₂)₂DPP catalyzed asymmetric Michael reactions of Hagemann's esters with nitroolefins at ambient conditions. This novel asymmetric Michael reaction proceeds in good yields with high selectivity using L-(3,5-Me₂)₂DPP as a catalyst. Furthermore, we demonstrated the application of chiral Michael products in the synthesis of highly functionalized cyclohexenones.

2. Introduction

The selective carbon-carbon bond formation is the basic tool for the construction of the molecular frameworks of organic molecules by synthesis. One of the fundamental processes for C–C bond formation is reactions involving enamine intermediates.

Enamines are the intermediates, generated from the reactions of carbonyl compounds with primary or secondary amines as nucleophiles and these enamines have long been recognised as key intermediates in organic synthesis.

Enamines are genarally used as preformed intermediates for acylation and alkylation of carbonyl compounds. The advantages of enamine reaction over the base-induced alkylation/acylation is no base or other catalyst is required so that it avoids self-condensation reactions of the carbonyl compound and another valuable feature is that it is regioselective. Based on the high regiochemical outcome of the alkylation and acylation reactions from enamine reactions, drives synthetic chemists paying attention to chemistry of enamines. The chemistry of enamines especially their use as nucleophiles in organic synthesis has been investigating since early 1950's.

As the research work described in this thesis deals with reactions involving push-pull dienamines, a brief overview of the reactions involving enamine intermediates are presented below.

Reactions with enamines was first reported by Strok *et al.* in 1954. In their landmark communication, they have used preformed enamines for the alkylation and acylation of ketones. In this new methodology, pyrrolidine enamine 3 generated from cyclohexanone 1 and pyrrolidine 2a reacts with methyl iodide in boiling methanol to yield methylated cyclohexanone 4, and reacts with acrylonitrile and methyl acrylate in boiling dioxane to furnish functionalized cyclohexanones 5 with very good yield as shown in eq. 2.²

In view of further scrutinizing the enamine reactions, pyrrolidine enamine 7 generated from 6 and 2a, was treated with acrylonitrile and methyl acrylate in boiling dioxane for 16 hours to furnish functionalized cyclopentanones 8 with good yields as shown in eq. 3. This enamine reaction is an alternative tool for classical base induced acylation and alkylation of carbonyl compounds.

In a continuation of development of chemistry of enamines, Woodward used enamine reaction in a keystep for the synthesis of vitamin- B_{12} . Total synthesis of vitamin- B_{12} is one of the most important and complicated naturally occurring molecule. One among the achievements of Woodward in the field of organic synthesis is the total synthesis of vitamin- B_{12} . In the course of synthesis of vitamin B_{12} functionalized cyclohexenone 11 was synthesized from compound 10 by pyrrolidine acetate in methanol as shown in eq. 4.³ However here the cyclization is catalyzed by pyrrolidine acetate and in this case the reaction is initiated by enamine formation at the less hindered side of the two carbonyl groups. This clearly shows the Woodward imagination of the potentiality of enamine catalysis.

With inspiration from Stork's work, in 1971 Wiechert *et al.* reported the intramolecular asymmetric aldol cyclodehydration of the achiral trione **13** using combination of amino acid and inorganic acid. In this communication they reported synthesis of optically active 7a-methyl-5,6,7,7a-tetrahydro-1,5-indandione **15** which is important starting material for total synthesis of steroids.⁴

This exceptionally elegant approach has also been applied to synthesize optically active Wieland-Miescher ketone **17** with good yield and excellent enantioselectivity as shown in eq. 6.⁴ These are very useful building blocks for construction of a broad variety of biologically active compounds such as steroids, terpenoids and taxol.

In 1974, Hajos-Parrish discovered the proline catalyzed intramolecular asymmetric aldol cyclodehydration reaction. The investigations by the Hajos group, which included optimization of the reaction conditions, provided detailed insight into the aldol cyclization step and formation of the important intermediate **18**. The best yield and enantioselectivity were obtained when a polar, aprotic solvent was used. In DMF the aldol cyclization of **13** into the ketol intermediate **18** proceeded in quantitative yield and with high enantiomeric excess 93.4% as shown in eq. 7.⁵ It is worthy of note that a small amount of L-proline is sufficient for effective catalysis under these conditions and subsequent dehydration of the ketol **18** gave the enone **15** in 99% yield with enantioselectivity 87.7%.

This approach has also attracted commercial attention because the use of economically attractive catalyst L-proline and also been applied to synthesize optically active Wieland-Miescher ketone 17 which can be used for easy access to the steroid precursors. They revealed that the carboxylic acid functionality and the pyrrolidine ring of proline both were essential for efficient asymmetric induction and this is one of the earliest example for organocatalysis through enamine intermediate. Although this is the first example of an asymmetric intramolecular aldol reaction using an organic molecule, L-proline as chiral catalyst, but neither its mechanism nor its potential for other reactions were realized.

In 1995, Barbas and Lerner co-workers reported the antibodies catalyzed asymmetric aldol reactions. They understand mechanism for this antibody catalysis of the reaction mimics as natural class I aldolase enzymes. Immunization with a reactive compound covalently trapped a lysine residue in the binding pocket of the antibody by formation of a stable enamine. The reaction mechanism for the formation of the covalent antibody-hapten complex was recruited to catalyze the aldol reaction. The antibody 33F12 used the amino group of lysine to form an enamine with acetone and used this enamine as a nascent carbon nucleophile to attack the specially designed aldehyde **20** which gives aldol product **21** by forming a new carbon-carbon bond as shown in eq. 9.⁶

With inspiration from nature's mechanism another report from Barbas group in 1996, is the decarboxylation of β -keto acids using catalytic antibodies via enamine intermediates. They have studied antibodies ability to catalyze the decarboxylation of structurally related β -keto acids. Both 38C2 and 33F12 aldolase antibodies were shown to efficiently catalyze the decarboxylation. Their studies support the role of an essential lysine residue in the active site of the antibodies and the formation of enamine intermediate in the mechanism. Investigation of the decarboxylation reaction of 2-{3'-(4''-acetamidophenyl)propyl}-acetoacetic acid **22**, to 6-(4'-acetamidophenyl)-2-

hexanone **25**, in the presence of ¹⁸O-labeled water by electrospray mass spectrometry revealed incorporation of ¹⁸O in the antibody-catalyzed reaction consistent with decarboxylation proceeding via an enamine mechanism as shown in eq. 10.⁷

After successful understanding and demonstration of enamine mechanism in antibody catalysis, in 1997 Barbas and Lerner co-workers reported the synthesis of optically active Wieland-Miescher ketone 17 by using biocatalyst (Ab 38C2) as shown in eq. 11.8 This study drives their interest in organocatalysis, and their comparative studies of aldolase antibodies with L-proline reveals that Hajos-Eder-Sauer-Wiechert reaction proceeds via an enamine reaction mechanism like aldolase antibodies. Mechanistically catalysis with antibody aldolases and the simple amino acid proline are very similar. This study served for launching their studies in organocatalysis.

For further understanding of comparative studies of aldolase antibodies with L-proline, Barbas and Lerner co-workers in 1999, reported that aldolase antibody 38C2 catalyzes the enantioselective aldol cyclodehydration of 4-substituted-2,6-

heptanediones **26** to give enantiomerically enriched 5-substituted-3-methyl-2-cyclohexen-1-ones **27** as shown in eq. 12. Enantioselectivities, and product purities are markedly similar to the L-proline catalyzed reactions.

In 2000, Barbas and his co-workers rediscovered the catalytic properties of L-proline by presenting the first enantioselective organocatalytic intermolecular aldol reaction of acetone with 4-nitrobenzaldehyde **28a** at room temperature for 4 hours furnished aldol product (*R*)-**30** in 68% yield and 76% ee with proposed transition state **29** as shown in eq. 13. Following this inspiring report, numerous innovative and increasingly sophisticated examples of the use of aminocatalysis have emerged as an explosive expansion of the field of "organocatalysis" thereby producing a "nearly endless" number of optically active building blocks by applying these ideas.

In 2001, Barbas *et al.* described another organocatalytic asymmetric Mannich type reaction. In this report the proline catalyzed reaction of *N*-PMP-protected α -imino ethyl glyoxylate **32a** with heptanal **31** furnished highly enantioselective functionalized protected amino acid **34** with good yield and excellent diastereoselectivity as shown in eq. 14.¹¹ This is the first report of unmodified aldehydes used as donors in catalytic asymmetric Mannich-type reactions and this reaction is an atom-economic.

After an initial understanding of enamine catalysis, in 2002 Ramachary and Barbas discovered the reactions involving *in situ* generation of dienamine intermediates named as 2-amino-1,3-diene from α,β -unsaturated ketones and catalytic amount of amines. They demonstrated for the first time amine-catalyzed direct Diels-Alder reactions of α,β -unsaturated ketone **36a** with dienophile **35a** via transition state **37** to provide cyclohexanone derivative **38** as shown in eq. 15. This dienamine catalysis has emerged as a powerful synthetic paradigm and has accelerated the development of new methods involving dienamine intermediates to make diverse chiral molecules.

In the same year Ramachary and Barbas demonstrated, for the first time amine-catalyzed self Diels-Alder (or double Michael) reaction of α,β -unsaturated ketone **36b** providing the synthesis of pro-chiral acyl-substituted cyclohexanone **39** as shown in eq. 16.¹³ These new catalytic Diels-Alder reactions afforded an attractive single-step route to functionlized cyclohexanone derivatives.

In continuation of their quest in organocatalytic assembly of multicomponent reactions, in 2003 same group reported, the first organocatalytic diastereospecific and enantioselective direct asymmetric domino Knoevenagel/Diels-Alder reaction that produced highly substituted spiro[5,5]undecane-1,5,9-trione 42 via proposed transition state 41 from commercially available 4-substituted-3- buten-2-one 36c, aldehyde 28a and 2,2-dimethyl-1,3-dioxane-4,6-dione (Meldrum's acid) 40 as shown in eq. 17.¹⁴ Spirocyclic ketones 42 are attractive intermediates in the synthesis of natural products and in medicinal chemistry, and are the starting materials for the synthesis of exotic amino acids which are used to modify the physical properties and biological activities of peptides, peptidomimetics and proteins.

In 2003, Ramachary and Barbas reported the first highly diastereoselective organocatalytic direct hetero-domino Knoevenagel-Diels-Alder-epimerization (K-DA-E) reactions that provide highly substituted prochiral spiro[cyclohexane-1,2'-indan]-1',3',4-trione **44a** from commercially available 4-substituted-3-buten-2-one **36d**, aldehyde **28b** and 1,3-indandione **43** as shown in eq. 18.¹⁵ Spirocyclic ketones **44a** are

the excellent starting materials for the synthesis of fenestranes and for the construction of graphite cuttings bearing a saddle-like, three-dimensionally distorted cores.

Following these inspiring "dienamine" mediated reports, in 2006 Jørgenson *et al.* reported organocatalytic enantioselective γ -functionalization of α,β -unsaturated aldehyde **45a** with diethyl azodicarboxylate **46** via transition state **48** to provide **49** with moderate yield and high enantioselectivity as shown in eq. 19.¹⁶ They discovered the catalytic amount of secondary amines can invert the usual reactivity of, α,β -unsaturated aldehydes enabling a direct γ -amination of the carbonyl compound using azodicarboxylate as the electrophilic nitrogen-source.

In 2006, Watanabe *et al.* demonstrated the use of proline to promote asymmetric self-condensation of α , β -unsaturated aldehyde **45b** to form trisubstituted cyclohexadiene **52** via diene **50** with good yield and moderate enantioselectivity as shown in eq. 20.¹⁷

This approach will allow diversification synthesis of these cyclohexadiene ring-fused homodimers in sufficient quantities for biological investigations.

In 2007, Barbas and his co-worker's reported highly enantiomerically enriched aza-Morita-Baylis-Hillman (aza-MBH) type products with β -substituted enal moiety **45c** and *N*-PMP-protected α -imino *iso*-propyl glyoxylate **32b** with transition state **53** to furnish functionalized amino acid **54** with good yield and high enantioselectivity as shown in eq. 21.¹⁸ They reported that this reaction proceeds through dienamine intermediate but not through a typical aza-MBH reaction route.

In 2007, Yamamoto *et al.* documented complete regioselective and efficient enantioselective nitroso Diels-Alder reaction by utilizing *in situ*-generated dienamines. The α , β -unsaturated ketone **55** with catalytic amount of **57** *in situ*-generates dienamine which employed as a diene precursor with nitrosobenzene **56a** to provide the cycloadduct **59** in moderate yield with good enantioselectivity as shown in eq. 22.

In 2007, Hong *et al.* documented highly enantioselective organocatalytic condensation of α,β -unsaturated aldehydes via formal [4 + 2] cycloaddition. They reported asymmetric condensation of α,β -unsaturated aldehydes **45d** and **45e** to form cyclohexadiene **62** by the formal [4 + 2] cycloaddition through *in situ* generated dienamine **60** as diene. This methodology allows the reaction to proceed at low temperature with moderate yield and higher enantioselectivity as shown in eq. 23.²⁰

In 2008, Ramachary *et al.* reported double cascade reactions based on the Barbas dienamine platform for the synthesis of highly stereoselective functionalized cyclohexanes for cardiovascular agents. They reported proline catalyzed three- and five-component cascade olefination-Diels-Alder-epimerization and olefination-Diels-Alder-

epimerization-olefination-hydrogenation reactions of readily available precursor enone **36d**, benzaldehyde **28b**, methyl cyanoacetate **63** and Hantzschester **67** to furnish highly substituted prochiral 1-cyano-4-oxo-2,6-diphenyl-cyclohexanecarboxylic acid methyl ester **66** and 1-cyano-4-(cyano-methoxycarbonyl-methyl)-2,6-diphenyl-cyclohexanecarboxylic acid methyl ester **68** in a highly diastereoselective fashion with excellent yields as shown in eq. 24.²¹ Prochiral *cis*-isomers **66** are excellent starting materials for the synthesis of cardiovascular agents and hypnotic active products.

In a similar time, Christmann *et al.* reported the synthesis of novel mono- and bicyclic scaffolds by γ -activation of tethered unsaturated dicarbonyl precursors. Herein, they used diphenylprolinol TMS ether to generate chiral electron-rich dienamine as a diene, which in turn direct the facial approach of α,β -unsaturated carbonyl moiety to furnish cycloadduct **71** which is converted to **72** by simple reduction with good yield and excellent enantioselectivity as shown in eq. 25.²²

In 2009, Tu *et al.* documented auto-catalysis based on the Barbas dienamine platform, a new domino autocatalytic reaction of imine **73** with Meldrum's acid **40** to furnish polycyclic dispiro[4.2.5.2]pentadecane-9,13-dione **76** derivatives, with remarkable diastereoselectivity in acidic condition. In this reaction byproduct acetohydrazide **77**, plays a critical role in the success of the reaction by serving as a self-catalyst for the Diels-Alder reaction between arylidene-Meldrum's acid **75** and Barbas dienamine **74** as shown in eq. 26.²³

In a similar time, Chen *et al.* reported the highly regio- and stereoselective synthesis of functionalized chiral hemiaminal **80** with 91% yield and >99% enantiomeric excess from N-tosyl-1-aza-1,3-butadiene **78** as a diene and *in situ* generated dienamine **79** as a dienophile via inverse-electron-demand aza-Diels-Alder reaction as shown in eq. 27. The resulting densely functionalized enantiomerically

pure hemiaminal derivative **80** is useful in the total synthesis of natural products and medicinal chemistry.

In 2009, Melchiorre *et al.* reported synthesis of spirocyclic oxindoles having multiple stereocenters through Barbas dienamine Platform. In this report, Barbas dienamine intermediate **83** which is *in situ* generated from α,β -unsaturated ketone **36d** reacts with dienophile **81** to afford the spiro-oxindole derivative **84** with good yield and moderate diastereoselectivity with high enantioselectivity as shown in eq. 28.²⁵ This spirocyclic oxindole core is found in a number of natural products and as well as medicinally relevant compounds.

In a similar time, Chen *et al.* reported that first direct chemo and regioselective Michael addition of γ , γ -disubstituted α , β -unsaturated aldehydes to nitroolefins via dienamine catalysis. The α , β -unsaturated aldehyde **45g** *in-situ* generates dienamine **85** which on treatment with nitroolefin **35b** furnished Michael adduct **86** which is converted to **87** by simple reduction with good yield and excellent enantioselectivity as shown in eq. 29.²⁶ These products have synthetic significance in medicinal chemistry.

In continuation of synthesis of highly functionalized molecules starting from the simple materials in one-pot,²⁷ research work has been carried out on the synthesis of functionalized molecules in a single step, and the results are presented in this thesis.

To begin with, starting from simple starting materials, reactions involving pushpull dienamines were developed¹ for the one-pot synthesis of functionalized molecules and the results are presented in the next sections.

3. SEQUENTIAL ONE-POT COMBINATION OF MULTI-REACTIONS THROUGH MULTI-CATALYSIS: A GENERAL APPROACH TO RAPID ASSEMBLY OF FUNCTIONALIZED PUSH-PULL OLEFINS AND PHENOLS

3.1 INTRODUCTION

Critical objectives in modern synthetic organic chemistry include the catalytic asymmetric assembly of simple and readily available precursor molecules into stereochemically and electronically complex compounds under sustainable reaction conditions as mimicking cellular reactions. In this regard, the development of one-pot sequential combination of multi-catalysis and multi-component reaction methodologies can provide expedient access to complex products from simple starting materials. Recently, amine- or amino acid-catalysis (organocatalysis) has emerged as a promising sustainable synthetic tool for the constructing combination of C-C, C-N, C-O, C-S, C-P, C-F and/or C-H bonds in a single operation with high diastereo- and enantioselectivity in a cascade or multi-component process. Penally in organocatalysis, structurally simple and stable chiral organoamines and amino acids facilitate iminium- and enamine-based transformations with carbonyl compounds and are used as catalysts in operationally simple and environmentally friendly cascade reactions.

As part of our research program to engineer direct combination of organocatalytic multi-component and multi-catalysis reactions, ^{27d-1} herein we report the organocatalytic regioselective direct cascade Claisen-Schmidt/iso-aromatization (CS/IA), Claisen-Schmidt/isomerization (CS/IA), Claisen-Schmidt/isomerization (CS/IA/I), Claisen-Schmidt/Michael (CS/M) and Michael addition that produce highly substituted 2-arylidene or 2-alkylidene cyclohexanones

(push-pull olefins) **89**, highly substituted push-pull phenols **90**, functionalized aldehydes **91**, highly functionalized (E)-1,3-dienes **92**, functionalized bis-enones **93** from commercially available Hagemann's esters **88**, aldehydes **28** and amines or amino acids **2** or **14** as shown in Scheme 1. Push-pull olefins and phenols **89/90** are attractive intermediates in the synthesis of natural products and in medicinal chemistry³⁰ also in organic synthesis (see Chart 1). Hence, their economical and environmental friendly preparation has continued to attract considerable synthetic interest in developing new methods for their synthesis.³¹

Scheme 1: Direct Organocatalytic Sequential One-pot Cascade Reactions Based on the Push-Pull Dienamine Platform

We envisioned that an amine- or amino acid would catalyze the cascade Claisen-Schmidt condensation of variety of aldehydes **28** with *in situ* generated push-pull dienamine (1-amino-1,3-butadiene)¹ intermediate from Hagemann's esters **88** and amine/amino acid **2/14** to form substituted push-pull olefins (3-arylidene Hagemann's ester) **89** in a highly regioselective manner, which then undergoes iso-aromatization to produce substituted push-pull phenols **90**, isomerization to produce substituted (*E*)-1,3-

dienes **92** or Michael addition with **88** to produce substituted bis-enones **93** under base-catalysis based on the electronic nature of aldehydes **28**.

Chart 1: Some Natural/Non-natural Products and Pharmaceuticals Containing CascadeCompounds Obtained from Push-Pull Dienamine Chemistry

3.2 RESULTS AND DISCUSSION

3.2.1 Direct Amino acid-Catalyzed Claisen-Schmidt Condensation: Reaction Optimization: We were surprised to find that the reaction of Hagemann's ester **88a** and 4-nitrobenzaldehyde **28a** with a catalytic amount of L-proline **14a** in DMSO at 25 °C or 75 °C for 48 h or 24 h respectively, didn't furnished the expected

products **89aa** or **90aa** as shown in Table 1, entries 1 and 2. Interestingly, the same reaction catalyzed by phenylalanine **14c** at 25 °C for 96 h furnished the Claisen-Schmidt (CS) product **89aa** as a 1.25:1 mixture of E/Z isomers in 40% yield and without formation of cascade product **90aa** (Table 1, entry 3).

Table 1: Optimization of the Direct Amino acid Catalyzed Cascade Claisen-Schmidt and Iso-aromatization reaction of **88a** and **28a**^a

O CHO Catalyst O OH
$$(20 \text{ mol}\%)$$
 $R + CO_2Et$ NO_2 CO_2Et CO_2Et $R + CO_2Et$ $R + CO_2E$

entry	catalyst	temp (°C)	time (h)	yield (%) ^b 89aa	yield (%) ^b 90 aa
1 ^c	proline 14a	25	48	_	
2 ^c	proline 14a	75	24	_	_
3^{d}	phenylalanine 14c	25	96	40	_
4 ^d	glycine 14d	25	24	73	_
5 ^{d,e}	glycine 14d	25	96	55	-
6 ^e	pyrrolidine 2a /AcOH	25	12	_	30

^a Reactions were carried out in DMSO (0.5 M) with the same proportions of Hagemann's ester **88a** and aldehyde **28a** in the presence of 20 mol% catalyst. ^b Yield refers to the column-purified product. ^c 70 to 85% of unreacted Hagemann's ester **88a** was isolated. ^d 1.25:1 Mixture of *E/Z* isomers were isolated. ^e DMF was used as the solvent.

In a similar manner, reaction of **88a** with **28a** under glycine **14d**-catalysis in DMSO at 25 °C for 24 h furnished the CS product **89aa** as a 1.25:1 mixture of E/Z isomers in 73% yield and without cascade product **90aa** (Table 1, entry 4). But unfortunately, same

reaction took longer reaction time with lesser yield in DMF as solvent (Table 1, entry 5). Interestingly, the cascade reaction catalyzed by bifunctional catalyst, pyrrolidine 2a/AcOH at 25 °C for 12 h furnished the Claisen-Schmidt/iso-aromatization (CS/IA) product 90aa in 30% yield and without isolation of CS product 89aa as shown in Table 1, entry 6. The optimal reaction conditions for push-pull olefin synthesis involved glycine 14d-catalysis at 25 °C in DMSO with equimolar quantities of 88a and 28a, which furnished the CS product 89aa in 73% yield (Table 1, entry 4). The regiochemistry of products 89aa and 90aa was established by NMR analysis.

3.2.2 Direct Amine-Catalyzed Cascade Claisen-Schmidt/Iso-aromatization **Reaction Optimization:** Based on the result of Table 1, entry 6, we screened several pyrrolidine-based catalysts 2 by monitoring the reaction yield and regioselectivity of the cascade reaction of 88a and 28a in DMF (Table 2). In the cascade Claisen-Schmidt/isoaromatization (CS/IA) reaction of ester 88a and 4-nitrobenzaldehyde 28a catalyzed directly by amine 2a, we found that the solvent had a significant effect on the rates and yields (Table 2). The results of this investigation indicated that the cascade CS/IA reaction catalyzed by amine 2a produced the product 90aa with good yields in aprotic dipolar solvents like DMF and DMSO (Table 2, entries 1-13) but did not furnish products 89aa and 90aa in protic polar solvents (H₂O, MeOH, EtOH, CHCl₃, CH₃CN, CH₂Cl₂), the aprotic polar solvent (THF) and in the ionic liquid [bmim]BF₄ (results not presented in Table 2). The amine-induced Claisen-Schmidt condensation is strongly solvent-dependent reaction. The first step, the formation of the 1-amino-1,3-butadiene from the keto ester 88a and amine, and its addition to the carbonyl (or imine) group is facilitated in solvents of high polarity and the second step, 1,2-elimination, is inhibited by protic solvents. Thus, dipolar aprotic solvents such as DMF and DMSO are especially useful for amine-catalyzed Claisen-Schmidt condensations.¹

Table 2: Optimization of the Direct Amine Catalyzed Cascade Claisen-Schmidt and Iso-aromatization reaction of **88a** and **28a**^a

O CHO Catalyst O OH (20 mol%)
$$R + CO_2Et$$
 CO_2Et $CO_$

entry	catalyst	temp (°C)	time (h)	yield (%) ^b 89aa	yield (%) ^b 90 aa
1	pyrrolidine 2a	25	6	_	65
2	pyrrolidine 2a	25	12	-	75
3 ^c	pyrrolidine 2a	25	12	-	77
4 ^d	pyrrolidine 2a	25	2	60	5
5	pyrrolidine 2a	60	2	_	61
6 ^e	pyrrolidine 2a	25	23	-	65
7	piperidine 2b	25	12	_	61
8 ^f	morpholine 2c	25	12	_	_
9	benzylamine 2d	25	24	30	_
10 ^f	aniline 2e	25	24	-	_
11 ^f	DMAP 2f	25	96	-	_
12 ^f	triethylamine 2g	25	12	-	-
13 ^f	DBU 2h	25	12	-	-

^a Reactions were carried out in DMF (0.5 M) with the same proportions of Hagemann's ester **88a** and aldehyde **28a** in the presence of 20 mol% catalyst. ^b Yield refers to the column-purified product. ^c 1.25 equivalents of ester **88a** was used. ^d 1:1 mixture of *E/Z* isomers were isolated. ^e DMSO was used as the solvent. ^f 70 to 85% of unreacted Hagemann's ester **88a** was isolated.

The structurally simple pyrrolidine **2a** catalyzed the cascade CS/IA reaction of **88a** and **28a** to produce **90aa** in 65% yield at 25 °C for 6 h and extension of the reaction time to 12 h furnished the **90aa** in 75% yield (Table 2, entries 1 and 2). Interestingly, the same reaction performed at 25 °C for 2 h furnished the **89aa** in 60% yield with 1:1 mixture of *E/Z* isomers and **90aa** in 5% yield (Table 2, entry 4). At 60 °C, the same

reaction with pyrrolidine 2a furnished the cascade CS/IA product 90aa in 61% yield in reduced time (Table 2, entry 5). As discussed in above, cascade CS/IA reaction of 88a and 28a under 2a-catalysis in DMSO at 25 °C for 23 h furnished the product 90aa in only 65% yield as shown in Table 2, entry 6. Piperidine 2b also catalyzed the cascade CS/IA reaction in 61% yield, but there is no reaction under morpholine 2c-catalysis (Table 2, entries 7 and 8). Primary amines like benzylamine 2d catalyzed the CS reaction of 88a and 28a to furnish 89aa in 30% yield, but there is no reaction under aniline 2e-catalysis (Table 2, entries 9 and 10). The *tert*-amines like DMAP 2f, triethylamine 2g and DBU 2h did not catalyzed the cascade CS or CS/IA reaction, which is strong evidence for the intermediate enamine formation during these reactions. The optimal reaction conditions for push-pull phenol synthesis involved pyrrolidine 2a catalysis at 25 °C in DMF with equimolar quantities of 88a and 28a, which furnished the cascade CS/IA product 90aa in 75% yield (Table 2, entry 2).

3.2.3 *Direct* Diamine-Catalyzed Cascade Claisen-Schmidt/Isoaromatization Reaction Optimization: After successful demonstration of glycine- and pyrrolidine-catalyzed cascade CS and CS/IA reactions, we further screened diamine-based catalysts to increase the reaction yield and regioselectivity of the cascade reaction of 88a and 28a (Table 3). The structurally simple diamine, (S)-1-(2pyrrolidinyl-methyl)pyrrolidine 2i catalyzed the cascade CS/IA reaction of 88a and 28a at 25 °C for 7 h in DMF to produce **90aa** in 75% yield (Table 3, entry 1). Interestingly, the same reaction in DMSO at 25 °C for 7 h furnished the cascade CS/IA product **90aa** in 80% yield and which is better-optimized condition compared to glycine- and pyrrolidine-catalysis (Table 3, entry 2). Surprisingly, cascade CS/IA reaction of 88a and **28a** under (S)-1-pyrrolidin-2-ylmethyl-piperidine **2j**-catalysis in DMF or DMSO at 25 °C for 8 h furnished the product **90aa** in 62-66% yield, which is not superior compared to 2i-catalysis (Table 3, entries 3 and 4). The optimized conditions for the CS/IA reaction of 88a and 28a in DMSO at 25 °C for 7 h to furnish 90aa with excellent conversions and yields required the presence of the catalytic amount of (S)-1-(2-pyrrolidinyl-methyl)pyrrolidine **2i** (Table 3, entry 2).

Table 3: Optimization of the Direct Diamine-Catalyzed Cascade Claisen-Schmidt and Iso-aromatization reaction of **88a** and **28a**^a

O CHO Catalyst (20 mol%)

Solvent (0.5 M)

Solvent (0.5 M)

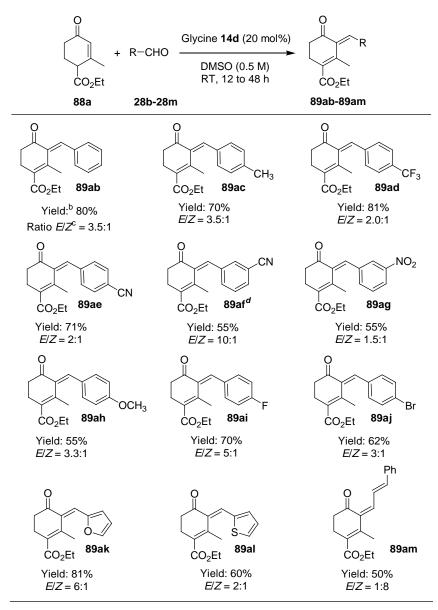
$$25 \, ^{\circ}\text{C}$$
 $CO_2\text{Et}$
 CO_2

entry	catalyst	solvent (0.5 M)	time (h)	yield (%) ^b 89aa	yield (%) ^b 90aa
1	diamine 2i	DMF	7	_	75
2	diamine 2i	DMSO	7	_	80
3	diamine 2j	DMF	8	=	62
4	diamine 2j	DMSO	8	-	66

^a Reactions were carried out in DMF or DMSO (0.5 M) with the same proportions of Hagemann's ester **88a** and aldehyde **28a** in the presence of 20 mol% catalyst. ^b Yield refers to the column-purified product.

3.2.4 Diversity-Oriented Synthesis of Push-Pull Olefins 89ab-am via Glycine-Catalysis: With an efficient amino acid-catalyzed CS protocol in hand, the scope of the glycine-catalyzed CS reactions was investigated with various aldehydes **28b-m**. A series of neutral, electron-donating, electron-withdrawing and α,β-unsaturated aldehydes **28b-m** were reacted with 1.0 equiv. of Hagemann's ester **88a** catalyzed by 20-mol% of glycine **14d** at 25 °C for 12-48 h in DMSO (Table 4). Interestingly, in all these reactions iso-aromatization did not taken place and only the CS products, 3-arylidene Hagemann's esters **89ab-89am** were isolated in moderate to

Table 4: Synthesis of Chemically Diverse Libraries of 3-Arylidene-Hagemann's Esters **89** via Glycine-Catalysis^a



^a All reactions were carried out in DMSO (0.5 M) with the same proportions of Hagemann's ester **88a** and aldehyde **28** in the presence of 20 mol% glycine. ^b Yield refers to the column-purified product. ^c *E/Z* ratio determined by NMR analysis. ^d Hagemann's ester **88a** (1.0 mmol) reacted with aldehyde **28f** (0.5 mmol) in the presence of 20 mol% glycine **14d** at 65 °C for 12 h.

good yields and with stereoselectivities favouring the *E*-isomers except in the case of α,β -unsaturated aldehyde **28m**.

The ethyl (E)-3-benzylidene-2-methyl-4-oxo-cyclohex-1-enecarboxylate **89ab** ethyl (E)-2-methyl-3-(4-methyl-benzylidene)-4-oxo-cyclohex-1-enecarboxylate 89ac were obtained as major isomers with excellent yields via 14d-catalyzed CS reaction of 88a with 28b/28c respectively (Table 4). Interestingly, the CS reaction of 88a with 4-trifluoromethyl-benzaldehyde 28d under glycine 14d-catalysis furnished the ethyl (E)-2-methyl-4-oxo-3-(4-trifluoromethyl-benzylidene)-cyclohex-1-enecarboxylate 89ad as major isomer in 81% yield (Table 4). The CS reaction of 88a with benzaldehydes 28e-g containing electron-withdrawing groups also under glycine 14dcatalysis furnished the CS products (E)-89ae-ag as major isomers with 10:1 to 1.5:1 dr ratio with good yields (Table 4). Interestingly, halogenated benzaldehydes 28i-j and heterocyclic aldehydes 28k-l also furnished the expected CS products 89ai-al with 88a under **14d**-catalysis in good yields with *E*-isomer as major (Table 4). Structure and regiochemistry of CS products 89ab-al were confirmed by ¹H NMR, ¹³C NMR [for example see Fig. 1] and mass analysis. Reaction of 88a with trans-cinnamaldehyde **28m** under glycine **14d** catalysis furnished the CS product **89am** in 50% yield with a 1:8 E/Z ratio. The regiochemistry of 89am was established based on a deuterium labeling experiment, nOe experiments and MMX calculations. ^{1a} The results in Table 4 demonstrate the broad scope of this novel CS methodology covering a structurally diverse group of aldehydes 28b-m with good yields and selectivity. 2-Arylidenecyclohexanones, 2,6-bis(arylidene)-cyclohexanones and related compounds were evaluated for anti-tumor, anti-inflammatory, anti-neoplastic, cytotoxic activity and also for the inhibition of mitochondrial function in yeast emphasizing the value of this amino-acid catalyzed CS approach. 30a-e In addition, generation of molecular diversity around the 2-arylidenecyclohexanone scaffolds may allow for the identification of more potent species.

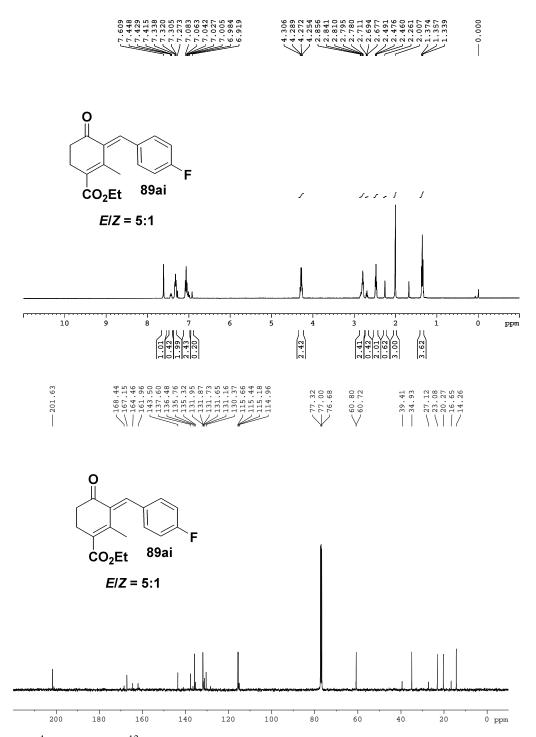
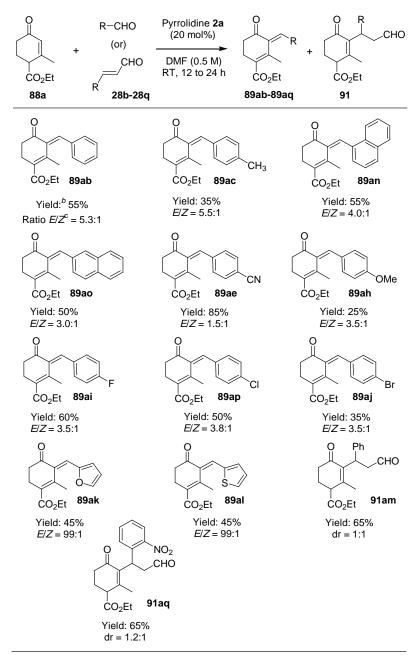


Figure-1: ¹H NMR and ¹³C NMR Spectrum of product **89ai**.

3.2.5 Diversity-Oriented Synthesis of Push-Pull Olefins 89ab-89ap and Functionalized Aldehydes 91am/91aq via Pyrrolidine-Catalysis: After successful demonstration of glycine 14d-catalyzed CS reactions with various aldehydes 28b-m, we further showed interest to screen similar reactions under pyrrolidine 2a-catalysis to test the iso-aromatization of *in situ* generated CS products 89ab-am by monitoring to the electronic nature of aldehydes 28 or/and basic nature of catalyst 2a in cascade CS/IA reactions (Table 5). As shown in Table 5, a series of neutral, electron-donating, electron-withdrawing and α,β -unsaturated aldehydes 28b-q were reacted with 1.0 equiv. of Hagemann's ester 88a catalyzed by 20 mol% of pyrrolidine 2a at 25 °C for 12-24 h in DMF. But unfortunately, in these reactions also iso-aromatization did not taken place and only the CS products, 3-arylidene Hagemann's esters 89ab-ap and Michael adducts 91am/91aq were isolated in moderate to good yields and with stereoselectivities favouring the *E*-isomers.

Compared to glycine-catalysis, pyrrolidine-catalysis generated the CS products 89ab-ac, 89an-ao with less yields and high *E*-selectivity from 88a and neutral benzaldehydes 28b-c, 28n-o at 25 °C for 12-24 h in DMF as shown in Table 5. But halogenated benzaldehydes 28i-j/p furnished the expected CS products 89ai-aj/ap with 88a under 2a-catalysis as similar to 14d-catalysis in moderate to good yields with *E*-isomer as major (Table 5). Interestingly, CS products 89ak-al generated with less yields and high *E*-selectivity (>99%) under the pyrrolidine-catalysis from 88a and heterocyclic aldehydes 28k-l at 25 °C for 24 h in DMF as shown in Table 5, which is different from the glycine-catalysis results. Reaction of 88a with *trans*-cinnamaldehyde 28m under pyrrolidine 2a catalysis furnished the Michael product 91am in 65% yield in a 1:1 diastereomeric ratio, which is different from the glycine-catalysis may be due to the nature of 28m as Michael acceptor under pyrrolidine-catalysis. Formation of Michael adducts from 88a and α,β-unsaturated aldehyde under pyrrolidine-catalysis is confirmed by one more example as shown in Table 5. Structure and regiochemistry of CS products 89ab-ap and Michael adducts 91am/91aq were confirmed by NMR and mass analysis.

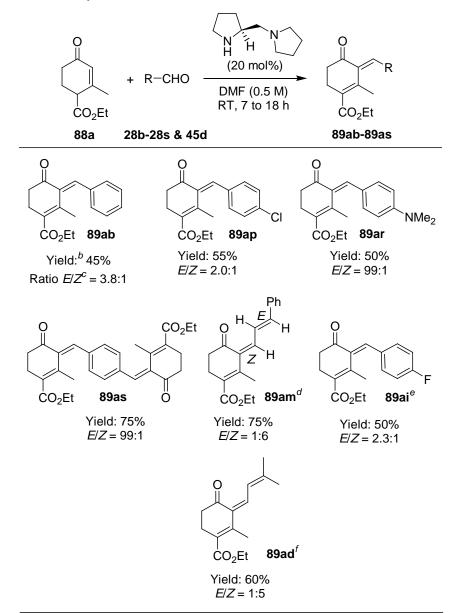
Table 5: Synthesis of Chemically Diverse Libraries of 3-Arylidene-Hagemann's Esters **89** via Pyrrolidine-Catalysis^a



^a All reactions were carried out in DMF (0.5 M) with the same proportions of Hagemann's ester **88a** and aldehyde **28** in the presence of 20 mol% pyrrolidine. ^b Yield refers to the column-purified product. ^c E/Z ratio determined by NMR analysis.

3.2.6 Diversity-Oriented Synthesis of Push-Pull Olefins 89ab-as via **Diamine-Catalysis:** After testing the glycine **14d**- and pyrrolidine **2a**-catalyzed CS reactions with various aldehydes 28b-q, we further showed interest to screen similar kind of reactions under diamine 2i-catalysis to look at the high-yielding formation of CS products 89 and CS/IA products 90 (Table 6). A series of neutral, electron-donating and α,β-unsaturated aldehydes 28b-s and 45d were reacted with 1.0 equiv. of Hagemann's ester **88a** catalyzed by 20 mol% of (S)-1-(2-pyrrolidinylmethyl)pyrrolidine 2i at 25 °C for 7-18 h in DMF or DMSO solvents. Interestingly, in these reactions also CS/IA products 90 are not furnished and only the CS products, 3arylidene Hagemann's esters 89ab-as were isolated in moderate to good yields with stereoselectivities favouring the E-isomers for simple benzaldehydes and favouring the Z-isomers for α,β -unsaturated aldehydes without formation of Michael adducts 91 (Table 6). Compared to glycine- and pyrrolidine-catalysis, diamine-catalysis generated the CS products 89ab, 89ai and 89ap with moderate yields and E-selectivity from 88a and neutral/halogenated benzaldehydes 28b, 28i and 28p at 25 °C for 7-18 h in DMF/DMSO as shown in Table 6. But 4-dimethylamino-benzaldehyde 28r and benzene-1,4-dicarbaldehyde 28s furnished the only CS products 89ar and 89as with **88a** under **2i**-catalysis in good yields with high *E*-selectivity as shown in Table 6. Interestingly, reaction of 88a with trans-cinnamaldehyde 28m under diamine 2icatalysis furnished the CS product (E, Z, E)-triene 89am in 75% yield with 1:6 E/Z ratio, which is different from the pyrrolidine-catalysis but similar to the glycinecatalysis. Formation of CS product, (E, Z, E)-triene from 88a and α,β -unsaturated aldehyde under diamine-catalysis is confirmed by one more example as shown in Table 6. Reaction of 2.0 equiv. of 88a with 3,3-dimethylacrolein 45d in DMF at 0 °C for 0.5 h under 2i-catalysis furnished the selective conjugated (E, Z, E)-triene 89ad in 60% yield with 1:5 E/Z ratio (Table 6). Structure and regiochemistry of CS products 89ab-as were confirmed by ¹H NMR, ¹³C NMR [for example see Fig. 2] and mass analysis and also finally confirmed by X-ray structure analysis on **89ai** as shown in Scheme **2**. 32

Table 6: Synthesis of Chemically Diverse Libraries of 3-Arylidene-Hagemann's ester **89** via Diamine-Catalysis^a



^a All reactions were carried out in DMF (0.5 M) with the same proportions of Hagemann's ester **88a** and aldehyde **28** in the presence of 20 mol% diamine **2i**. ^b Yield refers to the column-purified product. ^c *E/Z* ratio determined by NMR analysis. ^d Reaction time was 1 h. ^e DMSO was used as the solvent. ^f Hagemann's ester **88a** (1.0 mmol) reacted with aldehyde **45d** (0.5 mmol) in the presence of 20 mol% diamine **2i** at 0 °C in DMF for 0.5 h.

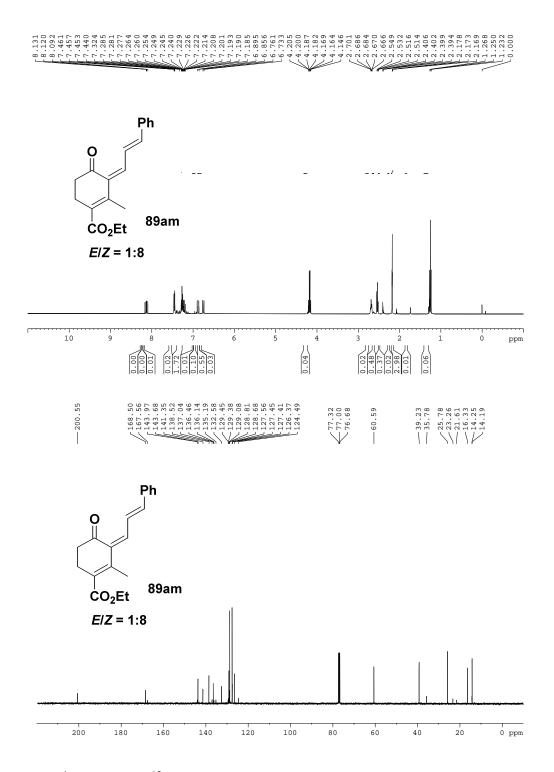


Figure-2: ¹H NMR and ¹³C NMR Spectrum of product **89am**.

Scheme 2: Crystal structure of 3-(4-fluoro-benzylidene)-2-methyl-4-oxo-cyclohex-1-enecarboxylic acid ethyl ester (**89ai**).

$$\equiv \bigcup_{\mathsf{CO}_2\mathsf{Et}}^{\mathsf{O}} \mathbb{S}\mathsf{9ai}$$

3.2.7 *Diversity-Oriented Synthesis of Push-Pull Phenols 90 via Diamine-Catalysis:* After thorough investigation of the glycine **14d**-, pyrrolidine **2a**- and diamine **2i**-catalyzed CS reactions of Hagemann's ester **88a** with various aldehydes **28a-s** [Tables 1-6], we further screened only benzaldehydes **28** containing an electron-withdrawing groups with variety of Hagemann's esters **88a-s** under diamine **2i**-catalysis to look at the high-yielding formation of CS/IA products **90** (Table 7). A series of benzaldehydes **28** containing an electron-withdrawing groups and α,β-unsaturation were reacted with 1.0 equiv. of Hagemann's esters **88a-s** catalyzed by 20 mol% of (*S*)-1-(2-pyrrolidinyl-methyl)pyrrolidine **2i** at 25 °C for 7-72 h in DMSO. Interestingly, in all these reactions expected CS/IA products or push-pull phenols **90** were furnished with moderate to very good yields as shown in Table 7. Functionalized push-pull phenols **90** are an important class of compounds, which are widely used as intermediates in the pharmaceuticals. ^{30f-1}

Table 7: Synthesis of Chemically Diverse Libraries of Highly Substituted Phenols **90** via Diamine-Catalysis^a

Table 7: Synthesis of Chemically Diverse Libraries of Highly Substituted Phenols **90** via Diamine-Catalysis (Continuation)^a

Compounds containing 2-alkyl-push-pull phenols **90** have found pharmaceutical applications as remedies for diabetes **D**, insulin resistance **E**, antiasthmatic LTD4 antagonist ICI-204/219 **F**, lasiodiplodin analogues **G** and also starting materials for the synthesis of natural products as shown in Chart 1. 30f-1 As such, the development of new and more general catalytic methods for their preparation is of significant interest. 31 L-Diamine-catalyzed cascade CS/IA reaction of **88a** with 4-cyanobenzaldehyde **28e** in DMSO at 25 °C for 7 h furnished the expected product **90ae** in 75% yield (Table 7, entry 2). The CS/IA reaction of 2-nitro- and 3-nitrobenzaldehydes **28t/g** with **88a** catalyzed by L-**2i** in DMSO at 25 °C for 7 h furnished the expected **90at/ag** with 75-60% yields respectively as shown in Table 7. After these interesting results, we decided to investigate the scope and limitations of the CS/IA reaction with other two aldehydes containing electron withdrawing groups **28d/f** with **88a** under L-diamine-catalysis at the ambient conditions (Table 7, entries 5-6). Interestingly, CS/IA reaction of 3-

^a All reactions were carried out in DMSO (0.5 M) with the same proportions of Hagemann's ester **88a** and aldehyde **28** in the presence of 20 mol% diamine **2i**. ^b Yield refers to the column-purified product. ^c Hagemann's ester **88** (1.0 mmol) reacted with aldehyde **28** (0.5 mmol) in the presence of 20 mol% diamine **2i** at 25 °C for 12-24 h. ^d Hagemann's ester **88q** (1.0 mmol) reacted with aldehyde **28a** (0.5 mmol) in the presence of 20 mol% diamine **2i** at 70 °C for 12 h.

cyanobenzaldehyde **28f** and 4-trifluoromethyl-benzaldehyde **28d** with **88a** under L-diamine-catalysis furnished the expected push-pull phenols **90af** and **90ad** in 60/70% yields respectively as shown in Table 7, entries 5-6.

After these interesting results, we further decided to investigate the scope and limitations of the CS/IA reaction with a range of Hagemann's esters 88a-s including chiral Hagemann's esters **88c-d**, simple ester **88e**, and 6-substituted Hagemann's esters **88f-r** with 4-nitrobenzaldehyde **28a**, 4-cyanobenzaldehyde **28e**, 3-(2-nitro-phenyl)propenal 28q and ethyl glyoxylate 28u under L-diamine-catalysis in DMSO at the ambient conditions to test the diversity nature of the CS/IA reaction (Table 7). Synthesis of chiral Hagemann's esters 88c, 88d described in experimental section as shown in Scheme E1 and Scheme E2.³³ As shown in Table 7, CS/IA reaction of *t*-butyl ester **88b** with 28a under 2i-catalysis for 24 h furnished the phenol 90ba with 75% yield (Table 7, entry 7). Interestingly, CS/IA reaction of 4-nitrobenzaldehyde 28a with 2.0 equiv. of chiral Hagemann's esters 88c-d under 2i-catalysis in DMSO at 25 °C for 24-12 h furnished the chiral push-pull phenols (+)-90ca in 80% yield and (-)-90da in 90% yield as shown in Table 7, entries 8/9. Diamine-catalyzed CS/IA reaction of 4nitrobenzaldehyde 28a with simple Hagemann's esters 88e-f and 88g-h³⁴ in DMSO at 25 °C for 10-24 h furnished the expected push-pull phenols **90ea-ha** in 60-85% yields respectively as shown in Table 7, entries 10-13. Interestingly, CS/IA reaction of 4nitrobenzaldehyde 28a with 6-methyl-Hagemann's ester 88i under 2i-catalysis in DMSO at 25 °C for 12 h furnished the highly functionalized push-pull phenol **90ia** in 65% yield as shown in Table 7, entry 14. Generality of the diamine-catalyzed CS/IA reaction of 6-substituted Hagemann's esters 88 with 4-nitrobenzaldehyde 28a or 4cyanobenzaldehyde 28e were confirmed by ten more examples with esters 88i-r containing different functional groups under 2i-catalysis and furnished the expected highly substituted push-pull phenols **90ia-ra** with 50-80% yields respectively as shown in Table 7. The products structures were confirmed by ¹H NMR, ¹³C NMR [for example see Fig. 3] and mass analysis.

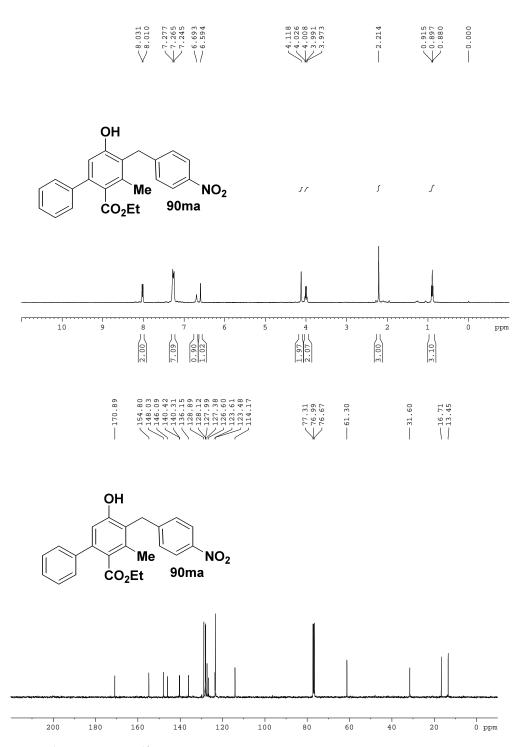


Figure-3: ¹H NMR and ¹³C NMR Spectrum of product **90ma**.

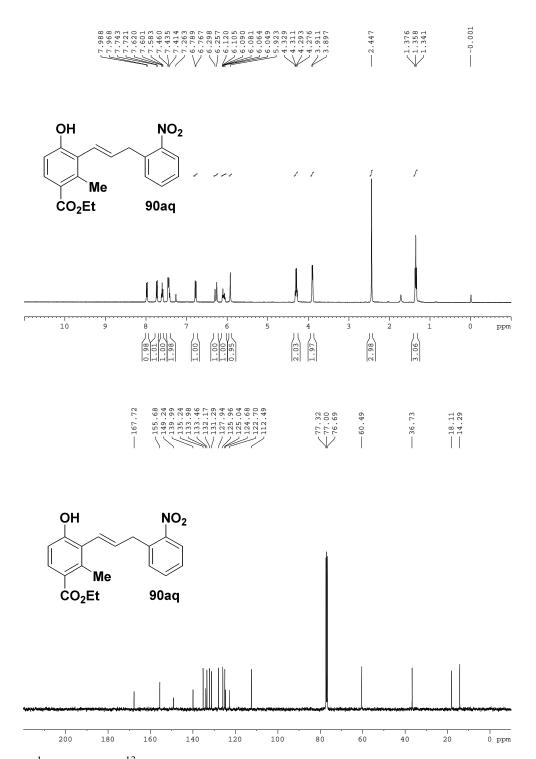


Figure-4: ¹H NMR and ¹³C NMR Spectrum of product **90aq**.

Interestingly, cascade reaction of 3-(2-nitro-phenyl)-propenal **28q** and ethyl glyoxylate **28u** with **88a** under L-diamine-catalysis furnished the novel products **90aq** in 65% yield via CS/IA/I and **90au** in 20% yield via CS/IA reactions as shown in Table 7, entries 25-27. Generality of the **2i**-catalyzed cascade CS/IA/I reaction was confirmed by one more example with **88s** and **28q** as shown in Table 7. Structure and regiochemistry of CS/IA/I products **90aq/sq** were confirmed by ¹H NMR, ¹³C NMR [for example see Fig. 4] and mass analysis and also finally confirmed by X-ray structure analysis on **90aq** as shown in Scheme 3.³⁵ Presently developed amine-catalyzed CS/IA/I reaction looks novel technology for the *ortho*-vinylation of *in situ* generated functionalized phenols compared to metal mediated *ortho*-vinylation of preformed phenols.³⁶ Functionalized push-pull phenols **90au-ra** are useful intermediates for the synthesis of analogues of remedies for diabetes **D**, insulin resistance **E**, antiasthmatic LTD4 antagonist ICI-204/219 **F**, lasiodiplodin analogues **G** and also for the synthesis of natural products as shown in Chart 1.^{30f-1} This CS/IA technology may be suitable to develop large number of diverse-compounds of **90** to screen and identify the suitable bioactive products.

Scheme 3: Crystal structure of 4-hydroxy-2-methyl-3-[3-(2-nitro-phenyl)-propenyl]-benzoic acid ethyl ester (**90aq**).

$$= \bigvee_{\mathsf{CO_2Et}}^{\mathsf{OH}} \bigvee_{\mathsf{NO_2}}^{\mathsf{NO_2}}$$

3.2.8 Diversity-Oriented Synthesis of Highly Functionalized (E)-1,3-Dienes 92 via Diamine-Catalysis: After thorough investigation of the diamine 2i-catalyzed CS, CS/IA and CS/IA/I reactions of Hagemann's esters 88a-s with various aldehydes 28a-u, we further showed interest to screen aliphatic aldehydes 28v-b' containing an αhydrogen with Hagemann's ester 88a under diamine 2i-catalysis to look at the effect of electronic factors of substrates on product formation (Table 8). A series of aliphatic aldehydes 28v-b' containing an α-hydrogen were reacted with 2.0 equiv. of Hagemann's ester 88a catalyzed by 20 mol% of (S)-1-(2-pyrrolidinylmethyl)pyrrolidine 2i at 25 °C for 0.5-24 h in DMSO (0.5 M). Surprisingly, in all these reactions unexpected CS/I products 92 were furnished in moderate to very good yields with high selectivity instead of CS products 89 as shown in Table 8.

Interestingly, cascade reaction of optically pure (R)-(+)-citronellal 28v with 2.0 equiv. of Hagemann's ester 88a under 2i-catalysis in DMSO at 25 °C for 24 h furnished the chiral functionalized (E)-1,3-diene (-)-92av in 65% yield with >99:1 ratio of E:Z isomers and 1:1 ratio of diastereomers as shown in Table 8, entry 1. Unexpected formation of product (-)-92av from 88a and 28v via 2i-catalysis can be explained through cascade Claisen-Schmidt/isomerization (CS/I) reactions and further reaction mechanism is discussed in the mechanistic insights. Diamine-catalyzed CS/I reaction of optically pure opposite enantiomer (S)-(-)-citronellal 28w with simple Hagemann's ester 88a in DMSO at 25 °C for 24 h furnished the chiral functionalized (E)-1,3-diene (+)-92aw in 65% yield with >99:1 ratio of E:Z isomers and 1:1 ratio of diastereomers as shown in Table 8, entry 2. Generality of the diamine-catalyzed CS/I reaction of Hagemann's ester **88a** with chiral aliphatic aldehyde³⁷ were confirmed by one more example with dimethyl 2-(3-oxo-1-phenyl-propyl)-malonate [93% ee, (-)-28x] containing an α -hydrogen under **2i**-catalysis and furnished the expected highly substituted chiral (E)-1,3-diene (+)-92ax in 50% yield with >99:1 ratio of E:Z isomers and 1:1 ratio of diastereomers as shown in Table 8, entry 3.

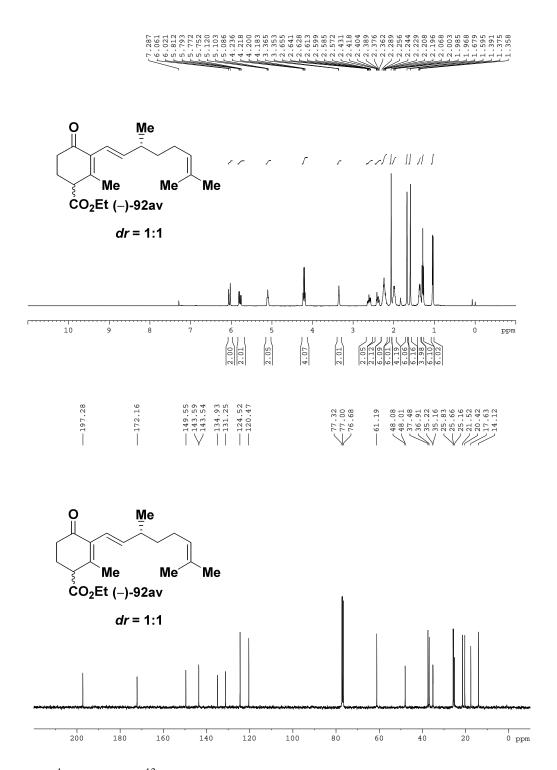


Figure-5: ¹H NMR and ¹³C NMR Spectrum of product **92av**.

After successful demonstration of diamine-catalyzed cascade CS/I reaction of **88a** with chiral aldehydes **28v-x**, we further investigated the scope and limitations of the cascade CS/I reaction with a range of achiral aliphatic aldehydes **28y-b'** containing an α -hydrogen under L-diamine-catalysis in DMSO at the ambient conditions to test the diversity nature of the CS/I reaction (Table 8). The products structures were confirmed by 1 H NMR, 13 C NMR [for example see Fig. 5] and mass analysis.

Table 8: Synthesis of Chemically Diverse Libraries of 1,3-Dienes **92** via Diamine-Catalysis^a

^a Hagemann's ester **88a** (1.0 mmol) reacted with aldehyde **28v-28b'** (0.5 mmol) in the presence of 20 mol% diamine **2i** at 25 $^{\circ}$ C in DMSO for 0.5 to 24 h. ^b Yield refers to the column-purified product. ^c E/Z and dr ratio determined by NMR analysis. ^d Reaction stirred at 0 $^{\circ}$ C in DMF for 0.5 h.

Cascade CS/I reaction of ester **88a** with 3-phenyl-propionaldehyde **28y** under **2i**-catalysis for 17 h furnished the achiral (*E*)-1,3-diene **92ay** in 80% yield with >99:1 ratio of *E:Z* isomers as shown in Table 8, entry 4. In a similar manner, cascade CS/I reaction of butyraldehyde **28z** and propionaldehyde **28a'** with 2.0 equiv. of Hagemann's ester **88a** under **2i**-catalysis in DMSO at 25 °C for 24/12 h furnished the achiral (*E*)-1,3-dienes **92az** in 50% yield with >99:1 ratio of *E:Z* isomers and **92aa'** in 70% yield with >99:1 ratio of *E:Z* isomers, respectively as shown in Table 8, entries 5/6. Interestingly, cascade CS/I reaction of Hagemann's ester **88a** with acetaldehyde **28b'** under **2i**-catalysis in DMSO is not clean reaction, but same reaction in DMF at 0 °C for 0.5 h furnished the only CS product **89ab'** in 40% yield with 1:>99 ratio of *E:Z* isomers instead of expected 1,3-diene **92ab'** as shown in Table 8, entry 7.

Amine-induced cascade CS/I reaction is first time to observe and certainly this CS/I technology suitable to develop large number of diverse-compounds of **92** to screen and identify the suitable intermediates for the bioactive and natural product synthesis.

3.2.9 Diversity-Oriented Synthesis of Highly Functionalized bis-Enones 93 via Piperidine-Catalysis: After the investigation of the amino acid 14 or amine 2 catalyzed CS, CS/IA, CS/IA/I and CS/I reactions of Hagemann's esters 88a-s with various aldehydes 28a-b', we further interested to screen simple formaldehyde 28c' with Hagemann's ester 88a under amine 2-catalysis (Table 9 and Scheme 4). Interestingly, reaction of 37% aqueous formaldehyde 28c' with 2.0 equiv. of Hagemann's ester 88a under proline 14a-catalysis in DMSO at 25 °C for 24 h furnished the functionalized bis-enone 93ac' in 55% yield with >99% de as shown in Table 9, entry 1. Unexpected formation of product 93ac' from 88a and 28c' via 14a-catalysis can be explained through cascade Claisen-Schmidt/Michael (CS/M) reactions and further information on the reaction mechanism is discussed in the mechanistic insights. Functionalized bis-enones 93 would be suitable intermediates for the synthesis of

terpenoid natural products³⁸ and for the development of high-yielding one-pot synthesis of **93** *via* amine-catalysis is very much needed.

Table 9: Optimization of the Direct Amine-Catalyzed Cascade Claisen-Schmidt and Michael reaction of **88a** and **28c'**^a

entry	catalyst (20 mol%)	solvent (0.16 M)	time (h)	yield (%) ^b 93ac'
1	proline 14a	DMSO	24	55
2	glycine 14d	DMSO	24	55
3 ^c	pyrrolidine 2a	DMSO	4	65
4	pyrrolidine 2a	DMSO	1	76
5	pyrrolidine 2a	DMF	1	70
6	piperidine 2b	DMSO	1	80
7	diamine 2i	DMSO	1	72

^a Reactions were carried out in solvent (0.16 M) with the 2.0 equiv. of Hagemann's ester **88a** to 37% aqueous formaldehyde **28c'** in the presence of 20 mol% catalyst. ^b Yield refers to the column-purified product. ^c 1:1 ratio of ester **88a** and 37% aqueous formaldehyde **28c'** was used.

For the high-yielding one-pot synthesis of *bis*-enone **93ac'** from simple substrates **88a** and **28c'**, we further screened catalysts and solvent effect on the cascade CS/M reaction as shown in Table 9. The cascade CS/M reaction of **88a** and **28c'** under glycine **14d**-catalysis in DMSO also as similar to **14a**-catalysis, but the same cascade reaction under pyrrolidine **2a**-catalysis furnished the product **93ac'** with improved yield (76%) as shown in Table 9, entries 2-4.

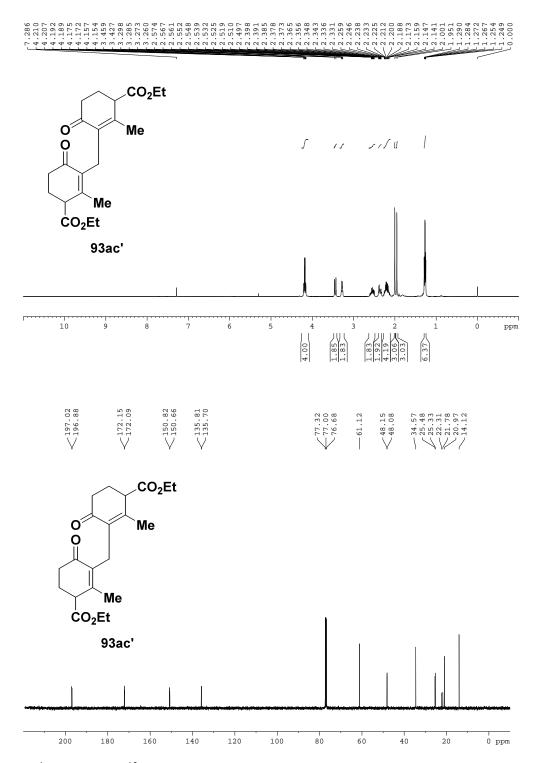


Figure-6: ¹H NMR and ¹³C NMR Spectrum of product **93ac'**.

DMSO looks better solvent for cascade CS/M reaction of **88a** and **28c'** compared to DMF under **2a**-catalysis (Table 9, entry 5). The CS/M reaction yield is increased to 80% under the piperidine **2b**-catalysis in DMSO for 1 h at 25 °C as shown in Table 9, entry 6. Diamine-catalyzed CS/M reaction of **28c'** with **88a** in DMSO at 25 °C for 1 h furnished the functionalized *bis*-enone **93ac'** in 72% yield with >99% de, which is not superior compared to piperidine-catalysis as shown in Table 9, entry 7. The optimized reaction conditions for *bis*-enone synthesis involved piperidine **2b**-catalysis at 25 °C in DMSO with 2 equiv. of **88a** and **28c'**, which furnished the cascade CS/M product **93ac'** in 80% yield with >99% de (Table 9, entry 6).

Scheme 4: Synthesis of Dienones **93** via Direct Amine-Catalyzed Cascade Claisen-Schmidt/Michael reactions^{a,b}

Piperidine 2b (20 mol%)

Me CO₂R¹

88

28c'

$$CO_{2}R^{1}$$

Me CO₂Et

$$CO_{2}R^{1}$$

93

$$CO_{2}R^{1}$$

Me CO₂Et

$$CO_{2}Et$$

$$CO_{3}Et$$

$$CO_{2}Et$$

$$CO_{2}Et$$

$$CO_{3}Et$$

$$CO_{4}E$$

$$CO_{5}Et$$

$$CO_$$

Generality of the piperidine-catalyzed CS/M reaction of formaldehyde **28c'** with Hagemann's esters **88** were confirmed by two more examples with *t*-butyl ester **88b** and methyl ester **88s** under **2b**-catalysis and furnished the expected highly substituted *bis*-

^a See Experimental Section. ^b Yield refers to the column purified product.

enones **93bc'** in 20% yield with >99% de and **93sc'** in 60% yield with >99% de as shown in Scheme 4. The products structures were confirmed by ¹H NMR, ¹³C NMR [for example see Fig. 6] and mass analysis.

3.2.10 Applications of the Push-Pull Dienamine Chemistry

A. Development of Sequential One-pot Combination of Cascade Reactions based on the CS/IA Platform: Functionalized phenols 90 are an important class of compounds, which are widely used as intermediates in the pharmaceuticals and natural product synthesis as shown in Chart 1.³⁰ As such, the development of new and more general one-pot catalytic methods for their high-yielding preparation is of significant interest.³¹ Hagemann's esters 88a-s were synthesized in good yields with minor modifications of known methods of direct piperidine- or KO^tBu-catalyzed cascade Knoevenagel/Michael/aldol condensation/decarboxylation (K/M/A/DC) reactions of alkyl acetoacetates and aldehydes (Table 10).^{1b} Herein, we utilized the direct sequential combination of piperidine-catalyzed cascade K/M/A/DC and diamine-catalyzed cascade CS/IA of ethyl acetoacetate, aldehydes and 4-nitrobenzaldehyde to furnish the highly functionalized phenols 90 with high yields in one-pot as shown in Table 10.

Cascade K/M/A/DC reaction of two equiv. of ethyl acetoacetate and benzaldehyde under piperidine-catalysis in EtOH at 80 °C for 5-6 h furnished the expected Hagemann's ester **88m** with >99% conversion, which on *in situ* treatment with 4-nitrobenzaldehyde **28a** at 25 °C in same solvent didn't furnished the expected product **90ma** in good yield; but removing the solvent EtOH by vacuum pump and adding solvent DMSO, 20 mol% of diamine **2i** and 4-nitrobenzaldehyde **28a** to the reaction mixture of cascade K/M/A/DC furnished the expected product **90ma** in 80% yield as shown in Table 10, entry 1.

Table 10: Sequential Combination of Cascade Knoevenagel/Michael/Aldol Condenasation/Decarboxylation and Cascade Claisen-Schmidt/Iso-aromatization Reactions in One-Pot^{a,b}

Successful sequential one-pot combination of two cascade K/M/A/DC and CS/IA reactions under piperidine-/diamine-catalysis was demonstrated by three more examples with good yields as shown in Table 10 and this one-pot synthetic strategy will show much impact on the synthesis of functionalized phenols **90** from simple substrates.

B. Base-Catalyzed Iso-aromatization of CS Products: After successful synthesis of functionalized push-pull phenols 90 from Hagemann's esters 88a-s and benzaldehydes

^a See Experimental Section. ^b Yield refers to the column purified product.

Scheme 5: Controlled Experiments on Base-induced Iso-aromatization of 3-Arylidene-Hagemann's ester 89 via Amine-Catalysis

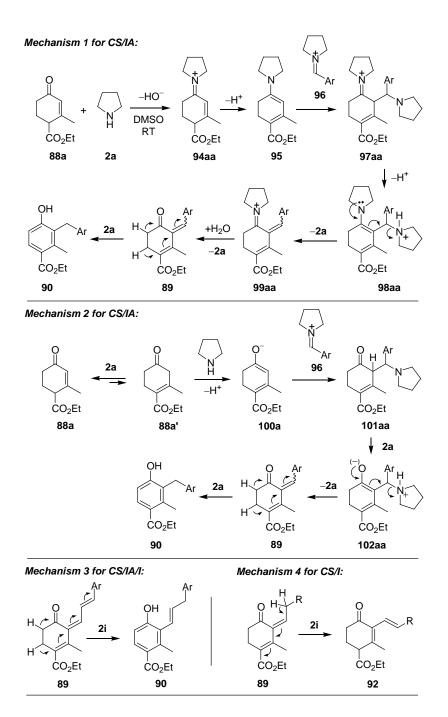
28 containing an electron-withdrawing group under diamine-catalysis, we thought of testing the how much basic nature of amine 2 and electronic nature of substrates 88/28 will control the iso-aromatization of CS products 89 as shown in Scheme 5. Interestingly, treatment of pure CS product 89aa with 20 mol% of diamine 2i in DMSO at 25 °C for 0.5 h furnished the phenol 90aa in 96% yield via iso-aromatization as shown in Scheme 5. In a similar manner, treatment of pure CS product 89ab with 20 mol% of piperidine 2b in DMSO at 70 °C for 12 h furnished the phenol 90ab in 86% yield via iso-aromatization as shown in Scheme 5. These results suggesting that both basic nature of amine 2, electronic nature of substrates 88/28 and also reaction temperature is important for the iso-aromatization of CS products 89.

With an efficient piperidine-catalyzed iso-aromatization protocol in hand, we continued our investigation for the synthesis of functionalized phenols 90 from Hagemann's esters 88a/m and benzaldehyde 28b under piperidine-catalysis in DMSO at 70 °C for 18 h through cascade CS/IA reactions as shown in Scheme 5. Cascade

products **90ab** and **90mb** were furnished in moderate yields by without showing much of electronic factors in CS/IA reaction as shown in Scheme 5. Functionalized phenols are of considerable importance in a variety of industries. These compounds **90** are widespread elements in natural products and have attracted much attention from a wide area of science, including physical chemistry, medicinal chemistry, natural product chemistry, synthetic organic chemistry and polymer science. As such, the development of new and more general catalytic methods for their preparation is of significant interest and our presently developed cascade chemistry will be useful to develop library of substituted phenols in very good yields with high selectivity.

3.2.11 *Mechanistic Insights:* The possible reaction mechanisms for the synthesis of push-pull olefins (3-arylidene-Hagemann's ester) **89**, push-pull phenols **90** and (E)-1,3dienes 92 via dienamine-catalysis are illustrated in Schemes 6-7 and Figure 8. First, reaction of amino acids (proline 14a, phenylalanine 14c or glycine 14d) or amines (pyrrolidine 2a, piperidine 2b, or (S)-1-(2-pyrrolidinyl-methyl)pyrrolidine 2i) with the aldehyde 28 generates the imine cation 96, an excellent electrophile that undergoes Mannich type reactions with the *in situ* generated push-pull dienamine 103 & 95 or dienolate 100a of Hagemann's ester 88a to generate Mannich products 97 and 101, respectively as shown in mechanism 1 and 2 of Scheme 6 (for the clarity purpose, we represented 2a as catalyst). Elimination reaction of amines 98 and 102 under basic conditions would furnish E/Z mixtures of push-pull olefin 89. Base-induced, electronically- and temperature-controlled iso-aromatization (IA) of the CS product 89 would then give push-pull phenol **90** as shown in mechanism 1 and 2 of Scheme 6. The formation of imine ion 96 and CS product 89 via Mannich and amine elimination type reactions supports our hypothesis that aldol products did not form in these reactions and also formation of the highly reactive push-pull dienamines 103 and 95 were established through NMR experiments as shown in Scheme 7 and Figure 8. We favor mechanism 1 based on in situ NMR experiments. The products structures were confirmed by ¹H NMR, ¹³C NMR [for example see Fig. 7] and mass analysis.

Scheme 6: Proposed Reaction Mechanisms for Push-Pull Dienamine Chemistry



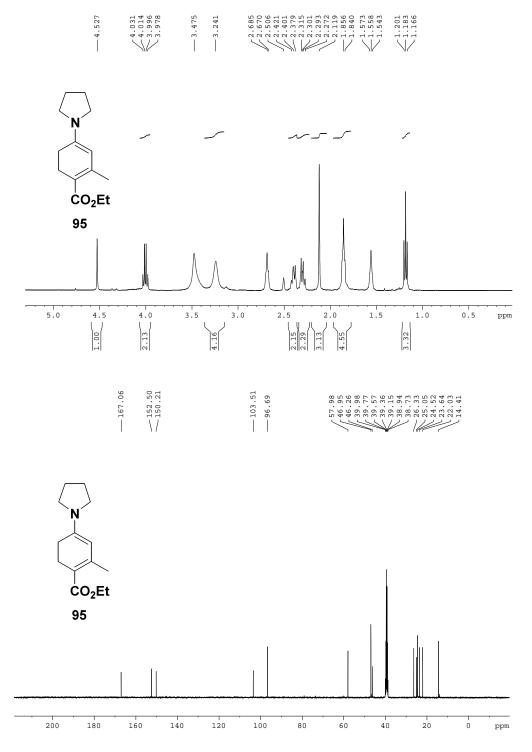


Figure-7: ¹H NMR and ¹³C NMR Spectrum of product **95**.

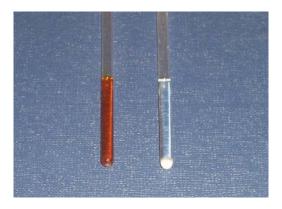
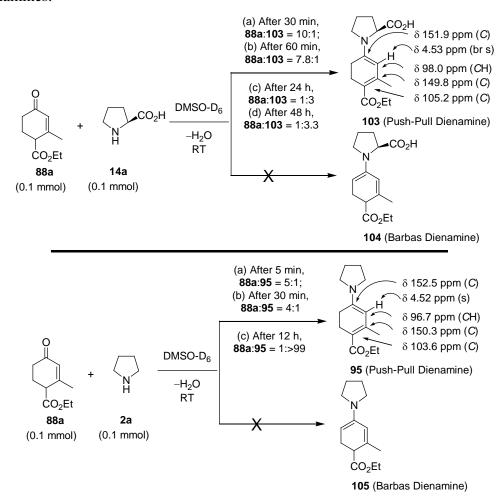


Figure-8: Picture of NMR samples recorded after three days [first NMR tube representing the mixture of **88a** and **14a** in DMSO-D₆ and second NMR tube representing the only undissolved catalyst **14a** in DMSO-D₆].

As shown in Scheme 7, in situ generation of novel push-pull dienamine 103 from 88a and 14a is slow process as compared to 95 from 88a and 2a and similar kind of reactivity patron represented in their reactions also as shown in Tables 1-10. Highly selective in situ formation of push-pull dienamines 103 and 95 over the Barbas dienamines (2-amino-1,3-butadienes)^{14,15b,19,21,23,25,39} **104** and **105** were established based on NMR and mass analysis as shown in Scheme 7. Presently discovered relatively stable push-pull dienamine 103 formation from Hagemann's ester 88a and amino acid, L-proline 14a is very good supportive to the recently proposed enamine mechanism in proline-catalyzed asymmetric reactions^{8,40} as revealed in Scheme 7 and Figure 8. Next, the possible mechanism for regioselective synthesis of cascade CS/IA/I and CS/I products, ortho-vinylated phenols 90 and (E)-1,3-dienes 92 through reaction of Hagemann's ester 88a, aldehydes 28 and diamine 2i is illustrated in mechanism 3 and 4 of Scheme 6. Treatment of in situ generated CS product, push-pull trienone (arylidene-Hagemann's ester) 89 under the amine-catalysis furnished the orthovinylated phenol 90 via IA/I reaction. In a similar manner, reaction of in situ generated CS product, push-pull dienone (alkylidene-Hagemann's ester) 89 under the basecatalysis furnished the unexpected (*E*)-1,3-dienes **92** via I reaction as shown in Scheme 6.

Scheme 7: NMR Experiment for the Detection of In Situ Generated Push-Pull Dienamines.



These two kinds of isomerizations (I) on **89** through diamine-catalysis were completely controlled by electronic factors and acidic nature of hydrogen's as shown in Scheme 6.

3.3 CONCLUSION

In this chapter, we have developed the sequential one-pot combination of amino acid, amine catalyzed direct cascade CS, CS/IA, CS/IA/I, CS/I, M, CS/M, K/M/A/DC/CS/IA reactions from simple substrates. This experimentally simple cascade approach can be used to construct highly substituted push-pull olefins and phenols with good yields in a selective fashion. We have demonstrated the *in situ* generation and application of novel push-pull dienamines in sequential cascade chemistry.

4. ORGANOCATALÝTIC CASCADE REACTIONS BASED ON PUSH-PULL DIENAMINE PLATFORM: SYNTHESIS OF HIGHLY SUBSTITUTED ANILINES

4.1 INTRODUCTION

Aryl-amines are of considerable importance in a variety of industries. As such, the development of new and more general methods for their preparation is of significant interest.⁴¹ Recently palladium catalysis emerging for the reactions of aryl halides with primary and secondary amines under the presence of strong base to provide general route to a variety of aryl-amines in good yields (Scheme 8).⁴²

Scheme 8: Synthesis of Highly Substituted Anilines

Herein, we discovered a metal-free, novel and green technology for the synthesis of highly substituted o-hydroxydiarylamines, o-pyrrolidin-1-yl-diarylamines and o-alkyloxydiarylamines by using direct organocatalytic cascade enamine amination/iso-aromatization (EA/IA) and enamine amination/iso-aromatization/alkylation (EA/IA/A) reactions from commercially available enones, nitrosobenzenes and alkyl halides (Scheme 8). Direct combination of amine-catalyzed cascade Knoevenagel/Michael/aldol condensation/decarboxylation (K/M/A/DC) and cascade enamine amination/iso-aromatization (EA/IA) of alkyl acetoacetates, aldehydes and nitrosoarenes was developed in one-pot as shown in Scheme 9. o-Hydroxydiarylamines are useful materials as additives for rubbers and plastics, antioxidants, antibacterial activity, anti-fibrillant activity and hair dyeing.⁴³

Scheme 9: Organocatalytic Cascade Approach to the Synthesis of Highly Substituted Anilines

In continuation of our discovery of in situ generation and application of novel push-pull dienamines^{1a} in tandem reactions, we initiated our studies of the cascade EA/IA reaction by screening a number of known and novel organocatalysts for the amination of a variety of Hagemann's esters 88 with different nitrosoarenes 56 as shown in Scheme 9. For developing this novel cascade EA/IA reaction, we need a library of Hagemann's esters 88 and we synthesized these esters 88 in good yields with minor modifications of known methods of direct piperidine- or KO^tBu-catalyzed cascade K/M/A/DC reactions (Scheme 9).⁴⁴

4.2 RESULTS AND DISCUSSION

We initiated our studies of the cascade EA/IA reaction by screening a number of known and novel organocatalysts for the amination of Hagemann's ester **88a** by 0.5 to 1.0 equivalents of nitrosobenzene **56a** as shown in Table 11.⁴⁵ Proline catalyzed the formation of o-hydroxydiarylamine **106aa** in moderate yields in DMSO and DMF solvents (Table 11, entries 1 and 2). Interestingly, catalyst diamine generated the cascade product **106aa** in very good yield in DMSO (Table 11, entry 3).

Table 11: Optimization of Direct Organocatalytic Cascade Synthesis of o-Hydroxydiarylamines^a

O CO ₂ E	+ N — S	atalyst HNR ₂ Solvent RT	OH H N P	+	YN. Ph	H H Ph N +	R N R CO ₂ Et
88a	56a		106aa	10	07aa	•	108aa
entry	catalyst (20 mol%)	solvent (0.3 M)	H. ester 88a (eq.)	time (h)	prod 106aa	lucts yield	d (%) ^b 108aa
1	Proline	DMSO	1.0	6	46	-	_
2	Proline	DMF	1.0	6	32	-	-
3	Diamine ^c	DMSO	2.0	1	90	_	_
4	Glycine	DMSO	2.0	36	30	_	-
5	Piperidine	DMSO	2.0	1	87	_	-
6	Morpholine	DMSO	2.0	1	88	_	_
7	Benzylamine	DMSO	2.0	1	83	-	_
8	Pyrrolidine	DMSO	2.0	1	88	4	4
9^d	Pyrrolidine	DMSO	2.0	1	85	2	2
10	Pyrrolidine	DMSO	1.0	1	70	2	2
11	Pyrrolidine	CH ₃ CN	2.0	10	40	10	10
12	Pyrrolidine	EtOH	2.0	10	35	10	10
13	106aa (5 mol%)	DMSO	2.0	72	35	_	_
14	-	DMSO	2.0	72	-	-	_

^a Reactions were carried out in solvent (0.3 M) with 1.0 to 2.0 equiv. of **88a** relative to the **56a** in the presence of 20 mol% of catalyst. ^b Yield refers to column purified product. ^c (s)-1-(2-pyrrolidinylmethyl)pyrrolidine. ^d 5 mol% of pyrrolidine used.

Secondary-amines like piperidine and morpholine catalysts also furnished the cascade product **106aa** in very good yields with excellent regio-selectivity in DMSO solvent (entries 5 and 6). Primary amine, benzyl amine also catalyzed the formation of cascade product **106aa** in good yield (entry 7). Simple amine, pyrrolidine catalyzed the cascade EA/IA reaction to produce **106aa** in 88% yield, which was accompanied by 1:1 regio isomers of o-pyrrolidin-1-yl-diarylamines **107aa** and **108aa** in 8% yield (entry 8). Amine-catalyzed cascade EA/IA reactions are solvent dependent and also autocatalyzed reactions as shown in Table 11, entries 10 to 14. 5 mol% of **106aa**-Catalyzed the cascade EA/IA reaction of **88a** and **56a** to produce, product **106aa** in 35% yield.

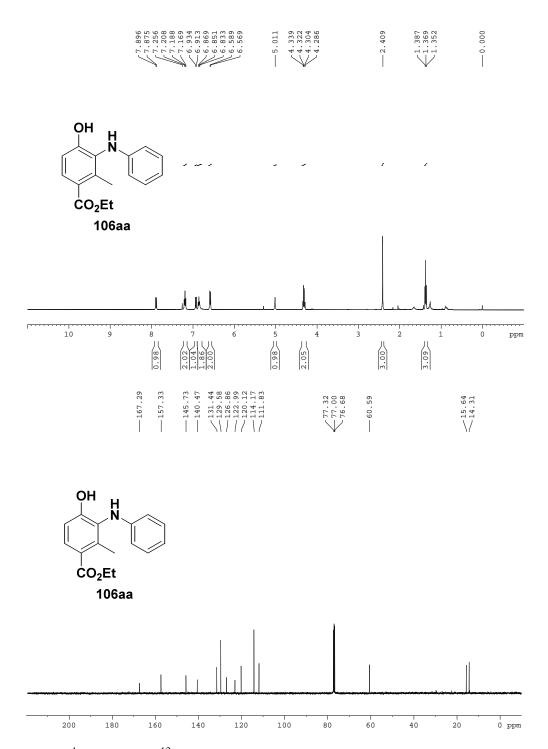


Figure-9: ¹H NMR and ¹³C NMR Spectrum of product **106aa**

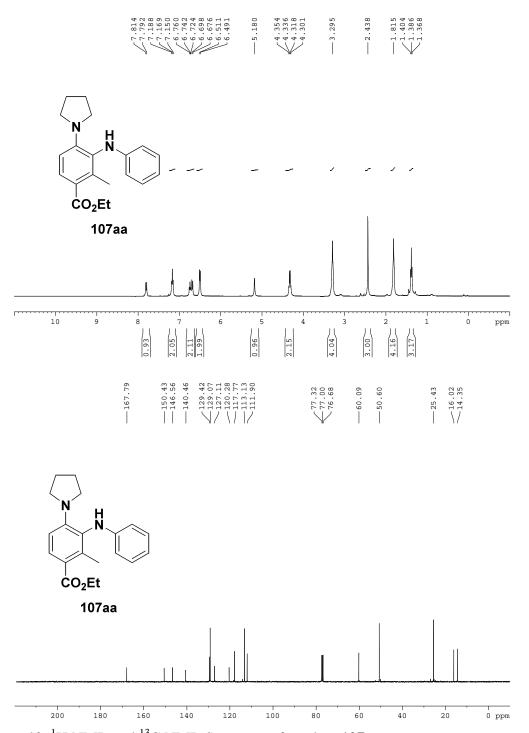


Figure-10: ¹H NMR and ¹³C NMR Spectrum of product 107aa

This is good demonstration for involvement of autocatalysis in present reactions (Table 11, entry 13). We envisioned the optimized condition to be 25 °C in DMSO under 5 mol% of pyrrolidine catalysis to furnish o-hydroxydiaryl-amine **106aa** in 85% yield (entry 9).

Table 12: Optimization of Direct Self-Catalyzed Cascade Synthesis of o-Aminediarylamines

In the investigation of EA/IA cascade reactions under pyrrolidine catalysis, product **106aa** was accompanied by interesting diamination products **107aa** and **108aa** with good conversion in EtOH via self-catalysis (Table 11, entry 12). To further exploit formation of this novel structures, we initiated our studies of the cascade EA/IA reaction by screening a number of known amines for the diamination of Hagemann's ester **88a** by 0.5 to 1.0 equivalents of both nitrosobenzene **56a** and amines **2** as shown in

^a Yield refers to the column purified product. ^b All reactants pyrrolidine (0.3 mmol), nitrosobenzene **56a** (0.3 mmol) and Hagemann's ester **88a** (0.6 mmol) were mixed at the same time in solvent (0.3 M) and stirred at room temperature. ^c To the mixture of Hagemann's ester **88a** (0.3 mmol) and amine **2** (0.6 mmol) in solvent (0.5 mL), added the 0.5 mL solution of nitrosobenzene **56a** (0.3 mmol) over the period of 0.5 h and stirred at RT. ^d To the mixture of Hagemann's ester **88a** (0.6 mmol), pyrrolidine (0.3 mmol) and MS 4A (300 mg) in solvent (0.5 mL), added the 0.5 mL solution of nitrosobenzene **56a** (0.3 mmol) over the period of 0.5 h and stirred at RT.

Table 12. Proline, diamine and glycine as reagent in cascade reactions, didn't furnish the expected diamination products 107/108 in EtOH via self-catalysis and produced only amination product 106aa (not shown in Table 12). Interestingly, pyrrolidine as reagent in self-catalyzed cascade reactions in DMSO furnished the amination product 106aa as major product and diamination products 107/108 as minor (Table 12, entry 1). Same reaction in protic solvents furnished the amination product 106aa and diamination 107aa/108aa products with poor regioselectivity (entries 2 and 3). Slow addition of nitrosobenzene 56a to Hagemann's ester 88a under pyrrolidine self-catalysis in EtOH furnished the expected o-pyrrolidin-1-yl-diarylamines 107aa and 108aa in 80% yield with good selectivity (entry 6). We envisioned the optimized condition to be slow addition of 56a to the mixture of 88a, 2a and MS 4A° at 25 °C in EtOH to furnish o-pyrrolidin-1-yl-diarylamines 107aa/108aa in 90% yield with 33:1 regioselectivity (entry 8). A mechanistic aspect of this cascade EA/IA reaction is discussed in next section.

Hagemann's esters 88a-b, 88i-s were prepared from alkyl acetoacetates and aldehydes with high yields in one-step according to literature procedures of Sangho Koo method^{44a} and Bhaduri method^{44b} with minor modifications as shown in Table A1. With an efficient organocatalytic cascade protocol in hand, the scope of the auto and selfcatalyzed EA/IA cascade reactions was investigated with various Hagemann's esters 88a-s and nitrosoarenes 56a-c. 46 A series of 6-substituted Hagemann's esters 88a-s were reacted with 0.5 equivalents of nitrosoarenes 56a-c catalyzed by 5 mol% of pyrrolidine or piperidine at 25 °C in DMSO (Table 13). The o-hydroxy-diarylamines 106 were obtained as single isomers with excellent yields. The reaction of 88a with 1methyl-2-nitroso-benzene **56b** furnished the o-hydroxydiarylamine **106ab** as a single isomer, in good yield (Table 13). Synthesis of o-hydroxydiarylamine 106ac from 88a, 56c and 2a at 25 °C taken longer reaction time (12 h), but reaction at 65 °C furnished the product **106ac** with good yield within 2.5 h (Table 13). Both aliphatic and aromatic substituted Hagemann's esters 88i-r were generated expected o-hydroxy-diarylamines 106 with nitrosoarenes 56a-c excellent yields (Table in 13).

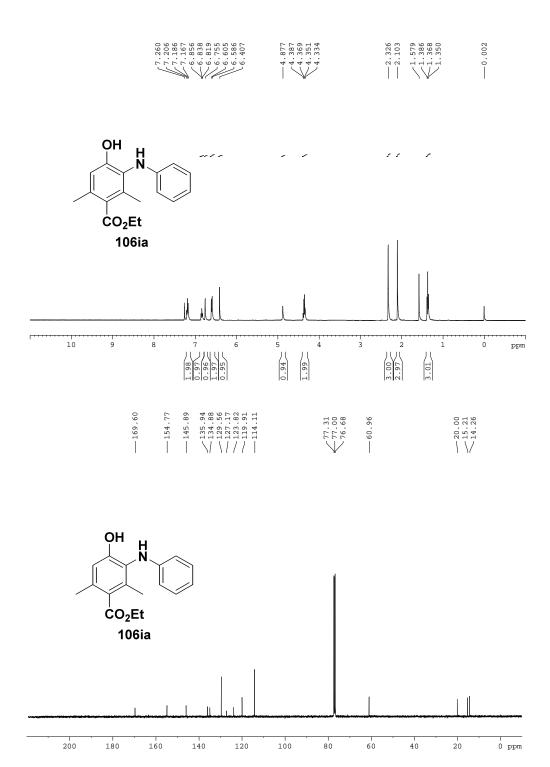


Figure-11: ¹H NMR and ¹³C NMR Spectrum of product **106ia**.

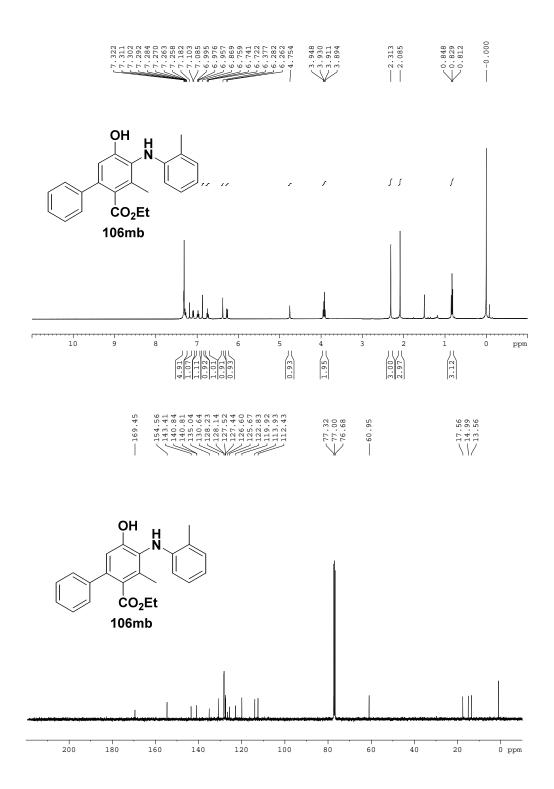


Figure-12: ¹H NMR and ¹³C NMR Spectrum of product **106mb**.

 $N^{\not\sim O}$ Pyrrolidine (5 mol%) DMSO (1 mL) RT, 0.5-1.0 h 88a-s CO₂R¹ 56а-с CO₂R¹ 106 OH ĊO₂Et ĊO₂tBu CO₂Me ĊO₂Et ĊO₂Et 85% (**106aa**) 85% (**106sa**) 80% (106ba) 86% (106ab) 90% (106ia) ĊO₂Et ĊO₂Et ĊO₂Et ĊO₂Et ĊO₂Et 85% (**106ka**) 95% (**106ja**) 90% (**106la**) 90% (**106lb**) 96% (106ma) OН ĊO₂Et ĊO₂Et ĊO₂Et ĊO₂Et 95% (**106pa**) MeO 93% (**106mb**) 90% (106na) 80% (106qa) ĊO₂Et ĊO₂Et CO₂Et ĊO₂Et 65% (106ac)^b 80% (**106oa**)^b 85% (**106ra**)^b 75% $(106ac)^{b,c}$

Table 13: Chemically Diverse Libraries of o-Hydroxydiarylamines^a

^a Yield refers to the column purified product. ^b Piperidine (5 mol%) used as catalyst. ^c Reaction performed at 65 °C.

Rate of EA/IA cascade reactions are accelerated by in situ generated products **106** and these reactions are ideal examples for the bio-mimetic auto-catalysis of functionalized amines in organic reactions.^{27b-c,47} Structure and regio-chemistry of o-hydroxy-diarylamines **106** were confirmed by ¹H NMR, ¹³C NMR [for example see Fig. 9,11-12] and mass analysis and also finally conformed by X-ray structure analysis on **106aa** as shown in scheme 10.⁴⁸ With the success of cascade synthesis of highly functionalized o-hydroxy-diarylamines **106**, we continued our investigation for generation of highly

functionalized diversity oriented library of cascade o-pyrrolidin-1-yl-diarylamines **107** under self-catalysis.

Scheme 10: Crystal structure of 4-Hydroxy-2-methyl-3-phenylamino-benzoic acid ethyl ester (**106aa**)

$$= \bigvee_{\text{CO}_2\text{Et}}^{\text{OH}} \bigvee_{\text{N}}^{\text{H}}$$
106aa

The results in Table 14 demonstrate the broad scope of this novel green methodology covering a structurally diverse group of Hagemann's esters **88a-s**, pyrrolidine and nitrosobenzenes **56a-c**. Cascade EA/IA reaction of Hagemann's esters **88b & 88s**, nitrosobenzene **56a** and pyrrolidine furnished the regioselective diamines **107ba** and **107sa** in 11:1 ratio with >85% yield (Table 14). Unexpectedly, cascade product **107ab** furnished with moderate yield in 3:1 isomeric ratio from **88a**, pyrrolidine and **56b**. 4-*N*,*N*-Dimethyl nitrosobenzene **56c** didn't furnished the expected cascade EA/IA product**107ac** at 25 °C, but given only <5% of expected **107ac** along with 40% of o-hydroxy-diarylamine **106ac** at 65 °C in EtOH, may be due to the electronic factor of NMe₂ group. Interestingly, all 6-substituted Hagemann's esters **88i-r** furnished expected o-pyrrolidin-1-yl-diarylamines **107ia-ra** with good yields as single isomer in self-catalyzed EA/IA cascade reactions as shown in Table 14. Structure and regiochemistry of o-pyrrolidin-1-yl-diarylamines **107** were confirmed by ¹H NMR, ¹³C NMR [for example see Fig. 10,13-14] and mass analysis and also finally confirmed by X-ray structure analysis on **107pa** as shown in scheme 11.⁴⁹

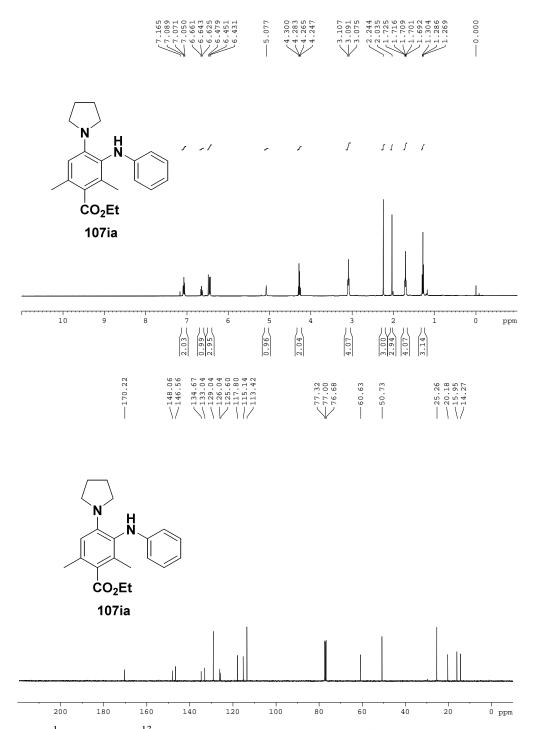


Figure-13: ¹H NMR and ¹³C NMR Spectrum of product 107ia

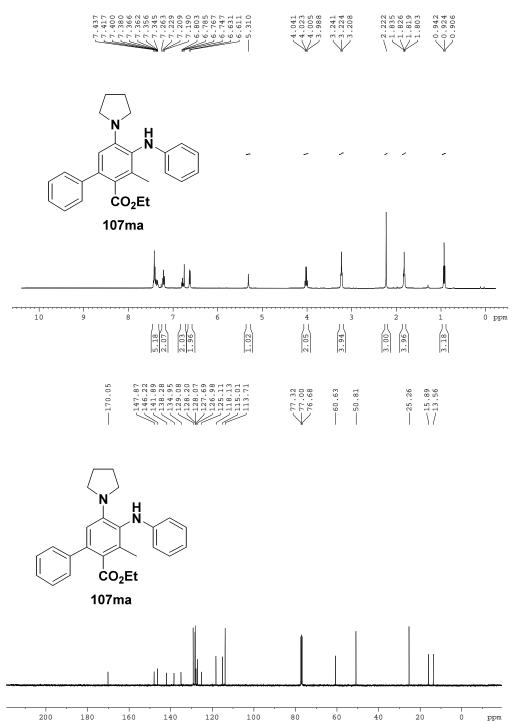


Figure-14: ¹H NMR and ¹³C NMR Spectrum of product **107ma**.

Table 14: Chemically Diverse Libraries of o-Pyrrolidin-1-yl-diarylamines^a

^a In all reactions, 10-35% of corresponding substituted o-hydroxy-diarylamines **106** were isolated. Yield refers to the column purified product. ^b 11:1 ratio of regio-isomers **107/108** are isolated. ^c 3:1 ratio of **107/108** are isolated. ^d Reaction performed at 65 °C for 2.5 h and 40% of product **106ac** were isolated.

Scheme 11: 4'-Chloro-3-methyl-4-phenylamino-5-pyrrolidin-1-yl-biphenyl-2-carboxylic acid ethyl ester (107pa).

$$= \bigvee_{\substack{\mathsf{CO}_2\mathsf{Et}\\ \mathbf{107pa}}} \overset{\mathsf{H}}{\mathsf{H}}$$

After successful demonstration of the piperidine-catalyzed cascade K/M/A/DC and EA/IA reactions, we decided to investigate the combination of these two cascade reactions in one-pot. Reaction of two equivalents of ethyl acetoacetate and benzaldehyde under piperidine-catalysis in EtOH at 80 °C for 3-6 h furnished the expected Hagemann's ester 88m, which on treatment with nitrosobenzene 56a at 25 °C in same solvent didn't furnished the expected o-hydroxydiarylamine 106ma in good yield; but removing the solvent EtOH by vacuum pump and adding solvent DMSO, 20 mol% of piperidine and nitrosobenzene 56a to the reaction mixture of cascade K/M/A/DC furnished the expected o-hydroxydiarylamine 106ma in good yield as shown in Table 15. Successful combination of two cascade K/M/A/DC and EA/IA reactions under piperidine-catalysis was demonstrated by two more examples as shown in Table 15 and this one-pot synthetic strategy will show much impact on synthesis of functionalized small molecules.

Table 15: Combination of Cascade Knoevenagel/Michael/Aldol Condenasation/Decarboxylation and Cascade Enamine Amination/Iso-aromatization Reactions in One-Pot^{a,b}

Table 16: Amine-Cs₂CO₃-Catalyzed Enamine Amination/Iso-aromatization/Alkylation Reactions in One-Pot^{a,b}

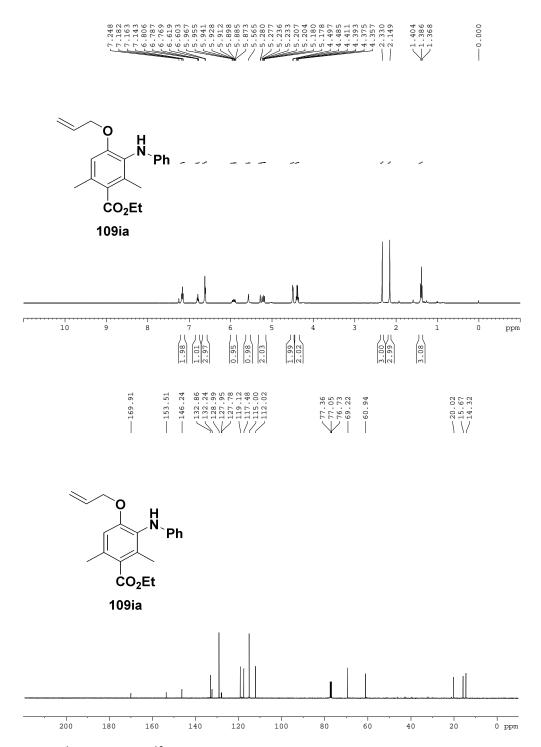


Figure-15: ¹H NMR and ¹³C NMR Spectrum of product 109ia.

With pharmaceutical applications in mind, we extended the two-component cascade EA/IA reactions into a novel amine/Cs₂CO₃-catalyzed three-component EA/IA/A reaction of **88a** and **88i**, **56a** with allyl and propargyl bromides in one-pot as shown in Table 16. o-Alkyloxy-diarylamines **109** were constructed in good yields with high selectivity as shown in Table 16. The products structures were confirmed by ¹H NMR, ¹³C NMR [for example see Fig. 15] and mass analysis. This method will be showing much impact on synthesis of carbazole alkaloids. ^{41a}

The possible reaction mechanism for regioselective synthesis of cascade products **106** and **107** through reaction of Hagemann's ester **88a**, nitrosobenzene **56** and pyrrolidine **2a** is illustrated in Scheme 13. First, reaction of pyrrolidine **2a** with Hagemann's ester **88a** generates the imine cation **94aa**, which will transform into both dienamines **95** (thermodynamic stable product, major) and **105** (kinetic product, minor) based on reaction conditions. The energy difference (ΔH) between the two dienamines **95** and **105** is 4.698 kcal/mol based on AM1 and 4.367 kcal/mol based on PM3 calculations. Minimized structures of thermodynamic stable and kinetic stable enamines **95** and **105** based on MOPAC calculations are depicted in the Scheme 12.⁵⁰

Scheme 12: Minimized structures of thermodynamic stable and kinetic stable enamines 95 and 105 based on MOPAC calculations

$$\equiv$$
 CO_2 Et 95 $\Delta H = -83.31689 \text{ kcal/mol}$ $\Delta H = -78.94943 \text{ kcal/mol}$

Since the difference in ΔH 's between the two dienamines of **95** and **105** are >4 kcal/mol, formation of thermodynamic stable dienamine **95** will be major under mild

organocatalysis conditions. Slow addition of nitrosoarenes **56** to the reaction mixture of **88** and **2a** controlled the formation of kinetic dienamine **105**, may be due the basic nature of nitrosoarenes **56** and this was supported by results in Tables 12 and 14. Reaction of push-pull dienamine **95** with **56** furnish the selectively nitroso aldol product **112**, which will give imine product **113** by losing hydroxide ion. Hydrolysis followed by iso-aromatization of imine product **113** converted into highly substituted o-hydroxydiarylamine **106** under amine catalysis. Imine product **113** was transformed into highly substituted o-pyrrolidin-1-yl-diarylamines **107** via iso-aromatization under

Scheme 13: Proposed Reaction Mechanism

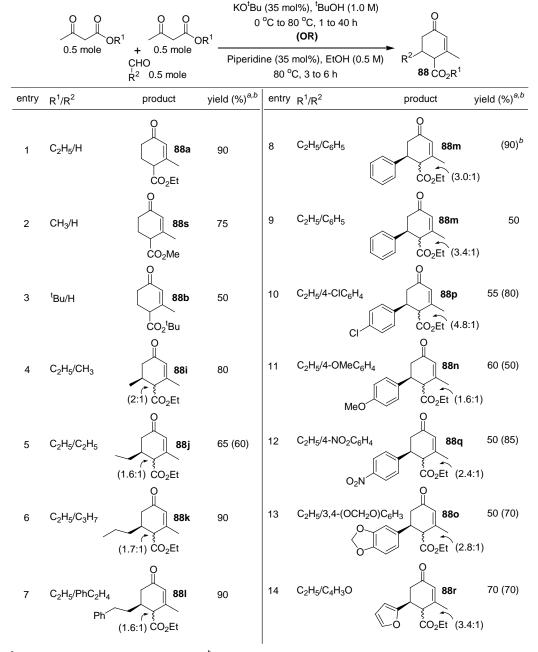
suitable conditions (EtOH and MS 4A°). Hydrolysis of imine 113 is more solvent dependent as shown in Tables 11 and 12, means this hydrolysis step fast in DMSO than in EtOH may be due to the more interactions with water. Similar manner, regio-isomer 108 also furnished from kinetic dienamine 105. As we discussed in this chapter, these cascade reactions are auto-catalyzed and product 106 can catalyze the nitroso aldol reaction of enolate 110 of 88a with 56 to form 117 via hydrogen bonding transition state 116, which will transform into expected product 106 through imine 114 as shown in Scheme 13 (see Table 11, entries 13 and 14).

4.3 CONCLUSION

In this chapter, we have described the metal-free synthesis of highly substituted anilines **106**, **107** and **109** from simple starting materials via cascade EA/IA and K/M/A/DC/EA/IA and EA/IA/A reactions under amine catalysis. The cascade reaction proceeds in good yields with high selectivity using pyrrolidine or piperidine as the catalyst. Furthermore, we have demonstrated the bio-mimetic auto and self-catalysis in amine-catalyzed cascade reactions.

ANNEXURE-I: DIRECT ORGANO- OR KO'BU-CATALYZED CASCADE SYNTHESIS OF HAGEMANN'S ESTERS.

Table A1: Direct Organo- and KO^tBu-Catalyzed Cascade Synthesis of Chemically Diverse Libraries of Hagemann's esters **88**.



^a Yield refers to the column purified product. ^b Yield in parentheses obtained from piperidine (35 mol%) catalysis.

5. AMINO ACID-CATALYZED CASCADE [3+2]-CYCLOADDITION/HYDROLYSIS REACTIONS BASED ON THE PUSH-PULL DIENAMINE PLATFORM: SYNTHESIS OF HIGHLY FUNCTIONALIZED MH-1,2,3-TRIAZOLES

5.1 INTRODUCTION

1,2,3-Triazoles are an important class of heterocycles, which display very large spectrum of biological activities and are widely used as pharmaceuticals and agrochemicals.⁵¹ Also compounds containing 1,2,3-triazoles have found industrial applications as corrosion inhibitors, lubricants, dyes, and photostabilizers.⁵¹ As such, the development of new and more general methods for their preparation is of significant interest.⁵² The conventional method to triazoles is the Huisgen 1,3-dipolar cycloaddition of alkynes with azides. Recent discovery of the novel technology of Cu^I-catalyzed [3+2]-cycloaddition reactions of terminal alkynes with organic azides to provide a general route to a variety of 1,4-disubstituted 1,2,3-triazoles in good yields, and it has become a paradigm of a "click chemistry" reaction.⁵³ The advent of click reaction technology triggered a burst of activity in the synthesis of a huge variety of differently substituted 1,2,3-triazoles as *in vitro* and *in vivo* conditions.⁵⁴

Recently, the copper-catalyzed azide-alkyne click reaction has proven extremely valuable for attaching small molecular probes to various biomolecules in a test tube or on fixed cells.⁵⁵ However, its use for biomolecule labeling in live cells or organisms is prohibited by the requirement of a cytotoxic copper catalyst.⁵⁵

As part of our program to engineer direct organocatalytic cascade reactions, ^{14b,27a-f,27h,56} herein we have discovered a copper-free, novel and green technology for the synthesis of highly substituted α-diazo compounds and *N*H-1,2,3-triazoles using organocatalytic cascade enamine amination/elimination (EA/E) and [3+2]-cycloaddition/hydrolysis ([3+2]-CA/H) reactions from commercially available activated enones, azides and amines/amino acid (eq. 30). In this chapter, first time we discovered the organocatalytic approach to the synthesis of *N*H-1,2,3-triazole products.

Over the last few years, we have been interested in amine mediated in situ generation and application of novel push-pull dienamines/push-pull dienols (**A** and **B**, see eq. 31) in cascade reactions from Hagemann's esters with nitrogenated species for the formation of C-N bonds. ^{1a-c,57} During our investigation for new coupling species for such processes, we decided to explore the potential ability of the organic azides to participate in an amine/amino acid-catalyzed coupling reaction. We expected that the coupling of an organic azide with in situ generated push-pull dienamines would lead to protected 1,2,3-triazoles. However, protected 1,2,3-triazoles were not detected and instead *N*H-1,2,3-triazoles were obtained under the standard reaction conditions. This unexpected cascade result represents a novel methodology for the preparation of *N*H-1,2,3-triazoles and a new reactivity for amino acid catalysts. Herein, we report our findings regarding these new amino acid-catalyzed cascade reactions.

5.2 RESULTS AND DISCUSSION

We initiated our preliminary studies of the cascade EA/E or [3+2]-CA/H reactions by screening a number of known and novel organocatalysts for the amination of

Hagemann's ester **88a** by 0.5 to 1.0 equivalents of p-toluenesulfonyl azide (TsN₃) **118a** and some representative results are shown in Table 17. Interestingly, reaction of **88a** with 1.0 equiv of TsN₃ **118a** in DMSO under 20-mol% of glycine **14d**-catalysis furnished the cascade [3+2]-CA/H product NH-1,2,3-triazole **121aa** as single product with only 25% yield (Table 17, entry 1). Same reaction under 20-mol% of L-Proline **14a**-catalysis also furnished the NH-1,2,3-triazole **121aa** as single product with 55% yield (Table 17, entry 2). Interestingly, reaction of **88a** with 1.0 equiv of **118a** under

Table 17: Preliminary study for the reaction optimisation. [a]

O CO ₂ Et	catalyst HNR ₂ - Ts-N ₃ 0.3 M DMSO RT	O N_2 CO_2Et	+	1-N N + O ₂ Et	HN-N N CO ₂ Et
88a	118a	119aa	12	0aa	121aa
Entry	Catalyst	<i>t</i> [h]		Yield[%] ^{[b}]
	[20 mol%]		119aa	120aa	121aa
1	glycine (14d)	96	_	=	25
2	proline (14a)	24	-	_	55
3	diamine (2i) ^[c]	0.75	83	_	-
4	piperidine (2b)	0.75	83	_	-
5	morpholine (2c)	0.75	73	_	-
6	pyrrolidine (2a)	1.0	67	_	-
7	benzylamine (2d)	0.75	83	_	-
8 ^[d]	2d	0.75	85	_	-
9 ^[e]	2d	0.70	90	-	-
10 ^[f]	2d	0.75	80	_	-
11 ^[g]	2d	0.75	70	_	-
12 ^[h]	2d	0.75	80	_	_
13	Et ₃ N (2g)	1.5	65	-	_
14	$Me_2NCH_2CH_2OH$ (2k)	1.5	71	-	-
15	DBU (2h)	1.5	58	_	_
16	DABCO (2I)	1.5	77	_	_
17	DMAP (2f)	1.5	67	=	_

[a] Reactions were carried out in solvent (0.3 M) with 1.0 equiv. of **88a** relative to the **118a** in the presence of 20 mol% of catalyst . Ts = p-toluenesulfonyl; [b] Yield refers to the column-purified product. [c] (S)-1-(2-Pyrrolidinyl methyl)pyrrolidine (**2i**). [d] 30 mol% of **2d** was used. [e] 40 mol% of **2d** was used. [f] 2.0 equiv. of **88a** was used. [g] DMF used as solvent. [h] NMP used as solvent.

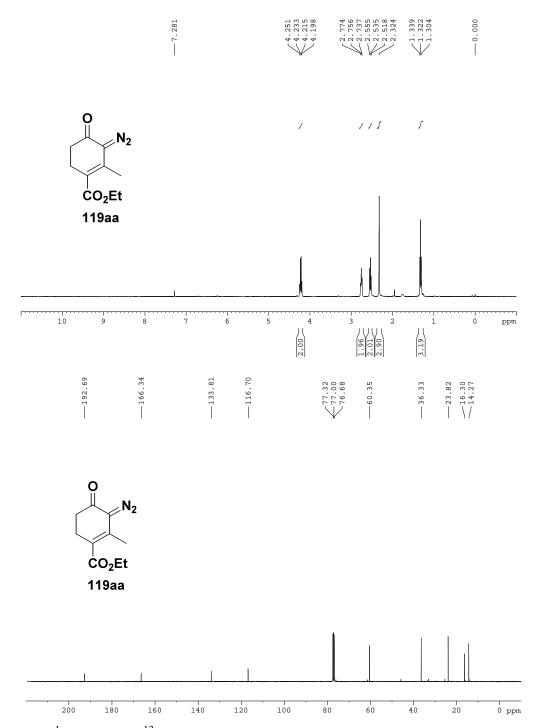


Figure-16: ¹H NMR and ¹³C NMR Spectrum of product **119aa**.

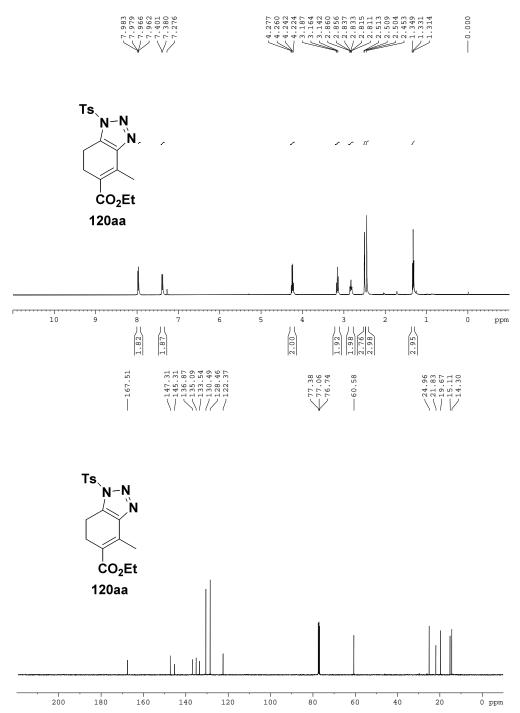


Figure-17: ¹H NMR and ¹³C NMR Spectrum of product **120aa**.

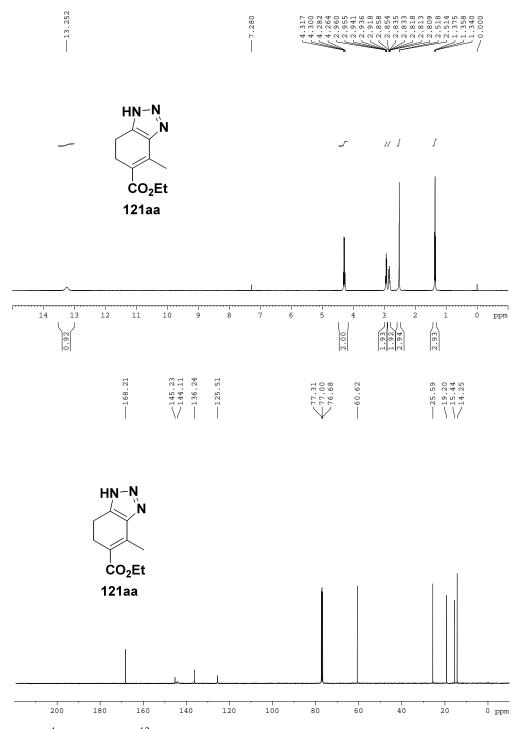


Figure-18: ¹H NMR and ¹³C NMR Spectrum of product **121aa**.

diamine 2i-catalysis generated the cascade EA/E diazo-product 119aa with 83% yield in DMSO and there are no products from the cascade [3+2]-CA/H sequence (Table 17, entry 3). Secondary-amines like piperidine 2b, morpholine 2c and pyrrolidine 2a catalysts also furnished the cascade EA/E diazo-product 119aa with good yields in DMSO solvent (entries 4, 5 and 6). Primary amine, benzylamine 2d also catalyzed the formation of cascade diazo-product 119aa in very good yield (entries 7-9). Benzylamine-catalyzed cascade EA/E reactions are solvent dependent reactions, working well in aprotic polar solvents like DMSO, DMF and NMP but only <15% conversion is observed in other solvents like EtOH, CH₃CN, CHCl₃, THF, H₂O and [bmim]BF₄ (Table 17, entries 11-12). Addition of 20-mol % of simple tertiary-amines such as Et₃N 2g, Me₂NCH₂CH₂OH 2k, DBU 2h, DABCO 2l and DMAP 2f as catalyst in DMSO at 25 °C for 1.5 h furnished the cascade EA/E diazo-product 119aa as single compound in 65-77% yields as shown in Table 17, entries 13-17. We envisioned the optimized condition to be 25 °C in DMSO under 20-40 mol% benzylamine 2d catalysis to furnish the highly substituted diazo-product **119aa** in 83-90% yield (Table 17, entries 7-9).

Reaction of **88a** with **118a** under amino acid **14a**-catalysis furnished the interesting [3+2]-CA/H cascade product **121aa** as single product (Table 17, entry 2). Further to improve the reaction yield, we initiated our studies of the cascade [3+2]-CA/H reaction by screening a number of reaction conditions for the coupling of Hagemann's ester **88a** by 0.5 to 1.0 equiv. of TsN₃ **118a** under amino acid **14a**-catalysis, and some representative results are shown in Table 18. Reaction of **88a** with 1.0 equiv. of TsN₃ **118a** in EtOH under 20-mol% of proline **14a**-catalysis furnished the cascade [3+2]-CA/H product *N*H-1,2,3-triazole **121aa** as single product with only 30% yield (Table 18, entry 1). Interestingly, same reaction in MeOH furnished the unhydrolyzed 1,2,3-triazole **120aa** as single product with 35% yield (Table 18, entry 2). Reaction of **88a** with 1.0 equiv. of **118a** under **14a**-catalysis in DMSO at 70 °C for 5 h furnished the cascade [3+2]-CA/H product **121aa** in 65% yield (Table 18, entry 4).

Table 18: Reaction optimisation for the NH-1,2,3-triazole synthesis.

	+ Ts-N O ₂ Et 88a 118	Solvent (>	Ts N-N CO ₂ E	N + C	N-N N CO ₂ Et
Entry	Solvent [0.5 M]	H. ester 88a [eq.]	Temp [°C]	Time [h]	Products 120aa	yield [%] ^[a] 121aa
1	EtOH	1.0	25	120	-	30
2	MeOH	1.0	25	120	35	=
3	DMSO	1.0	25	24	-	55
4	DMSO	1.0	70	5	-	65
5 ^[b]	DMSO	1.0	25	24	-	76
6 ^[c]	DMSO	1.0	25	24	-	90
7	DMF	1.0	25	24	55	10
8	NMP	1.0	25	24	60	10
9	NMP	2.0	25	19	70	10
10	DMSO	2.0	25	24	-	94
11	DMSO	2.0	80	5	-	91
12	DMSO	0.5	80	5	=	75

^[a] Yield refers to the column purified product. ^[b] 40 mol% of proline were used. ^[c] 50 mol% of proline were used.

Increasing the catalyst **14a** loading from 20-mol% to 50-mol% or substrate **88a** loading from 1.0 equiv. to 2.0 equiv. yield of cascade product **121aa** increased drastically from 55% to 90/94% at 25 °C for 24 h in DMSO solvent as shown in Table 18, entries 5-6 and 10-11. Interestingly, amino acid-catalyzed cascade [3+2]-CA/H reactions are also solvent dependent reactions, working well in aprotic polar solvents like DMSO, DMF and NMP but only <15% conversion is observed in other solvents like CH₃CN, CHCl₃, THF, H₂O and c-C₆H₁₂ (Results not presented in table 18). Cascade reaction of **88a** with **118a** under **14a**-catalysis in DMF at 25 °C for 24 h furnished the unhydrolyzed 1,2,3-triazole **120aa** in 55% yield accompanying with

product **121aa** in 10% yield as shown in Table 18, entry 7. Same cascade reaction in NMP as solvent furnished the products **120aa** and **121aa** in 60% and 10% yields respectively (Table 18, entry 8). We envisioned the optimized conditions to be addition of 2.0 equiv. of **88a** to TsN₃ **118a** under 20-mol% proline **14a**-catalysis in DMSO or NMP at 25 °C for 24 h to furnish the cascade products **121aa** and **120aa** in 94% and 70% yields respectively (Table 18, entry 9 and 10). The products structures were confirmed by ¹H and ¹³C NMR [for example see Fig. 16-18], mass analysis and structure and regiochemistry of 4-methyl-1-(toluene-4-sulfonyl)-6,7-dihydro-*1H*-benzotriazole-5-carboxylic acid ethyl ester **120aa** and 4-methyl-6,7-dihydro-*1H*-benzotriazole-5-carboxylic acid ethyl ester **121aa** was finally confirmed by X-ray structure analysis as shown in Scheme 14 and Scheme 15^{58,59}

Scheme 14: Crystal structure of 4-methyl-1-(toluene-4-sulfonyl)-6,7-dihydro-1H-benzotriazole-5-carboxylic acid ethyl ester (**120aa**).

$$= \begin{pmatrix} H_3C \\ O \\ N-N \\ N \\ CO_2Et \end{pmatrix}$$
120aa

Scheme 15: Crystal structure of 4-methyl-6,7-dihydro-1H-benzotriazole-5-carboxylic acid ethyl ester (**121aa**).

$$= \bigvee_{CO_2Et}^{HN-N}$$

Table 19: Effect of an azides 118a-g on the cascade EA/E and [3+2]-CA/H reactions. [a]

Condition A

	O + R-N ₃	— P	zylamine (40 mol' MSO (0.3 M), RT (or) Condition B roline (20 mol%) MSO (0.5 M), RT	O N ₂ + [R N-N N + CO ₂ Et	HN-N N CO ₂ Et	
	88a 118a-g		(213 11.),		120aa-ag	121aa	
	Condition	Α			Condition	n B	
Entry	R-N ₃ [1.0 equiv.]	Time [h]	Yield [%] ^[b] 119aa	R-N ₃ [0.5 equiv.]	Time [h]	Products yie	eld [%] ^[b] 121aa
1	4-CH ₃ C ₆ H ₄ SO ₂ N ₃ 118a	0.75	90	4-CH ₃ C ₆ H ₄ SO ₂ N ₃ 11	1 8a 24	-	94
2	CH ₃ SO ₂ N ₃ 118b	0.75	84	CH ₃ SO ₂ N ₃ 118b	12	-	78
3	C ₂ H ₅ OCON ₃ 118c	72	-	C ₂ H ₅ OCON ₃ 1180	48	-	30
4	$4\text{-NO}_2\text{C}_6\text{H}_4\text{SO}_2\text{N}_3 \textbf{118d}$	26	20	4-NO ₂ C ₆ H ₄ SO ₂ N ₃ 11	1 8d 48	_	30
5	$2\text{-NO}_2\text{C}_6\text{H}_4\text{SO}_2\text{N}_3 \textbf{118e}$	12	20	2-NO ₂ C ₆ H ₄ SO ₂ N ₃ 11	1 8e 48	_	30
6	$C_6H_5CH_2N_3$ 118f	48	-	$C_6H_5CH_2N_3$ 118f	48	-	_
7	(CH ₃) ₃ SiN ₃ 118g	72	=	(CH ₃) ₃ SiN ₃ 118g	48	-	_

[a] See Experimental section. [b] Yield refers to the column-purified product.

After successful demonstration of cascade EA/E and [3+2]-CA/H reactions from **88a** with TsN₃ **118a** under amine or amino acid-catalysis, we also investigated the amine or

amino acid-catalyzed cascade reaction of **88a** with the other azides⁶⁰ like MsN₃ **118b**, N₃CO₂Et **118c**, 4-NO₂C₆H₄SO₂N₃ (*p*NBSA) **118d**, 2-NO₂C₆H₄SO₂N₃ (*o*NBSA) **118e**, BnN₃ **118f** and TMSN₃ **118g**, but the diazotization or *N*H-1,2,3-triazole formation of the **88a** was inferior compared to TsN₃ **118a** as shown in Table 19, entries 1–7.

With the optimized reaction conditions in hand, the scope of the amino acid- and amine-catalyzed [3+2]-CA/H and EA/E cascade reactions was investigated. A series of substituted Hagemann's esters **88a-t** were reacted with 0.5 equivalents of azides **118a-b** catalyzed by 20-mol% of proline 14a at 25 °C in DMSO (Table 20). Both aliphatic and aromatic substituted Hagemann's esters 88 were furnished the expected NH-1,2,3triazoles 121 with excellent yields (Table 20). Yields of the cascade [3+2]-CA/H products 121 were increased by the prolonging reactions time up to 24 h and also water content of the DMSO solvent; and these are ideal examples for the bio-mimetic solvent induced cascade chemistry in organic reactions. For example, proline-mediated reaction of simple Hagemann's esters 88f, 88t and 88e⁶¹ with TsN₃ 118a in DMSO at 25 °C for 1 h furnished the expected unhydrolyzed 1,2,3-triazoles 120fa, 120ta and 120ea in 50%, 70% and 65% yields respectively as shown in Table 20, entries 11-13. But, unfortunately we are not observed the enantioselectivity in the L-proline 14a-mediated cascade reaction of 6-substituted Hagemann's esters 88 with TsN₃ 118a. Fascinatingly, reaction of 3,4-dihydro-1H-naphthalen-2-one 88u with azides TsN₃ 118a or MsN₃ 118b at 25 °C for 1 h in DMSO under amino acid-catalysis furnished the expected 1,2,3-triazoles 120ua/120ub in combination with 121ua/121ub in almost quantitative yields with excellent regioselectivity as shown in Table 20. The products structures were confirmed by ¹H and ¹³C NMR [for example see Fig. 19-21], mass analysis and regiochemistry of cascade products 120ua-ub/121ua-ub was finally confirmed by Xray structure analysis on **120ua** as shown in Scheme 16.⁶²

Table 20: Chemically diverse libraries of NH-1,2,3-triazoles 121

	+ Ts-N ₃	Proline 14a (20 mol%) DMSO (0.5 M) RT, 0.75 – 24 h	120 CO ₂ R	+ FG N 121 CO ₂ R	(1)
Entry	Product	Yield[%] ^[a]	Entry	Product	Yield[%] ^[a]
1	HN-N N CO ₂ Et 121	94 laa	8	HN-N N CO ₂ Et 1210a	70
2	HN-N N CO ₂ Me 12	90 1sa	9 MeO	HN-N N CO ₂ Et 121na	65 a
3	HN-N N CO ₂ ^t Bu 12	90 1ba	10 O ₂ N	HN-N N CO ₂ Et 121qa	55
4	HN-N N CO ₂ Et 12	65 Iia	11	Ts N-N N CO ₂ Et 120fa	50
5 Pr	HN-N CO ₂ Et 121	75 Ila	12	Ts N-N N CO ₂ Me 120ta	70
6	HN-N N CO ₂ Et 12	80 1 ma	13	Ts. N-N N CO ₂ Et 120ea	65 1
7	HN-N N O CO ₂ Et 12	70 Ira	14 ^[b] 15 ^{[b][c]}	Ts	80 80

[a] Yield refers to the column-purified product. For the entries 1-10, reaction time is 12-24 h and for the entries 11-15, time is 0.75-1.0 h. [b] In both reactions, 15-20% of corresponding hydrolyzed MH-1,2,3-triazole 121ua were isolated. [c] 118b was used instead of 118a.

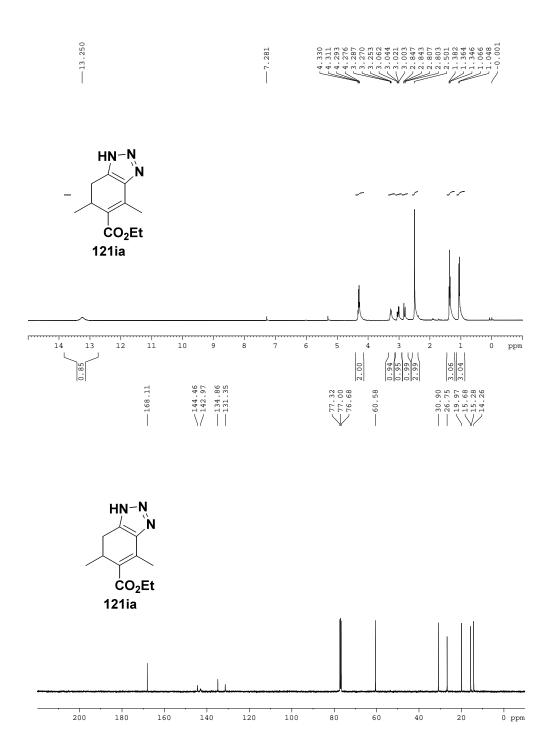


Figure-19: ¹H NMR and ¹³C NMR Spectrum of product 121ia.

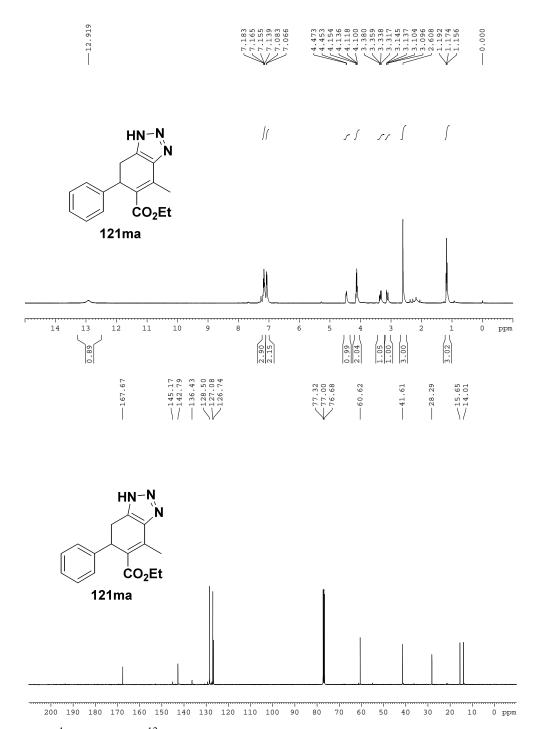


Figure-20: ¹H NMR and ¹³C NMR Spectrum of product **121ma**.

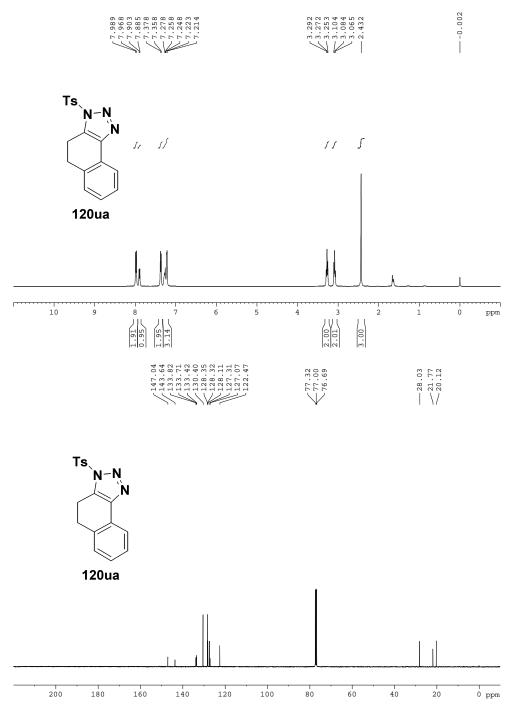


Figure-21: ¹H NMR and ¹³C NMR Spectrum of product **120ua**.

Cascade products **120ua/121ua** would be very interesting model analogues for the studies of inhibitors of hMetAP2 as shown in eq 31.

Scheme 16: Crystal structure of 3-(toluene-4-sulfonyl)-4,5-dihydro-3H-naphtho[1,2-d][1,2,3]triazole (**120ua**).

We also generated the highly functionalized diversity oriented library of cascade EA/E products 119 under primary amine-catalysis. The results in Table 21 demonstrate the broad scope of this novel green methodology covering a structurally diverse group of Hagemann's esters 88a-s and TsN₃ 118a with many of the yields obtained being very good compared to tertiary amine-catalysis. Cascade EA/E reaction of Hagemann's esters 88a-s, TsN₃ 118a under benzylamine 2d-catalysis furnished the diazo-products 119aa-119sa in good yields (Table 21). Structure and regiochemistry of cascade EA/E products 119 was also confirmed by X-ray structure analysis on 119sa as shown in Scheme 17.⁶³

Scheme 17: Crystal structure of 3-diazo-2-methyl-4-oxo-cyclohex-1-enecarboxylic acid methyl ester (119sa).

$$= \bigcup_{\substack{\mathsf{CO}_2\mathsf{Me}\\ \mathsf{119sa}}}^{\mathsf{N}_2}$$

Table 21: Chemically diverse libraries of diazo-products 119.[a]

[a] Yield refers to the column-purified product.

The reaction of *N*H-1,2,3-triazole **121aa** with 1.5 to 2.0 equiv. of DDQ **122** in DCM or benzene as solvent at different temparatures did not furnished the expected aromatized *N*H-1,2,3-triazole **123aa** as shown in Scheme 18.

Scheme 18: Reaction optimisation for the aromatized NH-1,2,3-triazole 123aa synthesis.

The possible reaction mechanism for the synthesis of cascade products 119, 120 and 121 through reaction of Hagemann's ester 88a, TsN₃ 118a and amine/amino acid is illustrated in Scheme 19. The reaction mechanism can be divided into three categories (I, II and III) as catalyzed by three different amines as shown in Scheme 19. In the first category (I), reaction of tertiary amines with Hagemann's ester 88a generates the push-pull dienol 110 based on reaction conditions. Reaction of push-pull dienol 110 with TsN₃ 118a furnish the selectively amination product 125, which will give product 126 via keto-enol tautomerism by treatment with basic amine. Rearrangement followed by elimination of toluene-4-sulfonamide of 126 converted into highly substituted diazo-product 119aa. In the second category (II), reaction of primary amine BnNH₂ 2d with Hagemann's ester 88a generates the push-pull dienamine 127, the which on treatment with TsN₃ 118a furnish the selectively amination product 128, which will transform into product 129 via imine-enamine tautomerism by treatment with basic amine. Rearrangement followed by elimination of toluene-4-sulfonamide of 129 and hydrolysis of resulting imine gives to the highly substituted diazo-product 119aa.

Scheme 19: Proposed reaction mechanism

In the third category (III), reaction of secondary amino acid proline 14a with Hagemann's ester 88a generates the push-pull dienamine 103, ^{1b} which on treatment with TsN₃ 118a furnish the selectively 7a-(2-carboxy-pyrrolidin-1-yl)-4-methyl-1-(toluene-4-sulfonyl)-3a,6,7,7a-tetrahydro-*1H*-benzotriazole-5-carboxylic acid ethyl ester 131 via concerted [3+2]-cycloaddition, which may transform into the product 120aa through rapid elimination of proline 14a. Possibility of weak interactions from acid group in transition state 130 will be the driving force to undergo concerted [3+2]-cycloaddition compare to 124/127 transition states. Solvent (DMSO) induced in situ hydrolysis of resulting 1,2,3-triazole 120aa gives to the *N*H-1,2,3-triazole 121aa in good yields as shown in Scheme 19. This in situ hydrolysis step is totally influenced by the water content of the DMSO solvent and not in the workup stage.

5.3 CONCLUSION

In summary, first time we have developed the synthesis of *N*H-1,2,3-triazole products **120** and **121** from simple starting materials via [3+2]-CA/H reactions under amino acid-catalysis. The cascade reaction proceeds in good yields with high selectivity using proline as the catalyst. Furthermore, we have demonstrated the bio-mimetic solvent induced hydrolysis in amino acid-catalyzed cascade reactions.

6. DIRECT ORGANOCATALYTIC ASYMMETRIC MICHAEL REACTIONS BASED ON THE PUSH-PULL DIENAMINE PLATFORM: SYNTHESIS OF HIGHLY SUBSTITUTED CHIRAL HAGEMANN'S ESTERS

6.1 INTRODUCTION

Asymmetric amino-catalysis has become one of the most important and wide spread area through iminium or enamine activation of carbonyl compounds.^{29h,64} Among them, the direct Michael addition of saturated carbonyl compounds to β-nitrostyrenes through the enamine activation via amino-catalysis providing an expedient access for the development of functionalized molecules.^{10b,29m,40b,65} Recently Barbas^{10b} and his coworkers discovered the novel technology for organocatalytic asymmetric Michael addition of aldehydes/ketones with nitroolefins to provide a general route to a variety of Michael adducts in good yields with high enantioselectivity, which is known as "Barbas-Michael" reaction. The advent of this enamine based Barbas-Michael technology triggered a burst of activity in the synthesis of a huge chiral pool of Michael adducts through bio-mimetic enamine chemistry. After this discovery, the development of more general studies in the field of asymmetric Barbas-Michael reaction of carbonyls with nitroolefins was developed intensively and various effective organocatalytic methodologies have been developed.^{29m,40b,65}

Interestingly, to the best of our knowledge asymmetric Michael addition of α , β -unsaturated carbonyl compounds having γ -hydrogen with nitroolefins via dienamine catalysis are not yet well studied. However, the amine-catalyzed Michael reaction of Hagemann's esters **88** with nitroolefins **35** was not known and the resulting products **133/134** have a wide range of uses in synthetic chemistry. Furthermore, there is no methodology available to prepare achiral compounds **133/134**. Herein, we have reporting a metal-free and novel technology for the asymmetric synthesis of substituted

2-alkyl-3-(2-nitro-1-aryl-ethyl)-4-oxo-cyclohex-2-enecarboxylic acid alkyl esters 133/134 using organocatalytic MA reactions from easily available Hagemann's esters 88, nitroolefins 35 and amines 2 as shown in Scheme 20. As such the development of new and more general catalytic asymmetric methods are having significant interest, because of the synthetic versatility of the Michael adducts.

Recently, Chen *et al.* reported²⁶ the direct asymmetric Michael addition of α,β -unsaturated aldehydes to nitroolefins through the inversion of the inherent electrophilicity of α,β -unsaturated aldehydes via dienamine catalysis, but the direct asymmetric Michael addition of α,β -unsaturated ketones having γ -hydrogen to nitroolefins via dienamine catalysis is not known. Herein, we are reporting a metal-free and novel technology for the asymmetric synthesis of highly substituted functionalized esters **133/134** via the activation of novel chiral push-pull dienamines from Hagemann's esters **88** and nitroolefins **35** as shown in Scheme 20.

As part of our research program to engineer direct organocatalytic cascade reactions,²⁷ we have been interested in amine mediated *in situ* generation and application of novel push-pull dienamines/push-pull dienols in cascade reactions from Hagemann's esters with reactive electrophilic species for the formation of C-C and C-N bonds in single or multiple steps.¹ During our investigation for new reactive species for such processes, we decided to explore the potential ability of the nitrostyrenes **35** to participate in an amine catalyzed Michael addition (MA) with Hagemann's esters **88** as shown in Scheme 20. This result represents a good methodology for the preparation of functionalized chiral cyclohexenones and new reactivity of chiral push-pull dienamines. Herein, we described our findings regarding these new asymmetric Michael reactions. *Scheme 20:* Direct asymmetric Michael reactions based on the push-pull dienamine

platform.

6.2 RESULTS AND DISCUSSION

We initiated our preliminary studies of the MA reactions by screening a number of known and novel organocatalysts for the Michael addition of Hagemann's ester 88a by 1.5 equivalents of β-nitrostyrene 35a and some representative results are shown in Table 22. Reaction of 88a with 1.5 equiv. of β-nitrostyrene 35a in DMSO solvent without catalyst furnished a 1:1 diastereomeric ratio of Michael products 133aa/134aa in 61% yield with 0% ee within 28 hours (Table 22, entry 1). With this interesting result, we screened other solvents like toluene, DCM, THF, CHCl₃, CH₃CN, EtOH, MeOH, acetone, H₂O, 1,4-dioxane and isopropanol without adding catalyst, but these reactions didn't furnished Michael products 133aa/134aa (results not presented in Table 22). We also screened various natural and unnatural amino acids such as (S)-(-)indoline-2-carboxylic acid, hydroxy L-proline, L-phenylalanine, L-tyrosine, L-serine, Lthreonine, L-leucine and O-tert-butyl-L-threonine for the Michael addition of Hagemann's ester 88a with β-nitrostyrene 35a in toluene but these amino acids didn't furnished Michael products 133aa/134aa (results not presented in Table 22). Same reaction under 30-mol% of L-proline **14a**-catalysis to furnish a 1.5:1 dr of Michael products 133aa/134aa in 30% yield with -12% ee and 10% ee in DMSO solvent (Table 22, entry 2). Same reaction of 88a with 1.5 equiv. of 35a under diamine 2i-catalysis in toluene to furnish a 1.3:1 dr of Michael products 133aa/134aa in 30% yield with 2% and 0% ee respectively (Table 22, entry 3). Bifunctional catalyst (S)-1-(2pyrrolidinylmethyl)pyrrolidine/AcOH also catalyzed the MA reaction of 88a with 35a in DMSO to furnish a 1.5:1 ratio of Michael products 133aa/134aa with good yield and 4% ee and -3% ee respectively (Table 22, entry 4). (S)-prolinol **2m** also catalyzed the MA reaction of 88a with 35a in CHCl₃ at 25 °C and 0 °C to furnish a 1.5:1 and 1.8:1 ratio of Michael products 133aa/134aa in 75% and 40% yields with 0/3% ee and -31/0% ee respectively (Table 22, entries 5-6). Same reaction under **2m**-catalysis in toluene to furnish a 1:1.6 ratio of Michael products 133aa/134aa in 60% yield with 11% ee and 2% ee, but under 2n-catalysis in same solvent to furnish a 1:1.6 ratio of

O

₽h

Michael products **133aa/134aa** in 60% yield with 26% ee and 24% ee respectively (Table 22, entries 7-8). We also tested a number of primary and secondary

Ö

₽h

Table 22: Effect of solvent and catalyst on the direct asymmetric Michael reaction^a

Ö

12^f

13

14

15

16

17

18

2q

2r

2s

2t

2u

2v

2w

Toluene

Toluene

Toluene

Toluene

Toluene

Toluene

Toluene

-	+ Ph	×1NO ₂ ———	est (30 mol%) vent (0.1 M)	-	NO ₂ +	NO ₂	
CO ₂ Et			RT	CO ₂ Et	Ē	O ₂ Et	
88a	3	5a		133aa	a	134aa	
Entry	Catalyst	Solvent	Time [h]	Yield [%] ^b	<i>dr</i> ratio ^c 133aa/134aa	ee ^c 133aa/134aa	_ a_
1	_	DMSO	28	61	1:1	0/0	
2	14a	DMSO	8	30	1.5:1	-12/10	
3	2i	Toluene	96	30	1.3:1	2/0	
4 ^d	2i	DMSO	2	60	1.5:1	4/–3	
5	2m	CHCl ₃	4	75	1.5:1	0/3	
6 ^e	2m	CHCl ₃	48	40	1.8:1	-31/0	
7	2m	Toluene	5	60	1:1.6	11/2	
8	2n	Toluene	60	60	1:1.6	26/24	
9	20	Toluene	48	_	-	_	
10	2p	Toluene	120	_	_	_	
11	47b	Toluene	72	_	_	_	

24

72

48

60

60

96

96

35

45

66

70

1.1:1

1.2:1

1:1.2

1:1.2

-49/-8

-55/-11

42/40

19/13

amines such as chiral thiourea 20, L-2-benzhydryl-pyrrolidine 2p, L-DPPOTMS 47b, L- $(3,5-(CF_3)_2)_2$ DPP 2s, Q-NH₂ 2v and CD-NH₂ 2w as catalysts for the MA reaction of 88a with 35a in toluene though they didn't furnished Michael products 133aa/134aa

^a Reactions were carried out in 0.1 M solvent with 1.5 equiv. of nitro styrene **35a** relative to the Hagemann's ester **88a** in the presence of 30 mol% of catalyst. ^b Yield refers to the column purified product. ^c dr and ee determined by HPLC analysis. ^d 30 mol% Aceticacid also used. ^e Reaction p erformed at 0 °C. ^f Reaction was carried out in 0.15 M solvent.

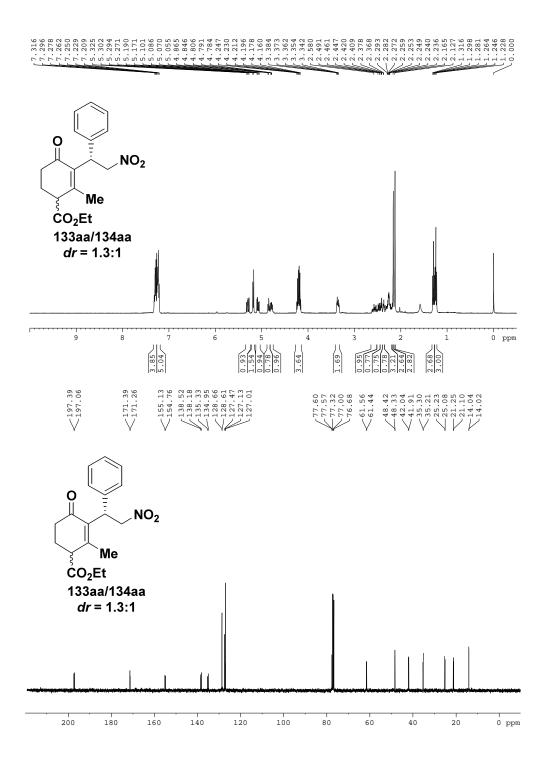


Figure-22: ¹H NMR and ¹³C NMR Spectrum of product **133aa/134aa**.

(Table 22, entries 9-11, 14, 17 & 18). The Michael reaction of **88a** with **35a** under D-DPP **2q**-catalysis in toluene to furnish a 1.1:1 ratio of Michael products **133aa/134aa** in 35% yield with -49/-8% ee respectively as shown in entry 12, Table 22. Interestingly, same reaction with D-(3,5-Me₂)₂DPP **2r** in toluene to furnish a 1.2:1 ratio of Michael products **133aa/134aa** in 45% yield with -55/-11% ee respectively (Table 22, entry 13). Quinine (Q) **2t** catalyzed the MA reaction of **88a** with **35a** in toluene to furnish a 1:1.2 ratio of Michael products **133aa/134aa** in 66% yield with 42/40% ee respectively (Table 22, entry 15).

Table 23: Effect of co-catalyst on the direct asymmetric Michael reaction^a

	+ O ₂ Et	Ph	O ₂ Co-cata	t 2 (30 mol alyst (30 mol oluene (0.1 RT	OI%) M)	Ph NO ₂ +	O Ph NO CO ₂ Et
8	8a	35a				133aa	134aa
_	Entry	Catalyst	Co-catalyst	Time [h]	Yield [%] ^b	<i>dr</i> ratio ^c 133aa/134aa	ее ^с 133aa/134aa
	1	2q	132a	72	50	1:1.3	-38/-31
	2	2q	132b	48	70	1:1.0	-52/-31
	3	2q	132c	36	44	1:1.4	-34/-30
	4 ^d	2q	20	72	69	1.3:1	-66/2
	5 ^d	2q	132d	96	54	1.2:1	–57/7
	6 ^e	2q	132b	72	76	1:1.1	-42/-30
	7	(S)-2r	132e	120	30	1:1.2	59/–4
	8	(<i>S</i>)-2r	132b	72	70	1.3:1	70/–7
	9^{d}	2r	20	72	66	1.2:1	-68/-12
	10	2r	132b	72	80	1.1:1	-69/-15

^a Reactions were carried out in solvent 0.1 M with 1.5 equiv. of nitrostyrene **35a** relative to the Hagemann's ester **88a** in the presence of 30 mol% of catalyst **2** and 30 mol% co-catal -yst. ^b Yield refers to the column purified product. ^c dr and ee determined by HPLC analys -is. ^d 10 mol% co-catalyst used. ^e Reaction performed at 50 °C.

To further improvement of ee/yield of Michael products 133aa/134aa, we also screened co-catalyst effect with urea 132a, thiourea 132b, dicyclohexyl thiourea 132c, chiral phosphoric acid 132d and L-taddol 132e (see Chart 3) on this MA reaction of

Hagemann's ester **88a** by 1.5 equivalents of β-nitrostyrene **35a** in toluene and some representative results are shown in Table 23. MA reaction of **88a** with 1.5 equiv. of **35a** under 30 mol% **2q**, along with 30 mol% urea **132a** as a co-catalyst furnished a 1:1.3 ratio of Michael products **133aa/134aa** in improved yield 50% with only -38% ee and -31% ee respectively (Table 23, entry 1).

Interestingly, same reaction conditions with 30-mol% thiourea 132b as cocatalyst furnished a 1:1 ratio of Michael products 133aa/134aa in 70% yield with -52/-31% ee (Table 23, entry 2). MA reaction of **88a** and **35a** with 30 mol% **2q** as catalyst and 30 mol% dicyclohexylthiourea 132c as co-catalyst to furnish a 1:1.4 ratio of Michael products 133aa/134aa in 44% yield with -34/-30% ee (Table 23, entry 3). MA reaction of 88a and 1.5 equiv. of 35a with 30 mol% 2q and 10 mol% chiral thiourea 2o as co-catalyst to furnish a 1.3:1 ratio of Michael products 133aa/134aa in good yield 69% with -66/2% ee (Table 23, entry 4). MA reaction of **88a** and 1.5 equiv. of **35a** with 30 mol% 2q and 10 mol% chiral phosphoric acid 132d as co-catalyst to furnish a 1.2:1 ratio of Michael products 133aa/134aa in 54% yield with -57/7% ee (Table 23, entry 5). But, under 2q catalysis with thiourea 132b as co-catalyst at 50 °C to furnish a 1.1:1 ratio of Michael products 133aa/134aa in 76% yield with diminished ee values -42/-30% as compared to room temperature reaction (Table 23, entry 6). MA reaction of 88a and 1.5 equiv. of 35a with 30 mol% L-(3,5-Me₂)₂DPP as a catalyst and L-taddol 132e as co-catalyst to furnish a 1:1.2 ratio of Michael products 133aa/134aa in less yield 30% with 59% ee and -4% ee respectively (Table 23, entry 7). Same reaction with 30 mol% D-(3,5-Me₂)₂DPP **2r** as catalyst and 10 mol% chiral thiourea **2o** as co-catalyst to furnish a 1.2:1 ratio of Michael products **133aa/134aa** in 66% yield with -68% ee and -12% ee respectively (Table 23, entry 9). We envisioned the optimized condition to be 25 °C in toluene using 30 mol% L-(3,5-Me₂)₂DPP as catalyst and 30 mol% thiourea 132b as co-catalyst to furnish highly substituted 1.3:1 ratio of Michael products 133aa/134aa in 70% yield with 70% ee and -7% ee respectively (Table 23, entry 8). The product structure and regiochemistry was confirmed by ¹H NMR, ¹³C NMR [for example see Fig. 22] and mass analysis.

Chart 2: Library of catalysts screened for the direct asymmetric Michael reactions.

Chart 3: Library of co-catalysts screened for the direct asymmetric Michael reactions.

After successful demonstration and understanding of the MA reaction of **88a** with **35a** under amine catalysis, the scope and generality of MA reaction was investigated with functionalized Hagemann's esters **88a-b**, **88h** and **88s** and β-nitrostyrenes⁶⁶ **35a-j**. There is no methodology available to prepare achiral Michael products **133/134**, herein we prepared the library generation of the synthesis of achiral Michael products **133/134** in good yields under pyrrolidine **2a** catalysis (Table 24). MA reaction of Hagemann's esters **88a-b**, **88h** and **88s** with structurally diverse group of electron-donating, electron-withdrawing, halogenated and heterocyclic substituted

Table 24: Synthesis of achiral Michael products 133/134.^a

^aReactions were carried out in DMSO (0.5 M) with 1.2 equiv. of **35** relative to the Hagemann's ester **88** in the presence of 20 mol% pyrrolidine **2a**. ^bYield refers to the column purified product. ^cDiastereomeric ratio determ -ined by NMR analysis.

β-nitrostyrenes **35a-j** were generated expected Michael products **133/134** in good to excellent yields with 1.0:1 to 6.6:1 ratio's as shown in Table 24. The structure and regio-chemistry of achiral Michael products **133/134** were confirmed by NMR and mass analysis.

With the optimized reaction conditions in hand, for the asymmetric MA reaction of **88a** with **35a** under amine catalysis, we decided to investigate the asymmetric MA reaction with other functionalized Hagemann's esters **88b**, **88h** and **88s** with other functionalized β-nitrostyrenes **35b-j** in toluene to study the asymmetric synthesis of **133/134** products. A series of β-nitrostyrenes **35** containing different functional groups were reacted with Hagemann's esters catalyzed by 30 mol% of L-(3,5-Me₂)₂DPP with 30 mol% thiourea **132b** as co-catalyst at 25 °C for 72 h in toluene to furnish asymmetric MA products **133/134** in good yields with good to moderate ee's as shown in Table 25.

Asymmetric MA reaction catalyzed by 30 mol% L-(3,5-Me₂)₂DPP with 30 mol% thiourea **132b** on simple Hagemann's ester **88s** with β-nitrostyrene **35a** in toluene at 25 °C to furnish a 1:1.6 ratio of Michael products **133sa/134sa** in 60% yield with 61% ee and 11% ee respectively (Table 25, entry 2). The asymmetric MA reaction of β-nitrostyrene **35a** with *tert*-butyl ester **88b** catalyzed by L-(3,5-Me₂)₂DPP with cocatalyst **132b** in toluene at 25 °C to furnish a 2.1:1 ratio of Michael products **133ba/134ba** in 61% yield with 84% ee and 40% ee respectively (Table 25, entry 3). After this interesting result with *tert*-butyl ester **88b**, we decided to investigate the scope and limitations of the asymmetric MA reaction with other two β-nitrostyrenes **35c/d** with **88b** under L-(3,5-Me₂)₂DPP catalysis at the ambient conditions (Table 25, entries 4-5). Interestingly, MA reaction of 2-chloro-β-nitrostyrene **35c** with **88b** under L-(3,5-Me₂)₂DPP-catalysis to furnish a 1.5:1 ratio of Michael products **133bc/134bc** in 35% yield with 83/33% ee respectively as shown in Table 25, entry 4. MA reaction of 2-(2-nitro-vinyl)-thiophene **35d** with **88b** under L-(3,5-Me₂)₂DPP-catalysis to furnish a 5.2:1 ratio of Michael products **133bd/134bd** in 66% yield with 69/11% ee respectively as

Table 25: Synthesis of chiral Michael products 133/134.^a

^a Reactions were carried out in Toulene (0.1 M) with 1.5 equiv. of **35** relative to Hagemann's ester **88** in the presence of 30 mol% catalyst (*S*)-2*r* and 30 mol% co-catalyst **132b**.^b Yield refers to the column purified product. ^c dr and ee determined by HPLC analysis. ^d Reaction time is 96 h. ^e 30 mol% 2*r* used as catalyst. ^f Reaction performed in 0.2 M solvent.

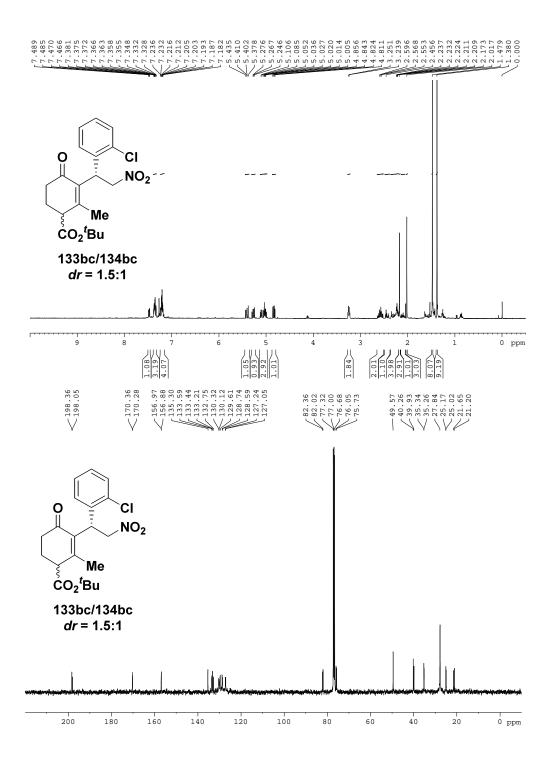


Figure-23: ¹H NMR and ¹³C NMR Spectrum of product 133bc/134bc.

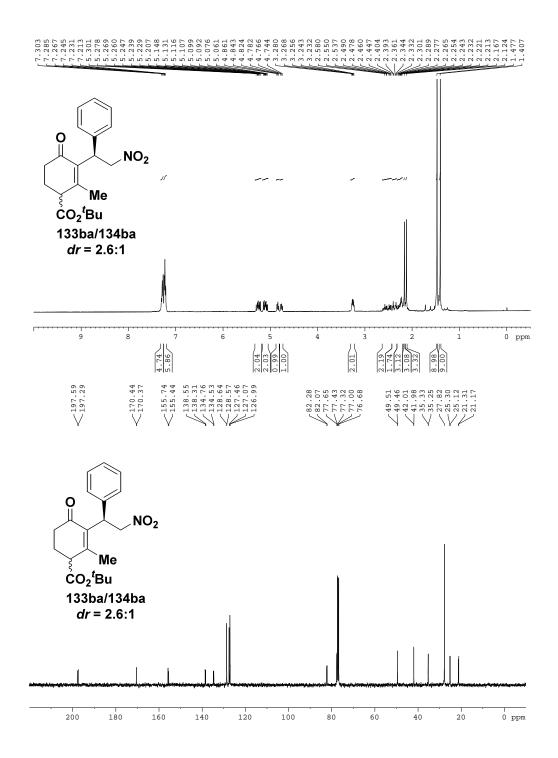


Figure-24: ¹H NMR and ¹³C NMR Spectrum of product 133ba/134ba.

shown in Table 25, entry 5. L-(3,5-Me₂)₂DPP-catalyzed asymmetric MA reaction of β-nitrostyrene **35a** with other Hagemann's ester **88h** in toluene at 25 °C to furnish a 2.0:1 ratio of Michael products **133ha/134ha** in 36% yield with 77% ee and 3% ee respectively (Table 25, entry 6). After these interesting results, we further decided to investigate the scope and limitations of the asymmetric MA reaction of Hagemann's ester **88a** with a range of β-nitrostyrenes **35** including functional groups containing electron-donating β-nitrostyrenes, electron-withdrawing β-nitrostyrenes, halogenated β-nitrostyrenes and heterocyclic β-nitrostyrenes under L-(3,5-Me₂)₂DPP catalysis with **132b** in toluene at the ambient conditions to test the diversity nature of the asymmetric MA reaction (Table 25).

Asymmetric MA reaction of Hagemann's ester 88a with 35e under L-(3,5-Me₂)₂DPP-catalysis to furnish a 1.2:1 ratio of Michael products 133ae/134ae in 45% yield with 78/17% ee respectively as shown in Table 25, entry 7. Asymmetric MA reaction of Hagemann's ester 88a with 35f under L-(3,5-Me₂)₂DPP-catalysis to furnish a 1.2:1 ratio of Michael products 133af/134af in 61% yield with 26/21% ee respectively as shown in Table 25, entry 8. Asymmetric MA reaction of Hagemann's ester 88a with halogenated β-nitrostyrene **35g** under L-(3,5-Me₂)₂DPP-catalysis to furnish a 1:1.4 ratio of Michael products 133ag/134ag in 60% yield with 44/15% ee respectively as shown in Table 25, entry 9. Asymmetric MA reaction of Hagemann's ester 88a with 35c under L-(3,5-Me₂)₂DPP-catalysis to furnish a 1.1:1 ratio of Michael products **133ac/134ac** in 63% yield with 60/4% ee respectively as shown in Table 25, entry 10. Interestingly, asymmetric MA reaction of 4-methoxy-β-nitrostyrene 35h with Hagemann's ester 88a under L-(3,5-Me₂)₂DPP-catalysis in toluene at 25 °C to furnish a 1:1.7 ratio of Michael products 133ah/134ah in 50% yield with 29/27% ee respectively as shown in Table 25, entry 11. Asymmetric MA reaction of 4-nitro-β-nitrostyrene 35i with Hagemann's ester 88a under L-(3,5-Me₂)₂DPP-catalysis in toluene at 25 °C to furnish a 1:1.7 ratio of Michael products 133ai/134ai in 74% yield with 42/27% ee respectively as shown in Table 25, entry 12. Generality of the L-(3,5-Me₂)₂DPP-catalyzed asymmetric MA reaction of Hagemann's esters 88 with β-nitrostyrenes 35 were confirmed by two more

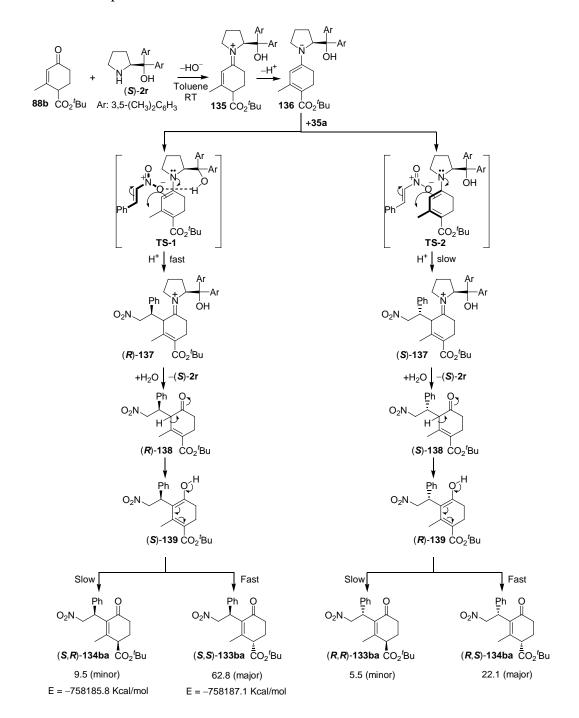
examples. Asymmetric Michael reaction of Hagemann's ester **88a** and heterocyclic β -nitrostyrene **35j** in toluene solvent to furnish a 1.1:1 ratio of Michael products **133aj/134aj** in 60% yield with 47/6% ee respectively as shown in Table 25, entry 13. Asymmetric Michael reaction of Hagemann's ester **88a** and **35d** in toluene solvent to furnish a 1.5:1 ratio of Michael products **133ad/134ad** in 60% yield with 41/12% ee respectively as shown in Table 25, entry 14.

To demonstrate the broad scope of this novel methodology for covering a structurally diverse group of Michael products 133/134, we also used D-(3,5-Me₂)₂DPP 2r/thiourea 132b as catalyst in this asymmetric MA reaction with two examples as shown in Table 25, entries 15-16. Asymmetric MA reaction of Hagemann's ester 88a and β-nirostyrene 35a under D-(3,5-Me₂)₂DPP-catalysis to furnish a 1.2:1 ratio of Michael products (+)-133aa/134aa in 73% yield with -70/-14% ee respectively as shown in Table 25, entry 15. Asymmetric MA reaction of *tert*-butyl Hagemann's ester 88b and β-nirostyrene 35a under D-(3,5-Me₂)₂DPP-catalysis to furnish a 2.6:1 ratio of Michael products (-)-133ba/134ba in 52% yield with -85/-8% ee respectively as shown in Table 25, entry 16. The products structures and regiochemistry were confirmed by ¹H NMR, ¹³C NMR [for example see Fig. 23-24] and mass analysis. The absolute configuration of products were established by correlation with Barbas-Michael reactions. ⁶⁵

The possible reaction mechanism for stereoselective synthesis of cascade products 133/134 through the reaction of *tert*-butyl Hagemann's ester 88b and β -nitrostyrene 35a under L-(3,5-Me₂)₂DPP catalysis is illustrated in Scheme 21. First, reaction of L-(3,5-Me₂)₂DPP with *tert*-butyl Hagemann's ester 88b generates the imine cation 135, which will transform into chiral push-pull dienamine 136 through proton elimination. Formation of the Michael product (*R*)-137 through *re*-face attack of β -nitrostyrene 35a with chiral push-pull dienamine 136 will be faster reaction as shown in TS-1. Formation of the Michael product (*S*)-137 through *si*-face attack of β -nitrostyrene 35a with chiral push-pull dienamine 136 will be slower reaction as shown in TS-2

based on the hydrogen bonding interactions with OH. In situ hydrolysis of both the Michael products (R)-137 and (S)-137

Scheme 21: Proposed reaction mechanism.



converted into highly substituted (R)-138 and (S)-138 under amine catalysis. Through the keto-enol tautomerism (R)-138 and (S)-138 will be converting to highly substituted (S)-139 and (S)-139 based on proton transfer reaction. Formation of the thermodynamically stable (S, S)-133ba product is faster reaction compared to formation of the kinetically stable (S, S)-134ba product. (S, S)-133ba product is a major isomer and (S, S)-134ba product is a minor isomer. The calculated heat of formation (S) for the (S, S)-133ba product is 1.3 kcal/mol more than the (S, S)-134ba product as shown in Scheme 21 & 22. This result also strongly suggests that thermodynamically product (S, S)-133ba formation is more favorable than product (S, S)-134ba formation as revealed by DFT calculations. Formation of the (S, S)-134ba product is faster reaction compared to formation of the (S, S)-133ba product as shown in Scheme 21. In the compound (S)-139 weak CH-S Interactions between the phenyl ring and olefinic methyl group and also anchoring effect of nitro group towards hydrogen facilitates the formation of (S, S)-134ba product as a major isomer. Formation of these four isomers are controlling the enantioselectivity of the reaction.

Scheme 22: Minimized structures of (S, S)-133ba and (S, R)-134ba based on DFT calculations.

$$= \bigcup_{\substack{(S,S)-133\text{ba $\bar{C}O_2$}^f\text{Bu} \\ E = -758187.1 \text{ Kcal/mol} }} \bigcup_{\substack{(S,R)-134\text{ba $\bar{C}O_2$}^f\text{Bu} \\ E = -758185.8 \text{ Kcal/mol} }}$$

The asymmetric MA reaction of β -nitrostyrene **35a** with *tert*-butyl ester **88b** catalyzed by L-(3,5-Me₂)₂DPP in toluene furnished a Michael products **133ba/134ba** with 84% ee and 40% ee and obtained the four possible isomers in a ratio of 9.5:62.8:5.5:22.1 (HPLC data). The reason for variation in the ee is because product (R,

S)-134ba is decreasing the ee value of product (S, R)-134ba. Another reason for decreasing the ee of product (S, R)-134ba may be due to the epimerization.

Scheme 23: Synthesis of functionalized cyclohexenone 142aa.

NO2 LiOH.H₂O (6 equiv.) MeOH/THF/H₂O (0.1 M)
$$3 h$$
, RT CO_2 H C

After successful demonstration of the L- $(3,5\text{-Me}_2)_2DPP$ catalyzed asymmetric MA reactions of Hagemann's esters **88** with β -nitrostyrenes **35**, we decided to study the utilization of the chiral Michael adducts **133/134** in the synthesis of functionalized chiral enones via acid/base catalysis⁶⁷ as shown in Scheme 23.

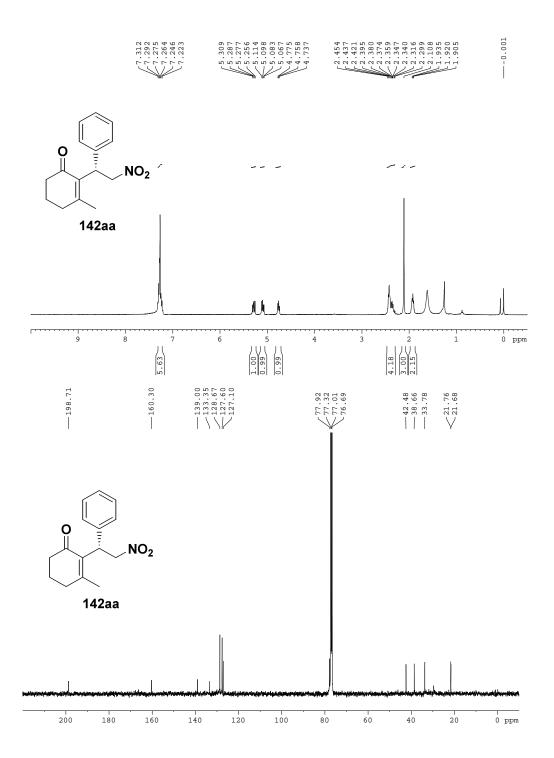


Figure-25: ¹H NMR and ¹³C NMR Spectrum of product **142aa**.

Highly functionalized cyclohexenones (±)-142aa, (R)-142aa and (S)-142aa were synthesized from (\pm) -133aa/134aa, (+)-133aa/134aa and 133ba/134ba in moderate yields with moderate ee's via acid/base catalysis as shown in Scheme 23. Reaction of 2-methyl-3-(2-nitro-1-phenylethyl)-4-oxo-cyclohex-2-enecarboxylic acid ethyl ester (±)-133aa/134aa with simple base (LiOH.H₂O) hydrolysis in MeOH/THF/H₂O solvent combination at RT furnished the keto acid (±)-140aa/141aa in 36% yield, which upon treatment with acid (HCl) in THF solvent furnished the decarboxylated product (±)-142aa in 42% yield as shown in Scheme 23. In a similar manner, (R)-142aa was synthesized from 2-methyl-3-(2-nitro-1-phenylethyl)-4-oxocyclohex-2-enecarboxylic acid ethyl ester (+)-133aa/134aa in 42% yield with -33% ee as shown in Scheme 23. Genaration of chiral cyclohexenones was studied by the synthesis of another enantiomer (S)-142aa as shown in Scheme 23. Reaction of 2methyl-3-(2-nitro-1-phenylethyl)-4-oxo-cyclohex-2-enecarboxylic acid tert-butyl ester 133ba/134ba with acidic (CH₃SO₃H) hydrolysis in DCM solvent at RT furnished the keto acid 140aa/141aa in 50% yield, which upon treatment with acid (HCl) in THF solvent furnished the decarboxylated product (S)-142aa in 60% yield with 44% ee as shown in Scheme 23. The product structure and regiochemistry was confirmed by ¹H NMR, ¹³C NMR [for example see Fig. 25] and mass analysis.

6.3 CONCLUSION

In summary, we have described chemo- and enantioselective process for the synthesis of highly substituted 2-alkyl-3-(2-nitro-1-arylethyl)-4-oxo-cyclohex-2-enecarboxylic acid alkyl esters 133/134 under amine catalysis. Here we described the L-(3,5-Me₂)₂DPP catalyzed asymmetric Michael reactions of Hagemann's esters 88 with nitroolefins 35 at ambient conditions. This novel asymmetric Michael reaction proceeds in good yields with high enantioselectivity using L-(3,5-Me₂)₂DPP as a catalyst along with thiourea 132b as a co-catalyst. Furthermore, we demonstrated the application of chiral Michael products in the synthesis of highly functionalized cyclohexenones.

7. REFERENCES

- (a) Ramachary, D. B.; Ramakumar, K.; Kishor, M. *Tetrahedron Lett.* 2005, 46, 7037–7042.
 (b) Ramachary, D. B.; Ramakumar, K.; Narayana, V. V. *J. Org. Chem.* 2007, 72, 1458–1463.
 (c) Ramachary, D. B.; Narayana, V. V.; Ramakumar, K. *Eur. J. Org. Chem.* 2008, 3907–3911.
 (d) Ramachary, D. B.; Ramakumar, K.; Narayana, V. V. *Chem.–Eur. J.* 2008, 14, 9143–9147.
 (e) Ramachary, D. B.; Narayana, V. V.; Prasad, M. S.; Ramakumar, K. *Org. Biomol. Chem.* 2009, 7, 3372–3378.
- 2. (a) Stork, G.; Terrell, R.; Szmuszkovicz, J. J. Am. Chem. Soc. **1954**, 76, 2029–2030. (b) Stork, G.; Landesman, H. K. J. Am. Chem. Soc. **1956**, 78, 5128–5129.
- (a) Woodward, R. B. Pure Appl. Chem. 1968, 17, 519–547. (b) Woodward, R. B. Pure Appl. Chem. 1973, 33, 145–177.
- (a) Eder, U.; Sauer, G.; Wiechert, R. Angew. Chem., Int. Ed. 1971, 10, 496–497.
 (b) Eder, U.; Wiechert, R.; Sauer, G. Ger. Offen. 1971, CODEN: GWXXBX DE 2014757 19711007, CAN 76:14180 (in German: 20 pp). (c) Wieland, P.; Miescher, K. Helv. Chim. Acta 1950, 33, 2215–2228. (d) Wieland, P.; Anner, G.; Miescher, K. Helv. Chim. Acta 1953, 36, 1803–1809.
- (a) Hajos, Z. G.; Parrish, D. R. Ger. Offen. 1971, CODEN: GWXXBX DE 2102623 19710729, CAN 76:59072 (in German: 42 pp). (b) Hajos, Z. G.; Parrish, D. R. J. Org. Chem. 1974, 39, 1615–1621.
- 6. Wagner, J.; Lerner, R. A.; Barbas, C. F. III. Science 1995, 270, 1797–1800.
- 7. Björnestedt, R.; Zhong, G.; Lerner, R. A.; Barbas, C. F. III. *J. Am. Chem. Soc.* **1996**, *118*, 11720–11724.
- 8. Zhong, G.; Hoffmann, T.; Lerner, R. A.; Danishefsky, S.; Barbas, C. F. III. *J. Am. Chem. Soc.* **1997**, *119*, 8131–8132.
- 9. List, B.; Lerner, R. A.; Barbas, C. F. III. Org. Lett. 1999, 1, 59-61.
- (a) List, B.; Lerner, R. A.; Barbas, C. F. III. J. Am. Chem. Soc. 2000, 122, 2395–2396.
 (b) Sakthivel, K.; Notz, W.; Bui, T.; Barbas, C. F. III. J. Am. Chem. Soc. 2001, 123, 5260–5267.

- 11. Córdova, A.; Watanabe, S.-i.; Tanaka, F.; Notz, W.; Barbas, C. F. III. *J. Am. Chem. Soc.* **2002**, *124*, 1866–1867.
- 12. Thayumanavan, R.; Ramachary, D. B.; Sakthivel, K.; Tanaka, F.; Barbas, C. F. III. *Tetrahedron Lett.* **2002**, *43*, 3817–3820.
- 13. Ramachary, D. B.; Chowdari, N. S.; Barbas, C. F. III. *Tetrahedron Lett.* **2002**, *43*, 6743–6746.
- (a) Ramachary, D. B.; Chowdari, N. S.; Barbas, C. F. III. *Angew. Chem., Int. Ed.* 2003, 42, 4233–4237. (b) Ramachary, D. B.; Barbas, C. F. III. *Chem.–Eur. J.* 2004, 10, 5323–5331.
- (a) Ramachary, D. B.; Chowdari, N. S.; Barbas, C. F. III. Synlett 2003, 12, 1910–1914.
 (b) Ramachary, D. B.; Anebouselvy, K.; Chowdari, N. S.; Barbas, C. F. III. J. Org. Chem. 2004, 69, 5838–5849.
- Bertelsen, S.; Marigo, M.; Brandes, S.; Dinér, P.; Jørgensen, K. A. *J. Am. Chem. Soc.* 2006, *128*, 12973–12980.
- 17. Bench, B. J.; Liu, C.; Evett, C. R.; Watanabe, C. M. H. *J. Org. Chem.* **2006**, *71*, 9458–9463.
- 18. Utsumi, N.; Zhang, H.; Tanaka, F.; Barbas, C. F. III. *Angew. Chem., Int. Ed.* **2007**, *46*, 1878–1880.
- 19. Momiyama, N.; Yamamoto, Y.; Yamamoto, H. J. Am. Chem. Soc. **2007**, 129, 1190–1195.
- 20. Hong, B.-C.; Wu, M.-F.; Tseng, H.-C.; Huang, G.-F.; Su, C.-F.; Liao, J.-H. *J. Org. Chem.* **2007**, *72*, 8459–8471.
- 21. Ramachary, D. B.; Reddy, Y. V.; Prakash, B. V. *Org. Biomol. Chem.* **2008**, *6*, 719–726.
- 22. de Figueiredo, R. M.; Fröhlich, R.; Christmann, M. *Angew. Chem., Int. Ed.* **2008**, *47*, 1450–1453.
- 23. Jiang, B.; Hao, W.-J.; Zhang, J.-P.; Tu, S.-J.; Shi, F. Org. Biomol. Chem. **2009**, 7, 2195-2201.

- 24. Han, B.; He, Z.-Q.; Li, J.-L.; Li, R.; Jiang, K.; Liu, T.-Y.; Chen, Y.-C. *Angew. Chem.*, *Int. Ed.* **2009**, *48*, 5474–5477.
- 25. (a) Wu, L.-Y.; Bencivenni, G.; Mancinelli, M.; Mazzanti, A.; Bartoli, G.; Melchiorre, p. *Angew. Chem.*, *Int. Ed.* **2009**, 48, 7196–7199. (b) Bencivenni, G.; Wu, L.-Y.; Mazzanti, A.; Giannichi, B.; Pesciaioli, F.; Song, M.-P.; Bartoli, G.; Melchiorre, P. *Angew. Chem.*, *Int. Ed.* **2009**, 48, 7200–7203.
- 26. Han, B.; Xiao, Y.-C.; He, Z.-Q.; Chen, Y.-C. Org. Lett. 2009, 11, 4660–4663.
- 27. Synthesis of highly functionalized molecules starting from the simple materials in one-pot (a) Ramachary, D. B.; Kishor, M.; Ramakumar, K. Tetrahedron Lett. **2006**, 47, 651–656. (b) Ramachary, D. B.; Kishor, M.; Reddy, G. B. Org. Biomol. Chem. 2006, 4, 1641–1646. (c) Ramachary, D. B.; Reddy, G. B. Org. Biomol. Chem. 2006, 4, 4463–4468. (d) Ramachary, D. B.; Kishor, M. J. Org. Chem. 2007, 72, 5056–5068. (e) Ramachary, D. B.; Kishor, M.; Reddy, Y. V. Eur. J. Org. Chem. 2008, 975–998. (f) Ramachary, D. B.; Kishor, M. Org. Biomol. Chem. 2008, 6, 4176-4187. (g) Ramachary, D. B.; Reddy, Y. V.; Kishor, M. Org. Biomol. Chem. 2008, 6, 4188-4197. (h) Ramachary, D. B.; Sakthidevi, R. Org. Biomol. Chem. 2008, 6, 2488–2492. (i) Ramachary, D. B.; Venkaiah, C.; Reddy, Y. V.; Kishor, M. Org. Biomol. Chem. 2009, 7, 2053-2062. (j) Ramachary, D. B.; Sakthidevi, R. Chem. –Eur. J. 2009, 15, 4516–4522. (k) Ramachary, D. B.; Reddy, Y. V. J. Org. Chem. 2010, 75, 74-85. (l) Ramachary, D. B.; Rumpa, M.; Venkaiah, C. Eur. J. Org. Chem. 2010, 3205-3210. (m) Ramachary, D. B.; Rumpa, M.; Venkaiah, C. Org. Biomol. Chem. **2010**, 8, 321–325. (n) Ramachary, D. B.; Kishor, M. Org. Biomol. Chem. **2010**, 8, 2859–2867.
- For selected recent reviews on general cascade and multi-component reactions, see: (a) Nicolaou, K. C.; Montagnon, T.; Snyder, S. A. Chem. Commun. 2003, 551-564. (b) Wasilke, J.-C.; Obrey, S. J.; Baker, R. T.; Bazan, G. C. Chem. Rev. 2005, 105, 1001-1020. (c) Ramon, D. J.; Yus, M. Angew. Chem., Int. Ed. 2005, 44, 1602-1634. (d) Tietze, L. F. Chem. Rev. 1996, 96, 115-136. (e) Tietze, L. F.;

- Haunert, F. Stimulating Concepts in Chemistry (eds Vogtle, F.; Stoddart, J. F.; Shibasaki, M.) 39-64 (Wiley-VCH, Weinheim, 2000). (f) Tietze, L. F.; Modi, A. Med. Res. Rev. 2000, 20, 304-322. (g) Mayer, S. F.; Kroutil, W.; Faber, K. Chem. Soc. Rev. 2001, 30, 332-339. (h) Cane, D. E. Chem. Rev. 1990, 90, 1089-1103.
- 29. For selected recent reviews on organocatalytic cascade and multi-component reactions, see: (a) Guo, H.-C.; Ma, J.-A. Angew. Chem., Int. Ed., 2006, 45, 354-366. (b) Pellissier, H. Tetrahedron 2006, 62, 2143-2173. (c) Chapman, C. J.; Frost, C. G. Synthesis 2007, 1-21. (d) Gerencser, J.; Dorman, G.; Darvas, F. QSAR & Combinatorial Science 2006, 25, 439-448. (e) Guillena, G.; Ramon, D. J.; Yus, M. Tetrahedron Asymmetry 2007, 18, 693-700. (f) Enders, D.; Grondal, C.; Huettl, M. R. M. Angew. Chem., Int. Ed., 2007, 46, 1570-1581. (g) Erkkilä, A.; Majander, I.; Pihko, P. M. Chem. Rev., 2007, 107, 5416-5470. (h) Notz, W.; Tanaka, F.; Barbas, C. F. III. Acc. Chem. Res. 2004, 37, 580-591. For selected recent papers on organocatalytic cascade and multi-component reactions, see: (i) Chowdari, N. S.; Ramachary, D. B.; Barbas, C. F. III. Org. Lett. 2003, 5, 1685-1688. (j) Yang, J. W.; Hechavarria Fonseca, M. T.; List, B. J. Am. Chem. Soc. 2005, 127, 15036-15037. (k) Huang, Y.; Walji, A. M.; Larsen, C. H.; MacMillan, D. W. C. J. Am. Chem. Soc. 2005, 127, 15051-15053. (1) Marigo, M.; Bertelsen, S.; Landa, A.; Jorgensen, K. A. J. Am. Chem. Soc. 2006, 128, 5475-5479. (m) Enders, D.; Huettl, M. R. M.; Grondal, C.; Raabe, G. Nature **2006**, 441, 861-863. (n) Rios, R.; Sunden, H.; Ibrahem, I.; Zhao, G. -L.; Cordova, A. Tetrahedron Lett. 2006, 47, 8679-8682. (o) Wang, W.; Li, H.; Wang, J.; Zu, L. J. Am. Chem. Soc. 2006, 128, 10354-10355. (p) Tejedor, D.; Gonzalez-Cruz, D.; Santos-Exposito, A.; Marrero-Tellado, J. J.; de Armas, P.; Garcia-Tellado, F. Chem.-Eur. J. 2005, 11, 3502-3510. (q) Rueping, M.; Antonchick, A. P.; Theissmann, T. Angew. Chem., Int. Ed. 2006, 45, 3683-3686. (r) Ishikawa, H.; Suzuki, T.; Hayashi, Y. Angew. Chem., Int. Ed. 2009, 48, 1304-

- 1307. (s) Zhu, D.; Lu, M.; Chua, P. J.; Tan, B.; Wang, F.; Yang, X.; Zhong, G. *Org. Lett.* **2008**, *10*, 4585-4588.
- 30. For the applications of olefins, see: (a) Dimmock, J. R.; Hamon, N. W.; Hindmarsh, K. W.; Sellar, A. P.; Turner, W. A.; Rank, G. H.; Robertson, A. J. J. Pharm. Sci. 1976, 65, 538-543. (b) Dimmock, J. R.; Sidhu, K. K.; Chen, M.; Li, J.; Quail, J. W.; Allen, T. M.; Kao, G. Y. J. Pharm. Sci. 1994, 83, 852-858. (c) Dimmock, J. R.; Baker, G. B.; Sutherland, R. G.; Can. J. Pharm. Sci. 1975, 10, 53-59. (d) Dimmock, J. R.; Jha, A.; Kumar, P.; Zello, G. A.; Quail, J. W.; Oloo, E. O.; Oucharek, J. J.; Pasha, M. K.; Seitz, D.; Sharma, R. K.; Allen, T. M.; Santos, C. L.; Manavathu, E. K.; De Clercq, E.; Balzarini, J.; Stables, J. P. Eur. J. Med. Chem. 2002, 37, 35-44. (e) Guilford, W. J.; Shaw, K. J.; Dallas, J. L.; Koovakkat, S.; Lee, W.; Liang, A.; Light, D. R.; McCarrick, M. A.; Whitlow, M.; Ye, B.; Morrissey, M. M. J. Med. Chem. 1999, 42, 5415-5425. For the applications of phenols, see: (f) Sato, M.; Kakinuma, H.; Asanuma, H. PCT Int. Appl. 2004, 106 pp. CODEN: PIXXD2 WO 2004014931 A1 2004:143171, CAN 140:199631 (patent written in English). (g) Boije, M.; Faegerhag, J.; Lindstedt Alstermark, E. -L.; Ohlsson, B. PCT Int. Appl. 2001, 49 pp. CODEN: PIXXD2 WO 2001040172 A1 2001:416889, CAN 135:33373 (patent written in English). (h) Campbell, E.; Martin, J. J.; Bordner, J.; Kleinman, E. F. J. Org. Chem. 1996, 61, 4806-4809. (i) Yoshino, T.; Ng, F.; Danishefsky, S. J. J. Am. Chem. Soc. 2006, 128, 14185-14191. (j) Delmau, L. H.; Bryan, J. C.; Hay, B. P.; Engle, N. L.; Sachleben, R. A.; Moyer, B. A. ACS Symp. Ser. 2000, 757, 86-106. (k) Morgan, E. D.; Jackson, B. D.; Ollett, D. G.; Sales, G. W. J. Chem. Ecol. 1990, 16, 3493-3510. (1) Bose, G.; Hong Nguyen, V. T.; Ullah, E.; Lahiri, S.; Gorls, H.; Langer, P. J. Org. Chem. 2004, 69, 9128-9134.
- 31. (a) Heathcock, C. H. In *Comprehensive Organic Synthesis*; Trost, B. M.; Fleming, I.; Heathcock, C. H.; Eds.; Pergamon Elsevier: Oxford, 1991; Vol. 2, p 133. (b) Perlmutter, P. *Conjugate Addition Reactions in Organic Synthesis*; Pergamon Elsevier: Oxford, 1992; Chapters 2 and 3. (c) Jung, M. E. In

- Comprehensive Organic Synthesis; Trost, B. M.; Fleming, I.; Semmelhack, M. F.; Eds.; Pergamon Elsevier: Oxford, 1991; Vol. 4, p 1. (d) Lee, V. J. In Comprehensive Organic Synthesis; Trost, B. M.; Fleming, I.; Semmelhack, M. F.; Eds.; Pergamon Elsevier: Oxford, 1991; Vol. 4, pp 69 and 139. (e) Kozlowski, J. A. In Comprehensive Organic Synthesis; Trost, B. M.; Fleming, I.; Semmelhack, M. F.; Eds.; Pergamon Elsevier: Oxford, 1991; Vol. 4, p 169. (f) Basavaiah, D.; Dharma Rao, P.; Suguna Hyma, R. Tetrahedron 1996, 52, 8001-8062. (g) Ciganek, E. Org. React. 1997, 51, 201. (h) Basavaiah, D.; Jaganmohan Rao, A.; Satyanarayana, T. Chem. Rev. 2003, 103, 811-891.
- 32. X-ray crystal data of **89ai**: $C_{17}H_{17}FO_3$; MW = 288.31, Monoclinic, space group $P2_1/c$, with a = 9.6423(11) Å, b = 20.332(2) Å, c = 7.4807(9) Å, α = 90°, β = 97.331°, γ = 90°. CCDC-777620 contains the supplementary crystallographic data for this crystal structure.
- 33. Srikrishna, A.; Ramachary, D. B. *Indian Journal of Chemistry* **2005**, *44B*, 751–761.
- 34. Brunke, E. J.; Hammerschmidt, F. J.; Struwe, H. *Tetrahedron* **1981**, *37*, 1033–1038.
- 35. X-ray crystal data of **90aq**: $C_{19}H_{19}NO_5$; MW = 341.35, Triclinic, space group p-1, with a = 8.025(3) Å, b = 8.415(3) Å, c = 14.351(6) Å, $\alpha = 87.633^{\circ}$, $\beta = 83.514^{\circ}$, $\gamma = 61.76^{\circ}$. CCDC-777621 contains the supplementary crystallographic data for this crystal structure.
- 36. (a) Chen, W.; Xu, L.; Xiao, J. Tetrahedron Lett. 2001, 42, 4275-4278. (b) Masllorens, J.; Moreno-Manas, M.; Pla-Quintana, A.; Roglans, A. Org. Lett. 2003, 5, 1559-1561. (c) Dams, M.; De Vos, D. E.; Celen, S.; Jacobs, P. A. Angew., Chem. Int. Ed. 2003, 42, 3512-3515. (d) Yokota, T.; Tani, M.; Sakaguchi, S.; Ishii, Y. J. Am. Chem. Soc. 2003, 125, 1476-1477. (e) Hwang, L. K.; Na, Y.; Lee, J.; Do, Y.; Chang, S. Angew. Chem., Int. Ed. 2005, 44, 6166-6169. (f) Borhade, S. R.; Waghmode, S. B. Tetrahedron Lett. 2008, 49, 3423-

- 3429. (g) Zhang, Y. -H.; Shi, B. -F.; Yu, J. -Q. J. Am. Chem. Soc. **2009**, 131, 5072-5074.
- 37. Ma, A.; Zhu, S.; Ma, D. Tetrahedron Lett. 2008, 49, 3075–3077.
- (a) Franck, B.; Scharf, V.; Schrameyer, M. Angew. Chem., Int. Ed. 1974, 13, 136-137.
 (b) Mandell, L.; Daley, R. F.; Day, R. A., Jr. J. Org. Chem. 1976, 41, 4087-4089.
 (c) Chandler, M.; Mincione, E.; Parsons, P. J. J. Chem. Soc., Chem. Commun. 1985, 1233-1234.
 (d) Handy, S. T.; Omune, D. Org. Lett. 2005, 7, 1553-1555.
- 39. For the recent papers on Barbas dienamines, see: (a) Ramachary, D. B.; Barbas,
 C. F. III. *Org. Lett.* 2005, 7, 1577-1580. (b) Aznar, F.; García, A. –B.; Cabal, M. –P. *Adv. Synth. Catal.* 2006, *348*, 2443-2448.
- (a) Allemann, C.; Gordillo, R.; Clemente, F. R.; Cheong, P. H-Y.; Houk, K. N. Acc. Chem. Res. 2004, 37, 558-569.
 (b) Zhu, X.; Tanaka, F.; Lerner, R. A.; Barbas, C. F. III; Wilson, I. A. J. Am. Chem. Soc. 2009, 131, 18206-18207.
- (a) Knolker, H. J.; Reddy, K. R. Chem. Rev. 2002, 102, 4303-4427. (b) Guram,
 A. S.; Buchwald, S. L. J. Am. Chem. Soc. 1994, 116, 7901-7902. (c) March, J. Advanced Organic Chemistry, 4th ed., Wiley, New York, 1992, pp 641-677.
- 42. (a) Hartwig, J. F. Angew. Chem., Int. Ed. 1998, 37, 2046-2067. (b) Wolfe, J. P.; Buchwald, S. L. J. Org. Chem. 1997, 62, 6066-6068. (c) Louie, J.; Driver, M. S.; Hamann, B. C.; Hartwig, J. F. J. Org. Chem. 1997, 62, 1268-1273. (d) Wolfe, J. P.; Buchwald, S. L. J. Org. Chem. 1997, 62, 1264-1267. (e) Marcoux, J. F.; Wagaw, S.; Buchwald, S. L. J. Org. Chem. 1997, 62, 1568-1569. (f) Wolfe, J. P.; Buchwald, S. L. J. Org. Chem. 1996, 61, 1133-1135. (g) Wolfe, J. P.; Wagaw, S.; Buchwald, S. L. J. Am. Chem. Soc. 1996, 118, 7215-7216. (h) Driver, M. S.; Hartwig, J. F. J. Am. Chem. Soc. 1996, 118, 7217-7218. (i) Guram, A. S.; Rennels, R. A.; Buchwald, S. L. Angew. Chem., Int. Ed. 1995, 34, 1348-1350.

- 43. (a) Gaina, L.; Lovasz, T.; Cristea, C.; Silberg, I. A.; Deleanu, C. Revue Roumaine de Chimie 2003, 48, 549-554. (b) Suzuki, T.; Goto, F. Jpn. Kokai Tokkyo Koho 2003, 6 pp. CODEN: JKXXAF JP 2003096032 A2 20030403, CAN 138:271374 (Patent written in Japanese). (c) Yamamura, T.; Suzuki, K.; Yamaguchi, T.; Nishiyama, T. Bull. Chem. Soc. Jpn. 1997, 70, 413-419. (d) Wahlig, H.; Henning, H. M.; Hepding, L.; Bertram, H. J. Arzneimittel-Forschung 1962, 12, 1123-1127. (e) Schmitt, J.; Suquet, M.; Callet, G.; Raveux, R. Bulletin de la Societe Chimique de France 1962, 444-455. (f) Protiva, M.; Jilek, J. O.; Kolinsky, J.; Sustr, M.; Urban, J. Chemicke Listy pro Vedu a Prumysl 1949, 43, 254-257. (g) Bolton, P. D.; Glenn, R. W. U.S. Pat. Appl. Publ. 2006, 17 pp. CODEN: USXXCO US 2006032002 A1 20060216, CAN 144:239220 (Patent written in English).
- 44. (a) Chong, B. D.; Ji, Y. I.; Oh, S. S.; Yang, J. D.; Baik, W.; Koo, S. J. Org. Chem. 1997, 62, 9323-9325. (b) Shiv Kumar, S. N.; Bhaduri, A. P. Indian J. Chem. 1983, 22B, 524-525. (c) Ahmed, M. G.; Ahmed, S. A.; Romman, U. K. R.; Akhter, K.; Chowdhury, M. A.; Kiuchi, F. Indian J. Chem. 2001, 40B, 710-712. (d) Banerjee, M.; Mukhopadhyay, R.; Achari, B.; Banerjee, A. Kr. J. Org. Chem. 2006, 71, 2787-2796. (e) Zhang, Shi-Wei; Mitsudo, Take-aki; Kondo, T.; Watanabe, Y. J. Organometallic Chem. 1995, 485, 55-62. (f) McCurry, P. M., Jr.; Singh, R. K. Synthetic Communications 1976, 6, 75-79. (g) Brunke, E. J.; Bielstein, H.; Kutschan, R.; Rehme, G.; Schuetz, H. J.; Wolf, H. Tetrahedron 1979, 35, 1607-1613. (h) Smith, W. T., Jr.; Eftax, D. S. P. J. Org. Chem. 1956, 21, 174-176. (i) Smith, W. T., Jr.; Eftax, D. S. P. J. Am. Chem. Soc. 1953, 75, 4356-4357.
- 45. (a) Four decades ago Mukaiyama and Moskal research groups independently reported the KOH catalyzed nitroso aldol reaction of active methylenes with nitrosobenzene to produce azomethine derivatives, see: (1) Nohira, H.; Sato, K.; Mukaiyama, T. *Bull. Chem. Soc. Jpn.* **1963**, *36*, 870-872. (2) Mirek, J.; Moskal, J.; Moskal, A. *Tetrahedron* **1975**, *31*, 2145-2149. (b) For an early reaction of

- aromatic nitroso-compounds with enamines see: Lewis, J. W.; Myers, P. L.; Ormerod, J. A. *J. Chem. Soc. Perkin Trans. 1* **1972**, 2521-2524. (c) For an excellent review of reactions with nitroso compounds see: Yamamoto, H.; Momiyama, N. *Chem. Commun.* **2005**, 3514-3525. (d) For an electrophilic aromatic substitution with nitrosoarenes see: (1) Sokolov, A. V. *International Journal of Quantum Chemistry* **2004**, *100*, 1-12. (2) Hubig, S. M.; Kochi, J. K. *J. Am. Chem. Soc.* **2000**, *122*, 8279-8288. (3) Eckert-Maksic, M.; Hodoscek, M.; Kovacek, D.; Maksic, Z. B.; Primorac, M. *Theochem* **1997**, *417*, 131-143. (4) White, E. H. *Tetrahedron Lett.* **1997**, *38*, 7649-7652. (5) Corbett, M. D.; Corbett, B. R. *Biochemical Pharmacology* **1986**, *35*, 3613-21. (6) Bowden, W. L.; Little, W. F.; Meyer, T. J. *J. Am. Chem. Soc.* **1974**, *96*, 5605-5607. (7) Abramovitch, R. A.; Challand, S. R.; Scriven, E. F. V. *J. Org. Chem.* **1972**, *37*, 2705-2710. (8) Shudo, K. *Yakugaku Zasshi* **1982**, *102*, 111-126.
- 46. Many of nitrosoarenes 56 are commercially available and suitable methods are known to prepare them. For excellent methods for synthesis of nitroso compounds see: (a) Priewish, B.; Ruck-Braun, K. J. Org. Chem. 2005, 70, 2350-2352. (b) Defoin, A. Synthesis 2004, 706-710. (c) Porta, F.; Prati, L. J. Mol. Catal. 2000, 157, 123-129. (d) Tollari, S.; Cuscela, M.; Porta, F. J. Chem. Soc., Chem. Commun. 1993, 1510-1511.
- 47. (a) Iwamura, H.; Mathew, S. P.; Blackmond, D. G. J. Am. Chem. Soc. 2004, 126, 11770-11771. (b) Iwamura, H.; Wells, D. H.; Mathew, S. P.; Klussmann, M.; Armstrong, A.; Blackmond, D. G. J. Am. Chem. Soc. 2004, 126, 16312-16313. (c) Mathew, S. P.; Iwamura, H.; Blackmond, D. G. Angew. Chem., Int. Ed. 2004, 43, 3317-3321.
- 48. X-ray crystal data of **106aa**: $C_{16}H_{17}NO_3$; MW = 271.31, Monoclinic, space group $P2_1/c$, with a = 10.9755(12) Å, b = 8.2031(9) Å, c = 16.2285(18) Å, α = 90°, β = 107.064°, γ = 90°. CCDC-611665 contains the supplementary crystallographic data for this crystal structure.

- 49. X-ray crystal data of **107pa**: $C_{26}H_{27}ClN_2O_2$; MW = 434.95, Triclinic, space group p-1, with a = 10.602(2) Å, b = 10.735(2) Å, c = 11.281(2) Å, α = 77.344°, β = 67.507°, γ = 83.509°. CCDC-611666 contains the supplementary crystallographic data for this crystal structure.
- 50. Heat of formations (ΔH) of enamines **95** and **105** are calculated based on PM3 (MOPAC) calculations in *CS Chem3D Ultra* using wave function as closed shell (restricted) and minimize energy to minimum RMS Gradient of 0.100.
- (a) Kallander, L. S.; Lu, Q.; Chen, W.; Tomaszek, T.; Yang, G.; Tew, D.; Meek, T. D.; Hofmann, G. A.; Schulz-Pritchard, C. K.; Smith, W. W.; Janson, C. A.; Ryan, M. D.; Zhang, G. F.; Johanson, K. O.; Kirkpatrick, R. B.; Ho, T. F.; Fisher, P. W.; Mattern, M. R.; Johnson, R. K.; Hansbury, M. J.; Winkler, J. D.; Ward, K. W.; Veber, D. F.; Thompson, S. K. J. Med. Chem. 2005, 48, 5644–5647. (b) Palmer, L. M.; Janson, C. A.; Smith, W. W. PCT Int. Appl. 2005, p. 347. CODEN: PIXXD2 WO 2005016237 A2 20050224, CAN: 142:256748 (patent written in English). (c) Tome, A. C. Sci. Synth. 2004, 13, 415–601. (d) Kallander, L. S.; Thompson, S. K. PCT Int. Appl. 2001, p. 44. CODEN: PIXXD2 2001078723 A1 20011025, CAN: 135:331429 (patent written in English). (e) Melo, J. O. F.; Donnici, C. L.; Augusti, R.; Lopes, M. T. P.; Mikhailovskii, A. G. Heterocycl. Commun. 2003, 9, 235–238. f) Baures, P. W. Org. Lett. 1999, 1, 249–252.
- (a) Krivopalov, V. P.; Shkurko, O. P. *Russ. Chem. Rev.* 2005, 74, 339–379. (b) for the reactions of organic azides with preformed enamines, see: Brunner, M.; Maas, G.; Klaerner, F. G. *Helv. Chim. Acta.* 2005, 88, 1813–1825, and references therein. (c) Barluenga, J.; Valdes, C.; Beltran, G.; Escribano, M.; Aznar, F. *Angew. Chem., Int. Ed.* 2006, 45, 6893–6896. (d) Aucagne, V.; Leigh, D. A. *Org. Lett.* 2006, 8, 4505–4507. (e) Jin, T.; Kamijo, S.; Yamamoto, Y. *Eur. J. Org. Chem.* 2004, 3789–3791. (f) Kamijo, S.; Huo, Z.; Jin, T.; Kanazawa, C.; Yamamoto, Y. *J. Org. Chem.* 2005, 70, 6389–6397.

- (a) Tornoe, C. W.; Christensen, C.; Meldal, M. J. Org. Chem. 2002, 67, 3057–3064. (b) Rostovtsev, V. V.; Green, L. G.; Fokin, V. V.; Sharpless, K. B. Angew. Chem., Int. Ed. 2002, 41, 2596–2599. (c) Lee, L. V.; Mitchell, M. L.; Huang, S. J.; Fokin, V. V.; Sharpless, K. B.; Wong, C. H. J. Am. Chem. Soc. 2003, 125, 9588–9589. (d) Speers, A. E.; Adam, G. C.; Cravatt, B. F. J. Am. Chem. Soc. 2003, 125, 4686–4687. (e) Lutz, J. F.; Zarafshani, Z. Adv. Drug Delivery Rev. 2008, 60, 958–970. (f) Nandivada, H.; Jiang, X.; Lahann, J. Adv. Mater. 2007, 19, 2197–2208. (g) Binder, W. H.; Kluger, C. Curr. Org. Chem. 2006, 10, 1791–1815. (h) Whiting, M.; Muldoon, J.; Lin, Y. C.; Silverman, S. M.; Lindstrom, W.; Olson, A. J.; Kolb, H. C.; Finn, M. G.; Sharpless, K. B.; Elder, J. H.; Fokin, V. V. Angew. Chem., Int. Ed. 2006, 45, 1435–1439. (i) see also: QSAR Comb. Sci. 2007, 26, Issues 11–12. Special edition devoted to click chemistry.
- 54. (a) Tornoe, C.W.; Sanderson, S. J.; Mottram, J. C.; Coombs, G. H.; Meldal, M. J. Comb. Chem. 2004, 6, 312–324. (b) Speers, A. E.; Cravatt, B. F. Chem. Biol. 2004, 11, 535–546. (c) Dondoni, A.; Marra, A. J. Org. Chem. 2006, 71, 7546–7557. (d) Whiting, M.; Tripp, J. C.; Lin, Y.-C.; Lindstrom, W.; Olson, A. J.; Elder, J. H.; Sharpless, K. B.; Fokin, V. V. J. Med. Chem. 2006, 49, 7697–7710. (e) El Akri, K.; Bougrin, K.; Balzarini, J.; Faraj, A.; Benhida, R. Bioorg. Med. Chem. Lett. 2007, 17, 6656–6659. (f) Moses, J. E.; Moorhouse, A. D. Chem. Soc. Rev. 2007, 36, 1249–1262. (g) Buck, S. B.; Bradford, J.; Gee, K. R.; Agnew, B. J.; Clarke, S. T.; Salic, A. Bio Techniques 2008, 44, 927–929. (h) Tron, G. C.; Pirali, T.; Billington, R. A.; Canonico, P. L.; Sorba, G.; Genazzani, A. A. Med. Res. Rev. 2008, 28, 278–308. (i) Nakane, M.; Ichikawa, S.; Matsuda, A. J. Org. Chem. 2008, 73, 1842–1851.
- (a) "Copper-free click chemistry": Arnaud, C. H. *Chem. Eng. News* **2007**, *85*, 15–15.
 (b) Baskin, J. M.; Prescher, J. A.; Laughlin, S. T.; Agard, N. J.; Chang, P. V.; Miller, I. A.; Lo, A.; Codelli, J. A.; Bertozzi, C. R. *Proc. Natl. Acad. Sci.* USA **2007**, *104*, 16793–16797.
 (c) Johnson, J. A.; Baskin, J. M.; Bertozzi, C. R.;

- Koberstein, J. T.; Turro, N. J. *Chem. Commun.* **2008**, 3064–3066. (d) Baskin, J. M.; Bertozzi, C. R.; QSAR *Comb. Sci.* **2007**, *26*, 1211–1219. (e) Geng, J.; Lindqvist, J.; Mantovani, G.; Chen, G.; Sayers, C. T.; Clarkson, G. J.; Haddleton, D. M. QSAR *Comb. Sci.* **2007**, *26*, 1220–1228.
- 56. Ramachary, D. B.; Reddy, G. B.; Rumpa, M. *Tetrahedron Lett.* **2007**, *48*, 7618–7623.
- 57. Ramachary, D. B.; Narayana, V. V.; Ramakumar, K. *Tetrahedron Lett.* **2008**, 49, 2704–2709.
- 58. X-ray crystal data of **120aa**: $C_{17}H_{19}N_3O_4S$; MW = 361.42, Monoclinic, space group *C* 2/c, with a = 8.0711(9) Å, b = 13.3560(16) Å, c = 32.165(4) Å, α = 90°, β = 90.516°, γ = 90°. CCDC-689189 contains the supplementary crystallographic data for this crystal structure.
- 59. X-ray crystal data of **121aa**: $C_{10}H_{13}N_3O_2$; MW = 207.23, Monoclinic, space group $P2_1/c$, with a = 13.520(9) Å, b = 8.995(6) Å, c = 9.344(6) Å, α = 90°, β = 107.868°, γ = 90°. CCDC-689190 contains the supplementary crystallographic data for this crystal structure.
- 60. (a) Pollex, A.; Hiersemann, M. Org. Lett. 2005, 7, 5705–5708. (b) Danheiser, L. R.; Miller, F. R.; Brisbois, G. R.; Park, Z. S. J. Org. Chem. 1990, 55, 1959–1964. (c) Brodsky, H. B.; Dubois, J. Org. Lett. 2004, 6, 2619–2621. (d) Lwowski, W.; Mattingly Jr, T. W. J. Am. Chem. Soc. 1965, 87, 1947–1958.
- (a) Taniguchi, Y.; Inanaga, J.; Yamaguchi, M. Bull. Chem. Soc. Jpn. 1981, 54, 3229–3230.
 (b) Yamamoto, Y.; Nunokawa, K.; Ohno, M.; Eguchi, S. Synthesis 1996, 949–953.
 (c) Ackland, D. J.; Pinhey, J. T. J. Chem. Soc. Perkin Trans. 1 1987, 2689–2694.
- 62. X-ray crystal data of **120ua**: $C_{17}H_{15}N_3O_2S$; MW = 325.39, Triclinic, space group p-1, with a = 6.7103(13) Å, b = 8.1812(16) Å, c = 14.451(3) α Å= 88.29°, β = 80.45°, γ = 82.94°. CCDC-690284 contains the supplementary crystallographic data for this crystal structure.

- 63. X-ray crystal data of **119sa**: $C_9H_{10}N_2O_3$; MW = 194.19, Monoclinic, space group $P2_1/c$, with a = 12.798(6) Å, b = 5.436(3) Å, c = 13.476(6) Å, α = 90°, β = 91.746°, γ = 90°. CCDC-689188 contains the supplementary crystallographic data for this crystal structure.
- 64. For selected reviews on asymmetric aminocatalysis, see: (a) List, B. Synlett.
 2001, 11, 1675–1686. (b) List, B. Acc. Chem. Res. 2004, 37, 548–557. (c) List, B. Chem. Commun. 2006, 819–824. (d) Sulzer-Mossé, S.; Alexakis, A. Chem. Commun. 2007, 3123–3135. (e) Erkkila, A.; Majander, I.; Pihko, P. M. Chem. Rev. 2007, 107, 5416–5470. (f) Mukherjee, S.; Yang, J. W.; Hoffmann, S.; List, B. Chem. Rev. 2007, 107, 5471–5569. (g) Barbas, C. F. III. Angew. Chem., Int. Ed. 2008, 47, 42–47. (h) Melchiorre, P.; Marigo, M.; Carlone, A.; Bartoli, G. Angew. Chem., Int. Ed. 2008, 47, 6138–6171. (i) Melchiorre, P. Angew. Chem., Int. Ed. 2009, 48, 1360–1363. (j) Bertelsen, S.; Jørgensen, K. A. Chem. Soc. Rev. 2009, 38, 2178–2189.
- 65. For selected recent papers on enamine-based Michael reaction of carbonyl compounds with β-nitrostyrenes, see: (a) Betancort, J. M.; Sakthivel, K.; Thayumanavan, R.; Barbas, C. F. III. *Tetrahedron Lett.* **2001**, *42*, 4441–4444. (b) Betancort, J. M.; Barbas, C. F. III. *Org. Lett.* **2001**, *3*, 3737–3740. (c) List, B.; Pojarliev, P.; Martin, H. J. *Org. Lett.* **2001**, *3*, 2423–2425. (d) Mase, N.; Thayumanavan, R.; Tanaka, F.; Barbas, C. F. III. *Org. Lett.* **2004**, *6*, 2527–2530. (e) Betancort, J. M.; Sakthivel, K.; Thayumanavan, R.; Tanaka, F.; Barbas, C. F. III. *Synthesis* **2004**, 1509–1521. (f) Cobb, A. J. A.; Longbottom, D. A.; Shaw, D. M.; Ley, S. V. *Chem. Commun.* **2004**, 1808–1809. (g) Hayashi, Y.; Gotoh, H.; Hayashi, T.; Shoji, M. *Angew. Chem., Int. Ed.* **2005**, *44*, 4212–4215. (h) García-García, P.; Ladépêche, A.; Halder, R.; List, B. *Angew. Chem., Int. Ed.* **2008**, *47*, 4719–4721. (i) Dinér, P.; Kjærsgaard, A.; Lie, M. A.; Jørgensen, K. A. *Chem.–Eur. J.* **2008**, *14*, 122–127. (j) Uehara, H.; Barbas, C. F. III. *Angew. Chem., Int. Ed.* **2009**, *48*, 9848–9852.

- 66. (a) Mampreian, D. M.; Hoveyda, A. H.
 Org. Lett. 2004, 6, 2829–2832. (b) Daniel, D.; René, R. Synthesis, 1984, 12, 1054–1057.
- 67. (a) Taber, D. F.; Sikkander, M. I.; Storck, P. H. J. Org. Chem. 2007, 72, 4098–4101. (b) Yang, Y-Q.; Chai, Z.; Wang, H-F.; Chen, X-K.; Cui, H-F.; Zheng, C-W.; Xiao, H.; Li, P.; Zhao, G. Chem.–Eur. J. 2009, 15, 13295–13298.

8. EXPERIMENTAL SECTION

1: Synthesis of chiral Hagemann's esters 88c and 88d:

First step: A magnetically stirred solution of the Hagemann's ester **88a** (12 mmol), ethylene glycol (60 mmol) and a catalytic amount of *p*-TSA in dry benzene (75 mL) was refluxed for 22 h with a dean-stark water trap. Solvent benzene was distilled off; saturated aq. NaHCO₃ solution (15 mL) was added to the residue and extracted with ether (2 x 30 mL). The combined organic layers were dried (Na₂SO₄), filtered and concentrated. Pure product was obtained by column chromatography (silica gel, mixture of hexane/ethyl acetate) as shown in Scheme E1.

Second step: A magnetically stirred solution of ketal ester (8 mmol), in methanol (15 ml) and 10% aq. NaOH (18 mL) was refluxed for 6 h. The reaction mixture was cooled and washed with DCM (2 x 10 mL). The *p*H of the reaction mixture was adjusted to 2 by stirring crude reaction mixture with 3 mL of 3N aq. HCl in THF (10 mL) for 0.5 h at 25 °C to furnish the keto-acid, which is extracted with dichloromethane (2 x 20 mL). The combined organic layers were dried with Na₂SO₄, and pure product was obtained by column chromatography (silica gel, pure ethyl acetate) as shown in Scheme E1.

Third step-1: A 10 mL oven-dried round bottom flask equipped with a stirring bar was charged with keto acid (2 mmol) and DMAP (25 mol-%) in dry CH₂Cl₂ (10 mL) and added ethyl L-lactate (2 mmol) and DCC (2 mmol). The reaction mixture was stirred under N₂ at room temperature for 4 h. The crude reaction mixture was worked up with aqueous NH₄Cl solution and the aqueous layer was extracted with dichloromethane (2 x 20 mL). DCU was removed by filtration and combined organic layers were dried (Na₂SO₄), filtered and concentrated. Pure product **88c** was obtained by column chromatography (silica gel, mixture of hexane/ethyl acetate) as shown in Scheme E1.

Scheme E1: Synthesis of Hagemann's ester 88c

Third step-2: A 10 mL oven-dried round bottom flask equipped with a stirring bar was charged with keto acid (2 mmol) and DMAP (25 mol-%) in dry CH₂Cl₂ (10 mL) and added (1*R*, 2*S*, 5*R*)-(–)-menthol (2 mmol) and DCC (2 mmol). The reaction mixture was stirred under N₂ at room temperature for 3 h. The crude reaction mixture was worked up with aqueous NH₄Cl solution and the aqueous layer was extracted with dichloromethane (2 x 20 mL). DCU was removed by filtration and combined organic layers were dried (Na₂SO₄), filtered and concentrated. Pure product **88d** was obtained by column chromatography (silica gel, mixture of hexane/ethyl acetate) as shown in Scheme E2.

Scheme E2: Synthesis of Hagemann's ester 88d

(S)-1-ethoxy-1-oxopropan-2-yl 2-methyl-4-oxocyclohex-2-enecarboxylate (88c):

Purified by column chromatography using EtOAc/hexane and isolated as a liquid. $[\alpha]^{25}_D = -16.0$ (c 0.3, CHCl₃) IR (neat): v_{max} 2926, 2852, 1741 (O-C=O), 1672 (C=O), 1252, 1211, 1177, 1134, 1097 cm⁻¹; ¹H NMR (CDCl₃, 1:1 ratio of diastereomers) δ 5.99 (2H, s, 2 X olefinic-H), 5.12 (2H, q, J = 7.2 Hz), 4.20 (4H, q, J = 6.8 Hz), 3.34 (2H, brs), 2.67-2.54 (2H, m), 2.43-2.24 (6H, m), 2.07 (3H, s, olefinic-CH₃), 2.06 (3H, s, olefinic-CH₃), 1.52 (6H, d, J = 7.2 Hz), 1.27 (6H, t, J = 7.2 Hz, 2 X OCH₂CH₃); ¹³C NMR (CDCl₃, DEPT-135, 1:1 ratio of diastereomers) δ 198.3 (2 X C, C=O), 171.1 (C, O-C=O), 171.0 (C, O-C=O), 170.3 (C, O-C=O), 170.2 (C, O-C=O), 156.5 (2 X C), 128.73 (CH), 128.66 (CH), 69.4 (CH), 69.3 (CH), 61.6 (2 X CH₂), 45.6 (2 X CH), 34.1 (CH₂), 34.0 (CH₂), 26.1 (CH₂), 26.0 (CH₂), 23.5 (CH₃, olefinic-CH₃), 23.4 (CH₃, olefinic-CH₃), 16.8 (2 X CH₃), 14.1 (2 X CH₃, OCH₂CH₃).

(1R,2S,5R)-2-isopropyl-5-methylcyclohexyl

2-methyl-4-oxocyclohex-2-

enecarboxylate (88d): Purified by column chromatography using EtOAc/hexane and isolated as a liquid. [α]²⁵_D = -62.4 (*c* 0.3, CHCl₃) IR (neat): v_{max} 2982, 2361, 2336, 1730 (O-C=O), 1673 (C=O), 1376, 1254, 1186, 1037, 857cm⁻¹; ¹H NMR (CDCl₃, 1:1 ratio of diastereomers) δ 5.95 (2H, s, 2 X olefinic-*H*), 4.76-4.69 (2H, m), 3.24 (2H, t, *J* = 4.4 Hz), 2.57-2.47 (2H, m), 2.37-2.30 (4H, m), 2.25-2.16 (2H, m), 2.00 (6H, s, 2 X olefinic-CH₃), 1.97-1.96 (2H, m), 1.84-1.78 (4H, m), 1.70-1.66 (4H, m), 1.50-1.36 (4H, m), 1.09-0.96 (4H, m), 0.91-0.87 (14H, m), 0.74 (6H, dd, *J* = 6.8, 1.6 HZ); ¹³C NMR (CDCl₃, DEPT-135, 1:1 ratio of diastereomers) δ 198.4 (2 X C, C=O), 171.1 (2 X C, O-C=O), 157.22 (C), 157.15 (C), 128.4 (CH), 128.3 (CH), 75.44 (CH), 75.37 (CH), 46.8 (2 X CH), 46.5 (CH), 46.3 (CH), 40.8 (CH₂), 40.6 (CH₂), 34.3 (CH₂), 34.2 (CH₂), 34.1 (2 X CH₂), 31.3 (2 X CH), 26.3 (2 X CH), 26.2 (2 X CH₂), 23.43 (CH₃), 23.39 (CH₃), 23.1 (2 X CH₂), 21.9 (2 X CH₃), 20.7 (2 X CH₃), 16.0 (2 X CH₃).

2a: Glycine-catalyzed Claisen-Schmidt reactions: In an ordinary glass vial equipped with a magnetic stirring bar, to 0.5 mmol of the Hagemann's ester 88a was added 1.0 mL of DMSO solvent, and then the catalyst glycine **14d** (0.1 mmol, 7.5 mg) was added and then 0.5 mmol of aldehyde 28b-28m was added in one-portion and the reaction mixture was stirred at 25 °C for the time indicated in Table 4. The crude reaction mixture was worked up with aqueous NH₄Cl solution and the aqueous layer was extracted with dichloromethane (2 x 20 mL). The combined organic layers were dried (Na₂SO₄), filtered and concentrated. The pure products **89ab-89am** were obtained by column chromatography (silica gel, mixture of hexane/ethyl acetate).

2-Methyl-4-oxo-3-(4-trifluoromethyl-benzylidene)-cyclohex-1-enecarboxylic

ĊO₂Et

89ad

purified by column chromatography using EtOAc/hexane and isolated as a light yellow solid. Mp 54 °C; IR (neat): v_{max} 2984, 1703 (C=O and O-C=O), 1456, 1324, 1241, 1168, 1126, 1064, 1017 and 838 cm⁻¹; ¹H NMR (CDCl₃, 2:1 ratio of E/Z isomers) δ 7.59 (4H, d, J = 8.4 Hz), 7.53 (2H, d, J = 8.0 Hz), 7.44-7.40 (3H, m), 6.95 (1H, s, olefinic-H), 4.26 (4H, q, J =7.2 Hz, 2 x OC H_2 CH₃), 2.83-2.78 (4H, m), 2.68 (2H, t, J = 6.4 Hz), 2.47 (2H, t, J = 6.4Hz), 2.24 (3H, s, olefinic- CH_3), 1.94 (3H, s, olefinic- CH_3), 1.33 (3H, t, J = 7.2 Hz, OCH_2CH_3), 1.32 (3H, t, J = 7.2 Hz, OCH_2CH_3); ¹³C NMR (CDCl₃, 2:1 ratio of E/Zisomers) δ 201.4 (C, C=O), 200.5 (C, C=O), 168. 4 (C, O-C=O), 167.1 (C, O-C=O), 142.6 (C), 139.9 (C), 139.5 (C), 139.3 (C), 138.9 (C), 138.1 (C), 134.8 (CH), 134.4 (CH), 131.2 (C), 130.8 (CF₃, q, J = 33.0 Hz), 129.9 (2 x CH), 129.6 (C), 129.4 (2 x CH), 125.2 (2 x CH, q, J = 3.0 Hz), 124.9 (2 x CH, q, J = 3.0 Hz), 122.6 (C), 122.4 (C), 60.9 (CH₂, OCH₂CH₃), 60.8 (CH₂, OCH₂CH₃), 39.3 (CH₂), 34.7 (CH₂), 27.2 (CH₂), 23.1 (CH₂), 20.4 (CH₃, olefinic-CH₃), 16.5 (CH₃, olefinic-CH₃), 14.2 (2 x CH₃, OCH_2CH_3); LRMS m/z 338.30 (M⁺), calcd $C_{18}H_{17}F_3O_3$ 338.1130; Anal. calcd for

ethyl ester (89ad): Prepared following the procedure 2a and

C₁₈H₁₇F₃O₃ (338.11): C, 63.90; H, 5.06. Found: C, 63.85; H, 5.12%.

3-(3-Cyano-benzylidene)-2-methyl-4-oxo-cyclohex-1-enecarboxylic acid ethyl ester

CN CO₂Et

89af *E/Z* = 10:1

(89af): Prepared following the procedure 2a and purified by column chromatography using EtOAc/hexane and isolated as a light yellow solid. Mp 52 °C; IR (neat): v_{max} 2986, 2323, 1703 (C=O), 1671 (O-C=O), 1450, 1250, 1230, 1182, 1055, 685 cm⁻¹; ¹H NMR (CDCl₃,

10:1 ratio of E/Z isomers, major isomer) δ 7.64 (1H, m), 7.59-7.56 (2H, m), 7.42 (1H, t, J = 8.0 Hz), 6.91 (1H, s, olefinic-H), 4.28 (2H, q, J = 7.2 Hz, OC H_2 CH $_3$), 2.87-2.82 (2H, m), 2.70 (2H, br t, J = 6.4 Hz), 2.25 (3H, t, J = 2.0 Hz, olefinic- CH_3), 1.35 (3H, t, J = 7.2 Hz, OCH $_2$ CH $_3$); ¹³C NMR (CDCl $_3$, 10:1 ratio of E/Z isomers, major isomer) δ 200.2 (C, C=O), 168. 3 (C, O-C=O), 139.6 (C), 138.2 (C), 137.2 (C), 133.5 (CH), 133.4 (CH), 132.7 (CH), 131.6 (CH), 130.1 (C), 128.8 (CH), 118.5 (C), 112.4 (C), 61.0 (CH $_2$, OCH $_2$ CH $_3$), 39.3 (CH $_2$), 27.1 (CH $_2$), 16.5 (CH $_3$, olefinic- $_3$ CH $_3$), 14.2 (CH $_3$, OCH $_2$ CH $_3$); LRMS m/z 296.00 (M + H $_3$ +), calcd C $_1$ 8H $_1$ 7NO $_3$ 295.1208; HRMS m/z 318.1111 (M + Na), calcd for C $_1$ 8H $_1$ 7NO $_3$ Na 318.1106; Anal. calcd for C $_1$ 8H $_1$ 7NO $_3$ (295.12): C, 73.20; H, 5.80; N, 4.74. Found: C, 73.25; H, 5.76; N, 4.81%.

2-Methyl-3-(3-nitro-benzylidene)-4-oxo-cyclohex-1-enecarboxylic acid ethyl ester

O NO₂ CO₂Et (89ag): Prepared following the procedure 2a and purified by column chromatography using EtOAc/hexane and isolated as a light yellow oily liquid. IR (neat): v_{max} 2994, 2961, 1692 (C=O and O-C=O) 1527, 1440, 1352, 1306, 1380, 1341, 1185, 1056 cm⁻¹; ¹H

89ag $_{E/Z=1.5:1}$ C=O), 1527, 1440, 1352, 1306, 1280, 1241, 1185, 1056 cm⁻¹; ¹H NMR (CDCl₃, 1.5:1 ratio of E/Z isomers) δ 8.22-8.14 (4H, m), 7.69 (1H, d, J=7.6 Hz), 7.64 (1H, d, J=7.6 Hz), 7.63 (1H, s, olefinic-H), 7.57 (1H, t, J=8.0 Hz), 7.49 (1H, t, J=8.0 Hz), 6.99 (1H, s, olefinic-H), 4.29 (4H, q, J=7.2 Hz, 2 x OC H_2 CH₃), 2.88-2.83 (4H, m), 2.71 (2H, t, J=6.4 Hz), 2.52 (2H, t, J=6.4 Hz), 2.27 (3H, t, J=2.0 Hz, olefinic-C H_3), 1.96 (3H, s, olefinic-C H_3), 1.36 (6H, t, J=7.2 Hz, 2 x OC H_2 C H_3); ¹³C NMR (CDCl₃, 1.5:1 ratio of E/Z isomers) δ 201.1 (C, C=O), 200.2 (C, C=O), 168.3 (C, O-C=O), 167.0 (C, O-C=O), 148.2 (C), 148.0 (C), 141.8 (C), 139.7 (C), 139.5 (C),

138.3 (C), 137.6 (C), 137.1 (C), 135.3 (CH), 135.1 (CH), 133.5 (CH), 133.4 (CH), 132.0 (C), 130.3 (C), 129.4 (CH), 128.9 (CH), 124.3 (CH), 124.2 (CH), 123.7 (CH), 123.0 (CH), 61.1 (CH₂, OCH₂CH₃), 61.0 (CH₂, OCH₂CH₃), 39.3 (CH₂), 34.7 (CH₂), 27.1 (CH₂), 23.2 (CH₂), 20.4 (CH₃, olefinic-CH₃), 16.6 (CH₃, olefinic-CH₃), 14.3 (2 x CH₃, OCH₂CH₃); LRMS m/z 314.25 (M - H⁺), calcd C₁₇H₁₇NO₅ 315.1107; HRMS m/z 338.1005 (M + Na), calcd for C₁₇H₁₇NO₅Na 338.1005; Anal. calcd for C₁₇H₁₇NO₅ (315.11): C, 64.75; H, 5.43; N, 4.44. Found: C, 64.78; H, 5.39; N, 4.51%.

3-(4-Bromo-benzylidene)-2-methyl-4-oxo-cyclohex-1-enecarboxylic acid ethyl ester

(89aj): Prepared following the procedure 2a and purified by column chromatography using EtOAc/hexane and isolated as a yellow liquid. IR (neat): v_{max} 1692 (C=O and O-C=O), 1594, 1270, 1242, ĊO₂Et **89aj** *E/Z* = 3:1 1196, 1055, 652 cm⁻¹; ¹H NMR (CDCl₃, 3:1 ratio of E/Z isomers) δ 7.56 (1H, s, olefinic-H), 7.50 (2H, br d, J = 8.4 Hz), 7.44 (2H, br d, J = 8.4 Hz), 7.28 (2H, br d, J = 8.4 Hz), 7.20 (2H, br d, J = 8.0 Hz), 6.88 (1H, s, olefinic-H), 4.28 (2H, q, J = 7.2 Hz, OC H_2 CH₃), 4.27 (2H, q, J = 7.2 Hz, OC H_2 CH₃), 2.86-2.78 (4H, m), 2.70 $(2H, t, J = 6.4 \text{ Hz}), 2.48 (2H, t, J = 6.4 \text{ Hz}), 2.25 (3H, t, J = 2.0 \text{ Hz}, \text{olefinic-C}H_3), 2.00$ (3H, t, J = 1.2 Hz, olefinic-C H_3), 1.36 (3H, t, J = 7.2 Hz, OC H_2 C H_3), 1.34 (3H, t, J = 1.2 Hz, olefinic-C H_3), 1.36 (3H, t, J = 1.2 Hz, OC H_2 C H_3), 1.37 (3H, t, J = 1.2 Hz, OC H_3 C $H_$ 7.2 Hz, OCH₂CH₃); 13 C NMR (CDCl₃, 3:1 ratio of E/Z isomers) δ 201.4 (C, C=O), 200.8 (C, C=O), 168.4 (C, O-C=O), 167.1 (C, O-C=O), 143.1 (C), 140.5 (C), 138.1 (C), 137.1 (C), 135.5 (CH), 135.0 (CH), 134.4 (C), 134.0 (C), 131.6 (2 x CH), 131.3 (2 x CH), 131.2 (2 x CH), 131.1 (2 x CH), 130.6 (C), 128.8 (C), 123.6 (C), 122.8 (C), 60.8 (CH₂, OCH₂CH₃), 60.7 (CH₂, OCH₂CH₃), 39.3 (CH₂), 34.8 (CH₂), 27.2 (CH₂), 23.1 (CH₂), 20.3 (CH₃, olefinic-CH₃), 16.6 (CH₃, olefinic-CH₃), 14.24 (CH₃, OCH₂CH₃), 14.19 (CH₃, OCH₂CH₃); LRMS m/z 349.00 (M + H⁺), calcd $C_{17}H_{17}BrO_3$ 348.0361; Anal. calcd for C₁₇H₁₇BrO₃ (348.03): C, 58.47; H, 4.91. Found: C, 58.41; H, 4.95%.

2b: Pyrrolidine-catalyzed Claisen-Schmidt and Michael reactions: In an ordinary glass vial equipped with a magnetic stirring bar, to 0.5 mmol of the

Hagemann's ester **88a** was added 1.0 mL of DMF solvent, and then the catalyst pyrrolidine **2a** (0.1 mmol, 8.33 μ L) was added and then 0.5 mmol of aldehyde **28a-28q** was added in one-portion and the reaction mixture was stirred at 25 °C for the time indicated in Table 5. The crude reaction mixture was worked up with aqueous NH₄Cl solution and the aqueous layer was extracted with dichloromethane (2 x 20 mL). The combined organic layers were dried (Na₂SO₄), filtered and concentrated. The pure products **89aa-aq** and **91** were obtained by column chromatography (silica gel, mixture of hexane/ethyl acetate).

(E)-2-Methyl-3-(4-nitro-benzylidene)-4-oxo-cyclohex-1-enecarboxylic acid ethyl

NO₂

ester (89aa) and (*Z*)-2-Methyl-3-(4-nitro-benzylidene)-4-oxo-cyclohex-1-enecarboxylic acid ethyl ester (89aa): Prepared following the procedure 2b and purified by column chromatography using EtOAc/hexane and isolated as a light yellow 2932, 1709 (C=O and O-C=O), 1599, 1518, 1446, 1057, 898, 852, NMR (CDCl₃, 1.24:1 ratio of E/Z isomers) δ 8.23 (2H, d, J = 8.0

89aa $_{EZ=1.24:1}$ chromatography using EtOAc/hexane and isolated as a light yellow oil. IR (neat): v_{max} 2932, 1709 (C=O and O-C=O), 1599, 1518, 1446, 1057, 898, 852, 738, 696 cm⁻¹; 1 H NMR (CDCl₃, 1.24:1 ratio of E/Z isomers) δ 8.23 (2H, d, J=8.0 Hz), 8.17 (2H, d, J=8.0 Hz), 7.63 (1H, s, olefinic-H, E-isomer), 7.49 (4H, d, J=8.0 Hz), 6.98 (1H, s, olefinic-H, Z-isomer), 4.30 (4H, q, J=7.2 Hz, 2 x CO₂CH₂CH₃), 2.85 (4H, m, CH₂), 2.72 (2H, t, J=8.0 Hz, CH₂), 2.53 (2H, t, J=8.0 Hz, CH₂), 2.26 (3H, s, olefinic-CH₃), 1.95 (3H, s, olefinic-CH₃), 1.37 (6H, t, J=7.2 Hz, 2 x CO₂CH₂CH₃); 13 C NMR (CDCl₃, 1.24:1 ratio of E/Z isomers) δ 201.0 (C, C=O), 200.2 (C, C=O), 168.2 (C, O-C=O), 166.9 (C, O-C=O), 147.7 (C), 147.1 (C), 142.9 (C), 142.0 (C), 141.8 (C), 140.3 (C), 139.4 (C), 138.8 (C), 133.6 (CH), 133.4 (CH), 132.0 (C), 130.5 (C), 130.3 (2 x CH), 129.8 (2 x CH), 123.5 (2 x CH), 123.2 (2 x CH), 61.0 (CH₂, OCH₂CH₃), 60.99 (CH₂, OCH₂CH₃), 39.3 (CH₂), 34.5 (CH₂), 27.2 (CH₂), 23.2 (CH₂), 20.5 (CH₃, olefinic-CH₃), 16.5 (CH₃, olefinic-CH₃), 14.25 (CH₃), 14.21 (CH₃); HRMS (ESI-TOF) m/z 316.1172 (M + H⁺), calcd C₁₇H₁₇NO₅H⁺ 316.1186.

(*E*)-2-Methyl-3-(benzylidene)-4-oxo-cyclohex-1-enecarboxylic acid ethyl ester (89ab): Prepared following the procedure 2b and purified by column chromatography

using EtOAc/hexane and isolated as a light yellow oil. IR (neat): v_{max} 2980, 1697

ĊO₂Et 89ab

E/Z = 5.3:1

(C=O and O-C=O), 1601, 1446, 1057, 758, 698 cm⁻¹; ¹H NMR (CDCl₃, 5.3:1 ratio of E/Z isomers, Major isomer) δ 7.67 (1H, s, olefinic-H), 7.35 (5H, m, Ph-H), 4.27 (2H, q, J = 7.2 Hz, $CO_2CH_2CH_3$), 2.79 (2H, t, J =5.2 Hz, CH_2), 2.48 (2H, t, J = 6.8 Hz, CH_2), 2.01 (3H, s, olefinic- CH_3), 1.36 (3H, t, J = 7.2 Hz, $CO_2CH_2CH_3$); ¹³C NMR (CDCl₃, 5.3:1 ratio of

E/Z isomers, Major isomer) δ 201.8 (C, C=O), 167.2 (C, O-C=O), 144.0 (C), 137.8 (C), 137.1 (CH), 135.1 (C), 129.9 (2 x CH), 129.5 (C), 129.3 (CH), 128.3 (2 x CH), 60.7 (CH₂, OCH₂CH₃), 34.98 (CH₂), 23.0 (CH₂), 20.4 (CH₃, olefinic-CH₃), 14.2 (CH₃, OCH_2CH_3); HRMS (ESI-TOF) m/z 293.1157 (M + Na⁺), calcd $C_{17}H_{18}O_3Na^+$ 293.1156.

(E)-2-Methyl-3-(4-methyl-benzylidene)-4-oxo-cyclohex-1-enecarboxylic acid ethyl

CO₂Et

89ac

ester (89ac): Prepared following the procedure 2b and purified by column chromatography using EtOAc/hexane and isolated as light yellow oil. IR (neat): v_{max} 2978, 1701 (C=O and O-C=O), 1595, 1508, 1458, 1053, 814 cm⁻¹; ¹H NMR (CDCl₃, 5.5:1 ratio of E/Z

E/Z = 5.5:1isomers, Major isomer) δ 7.65 (1H, s, olefinic-*H*), 7.23 (2H, d, J = 8.0 Hz, Ph-*H*), 7.17 (2H, d, J = 8.0 Hz, Ph-H), 4.27 (2H, q, J = 7.2 Hz, CO₂C H_2 CH₃), 2.78 (2H, t, J = 6.8Hz, CH_2), 2.46 (2H, t, J = 6.8 Hz, CH_2), 2.37 (3H, s, Ar- CH_3), 2.03 (3H, s, olefinic- CH_3), 1.35 (3H, t, J = 7.2 Hz, $CO_2CH_2CH_3$); ¹³C NMR (CDCl₃, 5.5:1 ratio of E/Z isomers, Major isomer) δ 201.8 (C, C=O), 167.2 (C, O-C=O), 144.3 (C), 139.8 (C), 137.3 (CH), 136.99 (C), 132.1 (C), 130.1 (2 x CH), 129.0 (2 x CH), 129.6 (C), 60.6 (CH₂, OCH₂CH₃), 35.0 (CH₂), 23.0 (CH₂), 21.4 (CH₃, Ar-CH₃), 20.3 (CH₃, olefinic- CH_3), 14.2 (CH_3 , OCH_2CH_3); HRMS (ESI-TOF) m/z 285.1509 (M + H⁺), calcd $C_{18}H_{20}O_3H^+$ 285.1491.

(E)-2-Methyl-3-(1-naphthalidene)-4-oxo-cyclohex-1-

89an CO₂Et E/Z = 4.0:1 enecarboxylic acid ethyl ester (89an): Prepared following the procedure 2b and purified by column chromatography using EtOAc/hexane and isolated as a light yellow oil. IR (neat): v_{max} 2978, 1699 (C=O and O-C=O), 1597, 1506, 1444, 1057, 900, 862, 779, 733 cm⁻¹; ¹H NMR (CDCl₃, 4.0:1 ratio of E/Z isomers, Major isomer) δ 8.31 (1H, s), 8.01 (1H, m), 7.88 (2H, m), 7.57 (2H, m), 7.44 (1H, m), 7.31 (1H, br d, J = 7.2 Hz), 4.25 (2H, q, J = 7.2 Hz, CO₂CH₂CH₃), 2.85 (2H, br t, J = 6.8 Hz, CH₂), 2.57 (2H, t, J = 6.8 Hz, CH₂), 1.79 (3H, s, olefinic-*C*H₃), 1.33 (3H, t, J = 7.2 Hz, CO₂CH₂CH₃); ¹³C NMR (CDCl₃, 4.0:1 ratio of E/Z isomers, Major isomer) δ 201.0 (C, C=O), 167.3 (C, O-C=O), 143.6 (C), 138.8 (C), 135.9 (CH), 133.2 (C), 133.0 (C), 131.3 (C), 130.0 (C), 129.7 (CH), 128.5 (CH), 127.3 (CH), 126.6 (CH), 126.3 (CH), 125.0 (CH), 124.6 (CH), 60.6 (CH₂, OCH₂CH₃), 35.4 (CH₂), 23.2 (CH₂), 20.1 (CH₃, olefinic-*C*H₃), 14.2 (CH₃, OCH₂CH₃); HRMS (ESI-TOF) m/z 343.1331 (M + Na⁺), calcd C₂₁H₂₀O₃Na⁺ 343.1312.

(E)-2-Methyl-3-(2-naphthalidene)-4-oxo-cyclohex-1-enecarboxylic acid ethyl ester

O CO₂Et **89ao** E/Z = 3.0:1 (89ao): Prepared following the procedure **2b** and purified by column chromatography using EtOAc/hexane and isolated as a light yellow oil. IR (neat): v_{max} 3055, 2978, 1701 (C=O and O-C=O), 1601, 1504, 1452, 1057, 912, 858, 750 cm⁻¹; ¹H NMR

(CDCl₃, 3.0:1 ratio of E/Z isomers, Major isomer) δ 7.83-7.40 (8H, m, Ph-*H*), 4.29 (2H, q, J = 7.2 Hz, CO₂CH₂CH₃), 2.84 (2H, m, CH₂), 2.50 (2H, m, CH₂), 2.03 (3H, s, olefinic-*C*H₃), 1.26 (3H, t, J = 7.2 Hz, CO₂CH₂CH₃); ¹³C NMR (CDCl₃, 3.0:1 ratio of E/Z isomers, Major isomer) δ 201.7 (C, C=O), 167.2 (C, O-C=O), 143.9 (C), 137.8 (C), 137.1 (CH), 133.4 (C), 132.8 (C), 132.6 (C), 130.2 (CH), 130.1 (C), 128.3 (CH), 127.8 (CH), 127.7 (CH), 127.1 (CH), 126.8 (CH), 126.6 (CH), 60.6 (CH₂, OCH₂CH₃), 34.9 (CH₂), 23.1 (CH₂), 20.5 (CH₃, olefinic-*C*H₃), 14.2 (CH₃, OCH₂CH₃); HRMS (ESITOF) m/z 343.1331 (M + Na⁺), calcd C₂₁H₂₀O₃Na⁺ 343.1312.

(E)-2-Methyl-3-(4-cyano-benzylidene)-4-oxo-cyclohex-1-enecarboxylic acid ethyl ester (89ae) and (Z)-2-Methyl-3-(4-cyano-benzylidene)-4-oxo-cyclohex-1-enecarboxylic acid ethyl ester (89ae): Prepared

CN CO₂Et **89ae** E/Z = 1.5:1 following the procedure **2b** and purified by column chromatography using EtOAc/hexane and isolated as a light yellow oil. IR (neat): v_{max} 2984, 1701 (C=O and O-C=O), 1602, 1442, 1055, 839, 750 cm⁻¹; ¹H NMR (CDCl₃, 1.5:1 ratio of E/Z isomers) δ 7.67 (4H, d, J = 8.4 Hz), 7.61 (1H, br s, olefinic-H), 7.44 (4H, m, Ph-H), 6.95 (1H, s, olefinic-H), 4.30 (4H, q, J = 7.2 Hz, CO₂CH₂CH₃), 2.88 (4H, m, CH₂), 2.71 (2H, t, J = 6.8 Hz, CH₂), 2.51 (2H, t, J = 6.8 Hz), 2.25 (3H, s, olefinic-CH₃), 2.03 (3H, s, olefinic-CH₃), 1.36 (6H, t, J = 7.2 Hz, 2 x CO₂CH₂CH₃); ¹³C NMR (CDCl₃, 1.5:1 ratio of E/Z isomers) δ 201.0 (C, C=O), 200.3 (C, C=O), 168.1 (C, O-C=O), 166.8 (C, O-C=O), 142.0 (C), 140.7 (C), 139.9 (C), 139.8 (C), 139.4 (C), 138.4 (C), 134.0 (CH), 133.8 (CH), 131.9 (2 x CH), 131.6 (2 x CH), 130.1 (2 x CH), 130.0 (C), 129.6 (2 x CH), 128.3 (C), 118.7 (C), 118.2 (C), 112.4 (C), 111.4 (C), 60.9 (CH₂, OCH₂CH₃), 60.8 (CH₂, OCH₂CH₃), 39.2 (CH₂), 34.5 (CH₂), 27.1 (CH₂), 23.0 (CH₂), 20.4 (CH₃, olefinic-CH₃), 16.4 (CH₃, olefinic-CH₃), 14.2 (CH₃, OCH₂CH₃), 14.1 (CH₃, OCH₂CH₃).

(E)-2-Methyl-3-(4-methoxy-benzylidene)-4-oxo-cyclohex-1-enecarboxylic acid

O CO_2Et 89ah E/Z = 3.5:1 **ethyl ester (89ah):** Prepared following the procedure **2b** and purified by column chromatography using EtOAc/hexane and isolated as a light yellow oil. ¹H NMR (CDCl₃, 3.5:1 ratio of E/Z isomers, Major isomer) δ 7.63 (1H, s, olefinic-*H*), 7.29 (2H, d, *J* =

8.4 Hz, Ph-H), 6.89 (2H, d, J = 8.4 Hz, Ph-H), 4.28 (2H, q, J = 6.8 Hz, CO₂CH₂CH₃), 3.84 (3H, s, OCH₃), 2.78 (2H, br s, CH₂), 2.45 (2H, t, J = 6.8 Hz, CH₂), 2.07 (3H, s, olefinic-CH₃), 1.36 (3H, t, J = 6.8 Hz, CO₂CH₂CH₃); ¹³C NMR (CDCl₃, 3.5:1 ratio of E/Z isomers, Major isomer) δ 201.9 (C, C=O), 167.2 (C, O-C=O), 160.7 (C), 144.6 (C), 137.0 (CH), 135.9 (C), 132.1 (2 x CH), 129.3 (C), 127.3 (C), 113.5 (2 x CH), 60.6 (CH₂, OCH₂CH₃), 55.3 (CH₃, OCH₃), 35.1 (CH₂), 23.0 (CH₂), 20.3 (CH₃, olefinic-CH₃), 14.3 (CH₃, OCH₂CH₃).

$(E)\hbox{-2-Methyl-3-} (4-Fluoro-benzylidene)\hbox{-4-oxo-cyclohex-1-enecarboxylic} \ \ acid \ \ ethyl$

ester (89ai): Prepared following the procedure 2b and purified by

column chromatography using EtOAc/hexane and isolated as a light yellow oil. IR (neat): v_{max} 2972, 1695 (C=O and O-C=O), 1595, 1504 cm⁻¹; ¹H NMR (CDCl₃, 3.5:1 ratio of E/Z isomers, Major isomer) δ 7.61 (1H, s, olefinic-*H*), 7.32 (2H, d, J = 5.6 Hz, Ph-*H*), 7.08 (2H, d, J = 5.6 Hz, Ph-*H*), 4.29 (2H, q, J = 6.8 Hz, CO₂CH₂CH₃), 2.79 (2H, br s, CH₂), 2.48 (2H, t, J = 6.8 Hz, CH₂), 2.01 (3H, s, olefinic-*C*H₃), 1.36 (3H, t, J = 6.8 Hz, CO₂CH₂CH₃); ¹³C NMR (CDCl₃, 3.5:1 ratio of E/Z isomers, Major isomer) δ 201.6 (C, C=O), 167.1 (C, O-C=O), 161.9 (C), 143.5 (C), 137.6 (C), 135.7 (CH), 131.9 (CH), 131.8 (CH), 131.1 (C), 130.3 (C), 115.6 (CH), 115.4 (CH), 60.7 (CH₂, OCH₂CH₃), 34.9 (CH₂), 23.0 (CH₂), 20.3 (CH₃, olefinic-*C*H₃), 14.2 (CH₃, OCH₂*C*H₃); HRMS (ESI-TOF) m/z 289.1228 (M + H⁺), calcd C₁₇H₁₇FO₃H⁺ 289.1240.

(E)-2-Methyl-3-(4-Chloro-benzylidene)-4-oxo-cyclohex-1-enecarboxylic acid ethyl

O CI CO₂Et **89ap** EIZ = 3.8:1 **ester (89ap):** Prepared following the procedure **2b** and purified by column chromatography using EtOAc/hexane and isolated as a light yellow oil. IR (neat): v_{max} 2978, 1697 (C=O and O-C=O), 1593, 1485, 1442, 1053, 900, 831, 744 cm⁻¹; ¹H NMR (CDCl₃, 3.8:1 ratio

of E/Z isomers, Major isomer) δ 7.59 (1H, s, olefinic-H), 7.34 (2H, d, J = 8.4 Hz, Ph-H), 7.26 (2H, d, J = 8.4 Hz, Ph-H), 4.28 (2H, q, J = 6.8 Hz, CO₂CH₂CH₃), 2.81 (2H, t, J = 6.0 Hz, CH₂), 2.48 (2H, t, J = 6.0 Hz, CH₂), 2.00 (3H, s, olefinic-CH₃), 1.36 (3H, t, J = 6.8 Hz, CO₂CH₂CH₃); ¹³C NMR (CDCl₃, 3.8:1 ratio of E/Z isomers, Major isomer) δ 201.4 (C, C=O), 167.1 (C, O-C=O), 143.2 (C), 138.0 (C), 135.4 (CH), 134.9 (C), 133.5 (C), 131.1 (2 x CH), 130.6 (C), 128.6 (2 x CH), 60.8 (CH₂, OCH₂CH₃), 34.8 (CH₂), 23.0 (CH₂), 20.3 (CH₃, olefinic-CH₃), 14.2 (CH₃, OCH₂CH₃); HRMS (ESI-TOF) m/z 327.0768 (M + Na⁺), calcd C₁₇H₁₇ClO₃Na⁺ 327.0766.

(E)-2-Methyl-3-(Furan-2-vlmethylene)-4-oxo-cyclohex-1-enecarboxylic acid ethyl

ester (89ak): Prepared following the procedure 2b and purified by column chromatography using EtOAc/hexane and isolated as a light yellow oil. IR (neat): v_{max} 2980, 1699 (C=O and O-C=O), 1572, 1466, 1093, 1057, 887, 846, 756 cm⁻¹; ¹H NMR (CDCl₃, 99:1 ratio of E/Z isomers) δ 7.79 (1H, s, olefinic-*H*), 7.53 (1H, br s), 6.99 (1H, br s), 6.51 (1H, br s), 4.26 (2H, q, J = 6.8

Hz, CO₂CH₂CH₃), 2.78 (2H, t, J = 6.4 Hz, CH₂), 2.64 (2H, t, J = 6.4 Hz, CH₂), 2.28 (3H, s, olefinic-CH₃), 1.32 (3H, t, J = 6.8 Hz, CO₂CH₂CH₃); ¹³C NMR (CDCl₃, 99:1 ratio of E/Z isomers) δ 198.9 (C, C=O), 168.5 (C, O-C=O), 151.7 (C), 145.1 (CH), 141.5 (C), 130.9 (C), 127.8 (C), 124.1 (CH), 118.2 (CH), 113.1 (CH), 60.7 (CH₂, OCH₂CH₃), 38.8 (CH₂), 25.9 (CH₂), 16.7 (CH₃, olefinic-CH₃), 14.2 (CH₃, OCH₂CH₃).

(E)-2-Methyl-3-(Thiophen-2-ylmethylene)-4-oxo-cyclohex-1-enecarboxylic acid

ethyl ester (89al): Prepared following the procedure 2b and purified by column chromatography using EtOAc/hexane and isolated as a light yellow oil. IR (neat): v_{max} 2968, 1697 (C=O and O-C=O), 1454, 1242, 904, 709 cm⁻¹; ¹H NMR (CDCl₃, 99:1 ratio of E/Z isomers) δ 7.53 (1H, d, J = 4.8 Hz), 7.48 (1H, d, J = 3.6 Hz), 7.27 (1H, s, olefinic-H), 7.10 (1H, dd, J = 4.8, 3.6 Hz), 4.27 (2H, q, J = 7.2 Hz, CO₂CH₂CH₃), 2.76 (2H, t, J = 6.4 Hz, CH₂), 2.64 (2H, t, J = 6.4 Hz, CH₂), 2.32 (3H, s, olefinic-CH₃), 1.33 (3H, t, J = 7.2 Hz, CO₂CH₂CH₃); ¹³C NMR (CDCl₃, 99:1 ratio of E/Z isomers) δ 199.6 (C, C=O), 168.5 (C, O-C=O), 141.7 (C), 138.5 (CH), 138.4 (C), 133.2 (CH), 131.7 (CH), 129.2 (C), 127.2 (C), 126.9 (CH), 60.6 (CH₂, OCH₂CH₃), 38.4 (CH₂), 25.3 (CH₂), 16.7 (CH₃, olefinic-CH₃), 14.2 (CH₃, OCH₂CH₃).

2-Methyl-4-oxo-3-(3-oxo-1-phenyl-propyl)-cyclohex-2-enecarboxylic acid ethyl

ester (91am): Prepared following the procedure 2b and purified by column chromatography using EtOAc/hexane and isolated as an oil. IR (neat): v_{max} 2980, 2727, 1728 (C=O), 1666 (O-C=O), 1190, 1028, 860, 756 cm⁻¹; ¹H NMR (CDCl₃, 1:1 mixture of diastereomers) δ 9.76 (2H, s, *H*-C=O), 7.22 (10H, m, Ph-*H*), 4.76 (1H, t, *J* = 6.8 Hz), 4.66 (1H, t, *J* = 6.8 Hz), 4.18 (4H, q, *J* = 7.2 Hz, 2 x CO₂CH₂CH₃), 3.25 (5H, m), 2.56-2.15 (9H, m), 2.06 (3H, s, olefinic-CH₃), 2.03 (3H, s, olefinic-CH₃), 1.25 (6H, t, *J* = 7.2 Hz, 2 x CO₂CH₂CH₃); ¹³C NMR (CDCl₃, 1:1 mixture of diastereomers) δ 201.2 (C, C=O), 201.1 (C, C=O), 197.2 (C, C=O), 197.1 (C, C=O), 171.7 (C, O=C-O), 171.6 (C, O=C-O), 152.8 (C), 152.6 (C), 141.9 (C), 141.8 (C), 138.5 (C), 138.2 (C), 128.3 (2 x CH), 128.2 (2 x CH),

127.3 (2 x CH), 127.1 (2 x CH), 126.1 (CH), 126.07 (CH), 61.26 (CH₂), 61.25 (CH₂), 48.51 (CH), 48.50 (CH), 46.8 (CH), 46.6 (CH), 36.4 (CH₂), 36.2 (CH₂), 35.3 (CH₂), 35.2 (CH₂), 25.4 (CH₂), 25.2 (CH₂), 21.3 (CH₃), 21.2 (CH₃), 14.1 (CH₃), 14.0 (CH₃).

2-Methyl-3-[1-(2-nitro-phenyl)-3-oxo-propyl]-4-oxo-cyclohex-2-enecarboxylic acid

ethyl ester (91aq): Prepared following the procedure 2b and purified by column chromatography using EtOAc/hexane and isolated as a light yellow oily liquid. IR (neat): v_{max} 2980, 2930, 2728, 1730 (C=O), 1665 (O-C=O), 1532, 1368, 743 cm⁻¹; ¹H NMR (CDCl₃, 1.2:1 ratio of CO₂Et diastereomers) δ 9.80 (1H, s, H-C=O), 9.72 (1H, s, H-C=O), 7.71 (1H, d, J = 8.4 Hz), 7.59 (1H, d, J = 8.0 Hz), 7.55-7.48 (4H, m), 7.36-7.30 (2H, m), 5.10 (2H, t, J = 6.8 Hz), 4.18 (2H, q, J = 7.2 Hz, OC H_2 CH₃), 4.14 (2H, q, J = 7.2 Hz, OC H_2 CH₃), 3.47-3.35 (2H, m), 3.30 (1H, t, J = 4.4 Hz), 3.27 (1H, t, J = 4.4 Hz), 3.21-3.09 (2H, m), 2.54-2.46 (1H, m), 2.42-2.35 (2H, m), 2.25-2.17 (4H, m), 2.11-2.06 (1H, m), 2.04 (3H, s, olefinic- CH_3), 1.83 (3H, s, olefinic- CH_3), 1.27 (3H, t, J = 7.2 Hz, OCH_2CH_3), 1.22 (3H, t, J = 7.2 Hz, OCH₂CH₃); ¹³C NMR (CDCl₃, 1.2:1 ratio of diastereomers) δ 200.4 (C, H-C=O), 200.3 (C, H-C=O), 198.1 (C, C=O), 197.2 (C, C=O), 171.45 (C, O-C=O), 171.42 (C, O-C=O), 154.8 (C), 154.3 (C), 149.8 (C), 149.4 (C), 136.5 (C), 136.1 (2 x C), 134.8 (C), 132.6 (CH), 131.9 (CH), 130.3 (CH), 130.0 (CH), 127.4 (CH), 127.2 (CH), 124.2 (CH), 123.8 (CH), 61.5 (CH₂, OCH₂CH₃), 61.4 (CH₂, OCH₂CH₃), 48.4 (CH), 48.1 (CH), 46.6 (CH₂), 45.7 (CH₂), 34.9 (CH₂), 34.8 (CH₂), 33.4 (CH), 33.0 (CH), 25.0 (CH₂), 24.9 (CH₂), 21.4 (CH₃, olefinic-CH₃), 21.0 (CH₃, olefinic-CH₃), 14.1 (CH₃, OCH₂CH₃), 14.0 (CH₃, OCH₂CH₃); LRMS m/z 360.05 (M + H⁺), calcd $C_{19}H_{21}NO_6$ 359.1369; HRMS m/z 382.1268 (M + Na), calcd for $C_{19}H_{21}NO_6Na$ 382.1267; Anal. calcd for C₁₉H₂₁NO₆ (359.13): C, 63.50; H, 5.89; N, 3.90. Found: C, 63.41; H, 5.93; N, 3.95%.

2c: (S)-1-(2-Pyrrolidinylmethyl)pyrrolidine-catalyzed Claisen-Schmidt reactions: In an ordinary glass vial equipped with a magnetic stirring bar, to 0.5 mmol of the Hagemann's ester 88a was added 1.0 mL of DMF solvent, and then the catalyst

(S)-1-(2-pyrrolidinylmethyl)pyrrolidine **2i** (0.1 mmol, 16.3 μL) was added and then 0.5 mmol of aldehyde **28b-2s & 45d** was added in one-portion and the reaction mixture was stirred at 25 °C for the time indicated in Table 6. The crude reaction mixture was worked up with aqueous NH₄Cl solution and the aqueous layer was extracted with dichloromethane (2 x 20 mL). The combined organic layers were dried (Na₂SO₄), filtered and concentrated. The pure products **89ab-as** were obtained by column chromatography (silica gel, mixture of hexane/ethyl acetate).

3-(4-Dimethylamino-benzylidene)-2-methyl-4-oxo-cyclohex-1-enecarboxylic acid

O NMe₂ CO_2Et 89ar E/Z = 99:1 **ethyl ester (89ar):** Prepared following the procedure **2c** and purified by column chromatography using EtOAc/hexane and isolated as a yellow liquid. IR (neat): v_{max} 2926, 2855, 1705 (C=O), 1678 (O-C=O), 1595, 1522, 1370, 1055, 808 cm⁻¹; ¹H NMR (CDCl₃, 99:1 ratio of *E/Z* isomers) δ 7.66 (1H, s, olefinic-

H), 7.26 (2H, d, J = 8.8 Hz), 6.65 (2H, d, J = 8.8 Hz), 4.27 (2H, q, J = 7.2 Hz, OC H_2 CH $_3$), 3.03 (6H, s, N(C H_3) $_2$), 2.76 (2H, t, J = 6.0 Hz), 2.43 (2H, t, J = 6.4 Hz), 2.16 (3H, s, olefinic-C H_3), 1.36 (3H, t, J = 7.2 Hz, OCH $_2$ C H_3); ¹³C NMR (CDCl $_3$, 99:1 ratio of E/Z isomers) δ 202.0 (C, C=O), 167.4 (C, O-C=O), 151.2 (C), 145.9 (C), 138.4 (CH), 133.3 (C), 132.8 (2 x CH), 128.0 (C), 122.0 (C), 111.2 (2 x CH), 60.4 (CH $_2$, OCH $_2$ CH $_3$), 40.0 (2 x CH $_3$, N(CH $_3$) $_2$), 35.5 (CH $_2$), 23.0 (CH $_2$), 20.5 (CH $_3$, olefinic-CH $_3$), 14.4 (CH $_3$, OCH $_2$ CH $_3$); LRMS m/z 314.25 (M + H $^+$), calcd C1 $_9$ H23NO $_3$ 313.1678; Anal. calcd for C1 $_9$ H23NO $_3$ (313.16): C, 72.82; H, 7.40; N, 4.47. Found: C, 72.91; H, 7.35; N, 4.56%.

2-Methyl-3-(3-methyl-but-2-enylidene)-4-oxo-cyclohex-1-enecarboxylic acid ethyl

ester (89ad): Prepared following the procedure 2c and purified by column chromatography using EtOAc/hexane and isolated as a red colored liquid. IR (neat): v_{max} 2978, 2930, 1713 (C=O), 1684 (O-C=O), 1597, 1447, 89ad

F/Z = 1:5

1372, 1246 cm⁻¹; ¹H NMR (CDCl₃, 1:5 ratio of E/Z isomers, major isomer) δ 7.53 (1H, d, J = 12.8 Hz), 6.29 (1H, d, J = 12.8 Hz), 4.26 (2H, q, J = 7.2 Hz, OC H_2 CH₃), 2.66 (2H, br t, J = 6.4 Hz), 2.43 (2H, t, J = 6.4 Hz), 2.38 (3H, s, olefinic-C H_3), 1.98 (3H, s), 1.96 (3H, s), 1.34 (3H, t, J = 7.2 Hz, OCH₂CH₃); ¹³C NMR (CDCl₃, 1:5 ratio of E/Z isomers, major isomer) δ 201.5 (C, C=O), 167.7 (C, O-C=O), 150.9 (C), 143.9 (C), 133.5 (C), 133.2 (CH), 128.1 (C), 122.3 (CH), 60.5 (CH₂, OCH₂CH₃), 35.9 (CH₂), 27.6 (CH₃), 23.1 (CH₂), 21.6 (CH₃), 18.9 (CH₃, olefinic CH₃), 14.3 (CH₃, OCH₂CH₃); LRMS m/z 249.15 (M + H⁺), calcd C₁₅H₂₀O₃ 248.1412; Anal. calcd for C₁₅H₂₀O₃ (248.14): C, 72.55; H, 8.12. Found: C, 72.61; H, 8.08%.

(E, E)-bis-1,4-[2-Methyl-3-methylidenyl-4-oxo-cyclohex-1-enecarboxylic acid ethyl

ester]-benzene (89as): Prepared following the procedure 2c and purified by column chromatography using EtOAc/hexane and isolated as a light yellow solid and it has a C_2 symmetry. IR (neat): v_{max} 2970, 1701 (C=O and O-C=O), 1577, 1377, 1051, 831 cm⁻¹; ¹H NMR (CDCl₃, 99:1 ratio of E/Z isomers) δ 7.63 (2H, s, olefinic-H), 7.34 (4H, s, Ph-H), 4.29 (2H, q, J = 6.8 Hz, CO₂CH₂CH₃), 2.82 (4H, m, CH₂), 2.48 (4H, t, J = 6.0 Hz, CH₂), 2.02 (3H, s, olefinic-H), 1.36 (3H, t, H = 6.8 Hz, CO₂CH₂CH₃); ¹³C NMR (CDCl₃, 99:1 ratio of E/Z isomers) δ 201.6 (C, C=O), 167.1 (C, O-C=O), 143.4 (C), 138.5 (C), 136.2 (C), 135.9 (CH), 130.6 (C), 129.9 (4 x CH), 60.8 (CH₂, OCH₂CH₃), 34.8 (CH₂), 23.1 (CH₂), 20.4 (CH₃, olefinic-H₃), 14.3 (CH₃, OCH₂H₃); GCMS: m/z 462.20 (M⁺), calcd C₂₈H₃₀O₆ 462.2042.; base peak: m/z 416.15 (M-EtOH).

(E, Z, E)-2-Methyl-4-oxo-3-(3-phenyl-allylidene)-cyclohex-1-enecarboxylic acid

ethyl ester (89am): Prepared following the procedure 2c and purified by column chromatography using EtOAc/hexane and isolated as a light yellow solid. IR (neat): v_{max} 3078, 2980, 1693, 1572, 1552, CO_2Et 89am E/Z = 1:6 1446, 1055, 837, 752, 690 cm⁻¹; ¹H NMR (CDCl₃, 1:6 ratio of E/Z isomers, major isomer) δ 8.21 (1H, dd, J = 15.6, 11.6 Hz), 7.54-7.52 (2H, m), 7.37-7.28 (3H, m), 6.96 (1H, d, J = 15.6 Hz), 6.83 (1H, d, J = 11.6 Hz), 4.24 (2H, q, J = 7.2 Hz,

CO₂CH₂CH₃), 2.77 (2H, br t, J = 6.0 Hz, CH₂), 2.62 (2H, dt, J = 6.8, 0.8 Hz, CH₂), 2.26 (3H, t, J = 1.6 Hz, olefinic-CH₃), 1.34 (3H, t, J = 7.2 Hz, CO₂CH₂CH₃); ¹³C NMR (CDCl₃, 1:6 ratio of E/Z isomers, major isomer) δ 200.6 (C, C=O), 168.6 (C, O-C=O), 143.7 (CH), 141.4 (C), 138.5 (CH), 136.5 (C), 132.6 (C), 129.1 (CH), 128.7 (2 x CH), 127.6 (2 x CH), 127.5 (C), 126.4 (CH), 60.6 (CH₂), 39.3 (CH₂), 25.8 (CH₂), 16.4 (CH₃), 14.2 (CH₃); HRMS (ESI-TOF) m/z 319.1294 (M + Na⁺), calcd C₁₉H₂₀O₃+Na 319.1305.

(E, Z, E)-2-Methyl-4-oxo-3-[1-deutero-3-phenyl-allylidene]-cyclohex-1-

enecarboxylic acid ethyl ester (89am): Prepared following the procedure 2c and purified by column chromatography using EtOAc/hexane and isolated as a light yellow solid. ¹H NMR (CDCl₃) δ (CO₂Et ^{89am} 8.20 (1H, d, J = 16.0 Hz), 7.53(2H, br d, J = 7.2 Hz, Ph-H), 7.37-7.28 (3H, m, Ph-H), 6.95 (1H, d, J = 15.6 Hz), 4.25 (2H, q, J = 7.2 Hz, CO₂C H_2 CH₃), 2.77 (2H, br t, J = 6.0 Hz, C H_2), 2.61 (2H, t, J = 6.8 Hz, C H_2), 2.25 (3H, t, J = 1.6 Hz, olefinic-C H_3), 1.33 (3H, t, J = 6.8 Hz, CO₂C H_2 C H_3); ¹³C NMR (CDCl₃) δ 200.6 (C, C=O), 168.5 (C, O-C=O), 143.7 (CH), 141.3 (C), 136.5 (C), 132.5 (C), 129.1 (CH), 128.7 (2 x CH), 127.6 (2 x CH), 127.4 (C), 126.3 (CH), 60.6 (CH₂), 39.2 (CH₂), 25.8 (CH₂), 16.3 (CH₃), 14.2 (CH₃).

2d: (S)-1-(2-Pyrrolidinylmethyl)pyrrolidine-catalyzed cascade Claisen-Schmidt and iso-aromatization reactions: In an ordinary glass vial equipped with a magnetic stirring bar, to 0.5 mmol of the Hagemann's esters 88a-s was added 1.0 mL of DMSO solvent, and then the catalyst (S)-1-(2-pyrrolidinylmethyl)pyrrolidine 2i (0.1 mmol, 16.3 μ L) was added and then 0.5 mmol of aldehydes 28a-u was added in one-portion and the reaction mixture was stirred at 25 °C for the time indicated in Table 7. The crude reaction mixture was worked up with aqueous NH₄Cl solution and the aqueous layer was extracted with dichloromethane (2 x 20 mL). The combined organic layers were dried (Na₂SO₄), filtered and concentrated. The pure cascade products 90 were obtained by column chromatography (silica gel, mixture of hexane/ethyl acetate).

4-Hydroxy-2-methyl-3-(4-nitrobenzyl)-benzoic acid ethyl ester (90aa): Prepared

OH following the procedure **2d** and purified by column chromatography using EtOAc/hexane and isolated as a white solid. IR (neat): v_{max} 3339 (O-H), 2985, 1678 (O-C=O), 1595, 1578, 1519, 1345, 1292, 1263, 1174, 1153, 1108, 1049, 1007, 774, 749, 732 cm⁻¹; ¹H NMR (CDCl₃) δ 8.05 (2H, td, J = 8.8, 2.0 Hz), 7.71 (1H, d, J = 8.8 Hz), 7.25 (2H, d, J = 8.8 Hz), 6.76 (1H, d, J = 8.8 Hz) [Ar-H]; 6.24 (1H, s, O-H), 4.32 (2H, q, J = 7.2 Hz, CO₂CH₂CH₃), 4.19 (2H, s, -CH₂Ar), 2.44 (3H, s, olefinic-CH₃), 1.37 (3H, t, J = 6.8 Hz, CO₂CH₂CH₃); ¹³C NMR (CDCl₃) δ 168.3 (C, O-C=O), 156.9 (C), 148.1 (C), 146.2 (C), 141.0 (C), 130.9 (CH), 128.8 (2 x CH), 125.2 (C), 123.8 (C), 123.6 (2 x CH), 112.5 (CH), 60.8 (CH₂, OCH₂CH₃), 31.7 (CH₂), 17.0 (CH₃), 14.3 (CH₃); HRMS (ESI-TOF) m/z 316.1180 (M + H⁺), calcd for C₁₇H₁₇O₅NH⁺ 316.1179.

4-Hydroxy-2-methyl-3-(4-cyanobenzyl)-benzoic acid ethyl ester (90ae): Prepared

following the procedure **2d** and purified by column chromatography using EtOAc/hexane and isolated as a white solid. IR (neat): v_{max} 3335 (O-H), 2989, 1682 (O-C=O), 1579, 1267, 1153, 1049, 1006, 779, 756 cm⁻¹; ¹H NMR (CDCl₃) δ 7.71 (1H, d, J = 8.4 Hz), 7.49 (2H, d, J = 8.4 Hz), 7.22 (2H, d, J = 8.4 Hz), 6.78 (1H, d, J = 8.4 Hz) [Ar-H]; 6.10 (1H, s, O-H), 4.32 (2H, q, J = 6.8 Hz, CO₂CH₂CH₃), 4.16 (2H, s, -CH₂Ar), 2.42 (3H, s, olefinic-CH₃), 1.36 (3H, t, J = 6.8 Hz, CO₂CH₂CH₃); ¹³C NMR (CDCl₃) δ 168.6 (C, O-C=O), 157.6 (C), 146.3 (C), 140.9 (C), 132.0 (2 x CH), 130.8 (CH), 128.9 (2 x CH), 125.2 (C), 123.1 (C), 118.9 (C), 112.4 (CH), 109.0 (C), 60.8 (CH₂, OCH₂CH₃), 31.8 (CH₂, CH₂Ar), 16.97 (CH₃, olefinic-CH₃), 14.2 (CH₃); GCMS m/z 295.12 (M⁺), calcd for C₁₈H₁₇O₃N 295.1208.

4-Hydroxy-2, 6-dimethyl-3-(4-nitrobenzyl)-benzoic acid ethyl ester (90ia): Prepared

following the procedure **2d** and purified by column chromatography using EtOAc/hexane and isolated as a light yellow solid. IR (neat): v_{max} 3385 (O-H), 2976, 1716 (O-C=O), 1651, 1601, 1518, 1344, 1151, 1051, 734 cm⁻¹; ¹H NMR (CDCl₃) δ 8.06 (2H, dd, J =

8.8, 2.0 Hz), 7.28 (2H, br d, J = 8.8 Hz), 6.54 (1H, s) [Ar-H]; 6.07 (1H, s, O-H), 4.37 (2H, q, J = 7.2 Hz, CO₂CH₂CH₃), 4.11 (2H, s, -CH₂Ar), 2.25 (3H, s, olefinic-CH₃), 2.16 (3H, s, olefinic-CH₃), 1.37 (3H, t, J = 7.2 Hz, CO₂CH₂CH₃); ¹³C NMR (CDCl₃) δ 170.6 (C, O-C=O), 154.6 (C), 148.3 (C), 146.2 (C), 135.6 (C), 134.9 (C), 128.9 (2 x CH), 127.8 (C), 123.5 (2 x CH), 121.9 (C), 114.6 (CH), 61.1 (CH₂, OCH₂CH₃), 31.5 (CH₂, CH₂Ar), 19.6 (CH₃, olefinic-CH₃), 16.8 (CH₃, olefinic-CH₃), 14.2 (CH₃).

4-Hydroxy-2, 6-dimethyl-3-(4-cyanobenzyl)-benzoic acid ethyl ester (90ie):

Prepared following the procedure **2d** and purified by column chromatography using EtOAc/hexane and isolated as a white solid. IR (neat): v_{max} 3398 (O-H), 2986, 1718 (O-C=O), 1604, 1504, 1051, 852, 615 cm⁻¹; ¹H NMR (CDCl₃) δ 7.49 (2H, d, J = 8.4 Hz), 7.22 (2H, d, J = 8.4 Hz), 6.53 (1H, s) [Ar-H]; 6.07 (1H, s, O-H), 4.36 (2H, q, J = 7.2 Hz, CO₂CH₂CH₃), 4.07 (2H, s, -CH₂Ar), 2.24 (3H, s, olefinic-CH₃), 2.14 (3H, s, olefinic-CH₃), 1.37 (3H, t, J = 7.2 Hz, CO₂CH₂CH₃); ¹³C NMR (CDCl₃) δ 170.6 (C, O-C=O), 154.6 (C), 146.1 (C), 135.6 (C), 134.9 (C), 132.1 (2 x CH), 128.9 (2 x CH), 127.7 (C), 121.9 (C), 119.1 (C), 114.6 (CH), 109.4 (C), 61.0 (CH₂, OCH₂CH₃), 31.7 (CH₂, CH₂Ar), 19.6 (CH₃, olefinic-CH₃), 16.8 (CH₃, olefinic-CH₃), 14.2 (CH₃).

4-Hydroxy-2-methyl-3-(2-nitro-benzyl)-benzoic acid ethyl ester (90at): Prepared

OH NO2 following the procedure **2d** and purified by column chromatography using EtOAc/hexane and isolated as a yellow solid. Mp 122 °C; IR (neat): v_{max} 3369 (O-H), 1691 (O-C=O), 1585, 1524, 1352, 1295, 1265 cm⁻¹; ¹H NMR (CDCl₃) δ 7.93 (1H, dd, J = 8.0, 1.6 Hz), 7.74 (1H, d, J = 8.4 Hz), 7.36 (1H, dt, J = 7.6, 1.6 Hz), 7.30 (1H, dt, J = 7.6, 1.2 Hz), 6.84 (1H, br d, J = 7.6 Hz); 6.79 (1H, br s, O-H), 6.75 (1H, d, J = 8.8 Hz), 4.38 (2H, s, CH_2Ar), 4.32 (2H, q, J = 7.2 Hz, OCH_2CH_3), 2.38 (3H, s, Ar- CH_3), 1.37 (3H, t, J = 7.2 Hz, OCH_2CH_3); ¹³C NMR (CDCl₃, DEPT-135) δ 168.8 (C, O-C=O), 157.7 (C), 149.4 (C), 141.3 (C), 134.8 (C), 133.1 (CH), 131.2 (CH), 129.4 (CH), 126.9 (CH), 124.6 (CH), 124.6 (C), 123.6 (C), 112.6 (CH), 61.0 (CH₂, OCH_2CH_3), 28.3 (CH₂, CH_2Ar), 16.9

 $(CH_3, Ar-CH_3)$, 14.3 (CH_3, OCH_2CH_3) ; LRMS m/z 314.00 $(M - H^+)$, calcd $C_{17}H_{17}NO_5$ 315.1107; Anal. calcd for C₁₇H₁₇NO₅ (315.11): C, 64.75; H, 5.43; N, 4.44. Found: C, 64.81; H, 5.49; N, 4.40%.

4-Hydroxy-2-methyl-3-(3-nitro-benzyl)-benzoic acid ethyl ester (90ag): Prepared

NO₂ Me ĊO₂Et 90ag

following the procedure 2dand purified chromatography using EtOAc/hexane and isolated as a solid. Mp 150 °C; IR (neat): v_{max} 3358 (O-H), 1678 (O-C=O), 1581, 1526, 1350, 1292, 1178 cm⁻¹; ¹H NMR (CDCl₃) δ 8.02 (2H, br s), 7.74

(1H, d, J = 8.4 Hz), 7.46 (1H, d, J = 7.6 Hz), 7.39 (1H, t, J = 7.6 Hz), 6.74 (1H, d, J = 7.6 Hz)8.4 Hz); 6.04 (1H, s, O-H), 4.33 (2H, q, J = 7.2 Hz, OC H_2 CH $_3$), 4.20 (2H, s, C H_2 Ar), 2.48 (3H, s, Ar-C H_3), 1.37 (3H, t, J = 7.2 Hz, OCH₂C H_3); ¹³C NMR (CDCl₃, DEPT-135) δ 168.2 (C, O-C=O), 156.8 (C), 148.4 (C), 142.2 (C), 141.1 (C), 134.4 (CH), 131.0 (CH), 129.1 (CH), 125.4 (C), 124.0 (C), 123.1 (CH), 121.1 (CH), 112.6 (CH), 60.8 (CH₂, OCH₂CH₃), 31.4 (CH₂, CH₂Ar), 17.0 (CH₃, Ar-CH₃), 14.3 (CH₃, OCH₂CH₃); LRMS m/z 316.00 (M + H⁺), calcd $C_{17}H_{17}NO_5$ 315.1107; HRMS m/z 338.1002 (M + Na), calcd for C₁₇H₁₇NO₅Na 338.1005; Anal. calcd for C₁₇H₁₇NO₅ (315.11): C, 64.75; H, 5.43; N, 4.44. Found: C, 64.55; H, 5.38; N, 4.55%.

3-(3-Cyano-benzyl)-4-hydroxy-2-methyl-benzoic acid ethyl ester (90af): Prepared

Me ĊO₂Et 90af

following the procedure 2d and purified by column chromatography using EtOAc/hexane and isolated as a white solid. Mp 116 °C; IR (neat): v_{max} 3361 (O-H), 2229, 1687 (O-C=O), 1678, 1582, 1459, 1291, 1269, 1180, 1156, 1054 cm⁻¹; ¹H NMR (CDCl₃) δ 7.74 (1H, d, J = 8.4 Hz), 7.44 (1H, br s, O-H), 7.45 – 7.41 (3H, m), 7.32 (1H, t, J = 7.6 Hz), 6.82 (1H, d, J = 8.4 Hz); 4.34 (2H, q, J = 7.2 Hz, OC H_2 CH₃), 4.14 (2H, s, C H_2 Ar), 2.46 (3H, s, Ar-C H_3), 1.38 (3H, t, J = 7.2 Hz, OCH₂C H_3); ¹³C NMR (CDCl₃, DEPT-135) δ 168.7 (C, O-C=O), 157.6 (C), 141.8 (C), 140.8 (C), 133.0 (CH), 131.6 (CH), 130.9 (CH), 129.5 (CH), 128.9 (CH), 125.4 (C), 123.1 (C), 119.0 (C), 112.5 (CH), 111.7 (C), 60.8 (CH₂, OCH₂CH₃), 31.1 (CH₂, CH₂Ar), 17.0 (CH₃, Ar-CH₃), 14.2 (CH₃, OCH_2CH_3); LRMS m/z 294.00 (M - H⁺), calcd $C_{18}H_{17}NO_3$ 295.1208; Anal. calcd for C₁₈H₁₇NO₃ (295.12): C, 73.20; H, 5.80; N, 4.74. Found: C, 73.15; H, 5.88; N, 4.81%.

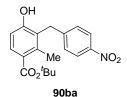
4-Hydroxy-2-methyl-3-(4-trifluoromethyl-benzyl)-benzoic acid ethyl ester (90ad):

Prepared following the procedure 2d and purified by column

Ме ĊO₂Et

chromatography using EtOAc/hexane and isolated as a white solid. ^{CF₃} Mp 120 °C; IR (neat): v_{max} 3347 (O-H), 1681 (O-C=O), 1580, 1326, 1262, 1156, 1112, 1068, 646 cm⁻¹; ¹H NMR (CDCl₃, at 25 90ad °C) δ 7.72 (1H, d, J = 8.4 Hz), 7.48 (2H, d, J = 8.0 Hz), 7.22 (2H, d, J = 8.0 Hz), 6.75 (1H, d, J = 8.8 Hz); 6.74(1H, s, O-H), $4.34(2H, q, J = 7.2 \text{ Hz}, OCH_2CH_3)$, 4.17(2H, s, O-H) CH_2Ar), 2.47 (3H, s, Ar- CH_3), 1.38 (3H, t, J = 7.2 Hz, OCH_2CH_3); ¹³C NMR (CDCl₃, DEPT-135, at 40 °C) δ 168.8 (C, O-C=O), 157.3 (C), 144.2 (C), 141.1 (C), 130.7 (CH), 128.4 (2 x CH), 128.3 (CF₃, q, J = 32.0 Hz), 126.0 (C), 125.2 (2 x CH, q, J = 4.0 Hz), 123.8 (C), 123.0 (C), 112.6 (CH), 60.9 (CH₂, OCH₂CH₃), 31.6 (CH₂, CH₂Ar), 17.0 $(CH_3, Ar-CH_3)$, 14.2 (CH_3, OCH_2CH_3) ; LRMS m/z 339.00 $(M + H^+)$, calcd $C_{18}H_{17}F_3O_3$ 338.1130; HRMS m/z 361.1310 (M + Na), calcd for $C_{18}H_{17}F_3O_3Na$ 361.1028; Anal. calcd for C₁₈H₁₇F₃O₃ (338.11): C, 63.90; H, 5.06. Found: C, 63.85; H, 5.11%.

4-Hydroxy-2-methyl-3-(4-nitro-benzyl)-benzoic acid *tert*-butyl ester (90ba):



Prepared following the procedure 2d and purified by column chromatography using EtOAc/hexane and isolated as a solid. Mp 98 °C; IR (neat): v_{max} 3387 (O-H), 2982, 2930, 1668 (O-C=O), 1593, 1516, 1344, 1285, 1132, 1011, 849 cm⁻¹; ¹H NMR (CDCl₃) δ 8.04 (2H, d, J = 8.8 Hz), 7.59 (1H, d, J = 8.4 Hz), 7.24 (2H, d, J = 8.4 Hz), 6.84 (1H, br)s, O-H), 6.71 (1H, d, J = 8.4 Hz), 4.17 (2H, s, CH_2Ar), 2.40 (3H, s, $Ar-CH_3$), 1.58 (9H,

s, 3 x CH₃, O-C(CH₃)₃); ¹³C NMR (CDCl₃, DEPT-135) δ 168.4 (C, O-C=O), 156.7 (C), 148.3 (C), 146.1 (C), 140.0 (C), 130.5 (CH), 128.8 (2 x CH), 125.5 (C), 125.1 (C), 123.5 (2 x CH), 112.4 (CH), 81.5 (C, O-C(CH₃)₃), 31.7 (CH₂, CH₂Ar), 28.2 (3 x CH₃, O-C(CH₃)₃), 17.1 (CH₃, Ar-CH₃); LRMS m/z 342.10 (M - H⁺), calcd C₁₉H₂₁NO₅ 343.1420; Anal. calcd for C₁₉H₂₁NO₅ (343.14): C, 66.46; H, 6.16; N, 4.08. Found: C, 66.35; H, 6.11; N, 4.21%.

4-Hydroxy-2-methyl-3-(4-nitro-benzyl)-benzoic acid 1-ethoxycarbonyl-ethyl ester

OH (90ca): Prepared following the procedure 2d and purified by column chromatography using EtOAc/hexane and isolated as a colorless liquid; $[\alpha]^{25}_{D} = +7.6$ (c 0.3, CHCl₃); IR (neat): v_{max} 3412 (O-H), 2984, 2938, 1717 (O-C=O), 1586, 1520, 1451, 1346, 1098, 754 cm⁻¹; ¹H NMR (CDCl₃) δ 8.08 (2H, d, J = 7.6 Hz), 7.80 (1H, d, J = 8.4 Hz), 7.26 (2H, d, J = 8.4 Hz), 6.73 (1H, d, J = 8.8 Hz), 6.45 (1H, s, O-H), 5.26 (1H, br q, J = 7.2 Hz, CH), 4.24 (2H, br q, J = 7.2 Hz, OCH₂CH₃), 4.18 (2H, br s, CH₂Ar), 2.45 (3H, s, Ar-CH₃), 1.60 (3H, d, J = 7.2 Hz), 1.30 (3H, t, J = 7.2 Hz, OCH₂CH₃); ¹³C NMR

Ar-C H_3), 1.60 (3H, d, J = 7.2 Hz), 1.30 (3H, t, J = 7.2 Hz, OCH₂C H_3); ¹³C NMR (CDCl₃, DEPT-135) δ 171.8 (C, O-C=O), 167.3 (C, O-C=O), 157.8 (C), 148.3 (C), 146.2 (C), 141.6 (C), 131.4 (CH), 128.9 (2 x CH), 125.4 (C), 123.6 (2 x CH), 122.2 (C), 112.7 (CH), 68.9 (CH), 61.7 (CH₂, OCH₂CH₃), 31.7 (CH₂, CH₂Ar), 17.1 (CH₃, Ar-CH₃), 17.0 (CH₃), 14.1 (CH₃, OCH₂CH₃); LRMS m/z 386.15 (M - H⁺), calcd C₂₀H₂₁NO₇ 387.1318; Anal. calcd for C₂₀H₂₁NO₇ (387.13): C, 62.01; H, 5.46; N, 3.62.

4-Hydroxy-2-methyl-3-(4-nitro-benzyl)-benzoic acid 2-isopropyl-5-methyl-

OH $Me \qquad NO_2$ $CO_2(+)\text{-menthyl}$ (-)-90da

Found: C, 62.15; H, 5.51; N, 3.67%.

cyclohexyl ester (90da): Prepared following the procedure **2d** and purified by column chromatography using EtOAc/hexane and isolated as a white solid. Mp 150 °C; $[\alpha]^{25}_D = -37.3$ (*c* **0.3**, **CHCl₃);** IR (neat): ν_{max} 3403 (O-*H*), 2957, 1672 (O-C=O), 1582,

1516, 1454, 1344, 1294, 1248, 1042, 858 cm⁻¹; 1 H NMR (CDCl₃) δ 8.09 (2H, d, J = 8.0

Hz), 7.71 (1H, d, J = 8.8 Hz), 7.28 (2H, d, J = 8.0 Hz), 6.73 (1H, d, J = 8.8 Hz); 5.79 (1H, s, O-H), 4.89 (1H, dt, J = 10.4, 4.0 Hz), 4.21 (2H, s, CH_2Ar), 2.45 (3H, s, $Ar-CH_3$), 2.11 (1H, br d, J = 11.6 Hz), 2.00-1.93 (1H, m), 1.74 (1H, br s), 1.70 (1H, br s), 1.51 (2H, t, J = 10.8 Hz), 1.15-1.04 (2H, m), 0.92 (1H, m), 0.91 (6H, d, J = 6.8 Hz, 2 x CH₃), 0.79 (3H, d, J = 6.8 Hz, CH_3); ^{13}C NMR (CDCl₃, DEPT-135) δ 167.7 (C, O-C=O), 156.6 (C), 148.1 (C), 146.3 (C), 141.0 (C), 130.6 (CH), 128.9 (2 x CH), 125.2 (C), 124.6 (C), 123.6 (2 x CH), 112.5 (CH), 74.7 (CH), 47.2 (CH), 41.0 (CH₂), 34.2 (CH₂), 31.7 (CH₂), 31.5 (CH), 26.3 (CH), 23.3 (CH₂), 22.0 (CH₃), 20.9 (CH₃), 17.1 (CH₃), 16.2 (CH₃); LRMS m/z 424.25 (M - H⁺), calcd $C_{25}H_{31}NO_5$ 425.2202; HRMS m/z 448.2107 (M + Na), calcd for $C_{25}H_{31}NO_5Na$ 448.2100; Anal. calcd for $C_{25}H_{31}NO_5$ (425.22): C, 70.57; H, 7.34; N, 3.29. Found: C, 70.45; H, 7.28; N, 3.35%.

4-Hydroxy-3-(4-nitro-benzyl)-benzoic acid ethyl ester (90ea): Prepared following the

OH NO₂ CO₂Et 90ea procedure **2d** and purified by column chromatography using EtOAc/hexane and isolated as a yellow solid. Mp 126 °C; IR (neat): v_{max} 3355 (O-*H*), 1688 (O-C=O), 1607, 1515, 1343, 1297 cm⁻¹; ¹H NMR (CDCl₃) δ 8.10 (2H, d, J = 8.8 Hz), 7.86 (1H, s),

7.84 (1H, dd, J = 10.0, 1.6 Hz), 7.37 (2H, d, J = 8.4 Hz), 6.82 (1H, d, J = 8.0 Hz), 6.23 (1H, br s, O-H), 4.35 (2H, q, J = 7.2 Hz, OC H_2 CH $_3$), 4.08 (2H, s, C H_2 Ar), 1.38 (3H, t, J = 7.2 Hz, OCH $_2$ CH $_3$); ¹³C NMR (CDCl $_3$, DEPT-135) δ 166.7 (C, O-C=O), 158.0 (C), 148.0 (C), 146.5 (C), 132.8 (CH), 130.5 (CH), 129.5 (2 x CH), 125.9 (C), 123.7 (2 x CH), 123.1 (C), 115.4 (CH), 61.1 (CH $_2$, OCH $_2$ CH $_3$), 36.0 (CH $_2$, CH $_2$ Ar), 14.4 (CH $_3$, OCH $_2$ CH $_3$); LRMS m/z 300.15 (M - H $^+$), calcd C $_{16}$ H $_{15}$ NO $_{5}$ 301.0950; Anal. calcd for C $_{16}$ H $_{15}$ NO $_{5}$ (301.09): C, 63.78; H, 5.02; N, 4.65. Found: C, 63.85; H, 5.08; N, 4.71%.

5-Hydroxy-4-(4-nitro-benzyl)-biphenyl-2-carboxylic acid ethyl ester (90fa):

Ph NO₂

Prepared following the procedure **2d** and purified by column chromatography using EtOAc/hexane and isolated as a liquid; IR (neat): v_{max} 3391 (O-H), 2990, 2926, 2853, 1719 (O-C=O),

1682, 1607, 1520, 1344, 1287, 766, 700 cm⁻¹; ¹H NMR (CDCl₃) δ 8.13 (2H, d, J = 9.2 Hz), 7.72 (1H, s, Ar-H), 7.41 (2H, d, J = 9.2 Hz), 7.35-7.33 (3H, m), 7.23 (2H, m), 6.71 (1H, s, Ar-H), 5.87 (1H, s, O-H), 4.10 (2H, s, CH₂Ar), 4.04 (2H, q, J = 7.2 Hz, OCH₂CH₃), 0.95 (3H, t, J = 7.2 Hz, OCH₂CH₃); ¹³C NMR (CDCl₃, DEPT-135) δ 168.2 (C, O-C=O), 156.0 (C), 147.9 (C), 146.5 (C), 144.1 (C), 141.1 (C), 133.5 (CH), 129.5 (2 x CH), 128.2 (2 x CH), 127.9 (2 x CH), 127.3 (CH), 124.8 (C), 123.7 (2 x CH), 123.4 (C), 117.8 (CH), 60.8 (CH₂, OCH₂CH₃), 35.7 (CH₂, CH₂Ar), 13.6 (CH₃, OCH₂CH₃); LRMS m/z 378.05 (M + H⁺), calcd C₂₂H₁₉NO₅ 377.1263; Anal. calcd for C₂₂H₁₉NO₅ (377.12): C, 70.02; H, 5.07; N, 3.71. Found: C, 70.15; H, 5.11; N, 3.75%.

5-Hydroxy-6-(4-nitro-benzyl)-biphenyl-2-carboxylic acid ethyl ester (90ga):

OH
Ph NO₂
CO₂Et
90ga

Prepared following the procedure **2d** and purified by column chromatography using EtOAc/hexane and isolated as a white solid. Mp 144 °C; IR (neat): v_{max} 3407 (O-*H*), 2976, 1694 (O-C=O), 1588, 1516, 1020, 941, 858, 766 cm⁻¹; ¹H NMR (CDCl₃) δ 7.98

(2H, d, J = 8.0 Hz), 7.77 (1H, d, J = 8.4 Hz), 7.31-7.27 (3H, m), 7.03-6.98 (4H, m), 6.86 (1H, d, J = 8.4 Hz), 6.69 (1H, s, O-H), 3.95 (2H, q, J = 7.2 Hz, OC H_2 CH $_3$), 3.92 (2H, s, C H_2 Ar), 0.89 (3H, t, J = 7.2 Hz, OC H_2 CH $_3$); ¹³C NMR (CDCl $_3$, DEPT-135) δ 168.5 (C, O-C=O), 157.2 (C), 148.9 (C), 146.0 (C), 145.0 (C), 139.6 (C), 130.5 (CH), 129.0 (2 x CH), 128.6 (2 x CH), 127.8 (2 x CH), 127.2 (CH), 125.0 (C), 124.1 (C), 123.2 (2 x CH), 114.3 (CH), 60.8 (CH $_2$, OC H_2 CH $_3$), 32.6 (CH $_2$, CH $_2$ Ar), 13.5 (CH $_3$, OC H_2 CH $_3$); LRMS m/z 376.45 (M - H $^+$), calcd C $_{22}$ H $_{19}$ NO $_5$ 377.1263; Anal. calcd for C $_{22}$ H $_{19}$ NO $_5$ (377.12): C, 70.02; H, 5.07; N, 3.71. Found: C, 70.15; H, 5.12; N, 3.76%.

4-Hydroxy-3-(4-nitro-benzyl)-2-propyl-benzoic acid ethyl ester (90ha): Prepared

OH
Pr
NO₂
CO₂Et
90ha

following the procedure **2d** and purified by column chromatography using EtOAc/hexane and isolated as a solid. Mp 134 °C; IR (neat): v_{max} 3360 (O-*H*), 2963, 2928, 1684 (O-C=O), 1578, 1516, 1470, 1346, 1262, 1009, 733 cm⁻¹; ¹H NMR (CDCl₃) δ

8.07 (2H, d, J = 8.4 Hz), 7.74 (1H, d, J = 8.4 Hz), 7.26 (2H, d, J = 8.0 Hz), 6.76 (1H, s, O-H), 6.75 (1H, d, J = 8.4 Hz), 4.33 (2H, q, J = 7.2 Hz, OC H_2 CH₃), 4.20 (2H, s, C H_2 Ar), 2.86 (2H, t, J = 7.6 Hz, ArC H_2 CH₂CH₃), 1.52-1.42 (2H, m, ArC H_2 CH₂CH₃), 1.37 (3H, t, J = 7.2 Hz, OCH₂CH₃), 0.94 (3H, t, J = 7.6 Hz, ArC H_2 CH₂CH₃); ¹³C NMR (CDCl₃, DEPT-135) δ 168.5 (C, O-C=O), 157.5 (C), 148.8 (C), 146.1 (C), 145.5 (C), 131.4 (CH), 128.8 (2 x CH), 124.6 (C), 123.5 (2 x CH), 123.1 (C), 112.7 (CH), 60.9 (CH₂, OCH₂CH₃), 32.3 (CH₂, CH₂Ar), 31.3 (CH₂, ArCH₂CH₂CH₃), 24.5 (CH₂, ArCH₂CH₂CH₃), 14.5 (CH₃, OCH₂CH₃), 14.2 (CH₃, ArCH₂CH₂CH₃); LRMS m/z 342.10 (M - H⁺), calcd C₁₉H₂₁NO₅ 343.1420; HRMS m/z 366.1317 (M + Na), calcd for C₁₉H₂₁NO₅Na 366.1318; Anal. calcd for C₁₉H₂₁NO₅ (343.14): C, 66.46; H, 6.16; N, 4.08. Found: C, 66.52; H, 6.21; N, 4.15%.

6-Ethyl-4-hydroxy-2-methyl-3-(4-nitro-benzyl)-benzoic acid ethyl ester (90ja):

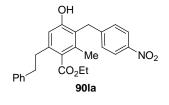
Prepared following the procedure 2d and purified by column chromatography using EtOAc/hexane and isolated as a solid. Mp NO_2 Ме 102 °C; IR (neat): ν_{max} 3401 (O-*H*), 2974, 2940, 1715 (O-C=O), ĊO₂Et 1696, 1601, 1520, 1344, 1152, 1051, 858 cm⁻¹; ¹H NMR $(CDCl_3)$ δ 8.06 (2H, d, J = 8.8 Hz), 7.26 (2H, d, J = 8.4 Hz), 6.60 (1H, s, Ar-H), 5.80 (1H, brs, O-H), 4.37 (2H, q, J = 7.2 Hz, OCH₂CH₃), 4.10 (2H, s, CH₂Ar), 2.54 (2H, q, J = 7.6 Hz, ArC H_2 CH₃), 2.15 (3H, s, Ar-C H_3), 1.37 (3H, t, J = 7.2 Hz, OCH₂C H_3), 1.18 (3H, t, J = 7.6 Hz, ArCH₂CH₃); ¹³C NMR (CDCl₃, DEPT-135) δ 170.8 (C, O-C=O), 154.7 (C), 148.2 (C), 146.2 (C), 141.1 (C), 135.4 (C), 128.9 (2 x CH), 127.5 (C), 123.5 (2 x CH), 122.0 (C), 113.0 (CH), 61.2 (CH₂, OCH₂CH₃), 31.6 (CH₂, CH₂Ar), 26.6 (CH₂, CH₂CH₃), 16.8 (CH₃, Ar-CH₃), 15.3 (CH₃, CH₂CH₃), 14.2 (CH₃, OCH_2CH_3); LRMS m/z 342.10 (M - H⁺), calcd $C_{19}H_{21}NO_5$ 343.1420; HRMS m/z 366.1318 (M + Na), calcd for $C_{19}H_{21}NO_5Na$ 366.1318; Anal. calcd for $C_{19}H_{21}NO_5$ (343.14): C, 66.46; H, 6.16; N, 4.08. Found: C, 66.55; H, 6.13; N, 4.12%.

4-Hydroxy-2-methyl-3-(4-nitro-benzyl)-6-propyl-benzoic acid ethyl ester (90ka):

Prepared following the procedure **2d** and purified by column chromatography using EtOAc/hexane and isolated as a solid. Mp 80 °C; IR (neat): v_{max} 3385 (O-*H*), 3056, 2963, 2870, 1717 (O-C=O), 1694, 1599, 1518, 1344, 1152, 1051, 858

cm⁻¹; ¹H NMR (CDCl₃) δ 8.04 (2H, d, J = 8.4 Hz), 7.25 (2H, d, J = 8.8 Hz), 6.53 (1H, s, Ar-H); 5.86 (1H, br s, O-H), 4.37 (2H, q, J = 7.2 Hz, OC H_2 CH₃), 4.10 (2H, s, C H_2 Ar), 2.48 (2H, t, J = 7.6 Hz, ArC H_2 CH₂CH₃), 2.15 (3H, s, Ar- CH_3), 1.63-1.53 (2H, m, ArC H_2 CH₂CH₃), 1.37 (3H, t, J = 7.2 Hz, OCH₂CH₃), 0.93 (3H, t, J = 7.2 Hz, ArCH₂CH₂CH₃); ¹³C NMR (CDCl₃, DEPT-135) δ 170.9 (C, O-C=O), 154.5 (C), 148.2 (C), 146.2 (C), 139.6 (C), 135.4 (C), 128.9 (2 x CH), 127.6 (C), 123.5 (2 x CH), 122.0 (C), 113.7 (CH), 61.2 (CH₂, OCH₂CH₃), 35.6 (CH₂, ArCH₂CH₂CH₃), 31.6 (CH₂, CH₂Ar), 24.3 (CH₂, ArCH₂CH₂CH₃), 16.8 (CH₃, Ar- CH_3), 14.2 (CH₃, OCH₂CH₃), 14.0 (CH₃, ArCH₂CH₂CH₃); LRMS m/z 356.30 (M - H⁺), calcd C₂₀H₂₃NO₅ 357.1576; Anal. calcd for C₂₀H₂₃NO₅ (357.15): C, 67.21; H, 6.49; N, 3.92. Found: C, 67.35; H, 6.41; N, 4.07%.

4-Hydroxy-2-methyl-3-(4-nitro-benzyl)-6-phenethyl-benzoic acid ethyl ester (90la):



Prepared following the procedure 2d and purified by column chromatography using EtOAc/hexane and isolated as a yellow liquid; IR (neat): ν_{max} 3410 (O-H), 2928, 1719 (O-C=O), 1694, 1599, 1518, 1454, 1344, 1150, 858 cm⁻¹; ¹H NMR

(CDCl₃) δ 8.06 (2H, d, J = 8.4 Hz), 7.30-7.10 (7H, m), 6.49 (1H, s, Ar-H), 5.72 (1H, br s, O-H), 4.38 (2H, q, J = 7.2 Hz, OC H_2 CH₃), 4.11 (2H, s, C H_2 Ar), 2.89-2.78 (4H, m), 2.17 (3H, s, Ar-C H_3), 1.36 (3H, t, J = 7.2 Hz, OCH₂C H_3); ¹³C NMR (CDCl₃, DEPT-135) δ 170.5 (C, O-C=O), 154.5 (C), 148.1 (C), 146.2 (C), 141.4 (C), 138.8 (C), 135.7 (C), 128.9 (2 x CH), 128.4 (2 x CH), 128.3 (2 x CH), 127.7 (C), 126.1 (CH), 123.6 (2 x CH), 122.4 (C), 113.8 (CH), 61.3 (CH₂, OCH₂CH₃), 37.5 (CH₂), 35.7 (CH₂), 31.6 (CH₂, CH₂Ar), 16.9 (CH₃, Ar-CH₃), 14.2 (CH₃, OCH₂CH₃); LRMS m/z 418.25 (M -

 H^+), calcd $C_{25}H_{25}NO_5$ 419.1733; HRMS m/z 442.1632 (M + Na), calcd for $C_{25}H_{25}NO_5Na$ 442.1631; Anal. calcd for $C_{25}H_{25}NO_5$ (419.17): C, 71.58; H, 6.01; N, 3.34. Found: C, 71.48; H, 6.11; N, 3.31%.

5-Hydroxy-3-methyl-4-(4-nitro-benzyl)-biphenyl-2-carboxylic acid ethyl ester (**90ma**): Prepared following the procedure **2d** and purified by column chromatography using EtOAc/hexane and isolated as a liquid; IR (neat): v_{max} 3403 (O-H), 2978, 2930, 1688 (O-C=O), 1597, 1518, 1344, 1263, 1171, 1053 cm⁻¹; ¹H NMR (CDCl₃) δ 8.02 (2H, d, J = 8.4 Hz), 7.28-7.24 (7H, m); 6.69 (1H, br s, O-H), 6.59 (1H, s, Ar-H), 4.12 (2H, s, CH_2Ar), 4.00 (2H, q, J = 7.2 Hz, OCH_2CH_3), 2.21 (3H, s, Ar- CH_3), 0.90 (3H, t, J = 7.2 Hz, OCH_2CH_3); ¹³C NMR (CDCl₃, DEPT-135) δ 170.9 (C, O-C=O), 154.8 (C), 148.0 (C), 146.1 (C), 140.4 (C), 140.3 (C), 136.2 (C), 128.9 (2 x CH), 128.1 (2 x CH), 128.0 (2 x CH), 127.4 (CH), 126.6 (C), 123.6 (C), 123.5 (2 x CH), 114.2 (CH), 61.3 (CH₂, OCH_2CH_3), 31.6 (CH₂, CH_2Ar), 16.7 (CH₃, $Ar-CH_3$), 13.4 (CH₃, OCH_2CH_3); LRMS m/z 390.15 (M - H⁺), calcd $C_{23}H_{21}NO_5$ 391.1420; Anal. calcd for $C_{23}H_{21}NO_5$ (391.14): C, 70.58; H, 5.41; N, 3.58. Found: C, 70.45; H, 5.45; N, 3.65%.

$5- Hydroxy-4'-methoxy-3-methyl-4-(4-nitro-benzyl)-biphenyl-2-carboxylic \\ acid$

ethyl ester (90na): Prepared following the procedure 2d and purified by column chromatography using EtOAc/hexane and isolated as a yellow liquid; IR (neat): v_{max} 3407 (O-H), 2984, 2934, 1721 (O-C=O), 1692,

1599, 1516, 1344, 1250, 1028 cm⁻¹; ¹H NMR (CDCl₃) δ 8.08 (2H, d, J = 8.4 Hz), 7.30 (2H, d, J = 8.4 Hz), 7.23 (2H, d, J = 8.4 Hz), 6.87 (2H, d, J = 8.4 Hz), 6.62 (1H, s, Ar-H), 6.17 (1H, br s, O-H), 4.16 (2H, s, CH₂Ar), 4.05 (2H, q, J = 7.2 Hz, OCH₂CH₃), 3.82 (3H, s, O-CH₃), 2.22 (3H, s, Ar-CH₃), 0.99 (3H, t, J = 7.2 Hz, OCH₂CH₃); ¹³C NMR (CDCl₃, DEPT-135) δ 170.6 (C, O-C=O), 159.1 (C), 154.5 (C), 148.0 (C), 146.2 (C), 139.9 (C), 136.1 (C), 132.9 (C), 129.2 (2 x CH), 129.0 (2 x CH), 127.2 (C), 123.6 (2 x CH), 123.2 (C), 114.2 (CH), 113.6 (2 x CH), 61.1 (CH₂, OCH₂CH₃), 55.3 (CH₃,

OCH₃), 31.7 (CH₂, CH₂Ar), 16.8 (CH₃, Ar-CH₃), 13.7 (CH₃, OCH₂CH₃); LRMS m/z 420.25 (M - H⁺), calcd $C_{24}H_{23}NO_6$ 421.1525; Anal. calcd for $C_{24}H_{23}NO_6$ (421.15): C, 68.40; H, 5.50; N, 3.32. Found: C, 68.45; H, 5.46; N, 3.37%.

6-Benzo[1,3]dioxol-5-yl-4-hydroxy-2-methyl-3-(4-nitro-benzyl)-benzoic acid ethyl ester (90oa): Prepared following the procedure **2d** and purified by column chromatography using EtOAc/hexane and isolated as a yellow liquid; IR (neat): v_{max} 3393 (O-*H*), 2982, 1717 (O-C=O), 1599, 1520, 1480, 1344, 1240, 1040, 858 cm⁻¹; ¹H NMR (CDCl₃) δ 8.05 (2H, d, J = 8.4 Hz), 7.28 (2H, d, J = 8.4 Hz), 6.76 (2H, d, J = 6.4 Hz), 6.75 (1H, s, Ar-*H*), 6.60 (1H, s, Ar-*H*), 6.60 (1H, br s, O-*H*), 6.00 (2H, s, OC*H*₂O), 4.14 (2H, s, C*H*₂Ar), 4.10 (2H, q, J = 7.2 Hz, OC*H*₂CH₃), 2.21 (3H, s, Ar-*CH*₃), 1.05 (3H, t, J = 7.2 Hz, OCH₂CH₃); ¹³C NMR (CDCl₃, DEPT-135) δ 170.7 (C, O-C=O), 154.8 (C), 148.1 (C), 147.5 (C), 147.1 (C), 146.3 (C), 139.8 (C), 136.1 (C), 134.4 (C), 129.0 (2 x CH), 127.0 (C), 123.6 (2 x CH, C), 121.7 (CH), 114.2 (CH), 108.8 (CH), 108.1 (CH), 101.1 (CH₂, OCH₂O), 61.3 (CH₂, OCH₂CH₃), 31.7 (CH₂, CH₂Ar), 16.8 (CH₃, Ar-CH₃), 13.8 (CH₃, OCH₂CH₃); LRMS m/z 434.30 (M - H⁺), calcd C₂₄H₂₁NO₇ 435.1318; HRMS m/z 458.1218 (M + Na), calcd for C₂₄H₂₁NO₇Na 458.1216; Anal. calcd for C₂₄H₂₁NO₇ (435.13): C, 66.20; H, 4.86; N, 3.22. Found: C, 66.15; H, 4.80; N, 3.28%.

4'-Chloro-5-hydroxy-3-methyl-4-(4-nitro-benzyl)-biphenyl-2-carboxylic acid ethyl

ester (90pa): Prepared following the procedure 2d and purified by column chromatography using EtOAc/hexane and isolated as a yellow liquid; IR (neat): v_{max} 3391 (O-H), 2978, 2934, 1713 (O-C=O), 1690, 1599, 1520, 1344,

1167, 829 cm⁻¹; ¹H NMR (CDCl₃) δ 8.06 (2H, d, J = 8.0 Hz), 7.30-7.28 (4H, m), 7.20 (2H, d, J = 7.6 Hz); 6.69 (1H, br s, O-H), 6.61 (1H, s, Ar-H), 4.15 (2H, s, CH₂Ar), 4.04 (2H, q, J = 7.2 Hz, OCH₂CH₃), 2.23 (3H, s, Ar-CH₃), 0.99 (3H, t, J = 7.2 Hz, OCH₂CH₃); ¹³C NMR (CDCl₃, DEPT-135) δ 170.6 (C, O-C=O), 154.9 (C), 147.9 (C),

146.2 (C), 138.9 (2 x C), 136.4 (C), 133.5 (C), 129.4 (2 x CH), 128.9 (2 x CH), 128.3 (2 x CH), 126.6 (C), 124.1 (C), 123.5 (2 x CH), 114.0 (CH), 61.4 (CH₂, OCH₂CH₃), 31.6 (CH₂, CH₂Ar), 16.8 (CH₃, Ar-CH₃), 13.6 (CH₃, OCH₂CH₃); LRMS m/z 426.00 (M + H⁺), calcd C₂₃H₂₀ClNO₅ 425.1030; Anal. calcd for C₂₃H₂₀ClNO₅ (425.10): C, 64.87; H, 4.73; N, 3.29. Found: C, 64.75; H, 4.76; N, 3.32%.

5-Hydroxy-3-methyl-4'-nitro-4-(4-nitro-benzyl)-biphenyl-2-carboxylic acid ethyl

OH Me NO₂ 90qa 910, 853, 735 cm⁻¹: ¹H NMF **ester (90qa):** Prepared following the procedure **2d** and purified by column chromatography using EtOAc/hexane and isolated as a solid. Mp 168 °C; IR (neat): v_{max} 3437 (O-H), 2976, 1715 (O-C=O), 1686, 1597, 1518, 1344,

910, 853, 735 cm⁻¹; ¹H NMR (CDCl₃) δ 8.19 (2H, d, J = 8.0 Hz), 8.07 (2H, d, J = 8.0 Hz), 7.45 (2H, d, J = 8.0 Hz), 7.31 (2H, d, J = 8.0 Hz), 6.71 (1H, br s, O-H), 6.68 (1H, s, Ar-H), 4.19 (2H, s, CH₂Ar), 4.05 (2H, q, J = 6.8 Hz, OCH₂CH₃), 2.26 (3H, s, Ar-CH₃), 0.99 (3H, t, J = 6.8 Hz, OCH₂CH₃); ¹³C NMR (CDCl₃, DEPT-135) δ 170.0 (C, O-C=O), 155.0 (C), 147.6 (C), 147.3 (C), 147.0 (C), 146.2 (C), 137.9 (C), 137.0 (C), 129.0 (2 x CH), 128.9 (2 x CH), 126.5 (C), 125.1 (C), 123.6 (2 x CH), 123.4 (2 x CH), 113.9 (CH), 61.5 (CH₂, OCH₂CH₃), 31.7 (CH₂, CH₂Ar), 16.8 (CH₃, Ar-CH₃), 13.6 (CH₃, OCH₂CH₃); LRMS m/z 435.10 (M – H⁺), calcd C₂₃H₂₀N₂O₇ 436.1271; Anal. calcd for C₂₃H₂₀N₂O₇ (436.12): C, 63.30; H, 4.62; N, 6.42. Found: C, 63.38; H, 4.55; N, 6.45%.

6-Furan-2-yl-4-hydroxy-2-methyl-3-(4-nitro-benzyl)-benzoic acid ethyl ester

OH
Me
NO₂

90ra

(90ra): Prepared following the procedure 2d and purified by column chromatography using EtOAc/hexane and isolated as a yellow liquid; IR (neat): v_{max} 3382 (O-H), 2984, 2938, 2255, 1694 (O-C=O), 1599, 1516, 1344, 1154, 1111, 1016,

860 cm⁻¹; ¹H NMR (CDCl₃) δ 7.99 (2H, d, J = 8.4 Hz), 7.36 (1H, s), 7.19 (2H, d, J = 8.0 Hz), 6.99 (1H, br s, O-H), 6.91 (1H, s), 6.39 (2H, m), 4.34 (2H, q, J = 7.2 Hz,

OC H_2 CH₃), 4.07 (2H, s, C H_2 Ar), 2.16 (3H, s, Ar-C H_3), 1.27 (3H, t, J = 7.2 Hz, OC H_2 C H_3); ¹³C NMR (CDCl₃, DEPT-135) δ 171.4 (C, O-C=O), 154.9 (C), 151.6 (C), 147.9 (C), 146.1 (C), 142.3 (CH), 135.9 (C), 128.8 (2 x CH), 127.8 (C), 124.3 (C), 124.1 (C), 123.4 (2 x CH), 111.5 (CH), 110.9 (CH), 107.4 (CH), 61.8 (CH₂, OC H_2 CH₃), 31.6 (CH₂, CH₂Ar), 16.4 (CH₃, Ar-CH₃), 13.9 (CH₃, OC H_2 CH₃); LRMS m/z 382.00 (M + H⁺), calcd C₂₁H₁₉NO₆ 381.1212; HRMS m/z 404.1110 (M + Na), calcd for C₂₁H₁₉NO₆Na 404.1110; Anal. calcd for C₂₁H₁₉NO₆ (381.12): C, 66.13; H, 5.02; N, 3.67. Found: C, 66.25; H, 5.09; N, 3.71%.

4-Hydroxy-2-methyl-3-[3-(2-nitro-phenyl)-propenyl]-benzoic acid ethyl ester

OH NO₂
Me CO₂Et

NO₂ (**90aq**): Prepared following the procedure **2d** and purified by column chromatography using EtOAc/hexane and isolated as a white solid. Mp 112 °C; IR (neat): v_{max} 3359 (O-*H*), 2981, 2925, 1677 (O-C=O), 1579, 1519, 1353, 1284, 1246, 1160, 1051 cm⁻¹; ¹H

NMR (CDCl₃) δ 7.98 (1H, d, J = 8.0 Hz), 7.73 (1H, d, J = 8.4 Hz), 7.60 (1H, t, J = 7.6 Hz), 7.44 (1H, d, J = 8.0 Hz), 7.43 (1H, t, J = 7.6 Hz), 6.78 (1H, d, J = 8.8 Hz), 6.28 (1H, d, J = 16.4 Hz), 6.08 (1H, dt, J = 16.4, 6.0 Hz), 5.92 (1H, br s, O-H), 4.30 (2H, q, J = 7.2 Hz, OC H_2 CH₃), 3.90 (2H, d, J = 6.0 Hz), 2.45 (3H, s, Ar-C H_3), 1.36 (3H, t, J = 7.2 Hz, OCH₂CH₃); ¹³C NMR (CDCl₃, DEPT-135) δ 167.7 (C, O-C=O), 155.7 (C), 149.2 (C), 140.0 (C), 135.2 (CH), 134.0 (C), 133.5 (CH), 132.2 (CH), 131.3 (CH), 127.9 (CH), 126.0 (CH), 125.0 (CH), 124.7 (C), 122.7 (C), 112.5 (CH), 60.5 (CH₂, OCH₂CH₃), 36.7 (CH₂, CH₂Ar), 18.1 (CH₃, Ar-CH₃), 14.3 (CH₃, OCH₂CH₃); LRMS m/z 342.00 (M + H⁺), calcd C₁₉H₁₉NO₅ 341.1263; HRMS m/z 364.1164 (M + Na), calcd for C₁₉H₁₉NO₅Na 364.1161; Anal. calcd for C₁₉H₁₉NO₅ (341.12): C, 66.85; H, 5.61; N, 4.10. Found: C, 66.91; H, 5.57; N, 4.18%.

4-Hydroxy-2-methyl-3-[3-(2-nitro-phenyl)-propenyl]-benzoic acid methyl ester

OH NO₂
Me CO₂Me

90sa

NO₂ (90sq): Prepared following the procedure 2d and purified by column chromatography using EtOAc/hexane and isolated as a

161

yellow solid. Mp 136 °C; IR (neat): v_{max} 3352 (O-*H*), 2955, 1736, 1672 (O-C=O), 1580, 1524, 1435, 1346, 1252, 1196, 1034, 984 cm⁻¹; ¹H NMR (CDCl₃) δ 7.99 (1H, dd, J = 9.2, 1.6 Hz), 7.74 (1H, d, J = 8.8 Hz), 7.61 (1H, dt, J = 8.4, 1.2 Hz), 7.45 (1H, d, J = 7.2 Hz), 7.43 (1H, dt, J = 8.4, 1.2 Hz), 6.78 (1H, d, J = 8.8 Hz); 6.27 (1H, d, J = 16.4 Hz), 6.09 (1H, td, J = 16.4, 6.0 Hz), 5.83 (1H, s, O-*H*), 3.91 (2H, dd, J = 6.0, 1.2 Hz), 3.84 (3H, s, OC*H*₃), 2.45 (3H, s, Ar-C*H*₃); ¹³C NMR (CDCl₃, DEPT-135) δ 168.0 (C, O-C=O), 155.7 (C), 149.3 (C), 140.2 (C), 135.4 (CH), 133.9 (C), 133.5 (CH), 132.2 (CH), 131.4 (CH), 128.0 (CH), 126.0 (CH), 125.1 (CH), 124.7 (C), 122.3 (C), 112.5 (CH), 51.7 (CH₃, OCH₃), 36.8 (CH₂, CH₂Ar), 18.1 (CH₃, Ar-CH₃); LRMS m/z 328.20 (M + H⁺), calcd C₁₈H₁₇NO₅ 327.1107; Anal. calcd for C₁₈H₁₇NO₅ (327.11): C, 66.05; H, 5.23; N, 4.28. Found: C, 66.12; H, 5.28; N, 4.21%.

3-Ethoxycarbonylmethyl-4-hydroxy-benzoic acid ethyl ester (90au): Prepared

OH following the procedure **2d** and purified by column chromatography using EtOAc/hexane and isolated as a liquid; IR (neat): ν_{max} 3370 (O- $_{O_2Et}$ H), 2982, 1713 (O-C=O), 1588, 1447, 1370, 1258, 1181, 1055, 783 cm⁻¹; 1 H NMR (CDCl₃) δ 7.71 (1H, d, J = 8.4 Hz), 7.37 (1H, br s, O- $_{I}$ H), 6.76 (1H, d, J = 8.8 Hz), 4.32 (2H, q, J = 7.2 Hz, OCH₂CH₃), 4.19 (2H, q, J = 7.2 Hz, OCH₂CH₃), 3.77 (2H, s, CH₂Ar), 2.56 (3H, s, Ar-CH₃), 1.37 (3H, t, J = 7.2 Hz, OCH₂CH₃), 1.28 (3H, t, J = 7.2 Hz, OCH₂CH₃); 13 C NMR (CDCl₃, DEPT-135) δ 173.2 (C, O-C=O), 168.0 (C, O-C=O), 157.9 (C), 140.4 (C), 131.3 (CH), 123.9 (C), 121.2 (C), 114.1 (CH), 61.8 (CH₂, OCH₂CH₃), 60.6 (CH₂, OCH₂CH₃), 32.9 (CH₂, CH₂Ar), 17.1 (CH₃, Ar-CH₃), 14.3 (CH₃, OCH₂CH₃), 14.1 (CH₃, OCH₂CH₃); LRMS m/z 265.20 (M - H⁺), calcd C₁₄H₁₈O₅ 266.1154; Anal. calcd for C₁₄H₁₈O₅ (266.11): C, 63.15; H, 6.81. Found: C, 63.22; H, 6.85%.

2e: (S)-1-(2-Pyrrolidinylmethyl)pyrrolidine-catalyzed cascade Claisen-Schmidt and isomerization reactions: In an ordinary glass vial equipped with a magnetic stirring bar, to 1.0 mmol of the Hagemann's ester **88a** was added 1.0 mL of DMSO

solvent, and then the catalyst (S)-1-(2-pyrrolidinylmethyl)pyrrolidine **2i** (0.1 mmol, 16.3 μ L) was added and then 0.5 mmol of aldehyde **28v-28b**' was added in one-portion and the reaction mixture was stirred at 25 °C for the time indicated in Table 8. The crude reaction mixture was worked up with aqueous NH₄Cl solution and the aqueous layer was extracted with dichloromethane (2 x 20 mL). The combined organic layers were dried (Na₂SO₄), filtered and concentrated. The pure cascade products **92** were obtained by column chromatography (silica gel, mixture of hexane/ethyl acetate).

(3R)-3-(3,7-Dimethyl-octa-1,6-dienyl)-2-methyl-4-oxo-cyclohex-2-enecarboxylic

Ме acid ethyl ester (92av): Prepared following the procedure 2e and purified by column chromatography using EtOAc/hexane Me and isolated as a colorless oily liquid. $[\alpha]^{25}_{D} = -43.6$ (c 0.54, ÇO₂Et (-)-92av **CHCl₃);** IR (neat): v_{max} 2960, 2912, 1730 (C=O), 1673 (O-E/Z = >99:1C=O), 1157 cm⁻¹; ¹H NMR (CDCl₃, 1:1 mixture of diastereomers, at room temperature) δ 6.02 (2H, d, J = 16.0 Hz), 5.76 (2H, dd, J = 16.0, 8.0 Hz), 5.08 (2H, t, J= 7.2 Hz), 4.19 (4H, q, J = 7.2 Hz, 2 x OC H_2 CH₃), 3.33 (2H, t, J = 4.8 Hz), 2.64-2.55 (2H, m), 2.38 (2H, td, J = 16.8, 5.6 Hz), 2.27-2.18 (4H, m) 2.05 $(6H, s, 2 \times s)$ CH_3), 1.98 (4H, m), 1.66 (6H, s, 2 x CH_3), 1.58 (6H, s, 2 x CH_3), 1.35 (4H, m), 1.27 (6H, t, J = 7.2 Hz, 2 x OCH₂CH₃), 1.02 (6H, d, J = 6.8 Hz, 2 x CHCH₃); ¹³C NMR (CDCl₃, 1:1 mixture of diastereomers, at room temperature) δ 197.3 (2 x C, C=O), 172.2 (2 x C, O-C=O), 149.6 (2 x C), 143.6 (CH), 143.5 (CH), 134.9 (2 x C), 131.2 (2 x C), 124.5 (2 x CH), 120.5 (2 x CH), 61.2 (2 x CH₂, OCH₂CH₃), 48.1 (CH), 48.0 (CH), 37.5 (2 x CH), 36.9 (2 x CH₂), 35.22 (CH₂), 35.16 (CH₂), 25.8 (2 x CH₂), 25.7 (2 x CH₃, olefinic-CH₃), 25.2 (2 x CH₂), 21.5 (2 x CH₃, olefinic-CH₃), 20.4 (2 x CH₃, olefinic-CH₃), 17.6 (2 x CH₃), 14.1 (2 x CH₃, OCH₂CH₃); ¹H NMR (CDCl₃, 1:1 mixture of diastereomers, at -40 °C) δ 6.07 (1H, d, J = 16.4 Hz), 6.06 (1H, d, J = 16.4Hz), 5.70 (1H, dd, J = 8.0, 4.0 Hz), 5.66 (1H, dd, J = 8.4, 4.0 Hz), 5.11 (2H, t, J = 6.4Hz), 4.22 (4H, q, J = 7.2 Hz, 2 x OC H_2 CH₃), 3.41 (2H, br s), 2.66-2.58 (2H, m), 2.45 (2H, td, J = 17.2, 4.8 Hz), 2.27-2.24 (4H, m) 2.10 (6H, s, 2 x olefinic-CH₃), 2.03-1.95

(4H, m), 1.70 $(6H, s, 2 \times CH_3)$, 1.61 $(6H, s, 2 \times CH_3)$, 1.38-1.30 (10H, m), 1.05 (6H, d, d)J = 6.4 Hz, 2 x CHC H_3); ¹³C NMR (CDCl₃, 1:1 mixture of diastereomers, at -40 °C) δ 198.16 (C, C=O), 198.13 (C, C=O), 172.4 (C, O-C=O), 172.3 (C, O-C=O), 150.5 (2 x C), 143.7 (CH), 143.6 (CH), 134.8 (2 x C), 131.6 (2 x C), 124.2 (2 x CH), 120.4 (CH), 120.3 (CH), 61.4 (2 x CH₂, OCH₂CH₃), 47.8 (CH), 47.6 (CH), 37.6 (CH), 37.5 (CH), 36.7 (2 x CH₂), 35.1 (CH₂), 34.9 (CH₂), 25.84 (2 x CH₂), 25.80 (2 x CH₃, olefinic- (CH_3) , 25.0 ((CH_2) , 24.9 ((CH_2) , 22.01 ((CH_3) olefinic- (CH_3)), 21.98 ((CH_3) olefinic- (CH_3)), 20.60 (CH₃, olefinic-CH₃), 20.55 (CH₃, olefinic-CH₃), 17.7 (2 x CH₃), 14.1 (2 x CH₃, OCH_2CH_3); LRMS m/z 319.30 (M + H⁺), calcd $C_{20}H_{30}O_3$ 318.2195; HRMS m/z 357.1843 (M + K), calcd for $C_{20}H_{30}O_3K$ 357.1832; Anal. calcd for $C_{20}H_{30}O_3$ (318.21): C, 75.43; H, 9.50. Found: C, 75.52; H, 9.48%.

(3S)-3-(3,7-Dimethyl-octa-1,6-dienyl)-2-methyl-4-oxo-cyclohex-2-enecarboxylic

acid ethyl ester (92aw): Prepared following the procedure 2e and purified by column chromatography using EtOAc/hexane and isolated as a colorless oily liquid. $[\alpha]^{25}_{D} = +45.6$ (c 0.43, **CHCl₃);** IR (neat): v_{max} 2960, 2912, 1730 (C=O), 1673 (O-C=O), 1157 cm⁻¹.

2-[3-(3-Ethoxycarbonyl-2-methyl-6-oxo-cyclohex-1-enyl)-1-phenyl-allyl]-malonic

MeO₂C
$$CO_2$$
Me

Me

CO₂Et

(+)-92ax

 $E|Z = >99:1$
 $dr = 1:1$

acid dimethyl ester (92ax): Prepared following the procedure 2e and purified by column chromatography using EtOAc/hexane and isolated as a yellow liquid. $[\alpha]^{25}_{D} = +19.8$ (c 0.90, CHCl₃); IR (neat): v_{max} 2953, 1752, 1721 (C=O), 1674 (O-C=O), 1601, 1437, 1161, 1024, 768, 702 cm⁻¹; ¹H NMR (CDCl₃, 1:1 mixture of **diastereomers**) δ 7.31-7.25 (8H, m, Ph-H), 7.20 (2H, t, J = 6.8 Hz), 6.18 (4H, m), 4.25-

4.10 (2H, m), 4.19 (4H, q, J = 7.2 Hz, 2 x OC H_2 CH₃), 3.92 (2H, d, J = 10.8 Hz), 3.75 $(6H, s, 2 \times OCH_3), 3.51 (6H, s, 2 \times OCH_3), 3.31 (2H, t, J = 5.2 Hz), 2.62-2.53 (2H, m),$ 2.35 (2H, td, J = 16.8, 4.8 Hz), 2.27-2.13 (4H, m), 1.98 (6H, s, 2 x olefinic-C H_3), 1.27 (6H, t, J = 6.8 Hz, 2 x OCH₂CH₃); ¹³C NMR (CDCl₃, 1:1 mixture of diastereomers) δ 196.58 (C, C=O), 196.55 (C, C=O), 171.9 (2 x C, 2 x O-C=O), 168.18 (C, O-C=O), 168.16 (C, O-C=O), 167.8 (2 x C, 2 x O-C=O), 151.0 (2 x C), 140.04 (C), 140.01 (C), 135.8 (CH), 135.7 (CH), 133.9 (2 x C), 128.6 (4 x CH), 127.9 (2 x CH), 127.8 (2 x CH), 127.1 (2 x CH), 124.4 (2 x CH), 61.3 (2 x CH₂, 2 x OCH₂CH₃), 57.6 (2 x CH), 52.6 (2 x CH₃, OCH₃), 52.4 (2 x CH₃, OCH₃), 49.92 (CH), 49.87 (CH), 48.1 (2 x CH), 35.2 (CH₂), 35.1 (CH₂), 25.12 (CH₂), 25.10 (CH₂), 21.29 (CH₃, olefinic-CH₃), 21.26 (CH₃, olefinic-CH₃), 14.1 (2 x CH₃, 2 x OCH₂CH₃); LRMS m/z 429.00 (M + H⁺), calcd C₂₄H₂₈O₇ 428.1835; HRMS m/z 451.1730 (M + Na), calcd for C₂₄H₂₈O₇Na 451.1733; Anal. calcd for C₂₄H₂₈O₇ (428.18): C, 67.28; H, 6.59. Found: C, 67.15; H, 6.65%.

2-Methyl-4-oxo-3-(3-phenyl-propenyl)-cyclohex-2-enecarboxylic acid ethyl ester

(92ay): Prepared following the procedure 2e and purified by column chromatography using EtOAc/hexane and isolated as a Ме colorless oily liquid. IR (neat): v_{max} 2923, 1728 (C=O), 1672 (O-ĊO₂Et C=O), 1158 cm⁻¹; ¹H NMR (CDCl₃) δ 7.30 (2H, t, J = 7.2 Hz), 7.23 (2H, d, J = 7.2 Hz), 7.20 (1H, t, J = 7.2 Hz), 6.26-6.05 (2H, m, olefinic-H), 4.20 (2H, q, d)J = 7.2 Hz, OC H_2 CH₃), 3.52 (2H, d, J = 6.4 Hz), 3.35 (1H, t, J = 4.4 Hz), 2.66-2.57 (1H, m), 2.40 (1H, td, J = 16.8, 5.2 Hz), 2.30-2.15 (2H, m), 2.04 (3H, s, olefinic-C H_3), 1.28 (3H, t, J = 7.2 Hz, OCH₂CH₃); ¹³C NMR (CDCl₃) δ 197.2 (C, C=O), 172.0 (C, O-C=O), 150.3 (C), 140.0 (C), 136.1 (CH), 134.5 (C), 128.5 (2 x CH), 128.4 (2 x CH), 126.0 (CH), 123.4 (CH), 61.2 (CH₂, OCH₂CH₃), 48.0 (CH), 40.0 (CH₂), 35.2 (CH₂), 25.2 (CH₂), 21.5 (CH₃, olefinic-CH₃), 14.1 (CH₃, OCH₂CH₃); LRMS m/z 299.25 (M + H^{+}), calcd $C_{19}H_{22}O_3$ 298.1569; HRMS m/z 337.1846 (M + K), calcd for $C_{19}H_{22}O_3K$ 337.1206; Anal. calcd for C₁₉H₂₂O₃ (298.15): C, 76.48; H, 7.43. Found: C, 76.51; H, 7.38%.

3-But-1-enyl-2-methyl-4-oxo-cyclohex-2-enecarboxylic acid ethyl ester (92az):

Prepared following the procedure **2e** and purified by column

165

92az *E/Z* = >99:1

CO₂Et

Ме

chromatography using EtOAc/hexane and isolated as a colorless oily liquid. IR (neat): v_{max} 2963, 2930, 1730 (C=O), 1674 (O-C=O), 1454, 1373, 1157, 1040, 970, 862, 779 cm⁻¹; ¹H NMR (CDCl₃) δ 6.07 (1H, d, J = 16.4 Hz), 5.97 (1H, td, J = 16.4, 6.4 Hz), 4.21 (2H, q, J = 7.2 Hz, OC H_2 CH₃), 3.35 (1H, t, J = 4.4 Hz), 2.66-2.58 (1H, m), 2.40 (1H, td, J = 16.8, 5.2 Hz), 2.29-2.16 (4H, m), 2.06 (3H, s, olefinic-C H_3), 1.29 (3H, t, J = 7.2 Hz, OCH₂CH₃), 1.06 (3H, t, J = 7.6 Hz); ¹³C NMR (CDCl₃) δ 197.5 (C, C=O), 172.2 (C, O-C=O), 149.7 (C), 139.6 (CH), 134.9 (C), 121.1 (CH), 61.2 (CH₂, OCH₂CH₃), 48.1 (CH), 35.3 (CH₂), 26.7 (CH₂), 25.2 (CH₂), 21.5 (CH₃, olefinic-CH₃), 14.2 (CH₃, OCH₂CH₃), 13.5 (CH₃, CH₂CH₃); LRMS m/z 237.20 (M + H⁺), calcd C₁₄H₂₀O₃ 236.1412; HRMS m/z 259.1308 (M + Na), calcd for C₁₄H₂₀O₃Na 259.1310; Anal. calcd for C₁₄H₂₀O₃ (236.14): C, 71.16; H, 8.53. Found: C, 71.22; H, 8.45%.

2-Methyl-4-oxo-3-propenyl-cyclohex-2-enecarboxylic acid ethyl ester (92aa'):

Prepared following the procedure **2e** and purified by column chromatography using EtOAc/hexane and isolated as a yellow liquid. IR (neat): v_{max} 2980, 2936, 2882, 1732 (C=O), 1674 (O-C=O), 1451, 1373, 92aa' [1350, 1157, 1040, 860 cm⁻¹; ¹H NMR (CDCl₃) δ 6.09 (1H, d, J = 16.0 Hz), 5.95 (1H, qd, J = 15.6, 6.4 Hz), 4.21 (2H, q, J = 7.2 Hz, OCH₂CH₃), 3.35 (1H, t, J = 4.8 Hz), 2.66-2.57 (1H, m), 2.40 (1H, td, J = 16.8, 5.2 Hz), 2.30-2.16 (2H, m), 2.05 (3H, s, olefinic-CH₃), 1.84 (3H, dd, J = 6.4, 1.2 Hz, HC=CHCH₃), 1.29 (3H, t, J = 7.2 Hz, OCH₂CH₃); ¹³C NMR (CDCl₃) δ 197.4 (C, C=O), 172.1 (C, O-C=O), 149.7 (C), 134.8 (C), 132.7 (CH), 123.3 (CH), 61.2 (CH₂, OCH₂CH₃), 48.0 (CH), 35.2 (CH₂), 25.2 (CH₂), 21.4 (CH₃, olefinic-CH₃), 19.1 (CH₃), 14.1 (CH₃, OCH₂CH₃); LRMS m/z 223.15 (M + H⁺), calcd C₁₃H₁₈O₃ 222.1256; HRMS m/z 245.1151 (M + Na), calcd for C₁₃H₁₈O₃Na 245.1154; Anal. calcd for C₁₃H₁₈O₃ (222.12): C, 70.24; H, 8.16. Found: C, 70.12; H, 8.22%.

3-Ethylidene-2-methyl-4-oxo-cyclohex-1-enecarboxylic acid ethyl ester (89ab'):

O Me Prepared following the procedure **2e** and purified by column

O Me CO₂Et 166

89ab'

E/Z = 1:>99

chromatography using EtOAc/hexane and isolated as a colorless oily liquid. IR (neat): v_{max} 2978, 1701 (C=O and O-C=O), 1241, 1184, 1069, 1021, 637 cm⁻¹; ¹H NMR (CDCl₃, 1:99 ratio of *E/Z* isomers) δ 6.44 (1H, q, J = 7.6 Hz, C=CHCH₃), 4.24 (2H, q, J = 7.2 Hz, OC H_2 CH₃), 2.72 (2H, t, J = 6.8 Hz), 2.56 (2H, t, J = 6.8 Hz), 2.19 (3H, d, J = 7.6 Hz, C=CHC H_3), 2.14 (3H, s, olefinic-C H_3), 1.32 (3H, t, J = 7.2 Hz, OCH₂C H_3); ¹³C NMR (CDCl₃, 1:99 ratio of *E/Z* isomers) δ 201.0 (C, C=O), 168.8 (C, O-C=O), 141.0 (C), 138.9 (CH, CHCH₃), 136.0 (C), 126.1 (C), 60.6 (CH₂, OCH₂CH₃), 39.3 (CH₂), 25.8 (CH₂), 16.7 (CH₃), 16.3 (CH₃), 14.3 (CH₃, OCH₂CH₃); LRMS m/z 209.15 (M + H⁺), calcd C₁₂H₁₆O₃ 208.1099; HRMS m/z 247.0715 (M + K), calcd for C₁₂H₁₆O₃K 247.0736; Anal. calcd for C₁₂H₁₆O₃ (208.10): C, 69.21; H, 7.74. Found: C, 69.32; H, 7.71%.

2f: Piperidine-catalyzed cascade Claisen-Schmidt and Michael reactions: In an ordinary glass vial equipped with a magnetic stirring bar, to 0.5 mmol of the Hagemann's ester 88 was added 1.5 mL of DMSO solvent, and then the catalyst piperidine 2b (0.05 mmol, 4.93 μL) was added and then 0.25 mmol of formaldehyde 28c' (37 wt% solution in water) was added in one-portion and the reaction mixture was stirred at 25 °C for the time indicated in Scheme 4. The crude reaction mixture was worked up with aqueous NH₄Cl solution and the aqueous layer was extracted with dichloromethane (2 x 20 mL). The combined organic layers were dried (Na₂SO₄), filtered and concentrated. The pure cascade products 93 were obtained by column chromatography (silica gel, mixture of hexane/ethyl acetate).

Diethyl 3,3'-methylenebis(2-methyl-4-oxocyclohex-2-enecarboxylate) (93ac'):

CO₂E

Me

CO₂Et

93ac'

dr. >99:1

Prepared following the procedure **2f** and purified by column chromatography using EtOAc/hexane and isolated as a colorless oily liquid. IR (neat): v_{max} 2928, 2859, 1728 (C=O), 1667 (O-C=O), 1622, 1512, 1451, 1372, 1157, 1042 cm⁻¹; ¹H NMR (CDCl₃, >99:1 ratio of diastereomers) δ 4.18 (2H, q, J = 7.2 Hz, OC H_2 CH₃), 4.17 (2H, q, J =

7.2 Hz, OCH_2CH_3), 3.44 (2H, d, J = 12.8 Hz), 3.28 (1H, t, J = 4.8 Hz), 3.27 (1H, t, J = 4.8 Hz), 3.28 (1H, t, J = 4.8 Hz), 3.27 (1H, t, J = 4.8 Hz), 3.28 (1H, t, J = 4.8 Hz), 3.27 (1H, t, J = 4.8 Hz), 3.28 (1H, t, J = 4.8 Hz), 3.27 (1H, t, J = 4.8 Hz), 3.27 (1H, t, J = 4.8 Hz), 3.28 (1H, t, J = 4.8 Hz), 3.27 (1H, t, J = 4.8 Hz), 3.28 (1H, t, J = 4.8 Hz), 3.27 (1H, t, J = 4.8 Hz), 3.28 (1H, t, J = 4.8 Hz), 3.27 (1H, t, J = 4.8 Hz), 3.28 (4.8 Hz), 2.59-2.50 (2H, m), 2.40-2.32 (2H, m) 2.26-2.13 (4H, m), 2.00 (3H, s, olefinic- CH_3), 1.95 (3H, s, olefinic- CH_3), 1.28 (3H, t, J = 7.2 Hz, OCH_2CH_3), 1.27 (3H, t, J = 7.2 Hz, OCH_2CH_3), 1.27 (3H, t, J = 7.2 Hz, OCH_2CH_3), 1.28 (3H, t, J = 7.2 Hz, OCH_2CH_3), 1.27 (3H, t, J = 7.2 Hz, OCH_2CH_3), 1.27 (3H, t, J = 7.2 Hz, OCH_2CH_3), 1.28 (3H, t, J = 7.2 Hz, OCH_2CH_3), 1.27 (3H, t, J = 7.2 Hz, OCH_2CH_3), 1.27 (3H, t, J = 7.2 Hz, OCH_2CH_3), 1.28 (3H, t, J = 7.2 Hz, OCH_2CH_3), 1.27 (3H, t, J = 7.2 Hz, OCH_2CH_3), 1.27 (3H, t, J = 7.2 Hz, OCH_2CH_3), 1.28 (3H, t, J = 7.2 Hz, OCH_2CH_3), 1.28 (3H, t, J = 7.2 Hz, OCH_2CH_3), 1.27 (3H, t, J = 7.2 Hz, OCH_2CH_3), 1.27 (3H, t, J = 7.2 Hz, OCH_2CH_3), 1.28 (3H, t, J = 7.2 Hz, OCH_2CH_3), 1.29 (3H, t, J = 7.2 Hz, OCH_2CH_3), 1.29 (3H, t, J = 7.2 Hz, OCH_2CH_3), 1.29 (3H, t, J = 7.2 Hz, OCH_2CH_3), 1.29 (3H, t, J = 7.2 Hz, OCH_2CH_3), 1.29 (3H, t, J = 7.2 Hz, OCH_2CH_3), 1.27 (3H, t, J = 7.2 Hz, OCH_2CH_3), 1.27 (3H, t, J = 7.2 Hz, OCH_2CH_3), 1.28 (3H, t, J = 7.2 Hz, OCH_2CH_3), 1.29 (3H, t, J = 7.2 Hz, OCH_2CH_3), 1.29 (3H, t, J = 7.2 Hz, OCH_2CH_3), 1.29 (3H, t, J = 7.2 Hz, OCH_2CH_3), 1.29 (3H, t, J = 7.2 Hz, OCH_2CH_3), 1.29 (3H, t, J = 7.2 Hz, OCH_2CH_3), 1.29 (3H, t, J = 7.2 Hz, OCH_2CH_3), 1.29 (3H, t, J = 7.2 Hz, OCH_2CH_3), 1.20 (3H, t, J = 7.2 Hz, OCH_2CH_3), 1.20 (3H, t, J = 7.2 Hz, OCH_2CH_3), 1.20 (3H, t, J = 7.2 Hz, OCH_2CH_3), 1.20 (3H, t, J = 7.2 Hz, OCH_2CH_3), 1.20 (3H, t, J = 7.2 Hz, OCH_2CH_3), 1.20 (3H, t, J = 7.2 Hz, OCH_2CH_3), 1.20 (3H, t, J = 7.2 Hz, OCH_2CH_3), 1.20 (3H, t, J = 7.2 Hz, J = 7.2 Hz, OCH_2CH_3), 1.20 (3H, t, J = 7.2 Hz, OCH_2CH_3), 1.20 (3H, t, J = 7.2 Hz, OCH_2CH_3), 1.20 (3H, t, J = 7.2 Hz, OCH_2CH_3), 1.20 (3H, t, J = 7.2 Hz, OCH_2CH_3), 1.20 (3H, t, J = 7.2 Hz, OCH_2CH_3), 1.20 (3H, t, J = 7.2 Hz, OCH_2CH_3), 1.20 (3H, t, J = 7.2 Hz, OCH_2CH_3), 1.20 (3H, t, J = 7.2 Hz, OCH_2CH_3), 1.20 (3H, t, J = 7.2 7.2 Hz, OCH₂CH₃); 13 C NMR (CDCl₃, >99:1 ratio of diastereomers) δ 197.0 (C, C=O), 196.9 (C, C=O), 172.2 (C, O-C=O), 172.1 (C, O-C=O), 150.8 (C), 150.7 (C), 135.8 (C), 135.7 (C), 61.1 (2 x CH₂, OCH₂CH₃), 48.2 (CH), 48.1 (CH), 34.6 (CH₂), 25.5 (CH₂), 25.3 (CH₂), 22.3 (CH₂), 21.8 (CH₂), 21.0 (2 x CH₃, olefinic-CH₃), 14.1 (2 x CH₃, OCH_2CH_3); LRMS m/z 377.25 (M + H⁺), calcd $C_{21}H_{28}O_6$ 376.1886; Anal. calcd for C₂₁H₂₈O₆ (376.18): C, 67.00; H, 7.50. Found: C, 67.49; H, 7.54%.

Di-tert-butyl-3,3'-methylenebis(2-methyl-4-oxocyclohex-2-enecarboxylate) (93bc'):

Ме ĊO₂^tBu dr: >99:1

CO₂^tBu Prepared following the procedure **2f** and purified by column chromatography using EtOAc/hexane and isolated as a colorless oily liquid. IR (neat): v_{max} 2928, 2857, 1724 (C=O), 1671 (O-C=O), 1454, 1370, 1256, 1152, 847 cm⁻¹; ¹H NMR (CDCl₃, >99:1 ratio of diastereomers) δ 3.43 (2H, d, J = 15.6 Hz), 3.20-3.15 (2H, m), 2.61-2.50 (2H, m), 2.36-2.32 (2H, m) 2.23-2.08 (4H, m), 2.01 (3H, s, olefinic- CH_3), 1.95 (3H, s, olefinic- CH_3), 1.46 (9H, s, $OC(CH_3)_3$), 1.45 (9H, s, $OC(CH_3)_3$); ¹³C NMR (CDCl₃, >99:1 ratio of diastereomers) δ 197.3 (C, C=O), 197.1 (C, C=O), 171.4 (C, O-C=O), 171.3 (C, O-C=O), 151.40 (C), 151.35 (C), 135.6 (C), 135.4 (C), 81.6 (2 x C, 2 x OC(CH₃)₃), 49.2 (2 x CH), 34.7 (CH₂), 28.0 (6 x CH₃, 2 x OC(CH₃)₃), 25.6 (CH₂), 25.4 (CH₂), 22.4 (CH₂), 21.7 (CH₂), 21.1 (CH₃, olefinic- CH_3), 21.0 (CH₃, olefinic- CH_3); LCMS m/z 431.45 (M - H⁺), calcd $C_{25}H_{36}O_6$ 432.2512; Anal. calcd for C₂₅H₃₆O₆ (432.25): C, 69.42; H, 8.39. Found: C, 69.35; H,

Dimethyl-3,3'-methylenebis(2-methyl-4-oxocyclohex-2-enecarboxylate) (93sc'):

CO₂Me Ме ĊO₂Me

8.41%.

Prepared following the procedure 2f and purified by column chromatography using EtOAc/hexane and isolated as a colorless oily

168

dr. >99:1

liquid. IR (neat): v_{max} 3052, 2957, 1732 (C=O), 1667 (O-C=O), 1622, 1435, 1346, 1271, 1161, 1078, 1042 cm⁻¹; ¹H NMR (CDCl₃, >99:1 ratio of diastereomers) δ 3.73 (6H, 2 x OC*H*₃), 3.45 (2H, br s), 3.31 (2H, t, J = 4.4 Hz), 2.57-2.48 (2H, m), 2.40-2.33 (2H, m), 2.26-2.13 (4H, m), 1.99 (3H, s, olefinic-C*H*₃), 1.95 (3H, s, olefinic-C*H*₃); ¹³C NMR (CDCl₃, >99:1 ratio of diastereomers) δ 197.0 (C, C=O), 196.9 (C, C=O), 172.7 (C, O-C=O), 172.6 (C, O-C=O), 150.8 (C), 150.6 (C), 135.9 (C), 135.8 (C), 52.24 (CH₃, OCH₃), 52.21 (CH₃, OCH₃), 48.0 (CH), 47.9 (CH), 34.6 (CH₂), 25.5 (CH₂), 25.3 (CH₂), 22.2 (CH₂), 22.0 (CH₂), 21.0 (CH₃, olefinic-CH₃), 20.98 (CH₃, olefinic-CH₃); GCMS m/z 348.30 (M⁺), calcd C₁₉H₂₄O₆ 348.1573; Anal. calcd for C₁₉H₂₄O₆ (348.15): C, 65.50; H, 6.94. Found: C, 65.63; H, 6.94%.

2g: **Sequential** combination of piperidine-catalyzed cascade **Knoevenagel/Michael/aldol** condensation/decarboxylation and (S)-1-(2pyrrolidinylmethyl)pyrrolidine-catalyzed Claisen-Schmidt/iso-aromatization reactions in one-pot: To a stirred solution of ethyl acetoacetate (2.0 mmol) and aldehyde (1.0 mmol) in EtOH (2 mL) was added a catalytic amount of piperidine 2b (0.35 mmol, 35 mol%) and the reaction mixture was stirred at 80 °C for 5-6 h. Solvent ethanol and piperidine was evaporated by vacuum pump then solvent DMSO (1 mL) was added and catalyst (S)-1-(2-pyrrolidinylmethyl)pyrrolidine 2i (0.1 mmol, 16.3 µL), aldehyde 28 (0.5 mmol) were added and the reaction mixture was stirred at 25 °C for 13-24 h. The crude reaction mixture was worked up with aqueous NH₄Cl solution and the aqueous layer was extracted with dichloromethane (2 x 20 mL). The combined organic layers were dried (Na₂SO₄), filtered and concentrated. Pure one-pot products 90 were obtained by column chromatography (silica gel, mixture of hexane/ethyl acetate).

Base-induced iso-aromatization of 3-arylidene-Hagemann's esters:

2h: Synthesis of 90aa: In an ordinary glass vial equipped with a magnetic stirring bar, to 0.3 mmol of the 2-methyl-3-(4-nitro-benzylidene)-4-oxo-cyclohex-1-enecarboxylic acid ethyl ester **89aa** was added 0.6 mL of DMSO solvent, and then the catalyst (S)-1-(2-pyrrolidinylmethyl)pyrrolidine **2i** (0.06 mmol, 9.7 μ L) was added and

the reaction mixture was stirred at 25 °C for 0.5 hour as indicated in Scheme 5. The crude reaction mixture was worked up with aqueous NH₄Cl solution and the aqueous layer was extracted with dichloromethane (2 x 20 mL). The combined organic layers were dried (Na₂SO₄), filtered and concentrated. Pure cascade product **90aa** was obtained by column chromatography (silica gel, mixture of hexane/ethyl acetate).

2i: Synthesis of 90ab: In an ordinary glass vial equipped with a magnetic stirring bar, to 0.18 mmol of the 3-benzylidene-2-methyl-4-oxo-cyclohex-1-enecarboxylic acid ethyl ester 89ab was added 0.36 mL of DMSO solvent, and then the catalyst piperidine 2b (0.036 mmol, 3.55 μL) was added and the reaction mixture was stirred at 70 °C for 12 hours as indicated in Scheme 5. The crude reaction mixture was worked up with aqueous NH₄Cl solution and the aqueous layer was extracted with dichloromethane (2 x 20 mL). The combined organic layers were dried (Na₂SO₄), filtered and concentrated. The pure cascade product 90ab was obtained by column chromatography (silica gel, mixture of hexane/ethyl acetate).

2j: Synthesis of 90ab and 90mb: In an ordinary glass vial equipped with a magnetic stirring bar, to 1.0 mmol of the Hagemann's ester **88a** or **88m** was added 1.0 mL of DMSO solvent, and then the catalyst piperidine **2b** (0.1 mmol, 9.87 μL) was added and then 0.5 mmol of benzaldehyde **28b** was added in one-portion and the reaction mixture was stirred at 70 °C for 18 hours as indicated in Scheme 5. The crude reaction mixture was worked up with aqueous NH₄Cl solution and the aqueous layer was extracted with dichloromethane (2 x 20 mL). The combined organic layers were dried (Na₂SO₄), filtered and concentrated. The pure cascade products **90ab** and **90mb** were obtained by column chromatography (silica gel, mixture of hexane/ethyl acetate).

3-Benzyl-4-hydroxy-2-methyl-benzoic acid ethyl ester (90ab): Prepared following

the procedure **2j** and purified by column chromatography using EtOAc/hexane and isolated as a white solid. Mp 102 °C; IR (neat): v_{max} 3362 (O-*H*), 2920, 2851, 1680 (O-C=O), 1638, 1580, 1454, 1267, 1177, 1051, 696 cm⁻¹; ¹H NMR (CDCl₃) δ 7.70 (1H, d, J = 8.4 Hz), 7.23 (2H,

t, J = 6.8 Hz), 7.16 (1H, t, J = 6.8 Hz), 7.11 (2H, d, J = 7.2 Hz), 6.72 (1H, d, J = 8.4 Hz); 6.00 (1H, s, O-H), 4.31 (2H, q, J = 7.2 Hz, OC H_2 CH $_3$), 4.11 (2H, s, C H_2 Ar), 2.48 (3H, s, Ar-C H_3), 1.36 (3H, t, J = 7.2 Hz, OCH $_2$ CH $_3$); ¹³C NMR (CDCl $_3$, DEPT-135) δ 168.6 (C, O-C=O), 157.1 (C), 141.1 (C), 139.5 (C), 130.4 (CH), 128.4 (2 x CH), 128.0 (2 x CH), 126.4 (C), 126.0 (CH), 123.6 (C), 112.6 (CH), 60.7 (CH $_2$, OCH $_2$ CH $_3$), 31.6 (CH $_2$, CH $_2$ Ar), 17.0 (CH $_3$, Ar-CH $_3$), 14.3 (CH $_3$, OCH $_2$ CH $_3$); LRMS m/z 271.00 (M + H $_3$), calcd C $_{17}$ H $_{18}$ O $_3$ 270.1256; Anal. calcd for C $_{17}$ H $_{18}$ O $_3$ (270.12): C, 75.53; H, 6.71. Found: C, 75.62; H, 6.68%.

4-Benzyl-5-hydroxy-3-methyl-biphenyl-2-carboxylic acid ethyl ester (90mb):

OH
Me
CO₂Et
90mb

Prepared following the procedure **2j** and purified by column chromatography using EtOAc/hexane and isolated as a yellow liquid; IR (neat): v_{max} 3403 (O-*H*), 2986, 2926, 1694 (O-C=O), 1589, 1452, 1263, 1173, 1053 cm⁻¹; ¹H NMR (CDCl₃) δ 7.35 (3H, m), 7.21-7.19 (3H, m); 6.68 (1H, br s, O-*H*), 4.11 (2H, s,

(5H, br s), 7.30-7.27 (3H, m), 7.21-7.19 (3H, m); 6.68 (1H, br s, O-*H*), 4.11 (2H, s, C*H*₂Ar), 4.01 (2H, q, J = 7.2 Hz, OC*H*₂CH₃), 2.30 (3H, s, Ar-C*H*₃), 0.92 (3H, t, J = 7.2 Hz, OCH₂C*H*₃); ¹³C NMR (CDCl₃, DEPT-135) δ 170.4 (C, O-C=O), 154.5 (C), 140.7 (C), 139.9 (C), 139.2 (C), 136.4 (C), 128.5 (2 x CH), 128.2 (6 x CH), 127.3 (CH), 127.1 (C), 126.1 (CH), 124.8 (C), 114.4 (CH), 61.0 (CH₂, OCH₂CH₃), 31.7 (CH₂, CH₂Ar), 16.8 (CH₃, Ar-CH₃), 13.6 (CH₃, OCH₂CH₃); LRMS m/z 347.00 (M + H⁺), calcd C₂₃H₂₂O₃ 346.1569; HRMS m/z 369.1466 (M + Na), calcd for C₂₃H₂₂O₃Na 369.1467; Anal. calcd for C₂₃H₂₂O₃ (346.15): C, 79.74; H, 6.40. Found: C, 79.65; H, 6.47%.

2-Methyl-4-pyrrolidin-1-yl-cyclohexa-1,3-dienecarboxylic acid ethyl ester (95):

This product was prepared in NMR tube as DMSO-D₆ solvent. IR (neat): v_{max} 1662, 1514, 1278, 1205, 1028, 823, 761 cm⁻¹; ¹H NMR (CDCl₃) δ 4.52 (1H, s, olefinic-*H*), 4.00 (2H, q, J = 7.2 Hz, CO₂CH₂CH₃), 3.24 (4H, m, 2 x CO₂Et CH₂), 2.40 (2H, br t, J = 8.8 Hz, CH₂), 2.30 (2H, br t, J = 8.8 Hz, CH₂), 2.12 (3H, s, olefinic-CH₃), 1.85 (4H, m, 2 x CH₂), 1.18 (3H, t, J = 7.2 Hz, CO₂CH₂CH₃); ¹³C NMR (CDCl₃) δ 167.1 (C, O-C=O), 152.5 (C), 150.3 (C), 103.6 (C), 96.7 (CH),

58.0 (CH₂), 47.0 (CH₂), 46.3 (CH₂), 26.4 (CH₂), 25.1 (CH₂), 24.6 (CH₂), 23.7 (CH₂), 22.1 (CH₃), 14.5 (CH₃); GCMS m/z 235.10 (M⁺), calcd C₁₄H₂₁NO₂ 235.1572.

1-(4-Ethoxycarbonyl-3-methyl-cyclohexa-1,3-dienyl)-pyrrolidine-2-carboxylic acidCO₂H

(103): This product was prepared *in situ* in NMR tube and DMSO-D₆ as solvent. IR (neat): v_{max} 3416 (COOH), 2257, 2130, 1651 (O-C=O), 1377, 1026, 999, 826 cm⁻¹; ¹H NMR (CDCl₃) δ 4.54 (1H, s, olefinic-*H*), 4.34-4.24 (1H, m), 4.01 (2H, q, J = 7.2 Hz, OC H_2 CH₃), 3.40-3.32 (2H, m), 103

2.43-2.26 (4H, m), 2.19-2.14 (2H, m), 2.10 (3H, s, olefinic-C H_3), 1.92-1.85 (2H, m), 1.18 (3H, t, J = 7.2 Hz, OCH₂CH₃); ¹³C NMR (CDCl₃) δ 174.3 (C, O-C=O), 167.3 (C, COOH), 152.0 (C), 149.8 (C), 105.2 (C), 98.0 (CH), 60.1 (CH), 58.4 (CH₂, OCH₂CH₃), 47.9 (CH₂), 30.1 (CH₂), 26.3 (CH₂), 23.9 (CH₂), 23.1 (CH₂), 22.2 (CH₃, olefinic-CH₃), 14.6 (CH₃, OCH₂CH₃); LRMS m/z 278.00 (M - H⁺), calcd C₁₅H₂₁NO₄ 279.1471.

3: General experimental procedures for the Hagemann's esters synthesis:

3a: ^tBuOK-Catalyzed cascade Knoevenagel/Michael/aldol condensation/decarboxylation reactions: To a stirred solution of β-keto ester (2 equiv.) and aldehyde (1 equiv.) in t-BuOH (1 M) was added a catalytic amount of ^tBuOK (0.10 equiv.) at 0 °C. The reaction mixture was stirred at that temperature for 60 min, and 0.25 equiv. of t-BuOK was added again. The mixture was then heated at reflux for 32-40 h. Upon cooling to room temperature, the mixture was quenched with 1 M HCl (10 mL) solution, diluted with dichloromethane (50 mL), washed with 1 M NaOH solution (20 mL) and brine (20 mL). The separated organic layer was dried over anhydrous Na₂SO₄, filtered, and concentrated under reduced pressure. Pure cascade products **88a-b,88i-s** were obtained by column chromatography (silica gel, mixture of hexane/ethyl acetate).

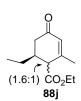
3b: Piperidine-catalyzed cascade Knoevenagel/Michael/aldol condensation/decarboxylation reactions: To a stirred solution of β -keto ester (4 mmol) and benzaldehyde (2 mmol) in EtOH (4 mL) was added a catalytic amount of piperidine (0.7 mmol, 35 mol%) and the reaction mixture was stirred at 80 °C for 3 h.

Upon cooling to room temperature, the mixture was quenched with aqueous NH₄Cl solution, diluted with diethyl ether (50 mL), washed with brine (10 mL). The separated organic layer was dried over anhydrous Na₂SO₄, filtered, and concentrated under reduced pressure. Pure cascade products **88j** and **88m-r** were obtained by column chromatography (silica gel, mixture of hexane/ethyl acetate).

2-Methyl-4-oxo-cyclohex-2-enecarboxylic acid methyl ester (88s): Purified by column chromatography using EtOAc/hexane and isolated as a liquid. IR (neat): v_{max} 2955, 1734 (O-C=O), 1670 (C=O), 1251, 1199, 1033, 779 cm⁻¹; co_2Me ¹H NMR (CDCl₃) δ 5.96 (1H, s, olefinic-*H*), 3.76 (3H, s, OC*H*₃), 3.30 (1H, t, *J* = 4.8 Hz), 2.55 (1H, m), 2.38 (2H, m), 2.23 (1H, m), 2.03 (3H, s, olefinic-C*H*₃); ¹³C NMR (CDCl₃, DEPT-135) δ 198.1 (C, C=O), 171.9 (C, O-C=O), 156.8 (C), 128.4 (CH), 52.4 (CH₃, O*CH*₃), 45.8 (CH), 34.2 (CH₂), 26.0 (CH₂), 23.4 (CH₃, olefinic-*CH*₃).

2-Methyl-4-oxo-cyclohex-2-enecarboxylic acid *tert*-butyl ester (88b): Purified by column chromatography using EtOAc/hexane and isolated as a liquid. IR (neat): ν_{max} 2978, 1718 (O-C=O), 1676 (C=O), 1454, 1369, 1251, 1151, 844 co₂/_{Bu} cm⁻¹; ¹H NMR (CDCl₃) δ 5.94 (1H, s, olefinic-*H*), 3.16 (1H, t, *J* = 5.2 Hz), 2.56 (1H, m), 2.30 (2H, m), 2.17 (1H, m), 2.02 (3H, s, olefinic-C*H*₃), 1.43 (9H, s, OC(C*H*₃)₃); ¹³C NMR (CDCl₃, DEPT-135) δ 198.5 (C, C=O), 170.7 (C, O-C=O), 157.6 (C), 128.1 (CH), 81.9 (C, OC(CH₃)₃), 47.1 (CH), 34.3 (CH₂), 28.0 (CH₃, OC(C*H*₃)₃), 26.2 (CH₂), 23.5 (CH₃, olefinic-*CH*₃).

2,6-Dimethyl-4-oxo-cyclohex-2-enecarboxylic acid ethyl ester (88i): Purified by column chromatography using EtOAc/hexane and isolated as a light yellow oil. IR (neat): v_{max} 2962, 1730 (O-C=O), 1668 (C=O), 1192, 1028, 910, (2:1) $\overset{\bullet}{\text{CO}_2\text{Et}}$ 854, 756 cm⁻¹; ^{1}H NMR (CDCl₃, 2.0:1 ratio of diastereomers, major isomer) δ 5.97 (1H, s, olefinic-H), 4.24 (2H, q, J = 7.2 Hz, OC H_2 CH₃), 3.03 (1H, d, J = 7.2 Hz), 2.59 (2H, m), 2.12 (1H, m), 1.97 (3H, s, olefinic-C H_3), 1.31



(3H, t, J = 7.6 Hz, OCH₂CH₃), 1.10 (3H, d, J = 6.0 Hz); ¹³C NMR (CDCl₃, DEPT-135, 2.0:1 ratio of diastereomers, major isomer) δ 197.9 (C, C=O), 171.8 (C, O-C=O), 155.8 (C), 127.9 (CH), 61.1 (CH₂, OCH₂CH₃), 54.4 (CH), 43.0 (CH₂), 32.7 (CH), 22.6 (CH₃, olefinic-CH₃), 19.7 (CH₃, CHCH₃), 14.1 (CH₃, OCH₂CH₃).

6-Ethyl-2-methyl-4-oxo-cyclohex-2-enecarboxylic acid ethyl ester (88j): Purified by column chromatography using EtOAc/hexane and isolated as a liquid. IR (neat): v_{max} 2972, 1734 (O-C=O), 1672 (C=O), 1186, 1030, 758, 700 cm⁻¹; ¹H NMR (CDCl₃, 1.6:1 ratio of diastereomers, major isomer) δ 5.95 (1H, s, olefinic-*H*), 4.20 (2H, q, J = 7.2 Hz, OC H_2 CH₃), 3.13 (1H, d, J = 6.4 Hz), 2.62 (2H, m), 2.12 (1H, m), 1.97 (3H, s, olefinic-C H_3), 1.48 (2H, m), 1.31 (3H, t, J = 7.2 Hz, OC H_2 CH₃), 0.93 (3H, t, J = 7.2 Hz, CH₂CH₃); ¹³C NMR (CDCl₃, DEPT-135, 1.6:1 ratio of diastereomers, major isomer) δ 198.1 (C, C=O), 171.9 (C, O-C=O), 155.8 (C), 127.9 (CH), 61.2 (CH₂, OCH₂CH₃), 52.3 (CH), 39.9 (CH₂), 39.1 (CH), 26.4 (CH₂), 22.8 (CH₃, olefinic-CH₃), 14.1 (CH₃, OCH₂CH₃), 10.9 (CH₃, CH₂CH₃).

2-Methyl-4-oxo-6-propyl-cyclohex-2-enecarboxylic acid ethyl ester (88k): Purified

by column chromatography using EtOAc/hexane and isolated as a liquid. IR (neat): v_{max} 2961, 1732 (O-C=O), 1674 (C=O), 1186, 1026 cm⁻¹; ¹H NMR (CDCl₃, 1.7:1 ratio of diastereomers, major isomer) δ 5.96 (1H, s, olefinic-H), 4.22 (2H, q, J = 7.2 Hz, OC H_2 CH₃), 3.10 (1H, d, J = 6.8 Hz), 2.61 (2H, m), 2.34 (1H, m), 1.97 (3H, s, olefinic-C H_3), 1.50-1.20 (4H, m), 1.29 (3H, t, J = 7.2 Hz, OCH₂CH₃), 0.91 (3H, t, J = 7.2 Hz, CH₂CH₂CH₃); ¹³C NMR (CDCl₃, DEPT-135, 1.7:1 ratio of diastereomers, major isomer) δ 198.1 (C, C=O), 171.8 (C, O-C=O), 155.8 (C), 128.0 (CH), 61.2 (CH₂, OCH₂CH₃), 52.7 (CH), 40.1 (CH₂), 37.2 (CH), 35.7 (CH₂), 22.9 (CH₃, olefinic-CH₃), 19.6 (CH₂), 14.1 (CH₃, OCH₂CH₃), 13.8 (CH₃, CH₂CH₂CH₃).

2-Methyl-4-oxo-6-phenethyl-cyclohex-2-enecarboxylic acid ethyl ester (881):

Purified by column chromatography using EtOAc/hexane and isolated as a liquid. IR (neat): v_{max} 3024, 2932, 2862, 1732 (O-C=O), 1670

(1.6:1) CO₂Et

(C=O), 1496, 1454, 1259, 1170, 1030, 750, 700 cm⁻¹; ¹H NMR (CDCl₃, 1.6:1 ratio of diastereomers, major isomer) δ 7.28-7.13 (5H, m, Ph-*H*), 5.96 (1H, s, olefinic-*H*), 4.20 (2H, q, J = 7.2 Hz, OCH₂CH₃), 3.14 (1H, d, J = 6.4 Hz), 2.69 (2H, m), 2.53 (2H, m), 2.29 (1H, m), 1.95 (3H, s, olefinic-CH₃), 1.72 (2H, m), 1.28 (3H, t, J = 6.8 Hz, OCH₂CH₃); ¹³C NMR (CDCl₃, DEPT-135, 1.6:1 ratio of diastereomers, major isomer) δ 197.6 (C, C=O), 171.5 (C, O-C=O), 155.6 (C), 141.1 (C), 128.4 (2 x CH), 128.3 (2 x CH), 128.0 (CH), 125.9 (CH), 61.3 (CH₂, OCH₂CH₃), 52.6 (CH), 40.1 (CH₂), 37.0 (CH), 35.4 (CH₂), 32.8 (CH₂), 23.0 (CH₃, olefinic-CH₃), 14.2 (CH₃, OCH₂CH₃).

2-Methyl-4-oxo-6-phenyl-cyclohex-2-enecarboxylic acid ethyl ester (88m): Purified

by column chromatography using EtOAc/hexane and isolated as a liquid. IR (neat): v_{max} 3032, 2982, 1732 (O-C=O), 1670 (C=O), 1496, 1454, 1180, 1032, 760, 700 cm⁻¹; ¹H NMR (CDCl₃, 3.0:1 ratio of diastereomers, major isomer) δ 7.36-7.20 (5H, m, Ph-*H*), 6.06 (1H, s, olefinic-*H*), 4.05 (2H, q, J = 7.2 Hz, OC H_2 CH₃), 3.64 (1H, m), 3.48 (1H, m), 2.68 (2H, m), 2.01 (3H, s, olefinic-C H_3), 1.14 (3H, t, J = 7.2 Hz, OC H_2 CH₃); ¹³C NMR (CDCl₃, DEPT-135, 3.0:1 ratio of diastereomers, major isomer) δ 197.2 (C, C=O), 171.2 (C, O-C=O), 156.2 (C), 140.9 (C), 128.7 (2 x CH), 128.4 (CH), 127.4 (CH), 127.2 (2 x CH), 61.1 (CH₂, OC H_2 CH₃), 54.3 (CH), 44.1 (CH), 42.8 (CH₂), 22.4 (CH₃, olefinic-CH₃), 13.9 (CH₃, OCH₂CH₃).

6-(4-Chloro-phenyl)-2-methyl-4-oxo-cyclohex-2-enecarboxylic acid ethyl ester

(88p): Purified by column chromatography using EtOAc/hexane and isolated as a solid. IR (neat): v_{max} 2980, 1734 (O-C=O), 1674 (C=O), 1493, 1014, 736 cm⁻¹; ¹H NMR (CDCl₃, 4.8:1 ratio of diastereomers, major isomer) δ 7.29 (2H, d, J = 7.2 Hz), 7.17 (2H, d, J = 7.2 Hz), 6.06 (1H, s, olefinic-H), 4.07 (2H, q, J = 7.2 Hz, OC H_2 CH₃), 3.64 (1H, m), 3.53 (1H, m), 2.67 (2H, m), 1.98 (3H, s, olefinic- CH_3), 1.12 (3H, t, J = 7.2 Hz, OC H_2 CH₃); ¹³C NMR (CDCl₃, DEPT-135, 4.8:1 ratio of diastereomers, major isomer) δ 196.7 (C, C=O), 171.0 (C, O-C=O), 156.0 (C), 139.3

(C), 133.1 (C), 128.9 (2 x CH), 128.6 (2 x CH), 128.4 (CH), 61.3 (CH₂, OCH₂CH₃), 54.1 (CH), 43.3 (CH), 42.6 (CH₂), 22.3 (CH₃, olefinic-CH₃), 13.9 (CH₃, OCH₂CH₃).

6-(4-Methoxy-phenyl)-2-methyl-4-oxo-cyclohex-2-enecarboxylic acid ethyl ester (**88n**): Purified by column chromatography using EtOAc/hexane and isolated as a gummy solid. IR (neat): v_{max} 2980, 2837, 1734 (O-C=O), 1672 (C=O), 1612, 1514, 1458, 1258, 1033, 769, 734 cm⁻¹; ¹H NMR (CDCl₃, 2.8:1 ratio of diastereomers, major isomer) δ 7.14 (2H, d, J = 7.2 Hz), 6.85 (2H, d, J = 7.2 Hz), 6.05 (1H, s, olefinic-H), 4.07 (2H, q, J = 7.2 Hz, OC H_2 CH₃), 3.79 (CH₃, s, OC H_3), 3.62 (1H, m), 3.53 (1H, m), 2.67 (2H, m), 1.98 (3H, s, olefinic-C H_3), 1.11 (3H, t, J = 7.2 Hz, OC H_2 CH₃); ¹³C NMR (CDCl₃, DEPT-135, 2.8:1 ratio of diastereomers, major isomer) δ 197.5 (C, C=O), 171.4 (C, O-C=O), 158.8 (C), 156.3 (C), 133.0 (C), 128.4 (CH), 128.2 (2 x CH), 114.1 (2 x CH), 61.1 (CH₂, OC H_2 CH₃), 55.4 (CH₃, OC H_3), 54.7 (CH), 43.3 (CH), 43.1 (CH₂), 22.3 (CH₃, olefinic-CH₃), 14.0 (CH₃, OCH₂CH₃).

2-Methyl-6-(4-nitro-phenyl)-4-oxo-cyclohex-2-enecarboxylic acid ethyl ester (88q):

Purified by column chromatography using EtOAc/hexane and isolated as a gummy solid. IR (neat): v_{max} 2982, 1736 (O- v_{o2}) C=O), 1668 (C=O), 1601, 1521, 1033, 746, 700 cm⁻¹; v_{max} NMR (CDCl₃, 2.4:1 ratio of diastereomers, major isomer) v_{max} 8.13 (2H, d, v_{max}) 7.45 (2H, d, v_{max}) 7.45 (2H, d, v_{max}) 8.13 (2H, d, v_{max}) 7.45 (2H, d, v_{max}) 7.45 (2H, d, v_{max}) 8.16 (1H, m), 2.70 (2H, m), 2.02 (3H, s, olefinic- v_{max}) 7.11 (3H, t, v_{max}) 7.2 Hz, OCH₂CH₃); v_{max} 1.3C NMR (CDCl₃, DEPT-135, 2.4:1 ratio of diastereomers, major isomer) v_{max} 1.15 (2H, d, v_{max}) 1.15 (2H, d, v_{max}) 1.16 (2H, d, v_{max}) 1.17 (3H, t, v_{max}) 1.17 (3H, t, v_{max}) 1.18 (2H, d, v_{max}) 1.19 (2H, d, v_{max}) 1.11 (3H, t, v_{max}) 1

53.6 (CH), 43.7 (CH), 42.1 (CH₂), 22.4 (CH₃, olefinic-*CH*₃), 14.0 (CH₃, OCH₂*CH*₃).

6-Benzo[1,3]dioxol-5-yl-2-methyl-4-oxo-cyclohex-2-

(C), 147.5 (C), 128.5 (CH), 128.3 (2 x CH), 124.1 (2 x CH), 61.6 (CH₂, OCH₂CH₃),

enecarboxylic acid ethyl ester (880): Purified by column chromatography using EtOAc/hexane and isolated as a liquid.

IR (neat): v_{max} 2986, 2910, 1730 (O-C=O), 1670 (C=O),

1506, 1037, 933 cm $^{-1}$; 1 H NMR (CDCl $_{3}$, 2.8:1 ratio of diastereomers, major isomer) δ

6.78-6.65 (3H, m, Ph-H), 6.04 (1H, s, olefinic-H), 5.94 (2H, s, OC H_2 O), 4.10 (2H, q, J = 7.2 Hz, OC H_2 CH₃), 3.58 (1H, m), 3.43 (1H, m), 2.63 (2H, m), 1.98 (3H, s, olefinic-C H_3), 1.14 (3H, t, J = 7.2 Hz, OC H_2 C H_3); ¹³C NMR (CDCl₃, DEPT-135, 2.8:1 ratio of diastereomers, major isomer) δ 197.1 (C, C=O), 171.2 (C, O-C=O), 156.1 (C), 147.8 (C), 146.7 (C), 134.7 (C), 128.4 (CH), 120.5 (CH), 108.3 (CH), 107.4 (CH), 101.1 (CH₂, OCH₂O), 61.2 (CH₂, OC H_2 CH₃), 54.6 (CH), 43.8 (CH), 43.1 (CH₂), 22.3 (CH₃, olefinic-CH₃), 14.1 (CH₃, OCH₂CH₃).

6-Furan-2-yl-2-methyl-4-oxo-cyclohex-2-enecarboxylic acid ethyl ester (88r):

Purified by column chromatography using EtOAc/hexane and isolated as a gummy solid. IR (neat): v_{max} 3119, 2982, 1732 (O-C₀C₀Et (3.4:1) (C=O), 1676 (C=O), 1504, 1440, 1186, 1016, 738 cm⁻¹; ¹H NMR (CDCl₃, 3.4:1 ratio of diastereomers, major isomer) δ 7.29 (1H, br s), 6.26 (1H, m), 6.04 (1H, m), 5.98 (1H, s, olefinic-H), 4.20 (2H, q, J = 7.2 Hz, OC H_2 CH₃), 3.83 (1H, m), 3.63 (1H, m), 2.75 (2H, m), 1.98 (3H, s, olefinic-C H_3), 1.25 (3H, t, J = 7.2 Hz, OCH₂CH₃); ¹³C NMR (CDCl₃, DEPT-135, 3.4:1 ratio of diastereomers, major isomer) δ 196.5 (C, C=O), 170.8 (C, O-C=O), 155.1 (C), 154.2 (C), 141.7 (CH), 128.1 (CH), 110.1 (CH), 106.0 (CH), 61.4 (CH₂, OC H_2 CH₃), 51.0 (CH), 38.9 (CH₂), 37.0 (CH), 23.0 (CH₃, olefinic-CH₃), 14.1 (CH₃, OCH₂CH₃).

4a: Pyrrolidine and auto-catalyzed two-component cascade enamine amination/iso-aromatization reactions: In an ordinary glass vial equipped with a magnetic stirring bar, to 0.6 mmol of the Hagemann's esters 88 was added 1.0 mL of solvent, and then the catalyst pyrrolidine 2a (0.015 mmol, 2.5 μL) was added and the reaction mixture was stirred at 25 °C for the 0.5 h; then 0.3 mmol of nitrosobenzene 56 was added in one-portion and the reaction mixture was stirred at 25 °C for the time indicated in Tables 11 and 13. The crude reaction mixture was directly loaded on silica gel column with or without aqueous work-up and pure cascade products 106 were obtained by column chromatography (silica gel, mixture of hexane/ethyl acetate).

ĊO₂Et

4-Hvdroxy-2-methyl-3-phenylamino-benzoic acid ethvl (106aa): Purified by column chromatography using EtOAc/hexane and isolated as a light yellow solid. Mp 112 °C; IR (neat): v_{max} 3400 (O-H and N-H), 2926, 1668 (O-C=O), 1601, 1496, 1342, 1055, 748 cm⁻¹; ¹H 106aa NMR (CDCl₃) δ 7.88 (1H, d, J = 8.4 Hz), 7.19 (2H, t, J = 8.0 Hz), 6.92 (1H, d, J = 8.8Hz), 6.85 (1H, t, J = 7.2 Hz), 6.58 (2H, d, J = 8.0 Hz), 5.01 (1H, s, N-H), 4.31 (2H, q, J = 8.0 Hz), 5.01 (1H, s, N-H), 4.31 (2H, q, J = 8.0 Hz), 6.85 (1H, t, J = 7.2 Hz), 6.85 (2H, d, J = 8.0 Hz), 5.01 (1H, s, N-H), 4.31 (2H, q, J = 8.0 Hz), 6.85 (1H, t, J = 7.2 Hz), 6.85 (2H, d, J = 8.0 Hz), 5.01 (1H, s, N-H), 4.31 (2H, q, J = 8.0 Hz), 6.85 (1H, t, J = 8.0 Hz), 6.85 (1H, t, J = 8.0 Hz), 6.85 (2H, d, J = 8.0 Hz), 6.85 (1H, t, J = 8.0 Hz), 6.85 (2H, d, J = 8.0 Hz), 6.85 = 6.8 Hz, OCH_2CH_3), 2.40 (3H, s, Ar-CH₃), 1.37 (3H, t, J = 6.8 Hz, OCH_2CH_3); ¹³C NMR (CDCl₃, DEPT-135) δ 167.3 (C, O-C=O), 157.3 (C), 145.7 (C), 140.4 (C), 131.4 (CH), 129.6 (2 x CH), 126.8 (C), 123.0 (C), 120.1 (CH), 114.1 (2 x CH), 111.8 (CH), 60.6 (CH₂, OCH₂CH₃), 15.6 (CH₃, Ar-CH₃), 14.3 (CH₃, OCH₂CH₃); LRMS m/z $270.10 \text{ (M} - \text{H}^{+})$, calcd $C_{16}H_{17}NO_3$ 271.1208; HRMS m/z 272.1271 (M + \square H⁺), calcd $C_{16}H_{17}NO_3H^+$ 272.1286; Anal. calcd for $C_{16}H_{17}NO_3$ (271.12): C, 70.83; H, 6.32; N, 5.16. Found: C, 70.773; H, 6.342; N, 5.329%.

4-Hydroxy-2-methyl-3-phenylamino-benzoic acid methyl ester (106sa): Purified by

CO₂Me 106sa

column chromatography using EtOAc/hexane and isolated as a colorless liquid. IR (neat): v_{max} 3362 (O-*H* and N-*H*), 3049, 2951, 1699 (O-C=O), 1601, 1496, 1435, 1055, 750, 692 cm⁻¹; ¹H NMR (CDCl₃) δ 7.89 (1H, d, J = 8.8 Hz), 7.12 (2H, t, J = 7.6 Hz), 6.93 (1H, d, J = 8.8Hz), 6.86 (1H, t, J = 7.2 Hz), 6.83 (1H, s, O-H), 6.58 (2H, d, J = 8.0 Hz), 4.99 (1H, s, N-H). 3.86 (3H, s. OCH₃), 2.42 (3H, s. Ar- CH_3); ¹³C NMR (CDCl₃, DEPT-135) δ 167.6 (C, O-C=O), 157.4 (C), 145.7 (C), 140.6 (C), 131.5 (CH), 129.6 (2 x CH), 126.8

(C), 122.6 (C), 120.2 (CH), 114.2 (2 x CH), 111.8 (CH), 51.7 (CH₃, OCH₃), 15.6 (CH₃, Ar- CH_3); LRMS m/z 258.15 (M + H⁺), calcd for $C_{15}H_{15}NO_3$ 257.1052; HRMS m/z 258.1125 (M + H⁺), calcd for $C_{15}H_{15}NO_3H^+$ 258.1130.

4-Hydroxy-2-methyl-3-phenylamino-benzoic acid tert-butyl ester (106ba): Purified

ĊO₂^tBu 106ba

by column chromatography using EtOAc/hexane and isolated as yellow oil. IR (neat): v_{max} 3366 (O-H and N-H), 2978, 1699 (O-C=O), 1601, 1496, 1397, 1140, 1049, 750, 692 cm⁻¹; ¹H NMR (CDCl₃) δ 7.80 (1H, d, J = 8.8 Hz), 7.19 (2H, t, J = 8.0 Hz), 6.90 (1H, d, J = 8.4 Hz), 6.85 (1H, t, J = 7.6 Hz), 6.75 (1H, s, O-H), 6.58 (2H, d, J = 8.4 Hz), 4.98 (1H, s, N-H), 2.38 (3H, s, Ar- CH_3), 1.57 (9H, s, 3 x CH₃, tert-CH₃); ¹³C NMR (CDCl₃, DEPT-135) δ 166.7 (C, O-C=O), 156.9 (C), 145.7 (C), 139.7 (C), 131.3 (CH), 129.6 (2 x CH), 126.6 (C), 124.8 (C), 120.1 (CH), 114.2 (2 x CH), 111.7 (CH), 80.8 (C, O- $C(CH_3)_3$), 28.3 (3 x CH₃, O- $C(CH_3)_3$), 15.7 (CH₃, Ar- CH_3); LRMS m/z 300.30 (M + H⁺), calcd for C₁₈H₂₁NO₃ 299.1521; HRMS m/z 300.1599 (M + H⁺), calcd for C₁₈H₂₁NO₃H⁺ 300.1599.

4-Hydroxy-2-methyl-3-o-tolylamino-benzoic acid ethyl ester (106ab): Purified by

column chromatography using EtOAc/hexane and isolated as a yellow solid. Mp 96 °C; IR (neat): v_{max} 3271 (O-H and N-H), 2926, 1670 (O-C=O), 1606, 1575, 1510, 1109, 1062, 746, 625 cm⁻¹; ¹H NMR (CDCl₃) δ 7.90 (1H, d, J = 8.8 Hz), 7.16 (1H, d, J = 7.6 Hz), 6.99 (1H, t, J = 7.6

Hz), 6.94 (1H, d, J = 8.8 Hz), 6.81 (1H, t, J = 7.2 Hz), 6.72 (1H, s, O-H), 6.18 (1H, d, J = 8.0 Hz), 4.83 (1H, s, N-H), 4.32 (2H, q, J = 7.2 Hz, OC H_2 CH₃), 2.39 (3H, s, Ar-CH₃), 2.37 (3H, s, Ar-CH₃), 1.38 (3H, t, J = 7.2 Hz, OCH₂CH₃); ¹³C NMR (CDCl₃, DEPT-135) δ 167.2 (C, O-C=O), 157.2 (C), 143.5 (C), 140.0 (C), 131.4 (CH), 130.6 (CH), 127.4 (CH), 127.0 (C), 123.0 (C), 122.8 (C), 119.9 (CH), 112.2 (CH), 111.8 (CH), 60.6 (CH₂, OCH₂CH₃), 17.6 (CH₃, Ar-CH₃), 15.5 (CH₃, Ar-CH₃), 14.4 (CH₃, OCH₂CH₃); LRMS m/z 286.30 (M + H⁺), calcd C₁₇H₁₉NO₃ 285.1365; HRMS m/z 286.1442 (M + H⁺), calcd C₁₇H₁₉NO₃H⁺ 286.1443.

4-Hydroxy-2,6-dimethyl-3-phenylamino-benzoic acid ethyl ester (106ia): Purified

by column chromatography using EtOAc/hexane and isolated as a light yellow solid. Mp 92 °C; IR (neat): v_{max} 3368 (O-H and N-H), 2962, 1718 (O-C=O), 1602, 1496, 1369, 1043, 750, 694 cm⁻¹; ¹H NMR (CDCl₃) δ 7.19 (2H, t, J = 8.0 Hz), 6.84 (1H, t, J = 7.2 Hz),

6.76 (1H, s), 6.59 (2H, d, J = 8.0 Hz), 6.41 (1H, s, O-H), 4.88(1H, s, N-H), 4.36 (2H, q, J = 7.2 Hz, OC H_2 CH₃), 2.32 (3H, s, Ar-C H_3), 2.10 (3H, s, Ar-C H_3), 1.37 (3H, t, J = 6.8 Hz, OC H_2 CH₃); ¹³C NMR (CDCl₃, DEPT-135) δ 169.6 (C, O-C=O), 154.8 (C), 145.9 (C), 135.9 (C), 134.9 (C), 129.6 (2 x CH), 127.1 (C), 123.8 (C), 119.9 (CH),

114.1 (2 x CH), 114.0 (CH), 61.0 (CH₂, OCH₂CH₃), 20.0 (CH₃, Ar-CH₃), 15.2 (CH₃, Ar-CH₃), 14.3 (CH₃, OCH₂CH₃); LRMS m/z 286.20 (M + H⁺), calcd C₁₇H₁₉NO₃ 285.1365; HRMS m/z 286.1425 (M + H⁺), calcd C₁₇H₁₉NO₃H⁺ 286.1443.

6-ethyl-4-Hydroxy-2-methyl-3-phenylamino-benzoic acid ethyl ester (106ja):

Purified by column chromatography using EtOAc/hexane and isolated as a solid. IR (neat): v_{max} 3387 (O-H and N-H), 2974, 1701 (O-C=O), 1658, 1602 1498, 1176, 1051 cm⁻¹; ¹H NMR (CDCl₃) δ 7.17 (2H, t, J = 8.0 Hz), 6.83 (1H, t, J = 7.6 Hz), 6.78 (1H, s, O-H), 6.59 (2H, d, J = 8.0 Hz), 6.41 (1H, s), 4.91 (1H, s, N-H), 4.35 (2H, q, J = 7.2 Hz, OC H_2 CH₃), 2.62 (2H, q, J = 8.0 Hz, C H_2 CH₃), 2.07 (3H, s, Ar-CH₃), 1.36 (3H, t, J = 6.8 Hz, OCH₂CH₃), 1.24 (3H, t, J = 7.6 Hz, CH₂CH₃); ¹³C NMR (CDCl₃, DEPT-135) δ 169.8 (C, O-C=O), 154.9 (C), 145.9 (C), 141.8 (C), 134.5 (C), 129.5 (2 x CH), 126.8 (C), 123.8 (C), 119.8 (CH), 114.1 (2 x CH), 112.5 (CH), 61.0 (CH₂, OC H_2 CH₃), 26.8 (CH₂, C H_2 CH₃), 15.4 (CH₃, Ar-CH₃), 15.2 (CH₃, OCH₂CH₃), 14.3 (CH₃, CH₂CH₃); LRMS m/z 298.15 (M – H⁺), calcd C₁₈H₂₁NO₃ 299.1521; HRMS m/z 300.1592 (M + \square H⁺), calcd C₁₈H₂₁NO₃H⁺ 300.1599.

4-Hydroxy-2-methyl-3-phenylamino-6-propyl-benzoic acid ethyl ester (106ka):

Purified by column chromatography using EtOAc/hexane and isolated as solid. Mp 58 °C; IR (neat): v_{max} 3371 (O-H and N-H), 3051, 2961, 1720 (O-C=O), 1660, 1602, 1498, 1176, 1053, 748, 694 cm⁻¹; ¹H NMR (CDCl₃) δ 7.17 (2H, t, J = 8.0 Hz), 6.82 (1H, t, J = 7.6 Hz), 6.76 (1H, s), 6.58 (2H, d, J = 8.0 Hz), 6.39 (1H, s, O-H), 4.88 (1H, s, N-H), 4.34 (2H, q, J = 7.6 Hz, OCH₂CH₃), 2.56 (2H, t, J = 8.0 Hz, ArCH₂CH₂CH₃), 2.08 (3H, s, Ar-CH₃), 1.64 (2H, m, ArCH₂CH₂CH₃), 1.36 (3H, t, J = 7.2 Hz, OCH₂CH₃), 0.96 (3H, t, J = 8.0 Hz, ArCH₂CH₂CH₃); ¹³C NMR (CDCl₃, DEPT-135) δ 169.7 (C, O-C=O), 154.7 (C), 145.8 (C), 140.4 (C), 134.5 (C), 129.5 (2 x CH), 127.0 (C), 123.8 (C), 119.8 (CH), 114.1 (2 x CH), 113.2 (CH), 61.0 (CH₂, OCH₂CH₃), 35.9 (CH₂, ArCH₂CH₂CH₃), 24.3 (CH₂, ArCH₂CH₂CH₃), 15.2 (CH₃, Ar-CH₃), 14.2 (CH₃,

OCH₂CH₃), 14.0 (CH₃, ArCH₂CH₂CH₃); LRMS m/z 312.20 (M – H⁺), calcd $C_{19}H_{23}NO_3$ 313.1678; HRMS m/z 314.1747 (M + H⁺), calcd $C_{19}H_{23}NO_3H^+$ 314.1756.

4-Hydroxy-2-methyl-6-phenethyl-3-phenylamino-benzoic acid ethyl ester (106la):

Purified by column chromatography using EtOAc/hexane and isolated as orange oil. IR (neat): v_{max} 3366 (O-H and N-H), 3026, 2986, 1699 (O-C=O), 1602, 1494, 1304, 1053, 750, 698 cm⁻¹; ¹H NMR (CDCl₃) δ 7.29 (2H, t, J = 8.0 Hz), 7.18 (5H, m, Ph-H), 6.82 (1H, t, J = 7.6 Hz), 6.76 (1H, s), 6.57 (2H, d, J = 8.0 Hz), 6.43 (1H, s, O-H), 4.93 (1H, s, N-H), 4.36 (2H, q, J = 6.8 Hz, OC H_2 CH₃), 2.89 (4H, m, PhC H_2 C H_2 Ar), 2.08 (3H, s, Ar-C H_3), 1.35 (3H, t, J = 7.2 Hz, OCH₂C H_3); ¹³C NMR (CDCl₃, DEPT-135) δ 169.7 (C, O-C=O), 154.8 (C), 145.8 (C), 141.5 (C), 139.5 (C), 134.8 (C), 129.5 (2 x CH), 128.4 (2 x CH), 128.3 (2 x CH), 126.9 (C), 126.0 (CH), 124.2 (C), 119.8 (CH), 114.0 (2 x CH), 113.3 (CH), 61.1 (CH₂, OC H_2 CH₃), 37.6 (CH₂), 36.0 (CH₂), 15.2 (CH₃, Ar-CH₃), 14.2 (CH₃, OCH₂CH₃); LRMS m/z 374.15 (M - H⁺), calcd

4-Hydroxy-2-methyl-6-phenethyl-3-o-tolylamino-benzoic acid ethyl ester (106lb):

 $C_{24}H_{25}NO_3$ 375.1834; HRMS m/z 398.1731 (M + Na⁺), calcd $C_{24}H_{25}NO_3Na$

398.1732; Anal. calcd for C₂₄H₂₅NO₃ (375.18): C, 76.77; H, 6.71; N, 3.73. Found: C,

Purified by column chromatography using EtOAc/hexane and isolated as a light yellow solid. Mp 82 °C; IR (neat): v_{max} 3383 (O-*H* and N-*H*), 3026, 2978, 1714 (O-C=O), 1587, 1498, 1365, 1111, 1051, 750, 700 cm⁻¹; ¹H NMR (CDCl₃) δ 7.29 (2H, t, J =

76.724; H, 6.733; N, 3.969%.

7.6 Hz), 7.21 (3H, m), 7.13 (1H, d, J = 7.6 Hz), 7.00 (1H, t, J = 7.6 Hz), 6.78 (1H, s), 6.77 (1H, t, J = 8.0 Hz), 6.36 (1H, s, O-H), 6.25 (1H, d, J = 8.0 Hz), 4.75 (1H, s, N-H), 4.35 (2H, q, J = 6.8 Hz, OC H_2 CH $_3$), 2.90 (4H, m, PhC H_2 CH $_2$ Ar), 2.34 (3H, s, Ar-CH $_3$), 2.06 (3H, s, Ar-CH $_3$), 1.35 (3H, t, J = 7.2 Hz, OCH $_2$ CH $_3$); ¹³C NMR (CDCl $_3$, DEPT-135) δ 169.6 (C, O-C=O), 154.6 (C), 143.6 (C), 141.5 (C), 139.4 (C), 134.4 (C), 130.5 (CH), 128.4 (2 x CH), 128.3 (2 x CH), 127.4 (CH), 127.0 (C), 126.0 (CH), 124.5

(C), 122.7 (C), 119.7 (CH), 113.4 (CH), 112.2 (CH), 61.1 (CH₂, O*CH*₂CH₃), 37.6 (CH₂), 36.0 (CH₂), 17.5 (CH₃, Ar-CH₃), 15.1 (CH₃, Ar-*C*H₃), 14.3 (CH₃, OCH₂C*H*₃); LRMS m/z 390.25 (M + H⁺), calcd C₂₅H₂₇NO₃ 389.1991; HRMS m/z 412.1888 (M + Na), calcd C₂₅H₂₇NO₃Na 412.1889; Anal. calcd for C₂₅H₂₇NO₃ (389.19): C, 77.09; H, 6.99; N, 3.60. Found: C, 77.245; H, 6.952; N, 3.891%.

5-Hydroxy-3-methyl-4-phenylamino-biphenyl-2-carboxylic acid ethyl ester

OH H N CO₂Et 106ma (106ma): Purified by column chromatography using EtOAc/hexane and isolated as yellow solid. Mp 88 °C; IR (neat): v_{max} 3371 (O-H and N-H), 2926, 1732 (O-C=O), 1668, 1602, 1498, 1157, 752, 700 cm⁻¹; ¹H NMR (CDCl₃) δ 7.38 (5H, m, Ph-

H), 7.22 (2H, t, J = 8.4 Hz), 6.93 (1H, s), 6.87 (1H, t, J = 7.2 Hz), 6.66 (2H, d, J = 8.0 Hz), 6.43 (1H, s, O-H), 5.00 (1H, s, N-H), 3.98 (2H, q, J = 6.8 Hz, OC H_2 CH₃), 2.20 (3H, s, Ar-C H_3), 0.90 (3H, t, J = 6.8 Hz, OCH₂C H_3); ¹³C NMR (CDCl₃, DEPT-135) δ 169.4 (C, O-C=O), 154.7 (C), 145.6 (C), 140.9 (C), 140.8 (C), 135.4(C), 129.6 (2 x CH), 128.2 (2 x CH), 128.1 (2 x CH), 127.4 (CH), 126.5 (C), 125.4 (C), 120.0 (CH), 114.2 (2 x CH), 113.9 (CH), 60.9 (CH₂, OC H_2 CH₃), 15.2 (CH₃, Ar-CH₃), 13.5 (CH₃, OCH₂C H_3); LRMS m/z 346.05 (M – H⁺), calcd C₂₂H₂₁NO₃ 347.1521; HRMS m/z 370.1415 (M + Na⁺), calcd C₂₂H₂₁NO₃Na⁺ 370.1419.

5-Hydroxy-3-methyl-4-o-tolylamino-biphenyl-2-carboxylic acid ethyl ester

OH H N CO₂Et 106mb

(106mb): Purified by column chromatography using EtOAc/hexane and isolated as a white solid. Mp 82 °C; IR (neat): v_{max} 3385 (*O-H* and *N-H*), 3057, 2978, 1712 (O-C=O), 1587, 1498, 1471, 1111, 1051, 750, 700 cm⁻¹; ¹H NMR (CDCl₃) δ 7.34

(5H, m, Ph-*H*), 7.17 (1H, d, J = 7.2 Hz), 7.05 (1H, t, J = 7.6 Hz), 6.95 (1H, s), 6.82 (1H, t, J = 7.2 Hz), 6.45 (1H, s, O-*H*), 6.35 (1H, d, J = 8.0 Hz), 4.83 (1H, s, N-*H*), 4.00 (2H, q, J = 7.6 Hz, OCH₂CH₃), 2.40 (3H, s, Ar-CH₃), 2.16 (3H, s, Ar-CH₃), 0.90 (3H, t, J = 7.6 Hz, OCH₂CH₃); ¹³C NMR (CDCl₃, DEPT-135) δ 169.4 (C, O-C=O), 154.5 (C), 143.4 (C), 140.9 (C), 140.8 (C), 135.0 (C), 130.6 (CH), 128.2 (2 x CH), 128.1 (2 x CH), 127.5 (CH), 127.4 (CH), 126.6 (C), 125.6 (C), 122.8 (C), 119.9 (CH), 113.9 (CH),

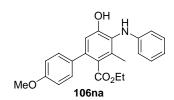
112.4 (CH), 60.9 (CH₂, O*CH*₂CH₃), 17.6 (CH₃, Ar-*C*H₃), 15.0 (CH₃, Ar-*C*H₃), 13.5 (CH₃, OCH₂C*H*₃); LRMS m/z 362.20 (M + H⁺), calcd C₂₃H₂₃NO₃ 361.1678; HRMS m/z 384.1557 (M + Na), calcd C₂₃H₂₃NO₃Na 384.1576; Anal. calcd for C₂₃H₂₃NO₃ (361.16): C, 76.43; H, 6.41; N, 3.88. Found: C, 76.539; H, 6.403; N, 3.985%.

4'-Chloro-5-hydroxy-3-methyl-4-phenylamino-biphenyl-2-carboxylic acid ethyl

ester (**106pa**): Purified by column chromatography using EtOAc/hexane and isolated as yellow oil. IR (neat): v_{max} 3389 (*O-H* and *N-H*), 2982, 1730 (O-C=O), 1670, 1597, 1493, 1199, 1091, 748, 694 cm⁻¹; ¹H NMR (CDCl₃) δ 7.39 (2H, d, *J*

= 8.4 Hz), 7.34 (2H, d, J = 8.4 Hz), 7.25 (2H, t, J = 8.4 Hz), 6.92 (1H, s), 6.90 (1H, t, J = 7.2 Hz), 6.68 (2H, d, J = 8.4 Hz), 6.54 (1H, s, O-H), 5.00 (1H, s, N-H), 4.05 (2H, q, J = 6.8 Hz, OC H_2 CH₃), 2.22 (3H, s, Ar-C H_3), 1.01 (3H, t, J = 7.2 Hz, OC H_2 C H_3); ¹³C NMR (CDCl₃, DEPT-135) δ 169.2 (C, O-C=O), 154.7 (C), 145.4 (C), 139.5 (C), 139.2 (C), 135.6 (C), 133.6 (C), 129.6 (2 x CH), 129.5 (2 x CH), 128.4 (2 x CH), 126.5 (C), 125.7 (C), 120.2 (CH), 114.3 (2 x CH), 113.8 (CH), 61.1 (CH₂, OC H_2 CH₃), 15.2 (CH₃, Ar-CH₃), 13.7 (CH₃, OCH₂C H_3); LRMS m/z 380.10 (M - H⁺), calcd C₂₂H₂₀ClNO₃ 381.1140; HRMS m/z 404.1014 (M + Na⁺), calcd C₂₂H₂₀ClNO₃Na⁺ 404.1029.

5-Hydroxy-4'-methoxy-3-methyl-4-phenylamino-biphenyl-2-carboxylic acid ethyl



ester (106na): Purified by column chromatography using EtOAc/hexane and isolated as yellow liquid. IR (neat): v_{max} 3385 (*O-H* and *N-H*), 2924, 1716 (O-C=O), 1604, 1502, 1406, 1248, 750, 694 cm⁻¹; ¹H NMR (CDCl₃) δ 7.31 (2H, d,

J = 8.4 Hz), 7.21 (2H, d, J = 8.0 Hz), 6.93 (1H, s), 6.90 (2H, d, J = 7.6 Hz), 6.85 (1H, t, J = 7.2 Hz), 6.65 (2H, d, J = 8.4 Hz), 6.53 (1H, s, O-H), 5.00 (1H, s, N-H), 4.04 (2H, q, J = 6.8 Hz, OCH₂CH₃), 3.84 (3H, s, OCH₃), 2.17 (3H, s, Ar-CH₃), 0.99 (3H, t, J = 6.8 Hz, OCH₂CH₃); ¹³C NMR (CDCl₃, DEPT-135) δ 169.7 (C, O-C=O), 159.1 (C), 154.6 (C), 145.7 (C), 140.3 (C), 135.2 (C), 133.1 (C), 129.6 (2 x CH), 129.2 (2 x CH), 126.6

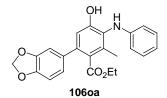
(C), 125.0 (C), 120.0 (CH), 114.2 (2 x CH), 113.8 (CH), 113.7 (2 x CH), 61.0 (CH₂, OCH₂CH₃), 55.3 (CH₃, OCH₃), 15.1 (CH₃, Ar-CH₃), 13.7 (CH₃, OCH₂CH₃); LRMS m/z 378.20 (M + H⁺), calcd $C_{23}H_{23}NO_4$ 377.1627; HRMS m/z 378.1698 (M + H⁺), calcd $C_{23}H_{23}NO_4H^+$ 378.1705.

5-Hydroxy-3-methyl-4'-nitro-4-phenylamino-biphenyl-2-carboxylic acid ethyl

ester (106qa): Purified by column chromatography using EtOAc/hexane and isolated as a yellow solid. Mp 180 °C; IR (neat): v_{max} 3404 (*O-H* and *N-H*), 2976, 1732 (O-C=O), 1684, 1593, 1516, 1055, 744, 692 cm⁻¹; ¹H NMR (CDCl₃) δ

8.27 (2H, d, J = 8.8 Hz), 7.55 (2H, d, J = 8.4 Hz), 7.24 (2H, t, J = 8.0 Hz), 6.92 (1H, s), 6.89 (1H, t, J = 7.2 Hz), 6.67 (2H, d, J = 8.0 Hz), 6.55 (1H, s, O-H), 5.00 (1H, s, N-H), 4.03 (2H, q, J = 7.2 Hz, OC H_2 CH $_3$), 2.23 (3H, s, Ar-C H_3), 0.97 (3H, t, J = 7.2 Hz, OC H_2 CH $_3$); ¹³C NMR (CDCl $_3$, DEPT-135) δ 168.7 (C, O-C=O), 154.9 (C), 147.5 (C), 147.2 (C), 145.1 (C), 138.5 (C), 136.2 (C), 129.7 (2 x CH), 129.1 (2 x CH), 126.6 (C), 126.3 (C), 123.5 (2 x CH), 120.5 (CH), 114.5 (2 x CH), 113.8 (CH), 61.2 (CH $_2$, OC H_2 CH $_3$), 15.3 (CH $_3$, Ar-CH $_3$), 13.7 (CH $_3$, OCH $_2$ CH $_3$); LRMS m/z 391.10 (M – H $^+$), calcd C $_{22}$ H $_{20}$ N $_2$ O $_5$ 392.1372; HRMS m/z 393.1467 \blacksquare M H $^+$), calcd C $_{22}$ H $_{20}$ N $_2$ O $_5$ H $^+$ 393.1450.

6-Benzo(1, 3)dioxol-5-yl-4-hydroxy-2-methyl-3-phenylamino-benzoic acid ethyl



ester (**1060a**): Purified by column chromatography using EtOAc/hexane and isolated as yellow oil. IR (neat): v_{max} 3371 (O-*H* and N-*H*), 3047, 2980, 1712 (O-C=O), 1602, 1500, 1039, 750, 694 cm⁻¹; ¹H NMR (CDCl₃) δ 7.22 (2H, t, *J*

= 8.0 Hz), 6.91-6.83 (5H, m, Ar-H), 6.67 (2H, d, J = 8.0 Hz), 6.00 (2H, s, OC H_2 O), 5.06 (1H, s, N-H), 4.13 (2H, q, J = 7.2 Hz, OC H_2 CH₃), 2.19 (3H, s, Ar-C H_3), 1.08 (3H, t, J = 7.6Hz, OC H_2 CH₃); ¹³C NMR (CDCl₃, DEPT-135) δ 169.5 (C, O-C=O), 154.5 (C), 147.5 (C), 147.0 (C), 145.7 (C), 140.2 (C), 135.2 (C), 134.7 (C), 129.6 (2 x CH), 126.6 (C), 125.3 (C), 121.7(CH), 119.9 (CH), 114.2 (2 x CH), 113.8 (CH), 108.8 (CH),

108.1 (CH), 101.1 (CH₂, O*C*H₂O), 61.0 (CH₂, O*CH*₂CH₃), 15.1 (CH₃, Ar-*C*H₃), 13.8 (CH₃, OCH₂C*H*₃); GCMS m/z 391 (M⁺), calcd C₂₃H₂₁NO₅ (391.14); Anal. calcd for C₂₃H₂₁NO₅ (391.14): C, 70.58; H, 5.41; N, 3.58. Found: C, 70.644; H, 5.416; N, 3.924%.

6-Furan-2-yl-4-hydroxy-2-methyl-3-phenylamino-benzoic acid ethyl ester (106ra):

Purified by column chromatography using EtOAc/hexane and isolated as a white solid. Mp 84 °C; IR (neat): v_{max} 3383 (O-H and N-H), 2928, 1722 (O-C=O), 1602, 1498, 1367, 1199, 1055, 746, 694 cm⁻¹; ¹H NMR (CDCl₃) δ 7.46 (1H, d, J = 1.2 Hz), 7.20 (2H, t, J = 8.4 Hz), 7.17 (1H, s), 6.85 (1H, t, J = 7.6 Hz), 6.62 (2H, d, J = 8.0 Hz), 6.56 (1H, d, J = 3.0 Hz), 6.46 (1H, br d, J = 3.2 Hz), 5.02 (1H, s, N-H), 4.30 (2H, q, J = 7.2 Hz, OC H_2 CH₃), 2.14 (3H, s, Ar-C H_3), 1.25 (3H, t, J = 7.2 Hz, OCH₂C H_3); ¹³C NMR (CDCl₃, DEPT-135) δ 169.7 (C, O-C=O), 154.6 (C), 151.9 (C), 145.5 (C), 142.6 (CH), 135.1 (C), 129.6 (2 x CH), 128.2 (C), 125.8 (C), 124.4 (C), 120.1 (CH), 114.2 (2 x CH), 111.6 (CH), 110.7 (CH), 107.7 (CH), 61.4 (CH₂, OC H_2 CH₃), 14.9 (CH₃, Ar-CH₃), 14.1 (CH₃, OCH₂C H_3); GCMS m/z 337 (M⁺), calcd C₂₀H₁₉NO₄ (337.13); Anal. calcd for C₂₀H₁₉NO₄ (337.13): C, 71.20; H, 5.68; N, 4.15. Found: C, 71.187; H, 5.689; N, 4.199%.

3-(4-Dimethylamino-phenylamino)-4-hydroxy-2-methyl-benzoic acid ethyl ester

(106ac): Purified by column chromatography using EtOAc/hexane and isolated as a dark greenish liquid. IR (neat): v_{max} 3391 (O-H and N-H), 2978, 1712 (O-C=O), 1589, 1516, 1053, 779 cm⁻¹; ^{1}H NMR (CDCl₃) δ 7.85 (1H, d, J = 8.4 Hz), 6.90 (1H, d, J = 8.8 Hz), 6.67 (2H, d, J = 6.8 Hz), 6.53 (2H, d, J = 8.8 Hz), 4.75 (1H, br s, N-H), 4.31 (2H, q, J = 7.2 Hz, OC H_2 CH₃), 2.83 (6H, s, N(C H_3)₂), 2.41 (3H, s, Ar-C H_3), 1.34 (3H, t, J = 7.2 Hz, OC H_2 CH₃); 13 C NMR (CDCl₃, DEPT-135) δ 167.3 (C, O-C=O), 157.3 (C), 145.6 (C), 140.0 (C), 137.0 (C), 130.9 (CH), 128.0 (C), 122.8 (C), 115.6 (2 x CH), 115.1 (2 x CH), 111.6 (CH), 60.5 (CH₂, OCH₂CH₃), 41.7 (CH₃, N(CH_3)₂), 15.6 (CH₃, Ar- CH_3),

14.3 (CH₃, OCH₂CH₃); LRMS m/z 314.1 (M⁺), calcd $C_{18}H_{22}N_2O_3$ (314.16); Anal. calcd for C₁₈H₂₂N₂O₃ (314.16): C, 68.77; H, 7.05; N, 8.91. Found: C, 68.825; H, 7.043; N, 9.142%.

4b: **Self-catalyzed** three-component cascade enamine amination/iso**aromatization reactions:** In an ordinary glass vial equipped with a magnetic stirring bar, to 0.3 mmol of the pyrrolidine, 0.6 mmol of Hagemann's esters 88 and 300 mg of MS 4A° was added 0.5 mL of solvent, and then the 0.5 mL solution of nitrosobenzene (0.3 mmol) **56** was added dropwise for 0.5 h and the reaction mixture was stirred at 25 °C for the time indicated in Tables 12 and 14. The crude reaction mixture was worked up with aqueous NH₄Cl solution and the aqueous layer was extracted with dichloromethane (2 x 20 mL). The combined organic layers were dried (Na₂SO₄), filtered and concentrated. Pure cascade products 107 were obtained by column chromatography (silica gel, mixture of hexane/ethyl acetate).

2-Methyl-3-phenylamino-4-pyrrolidin-1-yl-benzoic acid ethyl ester (107aa):

Purified by column chromatography using EtOAc/hexane and isolated as a solid. Mp 76 °C; IR (neat): v_{max} 3381 (N-H), 2970, 1705 (O-C=O), 1597, 1498, 1167, 1055, 748, 694 cm⁻¹; ¹H NMR (CDCl₃) δ 7.81 (1H, ĊO₂Et d, J = 8.8 Hz), 7.17 (2H, t, J = 8.0 Hz), 6.75 (1H, t, J = 7.2 Hz), 6.69 107aa

(1H, d, J = 8.8 Hz), 6.50 (2H, d, J = 8.4 Hz), 5.18 (1H, s, N-H), 4.33

 $(2H, q, J = 7.2 \text{ Hz}, OCH_2CH_3), 3.30 (4H, t, J = 6.4 \text{ Hz}), 2.44 (3H, s, Ar-CH_3), 1.82$ (4H, m), 1.39 (3H, t, J = 6.8 Hz, OCH₂CH₃); ¹³C NMR (CDCl₃, DEPT-135) δ 167.8 (C, O-C=O), 150.4 (C), 146.5 (C), 140.4 (C), 129.4 (CH), 129.0 (2 x CH), 127.1 (C), 120.2 (C), 117.7 (CH), 113.1 (2 x CH), 111.9 (CH), 60.1 (CH₂, OCH₂CH₃), 50.6 (2 x CH₂), 25.4 (2 x CH₂), 16.0 (CH₃, Ar-CH₃), 14.3 (CH₃, OCH₂CH₃); LRMS m/z 325.25

CO₂Et

 $(M + H^{+})$, calcd $C_{20}H_{24}N_{2}O_{2}$ 324.1838; HRMS m/z 325.1902 (M + H^{+}), calcd $C_{20}H_{24}N_{2}O_{2}H^{+}$ 325.1916; Anal. calcd for $C_{20}H_{24}N_{2}O_{2}$ (324.18): C, 74.04; H, 7.46; N, 8.64. Found: C, 74.038; H, 7.465; N, 8.508%.

108aa

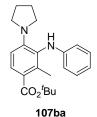
2-Methyl-5-phenylamino-4-pyrrolidin-1-yl-benzoic acid ethyl ester (108aa): Purified by column chromatography using EtOAc/hexane and isolated as liquid. IR (neat): v_{max} 3379 (N-H), 2966, 1707 (O-C=O), 1599, 1498, 1076, 748, 694 cm⁻¹; ¹H NMR (CDCl₃) δ 7.81 (1H, s), 7.19 (2H, t, J = 7.2 Hz), 6.78 (1H, t, J = 7.2 Hz), 6.70 (2H, d, J = 7.6 Hz), 6.61 (1H, s), 5.28 (1H, s, N-H), 4.26 (2H, q, J = 7.2 Hz, OC H_2 CH₃), 3.32 (4H, t, J = 6.4 Hz), 2.58 (3H, s, Ar-C H_3), 1.84 (4H, m), 1.32 (3H, t, J = 7.2 Hz, OCH₂C H_3); ¹³C NMR (CDCl₃, DEPT-135) δ 167.1 (C, O-C=O), 148.0 (C), 146.3 (C), 138.2 (C), 129.3 (CH), 129.2 (2 x CH), 126.9 (C), 119.2 (C), 118.5 (CH), 118.3 (CH), 114.4 (2 x CH), 60.0 (CH₂, OCH₂CH₃), 50.1 (2 x CH₂), 25.2 (2 x CH₂), 22.0 (CH₃, Ar-CH₃), 14.4 (CH₃, OCH₂C H_3); LRMS m/z 325.30 (M + H⁺), calcd C₂₀H₂₄N₂O₂ 324.1838.

2-Methyl-3-phenylamino-4-pyrrolidin-1-yl-benzoic acid methyl ester (107sa):

 Purified by column chromatography using EtOAc/hexane and isolated as a solid. Mp 96 °C; IR (neat): v_{max} 3395 (N-H), 2932, 1701 (O-C=O), 1602, 1508, 1055, 746, 694 cm⁻¹; ¹H NMR (CDCl₃) δ 7.79 (1H, d, J = 8.8 Hz), 7.17 (2H, t, J = 7.6 Hz), 6.74 (1H, t, J = 7.2 Hz), 6.68 (1H, d, J = 8.8 Hz), 6.50 (2H, d, J = 8.0 Hz), 5.17 (1H, s, N-H), 3.85 (3H, s,

OC H_3), 3.31 (4H, t, J = 6.4 Hz), 2.43 (3H, s, Ar-C H_3), 1.83 (4H, m); ¹³C NMR (CDCl₃, DEPT-135) δ 168.2 (C, O-C=O), 150.6 (C), 146.6 (C), 140.8 (C), 129.6 (CH), 129.3 (2 x CH), 127.2 (C), 119.9 (C), 117.9 (CH), 113.2 (2 x CH), 112.0 (CH), 51.5 (CH₃, OCH₃), 50.7 (2 x CH₂), 25.6 (2 x CH₂), 14.2 (CH₃, Ar-CH₃); LRMS m/z 311.20 (M + H⁺), calcd C₁₉H₂₂N₂O₂ 310.1839; HRMS m/z 311.1744 (M + H⁺), calcd C₁₉H₂₂N₂O₂H⁺ 311.1759.

2-Methyl-3-phenylamino-4-pyrrolidin-1-yl-benzoic acid tert-butyl ester (107ba):



Purified by column chromatography using EtOAc/hexane and isolated as a solid. Mp 110 °C; IR (neat): v_{max} 3385 (N-*H*), 2966, 1703 (O-C=O), 1599, 1498, 1415, 1149, 1080, 746, 692 cm⁻¹; ¹H NMR (CDCl₃) δ 7.67 (1H, d, J = 8.8 Hz), 7.16 (2H, t, J = 8.0 Hz), 6.72 (1H, t, J = 7.2

Hz), 6.66 (1H, d, J = 8.8 Hz), 6.48 (2H, d, J = 8.0 Hz), 5.15 (1H, s, N-H), 3.23 (4H, t, J = 6.4 Hz), 2.38 (3H, s, Ar-C H_3), 1.78 (4H, p, J = 6.4 Hz), 1.24 (9H, s, 3 x CH₃, OC(C H_3)₃); ¹³C NMR (CDCl₃, DEPT-135) δ 167.4 (C, O-C=O), 150.0 (C), 146.5 (C), 139.5 (C), 129.1 (CH), 129.0 (2 x CH), 127.5 (C), 122.5 (C), 117.8 (CH), 113.2 (2 x CH), 112.0 (CH), 80.1 (C, OC(CH₃)₃), 50.6 (2 x CH₂), 28.3 (3 x CH₃, OC(CH₃)₃), 25.4 (2 x CH₂), 16.1 (CH₃, Ar-CH₃); LRMS m/z 353.30 (M + H⁺), calcd C₂₂H₂₈N₂O₂ 352.2151; HRMS m/z 353.2216 (M + H⁺), calcd C₂₂H₂₈N₂O₂H⁺ 353.2229; Anal. calcd for C₂₂H₂₈N₂O₂ (352.21): C, 74.97; H, 8.01; N, 7.95. Found: C, 74.782; H, 8.049; N, 8.010%.

2-Methyl-4-pyrrolidin-1-yl-3-o-tolyl amino-benzoic acid ethyl ester (107ab):

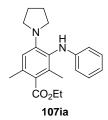
N H N CO₂Et

107ab

Purified by column chromatography using EtOAc/hexane and isolated as a light yellow solid. Mp 88 °C; IR (neat): v_{max} 3048 (N-H), 2968, 1703 (O-C=O), 1587, 1500, 1257, 1053, 748 cm⁻¹; ¹H NMR (CDCl₃) δ 7.77 (1H, d, J = 8.8 Hz), 7.12 (1H, d, J = 7.2 Hz), 7.00 (1H, t, J = 8.0 Hz), 6.75 (2H, m), 6.21 (1H, d, J = 7.6 Hz), 5.00 (1H, s, N-H), 4.30 (2H,

q, J = 7.2 Hz, OC H_2 CH₃), 3.28 (4H, t, J = 6.8 Hz), 2.35 (3H, s, Ar-C H_3), 2.30 (3H, s, Ar-C H_3), 1.80 (4H, m), 1.37 (3H, t, J = 7.2 Hz, OCH₂C H_3); ¹³C NMR (CDCl₃, DEPT-135) δ 167.9 (C, O-C=O), 150.2 (C), 144.5 (C), 139.8 (C), 130.1 (CH), 129.2 (CH), 127.9 (C), 126.9 (CH), 121.7 (C), 120.8 (C), 117.8 (CH), 112.2 (CH), 111.9 (CH), 60.2 (CH₂, OCH₂CH₃), 50.4 (2 x CH₂), 25.4 (2 x CH₂), 17.7 (CH₃, Ar-CH₃), 16.0 (CH₃, Ar-CH₃), 14.4 (CH₃, OCH₂C H_3); LRMS m/z 339.30 (M + H⁺), calcd C₂₁H₂₆N₂O₂ 338.1994; HRMS m/z 361.1892 (M + Na⁺), calcd C₂₁H₂₆N₂O₂Na 361.1892.

2,6-Dimethyl-3-phenylamino-4-pyrrolidin-1-yl-benzoic acid ethyl ester (107ia):



Purified by column chromatography using EtOAc/hexane and isolated as a solid. Mp 90 °C; IR (neat): v_{max} 3383 (N-H), 2970, 1705 (O-C=O), 1601, 1500, 1101, 1053, 748, 694 cm⁻¹; ¹H NMR (CDCl₃) δ 7.16 (2H, t, J = 7.6 Hz), 6.74 (1H, t, J = 7.2 Hz), 6.57 (1H, s), 6.53 (2H, d, J = 7.6 Hz), 5.17 (1H, s, N-H), 4.37 (2H, q, J = 7.2 Hz,

 OCH_2CH_3), 3.18 (4H, t, J = 6.8 Hz), 2.34 (3H, s, Ar-C H_3), 2.13 (3H, s, Ar-C H_3), 1.80

(4H, m), 1.38 (3H, t, J = 7.2 Hz, OCH₂CH₃); ¹³C NMR (CDCl₃, DEPT-135) δ 170.2 (C, O-C=O), 148.0 (C), 146.5 (C), 134.6 (C), 133.0 (C), 129.0 (2 x CH), 126.0 (C), 125.6 (C), 117.8 (CH), 115.1 (CH), 113.4 (2 x CH), 60.6 (CH₂, OCH₂CH₃), 50.7 (2 x CH₂), 25.3 (2 x CH₂), 20.2 (CH₃, Ar-CH₃), 15.9 (CH₃, Ar-CH₃), 14.3 (CH₃, OCH₂CH₃); LRMS m/z 339.20 (M + H⁺), calcd C₂₁H₂₆N₂O₂ 338.1994; HRMS m/z 339.2085 (M + H⁺), calcd C₂₁H₂₆N₂O₂H⁺ 339.2073; Anal. calcd for C₂₁H₂₆N₂O₂ (338.19): C, 74.52; H, 7.74; N, 8.28. Found: C, 74.551; H, 7.745; N, 8.255%.

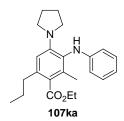
6-Ethyl-2-methyl-3-phenylamino-4-pyrrolidin-1-yl-benzoic acid ethyl ester (107ja):

N H N CO₂Et 107ja

Purified by column chromatography using EtOAc/hexane and isolated as a solid. IR (neat): v_{max} 3389 (N-H), 2968, 1714 (O-C=O), 1599, 1516, 1107, 1053, 748, 694 cm⁻¹; ¹H NMR (CDCl₃) δ 7.16 (2H, t, J = 7.6 Hz), 6.74 (1H, t, J = 7.2 Hz), 6.61 (1H, s), 6.54 (2H, d, J = 7.6 Hz), 5.18 (1H, s, N-H), 4.37 (2H, q, J = 7.2 Hz, OC H_2 CH₃), 3.12

(4H, t, J = 6.8 Hz), 2.64 (2H, q, J = 7.2 Hz, ArC H_2 CH₃), 2.12 (3H, s, Ar-C H_3), 1.80 (4H, m), 1.38 (3H, t, J = 7.2 Hz, OCH₂C H_3), 1.26 (3H, t, J = 7.2 Hz, ArCH₂C H_3); ¹³C NMR (CDCl₃, DEPT-135) δ 170.3 (C, O-C=O), 148.1 (C), 146.5 (C), 139.1 (C), 134.2 (C), 129.0 (2 x CH), 126.3 (C), 125.4 (C), 117.8 (CH), 113.7 (CH), 113.5 (2 x CH), 60.7 (CH₂, OCH₂CH₃), 50.7 (2 x CH₂), 27.2 (CH₂, ArCH₂CH₃), 25.2 (2 x CH₂), 15.9 (CH₃, Ar-CH₃), 15.8 (CH₃, OCH₂C H_3), 14.2 (CH₃, ArCH₂CH₃); LRMS m/z 353.25 (M + H⁺), calcd C₂₂H₂₈N₂O₂ 352.2151; HRMS m/z 375.2042 (M + Na⁺), calcd C₂₂H₂₈N₂O₂Na⁺ 375.2048.

2-Methyl-3-phenylamino-6-propyl-4-pyrrolidin-1-yl-benzoic acid ethyl ester



(107ka): Purified by column chromatography using EtOAc/hexane and isolated as a solid. IR (neat): v_{max} 3381 (N-H), 2957, 1720 (O-C=O), 1604, 1504, 1228, 1049, 746, 692 cm⁻¹; ¹H NMR (CDCl₃) δ 7.16 (2H, t, J = 8.4 Hz), 6.74 (1H, t, J = 7.2 Hz), 6.59 (1H, s), 6.54 (2H, d, J = 8.4 Hz), 5.18 (1H, s, N-H), 4.36 (2H, q, J = 7.2 Hz,

 OCH_2CH_3), 3.18 (4H, t, J = 6.4 Hz), 2.58 (2H, t, J = 8.0 Hz, $ArCH_2CH_2CH_3$), 2.12 (3H, s, $Ar-CH_3$), 1.80 (4H, m), 1.66 (2H, m, $ArCH_2CH_2CH_3$), 1.38 (3H, t, J = 7.2 Hz,

OCH₂CH₃), 0.99 (3H, t, J = 7.2 Hz, ArCH₂CH₂CH₃); ¹³C NMR (CDCl₃, DEPT-135) δ 170.3 (C, O-C=O), 147.9 (C), 146.5 (C), 137.6 (C), 134.2 (C), 129.0 (2 x CH), 126.3 (C), 125.6 (C), 117.8 (CH), 114.4 (CH), 113.5 (2 x CH), 60.6 (CH₂, OCH₂CH₃), 50.7 (2 x CH₂), 36.3 (CH₂, ArCH₂CH₂CH₃), 25.2 (2 x CH₂), 24.7 (CH₂, ArCH₂CH₂CH₃), 15.9 (CH₃, Ar-CH₃), 14.3 (CH₃, OCH₂CH₃), 14.2 (CH₃, ArCH₂CH₂CH₃); LRMS m/z 367.35 (M + H⁺), calcd C₂₃H₃₀N₂O₂ 366.2307; HRMS m/z 389.2200 (M + Na⁺), calcd C₂₃H₃₀N₂O₂Na⁺ 389.2205.

2-Methyl-6-phenethyl-3-phenylamino-4-pyrrolidin-1-yl-benzoic acid ethyl ester

(107la): Purified by column chromatography using EtOAc/hexane and isolated as a solid. Mp 54 °C; IR (neat): v_{max} 3389 (N-H), 2964, 1716 (O-C=O), 1602, 1498, 1053, 748, 698 cm⁻¹; ¹H NMR ĊO₂Et (CDCl₃) δ 7.32 (2H, t, J = 7.6 Hz), 7.23 (5H, m, Ph-H), 6.76 (1H, 107la t, J = 7.2 Hz), 6.56 (2H, d, J = 8.4 Hz), 6.52 (1H, s), 5.20 (1H, s, N-H), 4.40 (2H, q, J =6.8 Hz, OCH_2CH_3), 3.15 (4H, t, J = 6.4 Hz), 2.94 (4H, m), 2.16 (3H, s, Ar-C H_3), 1.80 (4H, m), 1.40 (3H, t, J = 7.2 Hz, OCH₂CH₃); ¹³C NMR (CDCl₃, DEPT-135) δ 170.2 (C, O-C=O), 147.9 (C), 146.4 (C), 141.9 (C), 136.6 (C), 134.4 (C), 129.0 (2 x CH), 128.4 (2 x CH), 128.3 (2 x CH), 126.6 (C), 125.8 (CH), 125.5 (C), 117.9 (CH), 114.7 (CH), 113.5 (2 x CH), 60.8 (CH₂, OCH₂CH₃), 50.7 (2 x CH₂), 38.1 (CH₂), 36.4 (CH₂), 25.2 (2 x CH₂), 16.0 (CH₃, Ar-CH₃), 14.3 (CH₃, OCH₂CH₃); LRMS m/z 427.05 (M – + H⁺), calcd H^+), calcd $C_{28}H_{32}N_2O_2$ 428.2464; HRMS m/z 429.2538 M $C_{28}H_{32}N_2O_2H^+$ 429.2542.

3-Methyl-4-phenylamino-5-pyrrolidin-1-yl-biphenyl-2-carboxylic acid ethyl ester

(107ma): Purified by column chromatography using EtOAc/hexane and isolated as a solid. Mp 118 °C; IR (neat): v_{max} 3335 (N-H), 2970, 1705 (O-C=O), 1601, 1498, 1053, 748, 700 cm⁻¹; ¹H NMR (CDCl₃) δ 7.38 (5H, m, Ph-H), 7.21 (2H, t, J = 8.0 Hz), 6.78 (1H, t, J = 7.2 Hz), 6.74 (1H, s), 6.62 (2H, d, J = 8.4

Hz), 5.31 (1H, s, N-*H*), 4.01 (2H, q, J = 7.2 Hz, OC H_2 CH₃), 3.22 (4H, t, J = 6.4 Hz), 2.22 (3H, s, Ar-C H_3), 1.81 (4H, m), 0.92 (3H, t, J = 6.8 Hz, OC H_2 C H_3); ¹³C NMR

(CDCl₃, DEPT-135) δ 170.0 (C, O-C=O), 147.8 (C), 146.2 (C), 141.9 (C), 138.3 (C), 134.9 (C), 129.1 (2 x CH), 128.2 (2 x CH), 128.0 (2 x CH), 127.7 (C), 127.0 (CH), 125.1 (C), 118.1 (CH), 115.0 (CH), 113.7 (2 x CH), 60.6 (CH₂, O*CH*₂CH₃), 50.8 (2 x CH₂), 25.2 (2 x CH₂), 15.9 (CH₃, Ar-*C*H₃), 13.6 (CH₃, OCH₂C*H*₃); LRMS m/z 401.30 (M + H⁺), calcd C₂₆H₂₈N₂O₂ 400.2151; HRMS m/z 401.2216 (M + H⁺), calcd C₂₆H₂₈N₂O₂H⁺ 401.2229.

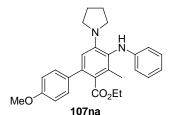
4'-Chloro-3-methyl-4-phenylamino-5-pyrrolidin-1-yl-biphenyl-2-carboxylic acid

CI CO₂Et

ethyl ester (107pa): Purified by column chromatography using EtOAc/hexane and isolated as a solid. Mp 104 °C; IR (neat): v_{max} 3369 (*N-H*), 1718 (O-C=O), 1601, 1496, 748, cm⁻¹; ¹H NMR (CDCl₃) δ 7.36 (4H, m, Ar-H), 7.20 (2H, t, J = 7.6 Hz), 6.78 (1H, t, J = 7.6 Hz), 6.68 (1H, s), 6.60 (2H, d, J

= 8.4 Hz), 5.30 (1H, s, N-H), 4.04 (2H, q, J = 7.2 Hz, OC H_2 CH₃), 3.22 (4H, t, J = 6.4 Hz), 2.20 (3H, s, Ar-C H_3), 1.82 (4H, m), 0.99 (3H, t, J = 7.2 Hz, OCH₂C H_3); ¹³C NMR (CDCl₃, DEPT-135) δ 169.8 (C, O-C=O), 147.9 (C), 146.1 (C), 140.3 (C), 136.9 (C), 135.1(C), 133.0 (C), 129.5 (2 x CH), 129.1 (2 x CH), 128.2 (2 x CH), 127.9 (C), 124.9 (C), 118.2 (CH), 114.8 (CH), 113.7 (2 x CH), 60.7 (CH₂, OC H_2 CH₃), 50.8 (2 x CH₂), 25.3 (2 x CH₂), 15.9 (CH₃, Ar-CH₃), 13.7 (CH₃, OCH₂C H_3); LRMS m/z 433.00 (M – H⁺), calcd C₂₆H₂₇ClN₂O₂ 434.1761; HRMS m/z 435.1821 (M + \square H⁺), calcd C₂₆H₂₇ClN₂O₂H⁺ 435.1839.

4'-Methoxy-3-methyl-4-phenylamino-5-pyrrolidin-1-yl-biphenyl-2-carboxylic acid



ethyl ester (107na): Purified by column chromatography using EtOAc/hexane and isolated as yellow oil. IR (neat): v_{max} 3383 (*N-H*), 2964, 1714 (O-C=O), 1602, 1500, 1141, 1053, 748, 694 cm⁻¹; ¹H NMR (CDCl₃) δ 7.33 (2H, d, J = 8.4

Hz), 7.19 (2H, t, J = 8.0 Hz), 6.92 (2H, d, J = 8.4 Hz), 6.76 (1H, t, J = 7.6 Hz), 6.70 (1H, s), 6.61 (2H, d, J = 7.6 Hz), 5.27 (1H, s, N-H), 4.04 (2H, q, J = 6.8 Hz, OC H_2 CH₃), 3.84 (3H, s, OC H_3), 3.19 (4H, t, J = 6.4 Hz), 2.18 (3H, s, Ar-C H_3), 1.80

(4H, m), 0.98 (3H, t, J = 7.2 Hz, OCH₂CH₃); ¹³C NMR (CDCl₃, DEPT-135) δ 170.2 (C, O-C=O), 158.8 (C), 147.8 (C), 146.3 (C), 137.7 (C), 134.6 (C), 134.3 (C), 129.3 (2 x CH), 129.1 (2 x CH), 127.6 (C), 125.4 (C), 118.1 (CH), 115.1 (CH), 113.7 (2 x CH), 113.5 (2 x CH), 60.6 (CH₂, OCH₂CH₃), 55.3 (CH₃, OCH₃), 50.8 (2 x CH₂), 25.3 (2 x CH₂), 15.9 (CH₃, Ar-CH₃), 13.8 (CH₃, OCH₂CH₃); LRMS m/z 427.25 (M \square -3), calcd C₂₇H₃₀N₂O₃ 430.2256; HRMS m/z 431.2325 (M + \square H[†]), calcd C₂₇H₃₀N₂O₃H[†] 431.2334; Anal. calcd for C₂₇H₃₀N₂O₃ (430.22): C, 75.32; H, 7.02; N, 6.51. Found: C, 75.376; H, 7.006; N, 6.548%.

3-methyl-4'-nitro-4-phenylamino-5-pyrrolidin-1-yl-biphenyl-2-carboxylic acid

ethyl ester (107qa): Purified by column chromatography using EtOAc/hexane and isolated as a yellow liquid. IR (neat): v_{max} 3387 (*N-H*), 2966, 1716 (O-C=O), 1602, 1500, 1105, 1080, 748, 694 cm⁻¹; ¹H NMR (CDCl₃) δ 8.25 (2H, d, J = 8.8 Hz), 7.56 (2H, d, J = 8.4 Hz), 7.20 (2H, t, J = 8.0 Hz),

6.79 (1H, t, J = 8.4 Hz), 6.66 (1H, s), 6.61 (2H, d, J = 8.0 Hz), 5.32 (1H, s, N-H), 4.02 (2H, q, J = 7.2 Hz, OC H_2 CH₃), 3.23 (4H, m), 2.21 (3H, s, Ar-C H_3), 1.82 (4H, m), 0.97 (3H, t, J = 7.2 Hz, OCH₂C H_3); ¹³C NMR (CDCl₃, DEPT-135) δ 169.4 (C, O-C=O), 148.9 (C), 148.0 (C), 146.8 (C), 145.8 (C), 136.0 (C), 135.6 (C), 129.1 (2 x CH), 129.0 (2 x CH), 128.6 (C), 124.6 (C), 123.4 (2 x CH), 118.5 (CH), 114.5 (CH), 113.8 (2 x CH), 60.9 (CH₂, OC H_2 CH₃), 50.8 (2 x CH₂), 25.3 (2 x CH₂), 16.0 (CH₃, Ar-CH₃), 13.7 (CH₃, OCH₂C H_3); LRMS m/z 446.20 (M + H⁺), calcd C₂₆H₂₇N₃O₄ 445.2002; HRMS m/z 446.2085 (M + H⁺), calcd C₂₆H₂₇N₃O₄H⁺ 446.2080; Anal. calcd for C₂₆H₂₇N₃O₄ (445.20): C, 70.09; H, 6.11; N, 9.43. Found: C, 70.036; H, 6.134; N, 9.381%.

6-Benzo[1,3]dioxol-5-yl-2-methyl-3-phenylamino-4-pyrrolidin-1-yl-benzoic acid

ethyl ester (1070a): Purified by column chromatography using EtOAc/hexane and isolated as a solid. IR (neat): ν_{max}

192

CO₂Et **107oa**

chromatography

using

3362 (*N-H*), 2964, 1712 (O-C=O), 1601, 1577, 1502, 748 cm⁻¹; 1 H NMR (CDCl₃) δ 7.21 (2H, t, J = 8.0 Hz), 6.94 (1H, d, J = 1.6 Hz), 6.90-6.81 (2H, m), 6.79 (1H, t, J = 8.0Hz), 6.71 (1H, br s), 6.62 (2H, d, J = 8.0 Hz), 6.00 (2H, s, OC H_2 O), 5.29 (1H, s, N-H), 4.11 (2H, q, J = 7.2 Hz, OC H_2 CH₃), 3.21 (4H, m), 2.20 (3H, s, Ar-C H_3), 1.83 (4H, m), 1.08 (3H, t, J = 7.2 Hz, OCH₂CH₃); ¹³C NMR (CDCl₃, DEPT-135) δ 170.1 (C, O-C=O), 147.8 (C), 147.4 (C), 146.8 (C), 146.2 (C), 137.7 (C), 135.8 (C), 134.7 (C), 129.1 (2 x CH), 127.7 (C), 125.4 (C), 121.7 (CH), 118.2 (CH), 115.0 (CH), 113.8 (2 x CH), 109.0 (CH), 108.1 (CH), 101.0 (CH₂, OCH₂O), 60.7 (CH₂, OCH₂CH₃), 50.9 (2 x CH₂), 25.3 (2 x CH₂), 15.9 (CH₃, Ar-CH₃), 13.9 (CH₃, OCH₂CH₃); GCMS m/z 443.95 (M), calcd $C_{27}H_{28}N_2O_4$ 444.21; Anal. calcd for $C_{27}H_{28}N_2O_4$ (444.2049): C, 72.95; H, 6.35; N, 6.30; O, 14.40. Found: C, 72.996; H, 6.398; N, 6.382%.

6-Furan-2-yl-2-methyl-3-phenylamino-4-pyrrolidin-1-yl-benzoic acid ethyl ester

(107ra): Purified by column EtOAc/hexane and isolated as a solid. IR (neat): v_{max} 3383 (N-H), 3051, 2972, 1716 (O-C=O), 1602, 1500, 1053, 736, 696 cm⁻¹; ¹H CO₂Et NMR (CDCl₃) δ 7.45 (1H, br s), 7.17 (2H, t, J = 7.2 Hz), 7.02 107ra

(1H, s), 6.77 (1H, t, J = 7.2 Hz), 6.58 (2H, d, J = 7.6 Hz), 6.50 (1H, d, J = 3.2 Hz), 6.44 (1H, m), 5.33 (1H, s, N-H), 4.29 (2H, q, J = 7.2 Hz, OCH_2CH_3), 3.18 (4H, m), 2.14 (3H, s, Ar-CH₃), 1.81 (4H, m), 1.25 (3H, t, J = 7.2 Hz, OCH_2CH_3); ¹³C NMR (CDCl₃, DEPT-135) δ 170.3 (C, O-C=O), 153.0 (C), 147.7 (C), 146.0 (C), 142.1 (CH), 134.1 (C), 129.1 (2 x CH), 129.0 (C), 125.8 (C), 123.9 (C), 118.5 (CH), 114.0 (2 x CH), 112.4 (CH), 111.5 (CH), 106.8 (CH), 61.1 (CH₂, OCH_2CH_3), 50.9 (2 x CH_2), 25.2 (2 x CH_2), 15.7 (CH_3 , $Ar-CH_3$), 14.1 (CH_3 , OCH₂CH₃); GCMS m/z 389.95 (M), calcd C₂₄H₂₆N₂O₃ 390.19; Anal. calcd for C₂₄H₂₆N₂O₃ (390.1943): C, 73.82; H, 6.71; N, 7.17; O, 12.29. Found: C, 73.715; H, 6.762; N, 7.041%.

4c: Piperidine-catalyzed combination of cascade Knoevenagel/Michael/aldol condenasation/decarboxylation and cascade enamine amination/iso-aromatization reactions in one-pot: To a stirred solution of ethyl acetoacetate (0.6 mmol) and aldehydes (0.3 mmol) in EtOH (1 mL) was added a catalytic amount of piperidine (0.1 mmol, 35 mol%) and the reaction mixture was stirred at 80 °C for 3 h. Solvent ethanol and piperidine was evaporated by vacuum pump; then catalyst piperidine (0.06 mmol, 20 mol%), nitrosobenzene **56a** (0.3 mmol) and solvent DMSO (1 mL) was added and the reaction mixture was stirred at 25 °C for 1-2 h. The crude reaction mixture was worked up with aqueous NH₄Cl solution and the aqueous layer was extracted with dichloromethane (2 x 20 mL). The combined organic layers were dried (Na₂SO₄), filtered and concentrated. Pure one-pot products **106** were obtained by column chromatography (silica gel, mixture of hexane/ethyl acetate).

4d: Pyrrolidine/Cs₂CO₃-catalyzed three-component enamine amination/iso-aromatization/alkylation reactions in one-pot: In an ordinary glass vial equipped with a magnetic stirring bar, to 0.6 mmol of the Hagemann's esters 88 was added 1.0 mL of solvent, and then the catalyst pyrrolidine (0.015 mmol, 2.5 μL) was added and the reaction mixture was stirred at 25 °C for the 0.5 h; then 0.3 mmol of nitrosobenzene 56a was added in one-portion and the reaction mixture was stirred at 25 °C for the time indicated in Table 16. To the reaction mixture, alkyl halide (0.39 mmol) and Cs₂CO₃ (0.45 mmol) was added and stirring continued at RT for 24 h. The crude reaction mixture was worked up with aqueous NH₄Cl solution and the aqueous layer was extracted with dichloromethane (2 x 20 mL). The combined organic layers were dried (Na₂SO₄), filtered and concentrated. Pure one-pot products 109 were obtained by column chromatography (silica gel, mixture of hexane/ethyl acetate).

4-Allyloxy-2-methyl-3-phenylamino-benzoic acid ethyl ester (109aa): Purified by column chromatography using EtOAc/hexane and isolated as a light yellow liquid. IR (neat): v_{max} 3387 (N-*H*), 2928, 1709 (O-C=O), 1601, 1498, 1267, 1055, 748, 694 cm⁻¹; ¹H NMR (CDCl₃) δ 7.77 (109aa) (1H, d, J = 8.8 Hz), 7.18 (2H, t, J = 8.4 Hz), 6.82 (1H, d, J = 8.8 Hz), 6.81 (1H, t, J = 8.8 Hz), 6.62 (2H, d, J = 7.6 Hz), 5.95 (1H, m, olefinic-*H*), 5.72 (1H, s,

N-*H*), 5.28 (1H, dd, J = 17.6 Hz, 4.0 Hz), 5.23 (1H, dd, J = 17.6 Hz, 4.0 Hz), 4.56 (2H, m), 4.36 (2H, q, J = 7.2 Hz, OC H_2 CH₃), 2.42 (3H, s, Ar-C H_3), 1.39 (3H, t, J = 6.8 Hz, OC H_2 CH₃); ¹³C NMR (CDCl₃, DEPT-135) δ 167.6 (C, O-C=O), 155.5 (C), 145.9 (C), 137.5 (C), 132.5 (CH), 130.6 (C), 129.0 (2 x CH), 128.0 (CH), 123.9 (C), 119.4 (CH), 117.8 (CH₂, CH=CH₂), 115.2 (2 x CH), 108.9 (CH), 69.1 (CH₂, OC H_2 CH=CH₂), 60.6 (CH₂, OC H_2 CH₃), 16.5 (CH₃, Ar-CH₃), 14.4 (CH₃, OC H_2 CH₃); LRMS m/z 312.20 (M + H⁺), calcd C₁₉H₂₁NO₃ 311.1524; HRMS m/z 334.1406 (M + Na⁺), calcd C₁₉H₂₁NO₃Na⁺ 334.1419.

2-methyl-3-phenylamino-4-prop-2-ynyloxy-benzoic acid ethyl ester (109ab):

Purified by column chromatography using EtOAc/hexane and isolated as a light yellow solid. Mp 60 °C; IR (neat): v_{max} 3381 (N-*H*), 3292 (C=C-*H*), 1714 (O-C=O), 1601, 1504, 1265, 1070, 777, 750 cm⁻¹; ¹H NMR (CDCl₃) δ 7.80 (1H, d, J = 8.8 Hz), 7.18 (2H, t, J = 8.0 Hz), 6.96 (1H, d, J = 8.4 Hz), 6.83 (1H, t, J = 7.2 Hz), 6.62 (2H, d, J = 7.6 Hz), 5.70 (1H, s, N-*H*), 4.72 (2H, d, J = 2.4 Hz), 4.36 (2H, q, J = 6.8 Hz, OC H_2 CH₃), 2.46 (1H, t, J = 2.4 Hz, OC H_2 C=C*H*), 2.42 (3H, s, Ar-C H_3), 1.39 (3H, t, J = 7.2 Hz, OC H_2 C H_3); ¹³C NMR (CDCl₃, DEPT-135) δ 167.5 (C, O-C=O), 154.5 (C), 145.6 (C), 137.7 (C), 130.7 (C), 129.0 (2 x CH), 127.8 (CH), 124.8 (C), 119.4 (CH), 115.1 (2 x CH), 109.4 (CH), 77.8 (CH), 76.2 (CH), C=CH), 60.6 (CH₂, OCH₃CH₃), 56.3 (CH₃, OCH₃C=CH), 16.4

(CDC1₃, DEP1-135) 8 167.5 (C, O-C=O), 154.5 (C), 145.6 (C), 137.7 (C), 130.7 (C), 129.0 (2 x CH), 127.8 (CH), 124.8 (C), 119.4 (CH), 115.1 (2 x CH), 109.4 (CH), 77.8 (C, C=CH), 76.2 (CH, C=CH), 60.6 (CH₂, OCH₂CH₃), 56.3 (CH₂, OCH₂C=CH), 16.4 (CH₃, Ar-CH₃), 14.3 (CH₃, OCH₂CH₃); LRMS m/z 310.10 (M + H⁺), calcd C₁₉H₁₉NO₃ 309.1365; HRMS m/z 332.1268 (M + Na⁺), calcd C₁₉H₁₉NO₃Na⁺ 332.1263.

4-Allyloxy-2,6-dimethyl-3-phenylamino-benzoic acid ethyl ester (109ia): Purified by

column chromatography using EtOAc/hexane and isolated as a light yellow solid. Mp 42 °C; IR (neat): v_{max} 3391 (N-*H*), 2928, 1720 (O-C₂Et C=O), 1602, 1498, 1265, 1053, 748, 694 cm⁻¹; ¹H NMR (CDCl₃) δ 7.16 (2H, t, J = 7.6 Hz), 6.79 (1H, d, J = 7.6 Hz), 6.62 (1H, s), 6.60 (2H, d, J = 7.6 Hz), 5.92 (1H, m, OCH₂C*H*=CH₂), 5.56 (1H, s, N-*H*), 5.25 (1H, br d, J

= 17.6 Hz), 5.19 (1H, br d, J = 17.6 Hz) [OCH₂CH=CH₂]; 4.49 (2H, m), 4.38 (2H, q, J = 7.2 Hz, OCH₂CH₃), 2.33 (3H, s, Ar-CH₃), 2.15 (3H, s, Ar-CH₃), 1.38 (3H, t, J = 7.2 Hz, OCH₂CH₃); ¹³C NMR (CDCl₃, DEPT-135) δ 169.8 (C, O-C=O), 153.4 (C), 146.2 (C), 132.8 (CH), 132.2 (C), 128.9 (2 x CH), 128.8 (C), 127.9 (C), 127.7 (C), 119.0 (CH), 117.4 (CH₂, CH=CH₂), 114.9 (2 x CH), 111.9 (CH), 69.2 (CH₂, OCH₂CH=CH₂), 60.9 (CH₂, OCH₂CH₃), 20.0 (CH₃, Ar-CH₃), 15.6 (CH₃, Ar-CH₃), 14.3 (CH₃, OCH₂CH₃); LRMS m/z 326.25 (M + H⁺), calcd C₂₀H₂₃NO₃ 325.1680; HRMS m/z 326.1769 (M + H⁺), calcd C₂₀H₂₃NO₃H⁺ 326.1756.

5a: Amino acid-catalyzed cascade [3+2]-CA/H reactions: In an ordinary glass vial equipped with a magnetic stirring bar, to 1.0 mmol of the Hagemann's esters **88** was added 1.0 mL of solvent, and then the catalyst proline **14a** (0.1 mmol, 20-mol%) was added and the reaction mixture was stirred at 25 °C for the 0.25 h; then 0.5 mmol of TsN₃ **118a** was added in one-portion and the reaction mixture was stirred at 25 °C for the time indicated in Tables 18 and 20. The crude reaction mixture was worked up with aqueous NH₄Cl solution and the aqueous layer was extracted with dichloromethane (2 x 20 mL). The combined organic layers were dried (Na₂SO₄), filtered and concentrated. Pure cascade products **120/121** were obtained by column chromatography (silica gel, mixture of hexane/ethyl acetate).

4-Methyl-6,7-dihydro-1H-benzotriazole-5-carboxylic acid ethyl ester (121aa): HN-N Purified by column chromatography using EtOAc/hexane and isolated as a white solid. Mp 80 °C; IR (neat): ν_{max} 3183 (N-*H*), 3054, 2978, 2903, 1703 co₂Et (O-C=O), 1609, 1530, 1368, 1294, 1055, 866 cm⁻¹; ¹H NMR (CDCl₃) δ 121aa 13.25 (1H, s, N-*H*), 4.29 (2H, q, *J* = 7.2 Hz, OC*H*₂CH₃), 2.96-2.92 (2H, m), 2.86-2.81 (2H, m), 2.52 (3H, br s, olefinic-C*H*₃), 1.36 (3H, t, *J* = 7.2 Hz, OCH₂C*H*₃); ¹³C NMR (CDCl₃, DEPT-135) δ 168.2 (C, O-C=O), 145.2 (C), 144.1 (C), 136.2 (C), 125.5 (C), 60.6 (CH₂, OCH₂CH₃), 25.6 (CH₂), 19.2 (CH₂), 15.4 (CH₃, olefinic-*C*H₃), 14.3 (CH₃, OCH₂CH₃); LRMS m/z 208.05 (M + H⁺), calcd C₁₀H₁₃N₃O₂ 207.1008;

Anal. calcd for $C_{10}H_{13}N_3O_2$ (207.10): C, 57.96; H, 6.32; N, 20.28. Found: C, 57.945; H, 6.325; N, 20.356%.

4-Methyl-6,7-dihydro-1H-benzotriazole-5-carboxylic acid methyl ester

(121sa): Purified by column chromatography using EtOAc/hexane and isolated as a white solid. Mp 88 °C; IR (neat): ν_{max} 3176 (N-*H*), 2952, 1709

(O-C=O), 1692, 1679, 1607, 1433, 1294, 1232, 1212, 1056 cm⁻¹; ¹H NMR

(CDCl₃) δ 13.58 (1H, s, N-*H*), 3.83 (3H, s, OC*H*₃), 2.96-2.92 (2H, m), 2.85-2.81 (2H, m), 2.52 (3H, br s, olefinic-C*H*₃); ¹³C NMR (CDCl₃, DEPT-135) δ 168.6 (C, O-C=O), 145.2 (C), 144.2 (C), 136.8 (C), 125.1 (C), 51.7 (CH₃, OCH₃), 25.6 (CH₂), 19.2 (CH₂), 15.5 (CH₃, olefinic-CH₃); LRMS m/z 194.00 (M + H⁺), calcd C₉H₁₁N₃O₂ 193.0851; HRMS m/z 216.0745 (M + Na⁺), calcd C₉H₁₁N₃O₂Na⁺ 216.0749; Anal. calcd for C₉H₁₁N₃O₂ (193.08): C, 55.95; H, 5.74; N, 21.75. Found: C, 55.925; H, 5.802; N, 20.872%.

4-Methyl-6,7-dihydro-1H-benzotriazole-5-carboxylic acid tert-butyl ester (121ba):

Purified by column chromatography using EtOAc/hexane and isolated as a white solid. Mp 68 °C; IR (neat): v_{max} 3180 (N-*H*), 2975, 1692 (O-C=O), 1453, 1367, 1347, 1299, 1241, 1160, 1053 cm⁻¹; ¹H NMR (CDCl₃) δ 12.97 (1H, s, N-*H*), 2.92 (2H, br t, J = 8.0 Hz), 2.78 (2H, br t, J = 8.0 Hz), 2.46 (3H, br s, olefinic-C*H*₃), 1.56 (9H, s, 3 x C*H*₃, O-C(*CH*₃)₃); ¹³C NMR (CDCl₃, DEPT-135) δ 167.7 (C, O-C=O), 145.4 (C), 144.2 (C), 134.3 (C), 127.4 (C), 81.2 (C, O-C(CH₃)₃), 28.3 (3 x CH₃, O-C(*CH*₃)₃), 25.9 (CH₂), 19.3 (CH₂), 15.3 (CH₃, olefinic-*C*H₃); LRMS m/z 236.00 (M + H⁺), calcd C₁₂H₁₇N₃O₂ 235.1321; Anal. calcd for C₁₂H₁₇N₃O₂ (235.13): C, 61.26; H, 7.28; N, 17.86. Found: C, 61.268; H, 7.252; N, 17.683%.

4,6-Dimethyl-6,7-dihydro-1H-benzotriazole-5-carboxylic acid ethyl ester (121ia):

Purified by column chromatography using EtOAc/hexane and isolated as a liquid. IR (neat): v_{max} 3185 (N-*H*), 2974, 1703 (O-C=O), 1692, 1678, 1666, CO₂Et 1295, 1197, 1045 cm⁻¹; ¹H NMR (CDCl₃) δ 13.25 (1H, s, N-*H*), 4.30 (2H, 121ia

q, J = 7.2 Hz, OC H_2 CH₃), 3.29-3.25 (1H, m), 3.03 (1H, dd, J = 16.0, 7.2 Hz), 2.83 (1H, dd, J = 16.0, 1.6 Hz), 2.50 (3H, s, olefinic-C H_3), 1.36 (3H, t, J = 7.2 Hz, OCH₂C H_3), 1.06 (3H, d, J = 7.2 Hz, CHC H_3); ¹³C NMR (CDCl₃, DEPT-135) δ 168.1 (C, O-C=O), 144.5 (C), 143.0 (C), 134.9 (C), 131.4 (C), 60.6 (CH₂, OCH₂CH₃), 30.9 (CH), 26.8 (CH₂), 20.0 (CH₃, CHCH₃), 15.3 (CH₃, olefinic-CH₃), 14.3 (CH₃, OCH₂CH₃); LRMS m/z 222.05 (M + H⁺), calcd C₁₁H₁₅N₃O₂ 221.1164; Anal. calcd for C₁₁H₁₅N₃O₂ (221.11): C, 59.71; H, 6.83; N, 18.99. Found: C, 59.629; H, 6.874; N, 18.667%.

4-Methyl-6-phenethyl-6,7-dihydro-1H-benzotriazole-5-carboxylic acid ethyl ester

HN-N (1211a): Purified by column chromatography using EtOAc/hexane and Ph isolated as a colorless liquid. IR (neat): v_{max} 3193 (N-H), 2910, 1698 (O-C=O), 1305, 1224, 1053, 980, 698 cm⁻¹; ¹H NMR (CDCl₃) δ 12.82 (1H, 121la s. N-H), 7.26-7.21 (2H, m), 7.16-7.11 (3H, m), 4.25 (2H, q, J = 7.2 Hz, OCH_2CH_3), 3.22-3.20 (1H, m), 3.10 (1H, dd, J = 16.4, 1.6 Hz), 2.96 (1H, dd, J = 16.4, 6.8 Hz), 2.71-2.66 (1H, m), 2.58-2.52 (1H, m), 2.51 (3H, s, olefinic-CH₃), 1.80-1.77 (1H, m), 1.61-1.58 (1H, m), 1.31 (3H, t, J = 7.2 Hz, OCH₂CH₃); ¹³C NMR (CDCl₃, DEPT-135) δ 168.0 (C, O-C=O), 144.9 (C), 143.3 (C), 141.6 (C), 135.6 (C), 130.6 (C), 128.3 (4 x CH), 125.8 (CH), 60.6 (CH₂, OCH₂CH₃), 35.3 (CH), 34.8 (CH₂), 33.1 (CH₂), 23.4 (CH₂), 15.7 (CH₃, olefinic-CH₃), 14.2 (CH₃, OCH₂CH₃); LRMS m/z $312.05 \text{ (M} + \text{H}^{+})$, calcd $C_{18}H_{21}N_{3}O_{2}$ 311.1634; HRMS m/z 312.1697 (M + H⁺), calcd $C_{18}H_{21}N_3O_2H^+$ 312.1712; Anal. calcd for $C_{18}H_{21}N_3O_2$ (311.16): C, 69.43; H, 6.80; N, 13.49. Found: C, 69.490; H, 6.832; N, 13.869%.

4-Methyl-6-phenyl-6,7-dihydro-1H-benzotriazole-5-carboxylic acid ethyl ester

(121ma): Purified by column chromatography using EtOAc/hexane and isolated as a gummy liquid; IR (neat): v_{max} 3183 (N-H), 2979, 1697 (O-C=O), 1307, 1269, 1225, 1053, 756, 699 cm⁻¹; ¹H NMR (CDCl₃) δ 12.91 (1H, s, N-H), 7.20-7.14 (3H, m), 7.07 (2H, d, J = 6.8 Hz), 4.46 (1H, br d, J = 7.6 Hz), 4.13 (2H, q, J = 7.2 Hz, OCH₂CH₃), 3.35 (1H, dd, J = 16.4, 8.4 Hz), 3.12 (1H, dd, J = 16.4, 2.8 Hz), 2.61 (3H, s, olefinic-CH₃), 1.17 (3H, t, J = 7.2 Hz,

OCH₂CH₃); ¹³C NMR (CDCl₃, DEPT-135) δ 167.7 (C, O-C=O), 145.2 (C), 142.8 (C), 136.4 (C), 128.5 (2 x CH), 127.1 (2 x CH), 126.7 (CH), 60.6 (CH₂, OCH₂CH₃), 41.6 (CH), 28.3 (CH₂), 15.6 (CH₃, olefinic-CH₃), 14.0 (CH₃, OCH₂CH₃) [2 quaternary carbons in the triazole ring are poor resolution even after more scans, may be due to the resonance]; LRMS m/z 284.00 (M + H⁺), calcd C₁₆H₁₇N₃O₂ 283.1321; HRMS m/z 284.1364 (M \square + H⁺) and 306.1220 (M \square + Na⁺), calcd C₁₆H₁₇N₃O₂H⁺ 284.1399 and C₁₆H₁₇N₃O₂Na⁺ 306.1219; Anal. calcd for C₁₆H₁₇N₃O₂ (283.13): C, 67.83; H, 6.05; N, 14.83. Found: C, 67.836; H, 6.080; N, 14.654%.

6-Furan-2-yl-4-methyl-6,7-dihydro-1H-benzotriazole-5-carboxylic acid ethyl ester

(121ra): Purified by column chromatography using EtOAc/hexane and isolated as a yellow liquid; IR (neat): v_{max} 3179 (N-H), 2979, 2904, 1703 (O-C=O), 1692, 1678, 1291, 1223, 1170, 1052, 736 cm⁻¹; ¹H NMR (CDCl₃) δ 12.78 (1H, s, N-H), 7.22 (1H, s), 6.12 (1H, d, J = 1.6 Hz), 5.84 (1H, d, J = 2.8 Hz), 4.60 (1H, br d, J = 7.2 Hz), 4.26 (2H, q, J = 7.2 Hz, OC H_2 CH₃), 3.38 (1H, br d, J = 16.4 Hz), 3.16 (1H, dd, J = 16.4, 7.2 Hz), 2.60 (3H, s, olefinic-C H_3), 1.30 (3H, t, J = 7.2 Hz, OCH₂C H_3); ¹³C NMR (CDCl₃, DEPT-135) δ 167.4 (C, O-C=O), 155.2 (C), 144.9 (C), 142.7 (C), 141.6 (CH), 138.0 (C), 125.9 (C), 109.9 (CH), 105.8 (CH), 60.8 (CH₂, OCH₂CH₃), 35.2 (CH), 24.9 (CH₂), 15.8 (CH₃, olefinic-CH₃), 14.2 (CH₃, OCH₂CH₃); LRMS m/z 274.05 (M + H⁺), calcd C₁₄H₁₅N₃O₃ 273.1113; Anal. calcd for C₁₄H₁₅N₃O₃ (273.11): C, 61.53; H, 5.53; N, 15.38. Found: C, 61.438; H, 5.583; N, 15.398%.

6-Benzo[1,3]dioxol-5-yl-4-methyl-6,7-dihydro-1H-benzotriazole-5-carboxylic acid

ethyl ester (1210a): Purified by column chromatography using EtOAc/hexane and isolated as a yellow gummy liquid; IR (neat): v_{max} 3241 (N-H), 2978, 2900, 1703 (O-C=O), 1485, 1441, 1368, 1290, 1228, 1122, 1095, 1039, 974, 811 cm⁻¹; ¹H NMR (CDCl₃) δ 12.87 (1H, s, N-H), 6.62 (1H, d, J = 8.0 Hz), 6.56-6.54 (2H, m), 5.86 (2H, d, J = 1.6 Hz, OCH₂O), 4.40 (1H, br d, J = 7.6 Hz, ArCH), 4.16 (2H, q, J = 7.2 Hz, OCH₂CH₃), 3.32

(1H, dd, J = 16.4, 8.4 Hz), 3.09 (1H, dd, J = 16.4, 2.8 Hz), 2.60 (3H, s, olefinic-C H_3), 1.23 (3H, t, J = 7.2 Hz, OCH₂C H_3); ¹³C NMR (CDCl₃, DEPT-135) δ 167.6 (C, O-C=O), 147.6 (C), 146.3 (C), 145.1 (C), 142.3 (C), 136.7 (C), 136.4 (C), 128.6 (C), 120.2 (CH), 108.2 (CH), 107.6 (CH), 100.8 (CH₂, OCH₂O), 60.7 (CH₂, OCH₂CH₃), 41.2 (CH), 28.4 (CH₂), 15.7 (CH₃, olefinic-CH₃), 14.1 (CH₃, OCH₂CH₃); LRMS m/z 328.00 (M + H⁺), calcd C₁₇H₁₇N₃O₄ 327.1219; HRMS m/z 328.1293 (M \square + H⁺), calcd C₁₇H₁₇N₃O₄H⁺ 328.1297; Anal. calcd for C₁₇H₁₇N₃O₄ (327.12): C, 62.38; H, 5.23; N, 12.84. Found: C, 62.412; H, 5.225; N, 12.872%.

6-(4-Methoxy-phenyl)-4-methyl-6,7-dihydro-1H-benzotriazole-5-carboxylic acid

ethyl ester (121na): Purified by column chromatography using HN-N EtOAc/hexane and isolated as a liquid; IR (neat): v_{max} 3190 (N-H), 2977, 1697 (O-C=O), 1606, 1510, 1462, 1303, 1225, 1179, ĊO₂Et MeO 1034, 832, 735 cm⁻¹; ¹H NMR (CDCl₃) δ 12.94 (1H, s, N-H), 6.99 121na (2H, d, J = 8.4 Hz), 6.72 (2H, d, J = 8.4 Hz), 4.42 (1H, br d, J = 6.8 Hz), 4.14 (2H, q, J) $= 7.2 \text{ Hz}, \text{ OC}H_2\text{CH}_3$), 3.72 (3H, s, OC H_3), 3.32 (1H, dd, J = 16.4, 8.4 Hz), 3.09 (1H, dd, J = 16.4, 2.4 Hz), 2.60 (3H, s, olefinic-C H_3), 1.20 (3H, t, J = 7.2 Hz, OCH₂C H_3); ¹³C NMR (CDCl₃, DEPT-135) δ 167.8 (C, O-C=O), 158.3 (C), 145.2 (C), 142.2 (C), 136.0 (C), 134.9 (C), 128.9 (C), 128.1 (2 x CH), 113.9 (2 x CH), 60.6 (CH₂, OCH₂CH₃), 55.1 (CH₃, OCH₃), 40.8 (CH), 28.4 (CH₂), 15.6 (CH₃, olefinic-CH₃), 14.1 (CH_3, OCH_2CH_3) ; LRMS m/z 314.00 $(M + H^+)$, calcd $C_{17}H_{19}N_3O_3$ 313.1426; HRMS m/z 336.1276 (M \square + Na⁺) and 314.1465 (M + H⁺), calcd $C_{17}H_{19}N_3O_3Na^+$ 336.1324 and $C_{17}H_{19}N_3O_3H^+$ 314.1505; Anal. calcd for $C_{17}H_{19}N_3O_3$ (313.14): C, 65.16; H, 6.11; N, 13.41. Found: C, 65.140; H, 6.142; N, 13.631%.

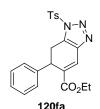
4-Methyl-6-(4-nitro-phenyl)-6,7-dihydro-1H-benzotriazole-5-carboxylic acid ethyl

ester (121qa): Purified by column chromatography using EtOAc/hexane and isolated as a yellow liquid; IR (neat): v_{max} 3201 co_2N (N-H), 2978, 1696 (O-C=O), 1601, 1519, 1463, 1346, 1312, 1268, 121qa

1226, 1053, 1017, 856 cm⁻¹; ¹H NMR (CDCl₃) δ 12.61 (1H, s, N-*H*), 8.06 (2H, d, J = 8.8 Hz), 7.24 (2H, d, J = 8.8 Hz), 4.58 (1H, br d, J = 7.6 Hz), 4.16 (2H, q, J = 7.2 Hz, OC H_2 CH₃), 3.41 (1H, dd, J = 16.4, 8.4 Hz), 3.11 (1H, dd, J = 16.8, 2.4 Hz), 2.66 (3H, s, olefinic-C H_3), 1.21 (3H, t, J = 7.2 Hz, OCH₂C H_3); ¹³C NMR (CDCl₃, DEPT-135) δ 167.2 (C, O-C=O), 150.4 (C), 146.9 (C), 145.1 (C), 142.0 (C), 138.4 (C), 128.0 (2 x CH), 127.0 (C), 123.9 (2 x CH), 60.9 (CH₂, OCH₂CH₃), 41.4 (CH), 27.9 (CH₂), 15.8 (CH₃, olefinic-CH₃), 14.1 (CH₃, OCH₂CH₃); LRMS m/z 329.00 (M + H⁺), calcd C₁₆H₁₆N₄O₄ 328.1172; HRMS m/z 329.1258 (M \square + H⁺), calcd C₁₆H₁₆N₄O₄H⁺ 329.1250; Anal. calcd for C₁₆H₁₆N₄O₄ (328.11): C, 58.53; H, 4.91; N, 17.06. Found: C, 58.411; H, 4.914; N, 16.889%.

4,5-Dihydro-3H-naphtho[1,2-*d*][1,2,3]triazole (121ua): Purified by column chromatography using EtOAc/hexane and isolated as a solid. Mp 112 °C; IR (neat): v_{max} 3037 (N-*H*), 2941, 2915, 2847, 2740, 2581, 1187, 1009, 768 cm⁻¹; ¹H NMR (CDCl₃) δ 12.86 (1H, s, N-*H*), 7.87 (1H, d, J = 7.6 Hz), 7.34-7.26 (3H, m), 3.12-3.03 (4H, m); ¹³C NMR (CDCl₃, DEPT-135) δ 144.0 (C), 143.1 (C), 136.1 (C), 128.6 (CH), 128.4 (CH), 127.4 (C), 127.2 (CH), 123.0 (CH), 29.1 (CH₂), 20.0 (CH₂); LRMS m/z 172.00 (M + H⁺), calcd C₁₀H₉N₃ 171.0796; Anal. calcd for C₁₀H₉N₃ (171.07): C, 70.16; H, 5.30; N, 24.54. Found: C, 70.163; H, 5.296; N, 24.481%.

4-Methyl-1-(toluene-4-sulfonyl)-6,7-dihydro-1H-benzotriazole-5-carboxylic acid Ts_{N-N} ethyl ester (120aa): Purified by column chromatography using EtOAc/hexane and isolated as a white solid. Mp 106 °C; IR (neat): v_{max} 1703 (O-C=O), 1610, 1385, 1278, 1207, 1192, 1054, 808 cm⁻¹; ¹H NMR (CDCl₃) δ CO₂Et 120aa 7.98 (2H, d, J = 8.0 Hz), 7.39 (2H, d, J = 7.6 Hz), 4.25 (2H, q, J = 7.2 Hz, OCH₂CH₃), 3.16 (2H, br t, J = 8.8 Hz), 2.84 (2H, br t, J = 8.4 Hz), 2.51 (3H, s, olefinic-CH₃), 2.45 (3H, s, Ar-CH₃), 1.33 (3H, t, J = 7.2 Hz, OCH₂CH₃); ¹³C NMR (CDCl₃, DEPT-135) δ 167.5 (C, O-C=O), 147.3 (C), 145.3 (C), 136.9 (C), 135.1 (C), 133.5 (C),



130.5 (2 x CH), 128.5 (2 x CH), 122.4 (C), 60.6 (CH₂, O*C*H₂CH₃), 25.0 (CH₂), 21.8 (CH₃, Ar-*C*H₃), 19.7 (CH₂), 15.1 (CH₃, olefinic-*C*H₃), 14.3 (CH₃, OCH₂*C*H₃); LRMS m/z 362.00 (M + H⁺), calcd C₁₇H₁₉N₃O₄S 361.1096; Anal. calcd for C₁₇H₁₉N₃O₄S (361.10): C, 56.49; H, 5.30; N, 11.63. Found: C, 56.450; H, 5.294; N, 11.612%.

6-Phenyl-1-(toluene-4-sulfonyl)-6,7-dihydro-1H-benzotriazole-5-carboxylic acid ethyl ester (120fa): Purified by column chromatography using EtOAc/hexane and isolated as a yellow liquid; IR (neat): v_{max} 2924, 1712 (O-C=O), 1371, 1297, 1229, 1161, 1121, 1030, 1008, 816 cm⁻¹; ¹H NMR (CDCl₃) δ 7.87 (1H, s, olefinic-*H*), 7.63 (2H, d, J = 8.0 Hz), 7.21-7.13 (5H, m, Ph-*H*), 7.02 (2H, d, J = 7.2 Hz), 4.46 (1H, br d, J = 8.8 Hz, PhC*H*), 4.17 (2H, m, OC*H*₂CH₃), 3.66 (1H, br d, J = 18.0 Hz), 3.54 (1H, br dd, J = 18.0, 9.2 Hz), 2.40 (3H, s, Ar-C*H*₃), 1.24 (3H, t, J = 7.2 Hz, OCH₂C*H*₃); ¹³C NMR (CDCl₃, DEPT-135) δ 165.6 (C, O-C=O), 147.0 (C), 142.7 (C), 140.9 (C), 134.6 (C), 133.4 (C), 131.7 (C), 130.4 (2 x CH), 128.8 (2 x CH), 128.0 (2 x CH), 127.2 (CH), 127.0 (CH), 126.6 (2 x CH), 61.1 (CH₂, O*C*H₂CH₃), 38.9 (CH), 29.3 (CH₂), 21.8 (CH₃, Ar-*C*H₃), 14.1 (CH₃, OCH₂*C*H₃); Anal. calcd for C₂₂H₂₁N₃O₄S (423.13): C, 62.40; H, 5.00; N, 9.92. Found: C, 62.395; H, 5.050; N, 10.055%.

1-(Toluene-4-sulfonyl)-6,7-dihydro-1H-benzotriazole-5-carboxylic acid methyl ester (120ta): Purified by column chromatography using EtOAc/hexane and isolated as a white solid. Mp 134 °C; IR (neat): v_{max} 2953, 1711 (O-C=O), 1434, 1395, 1295, 1201, 1194, 1184, 1093, 681 cm⁻¹; ¹H NMR (CDCl₃) δ 7.99 (2H, d, J = 8.4 Hz), 7.60 (1H, s, olefinic-H), 7.40 (2H, d, J = 8.4 Hz), 3.81 (3H, s, OC H_3), 3.26 (2H, t, J = 9.2 Hz), 2.87 (2H, dt, J = 9.2, 1.6 Hz), 2.46 (3H, s, Ar-C H_3); ¹³C NMR (CDCl₃, DEPT-135) δ 166.6 (C, O-C=O), 147.4 (C), 142.5 (C), 135.8 (C), 133.4 (C), 130.5 (2 x CH), 128.6 (2 x CH), 128.2 (C), 126.7 (CH), 52.1 (CH₃, OCH₃), 22.8 (CH₂), 21.8 (CH₃, Ar-CH₃), 19.7 (CH₂); LRMS m/z 334.00 (M + H⁺), calcd C₁₅H₁₅N₃O₄S 333.0783; Anal. calcd for C₁₅H₁₅N₃O₄S (333.07): C, 54.04; H, 4.54; N, 12.60. Found: C, 54.076; H, 4.526; N, 12.576%.

T_S N-N ethyl ester (120ea): Purified by column chromatography using EtOAc/hexane and isolated as a white solid. Mp 102 °C; IR (neat): v_{max} 2929, 1703 (O-C=O), 1447, 1395, 1336, 1295, 1200, 1194, 1184, 1094, 1069, 735, 681 cm⁻¹; ¹H NMR (CDCl₃) δ 7.91 (2H, d, J = 8.4 Hz), 7.52 (1H, t, J = 1.6 Hz, olefinic-H), 7.33 (2H, d, J = 8.0 Hz), 4.18 (2H, q, J = 7.2 Hz, OCH₂CH₃), 3.18 (2H, t, J = 9.2 Hz), 2.79 (2H, dt, J = 9.2, 1.6 Hz), 2.39 (3H, s, Ar-CH₃), 1.25 (3H, t, J = 7.2 Hz, OCH₂CH₃); ¹³C NMR (CDCl₃, DEPT-135) δ 166.1 (C, O-C=O), 147.4 (C), 142.5 (C), 135.8 (C), 133.4 (C), 130.5 (2 x CH), 128.5 (2 x CH and C), 126.4 (CH), 61.0 (CH₂, OCH₂CH₃), 22.8 (CH₂), 21.8 (CH₃, Ar-CH₃), 19.7 (CH₂), 14.2 (CH₃, OCH₂CH₃); LRMS m/z 348.00 (M + H⁺), calcd C₁₆H₁₇N₃O₄S 347.0940; Anal. calcd for C₁₆H₁₇N₃O₄S (347.09): C, 55.32; H, 4.93; N, 12.10. Found: C, 55.278; H, 4.951; N, 12.495%.

3-(Toluene-4-sulfonyl)-4,5-dihydro-3H-naphtho[1,2-*d*][1,2,3]triazole (120ua):

Purified by column chromatography using EtOAc/hexane and isolated as a white solid. Mp 140 °C; IR (neat): v_{max} 2924, 1589, 1446, 1389, 1339, 1195, 1172, 1121, 994 cm⁻¹; ¹H NMR (CDCl₃) δ 7.98 (2H, d, J = 8.4 Hz), 7.90 (1H, 120ua d, J = 7.2 Hz), 7.37 (2H, d, J = 8.0 Hz), 7.28-7.22 (3H, m), 3.27 (2H, t, J = 7.6 Hz), 3.09 (2H, t, J = 7.6 Hz), 2.43 (3H, s, Ar-CH₃); ¹³C NMR (CDCl₃, DEPT-135) δ 147.0 (C), 143.6 (C), 133.8 (C), 133.7 (C), 133.4 (C), 130.4 (2 x CH), 128.4 (2 x CH), 128.3 (CH), 128.1 (CH), 127.3 (CH), 127.1 (C), 122.5 (CH), 28.0 (CH₂), 21.8 (CH₃, Ar-CH₃), 20.1 (CH₂); LRMS m/z 326.00 (M + H⁺), calcd C₁₇H₁₅N₃O₂S 325.0885; Anal. calcd for C₁₇H₁₅N₃O₂S (325.08): C, 62.75; H, 4.65; N, 12.91. Found: C, 62.740; H, 4.644; N, 12.999%.

3-Methanesulfonyl-4,5-dihydro-3H-naphtho[1,2-*d***][1,2,3]triazole (120ub):** Purified by column chromatography using EtOAc/hexane and isolated as a white solid. Mp 124 °C; IR (neat): v_{max} 3018, 2933, 1384, 1371, 1340, 1333, 1195, 1180, 1008, 975, 956, 779, 764 cm⁻¹; ¹H NMR (CDCl₃) δ 7.92 (1H, d, J = 7.6

120ub

Hz), 7.31-7.24 (3H, m), 3.56 (3H, s, SO₂CH₃), 3.25 (2H, t, J = 7.6 Hz), 3.10 (2H, t, J = 7.6 Hz); ¹³C NMR (CDCl₃, DEPT-135) δ 143.5 (C), 134.1 (C), 133.5 (C), 128.5 (CH), 128.2 (CH), 127.3 (CH), 126.7 (C), 122.4 (CH), 43.0 (CH₃, SO₂CH₃), 27.9 (CH₂), 19.9 (CH₂); LRMS m/z 250.00 (M + H⁺), calcd C₁₁H₁₁N₃O₂S 249.0572; Anal. calcd for C₁₁H₁₁N₃O₂S (249.05): C, 53.00; H, 4.45; N, 16.86. Found: C, 53.008; H, 4.453; N, 16.937%.

5b: Amine-catalyzed cascade EA/E reactions: In an ordinary glass vial equipped with a magnetic stirring bar, to 0.12 mmol of the BnNH₂ 2d, and 0.3 mmol of Hagemann's esters 88 was added 0.5 mL of solvent, and then the 0.5 mL solution of TsN₃ 118a (0.3 mmol) was added dropwise for 0.5 h and the reaction mixture was stirred at 25 °C for the time indicated in Tables 17 and 21. The crude reaction mixture was worked up with aqueous NH₄Cl solution and the aqueous layer was extracted with dichloromethane (2 x 20 mL). The combined organic layers were dried (Na₂SO₄), filtered and concentrated. Pure cascade products 119 were obtained by column chromatography (silica gel, mixture of hexane/ethyl acetate).

3-Diazo-2-methyl-4-oxo-cyclohex-1-enecarboxylic acid ethyl ester (119aa): Purified

by column chromatography using EtOAc/hexane and isolated as a yellow solid. Mp 34 °C; IR (neat): v_{max} 2982, **2091** (**diazo**), 1699 (O-C=O), 1661, 1603, 1447, 1370, 1260, 1173, 1047, 864 cm⁻¹; ¹H NMR (CDCl₃) δ 4.22 (2H, q, J = 7.2Hz, OC H_2 CH₃), 2.76 (2H, t, J = 6.8 Hz), 2.54 (2H, t, J = 6.8 Hz), 2.32 (3H, s, olefinic-C H_3), 1.32 (3H, t, J = 7.2 Hz, OC H_2 CH₃); ¹³C NMR (CDCl₃, DEPT-135) δ 192.7 (C, C=O), 166.3 (C, O-C=O), 133.8 (C), 116.7 (C), 60.4 (CH₂, OC H_2 CH₃), 36.3 (CH₂), 23.8 (CH₂), 16.3 (CH₃, olefinic-CH₃), 14.3 (CH₃, OC H_2 CH₃) [one of the quaternary carbon attached to the diazo group is poor resolution even after more scans, may be due to the resonance]; Anal. calcd for C₁₀H₁₂N₂O₃ (208.08): C, 57.68; H, 5.81; N, 13.45. Found: C, 57.597; H, 5.795; N, 13.578%.

3-Diazo-2-methyl-4-oxo-cyclohex-1-enecarboxylic acid methyl ester (119sa):

Purified by column chromatography using EtOAc/hexane and isolated as a yellow solid. Mp 50 °C; IR (neat): v_{max} 3000, **2088** (**diazo**), 1696 (O-C=O), 1662, 1598, 1435, 1366, 1259, 1201, 1167, 1139 cm⁻¹; ¹H NMR (CDCl₃) δ 3.76 (3H, s, OCH₃), 2.75 (2H, t, J = 6.4 Hz), 2.53 (2H, t, J = 6.8 Hz), 2.33 (3H, s, olefinic-CH₃); ¹³C NMR (CDCl₃, DEPT-135) δ 192.5 (C, C=O), 166.7 (C, O-C=O), 134.3 (C), 116.3 (C), 51.4 (CH₃, OCH₃), 36.3 (CH₂), 23.8 (CH₂), 16.3 (CH₃, olefinic-CH₃) [one of the quaternary carbon attached to the diazo group is poor resolution even after more scans, may be due to the resonance]; Anal. calcd for C₉H₁₀N₂O₃ (194.07): C, 55.67; H, 5.19; N, 14.43. Found: C, 55.680; H, 5.204; N, 14.608%.

3-Diazo-2-methyl-4-oxo-cyclohex-1-enecarboxylic acid tert-butyl ester (119ba):

Purified by column chromatography using EtOAc/hexane and isolated as a yellow solid. Mp 40 °C; IR (neat): v_{max} 2975, **2087** (**diazo**), 1691 (O-C=O), 1678, 1665, 1599, 1456, 1366, 1267, 1207, 1157, 1137 cm⁻¹; ¹H NMR (CDCl₃) δ 2.72–2.68 (2H, m), 2.52 (2H, t, J = 7.2 Hz), 2.27 (3H, s, olefinic-CH₃), 1.52 (9H, s, 3 x CH₃, tert-Bu); ¹³C NMR (CDCl₃, DEPT-135) δ 193.0 (C, C=O), 165.9 (C, O-C=O), 132.0 (C), 118.6 (C), 80.8 [C, OC(CH₃)₃], 36.4 (CH₂), 28.3 [3 x CH₃, OC(CH₃)₃], 24.2 (CH₂), 16.3 (CH₃, olefinic-CH₃) [one of the quaternary carbon attached to the diazo group is poor resolution even after more scans, may be due to the resonance]; LRMS m/z 237.00 (M + H⁺), calcd C₁₂H₁₆N₂O₃ 236.1161; Anal. calcd for C₁₂H₁₆N₂O₃ (236.12): C, 61.00; H, 6.83; N, 11.86. Found: C, 60.901; H, 6.840; N, 11.941%.

3-Diazo-2,6-dimethyl-4-oxo-cyclohex-1-enecarboxylic acid ethyl ester (119ia):

Purified by column chromatography using EtOAc/hexane and isolated as a yellow liquid. IR (neat): v_{max} 2976, **2091** (**diazo**), 1693 (O-C=O), 1666, 1596, 1369, 1330, 1292, 1260, 1232, 1193, 1090, 1047 cm⁻¹; ¹H NMR 119ia

(CDCl₃) δ 4.24 (2H, q, J = 7.2 Hz, OC H_2 CH₃), 3.26 -3.23 (1H, m), 2.66 (1H, dd, J = 16.0, 6.4 Hz), 2.42 (1H, dd, J = 16.4, 2.0 Hz), 2.31 (3H, s, olefinic-C H_3), 1.33 (3H, t, J = 7.2 Hz, OCH₂C H_3), 1.09 (3H, d, J = 6.8 Hz, CHC H_3); ¹³C NMR (CDCl₃, DEPT-135) δ 192.4 (C, C=O), 166.2 (C, O-C=O), 132.5 (C), 122.2 (C), 60.3 (CH₂, OC H_2 CH₃), 43.6 (CH₂), 30.1 (CH), 19.6 (CH₃, CHCH₃), 16.5 (CH₃, olefinic-CH₃), 14.3 (CH₃, OCH₂CH₃) [one of the quaternary carbon attached to the diazo group is poor resolution even after more scans, may be due to the resonance]; LRMS m/z 223.00 (M + H⁺), calcd C₁₁H₁₄N₂O₃ 222.1004; Anal. calcd for C₁₁H₁₄N₂O₃ (222.10): C, 59.45; H, 6.35; N, 12.61. Found: C 59.440; H, 6.320; N, 12.602%.

3-Diazo-2-methyl-4-oxo-6-phenyl-cyclohex-1-enecarboxylicacid ethyl ester

(119ma): Purified by column chromatography using EtOAc/hexane and isolated as a yellow liquid. IR (neat): v_{max} 2982, 2929, 2091 (diazo), 1695 (O-C=O), 1666, 1596, 1369, 1247, 1170, 1137, 741 cm⁻¹; ¹H NMR (CDCl₃) δ 7.28–7.19 (3H, m), 7.15 (2H, d, J = 7.6 Hz), 4.47 (1H, br d, J = 7.2 Hz, PhCH), 4.13 (2H, q, J = 6.8 Hz, OCH2CH₃), 2.95 (1H, dd, J = 16.4, 7.2 Hz), 2.80 (1H, br d, J = 16.4 Hz), 2.43 (3H, s, olefinic-CH3), 1.19 (3H, t, J = 6.8 Hz, OCH₂CH₃); ¹³C NMR (CDCl₃, DEPT-135) δ 191.3 (C, C=O), 166.1 (C, O-C=O), 142.1 (C), 134.4 (C), 128.7 (2 x CH), 126.9 (CH), 126.8 (2 x CH), 119.7 (C), 60.5 (CH₂, OCH₂CH₃), 44.0 (CH₂), 40.4 (CH), 16.7 (CH₃, olefinic-CH₃), 14.1 (CH₃, OCH₂CH₃) [one of the quaternary carbon attached to the diazo group is poor resolution even after more scans, may be due to the resonance]; Anal. calcd for C₁₆H₁₆N₂O₃ (284.12): C, 67.59; H, 5.67; N, 9.85. Found: C, 67.582; H, 5.701; N, 9.991%.

3-Diazo-6-furan-2-yl-2-methyl-4-oxo-cyclohex-1-enecarboxylic acid ethyl ester

(119ra): Purified by column chromatography using EtOAc/hexane and isolated as a yellow liquid. IR (neat): v_{max} 2978, 2092 (diazo), 1734 (O-C=O), 1669, 1595, 1369, 1289, 1254, 1176, 1151, 1042, 739 cm⁻¹;

119ra

1H NMR (CDCl₃) δ 7.27 (1H, d, J = 2.8 Hz), 6.23 (1H, t, J = 2.8 Hz), 5.98 (1H, d, J = 2.8 Hz), 4.56 (1H, br d, J = 6.4 Hz), 4.23 (2H, q, J = 7.2 Hz,

OC H_2 CH₃), 2.92 (1H, dd, J = 16.4, 2.0 Hz), 2.81 (1H, dd, J = 16.4, 6.8 Hz), 2.39 (3H, s, olefinic-C H_3), 1.30 (3H, t, J = 7.2 Hz, OCH₂C H_3); ¹³C NMR (CDCl₃, DEPT-135) δ 191.0 (C, C=O), 165.7 (C, O-C=O), 154.7 (C), 141.9 (CH), 135.7 (C), 117.1 (C), 110.0 (CH), 105.5 (CH), 60.6 (CH₂, OCH₂CH₃), 40.7 (CH₂), 34.2 (CH), 16.8 (CH₃, olefinic-CH₃), 14.2 (CH₃, OCH₂CH₃) [one of the quaternary carbon attached to the diazo group is poor resolution even after more scans, may be due to the resonance]; LRMS m/z 273.00 (M - H⁺), calcd C₁₄H₁₄N₂O₄ 274.0954; Anal. calcd for C₁₄H₁₄N₂O₄ (274.09): C, 61.31; H, 5.14; N, 10.21. Found: C, 61.383; H, 5.184; N, 10.194%.

6a: Pyrrolidine-catalyzed Michael reactions: In an ordinary glass vial equipped with a magnetic stirring bar, to 0.5 mmol of the Hagemann's ester **88** was added 1.0 mL of DMSO solvent, and then the catalyst pyrrolidine **2a** (0.1 mmol, 8.19μL) was added and then 0.6 mmol of β-nitrostyrene **35** was added in one-portion and the reaction mixture was stirred at RT for the time indicated in Table 24. The crude reaction mixture was worked up with aqueous NH₄Cl solution and the aqueous layer was extracted with dichloromethane (2 x 20 mL). The combined organic layers were dried (Na₂SO₄), filtered and concentrated. Pure achiral Michael products **133/134** were obtained by column chromatography (silica gel, mixture of hexane/ethyl acetate).

6b: (*S*)-Bis-(3,5-dimethylphenyl)-pyrrolidin-2-yl-methanol catalyzed Michael reactions: In an ordinary glass vial equipped with a magnetic stirring bar, to 0.1 mmol of the Hagemann's esters **88** was added 1.0 mL of toluene solvent, and then the catalyst (*S*)-bis-(3,5-dimethylphenyl)-pyrrolidin-2-yl-methanol (*S*)-**2r** (0.03 mmol, 30 mol%), thiourea **132b** (0.03 mmol, 30 mol%) was added and then 0.15 mmol of β-nitrostyrene **35** were added in one-portion and the reaction mixture was stirred at RT for the time indicated in Table 25. The crude reaction mixture was worked up with aqueous NH₄Cl solution and the aqueous layer was extracted with dichloromethane (2 x 20 mL). The combined organic layers were dried (Na₂SO₄), filtered and concentrated. Pure chiral

Michael products 133/134 were obtained by column chromatography (silica gel, mixture of hexane/ethyl acetate).

6c: (*R*)-Bis-(3,5-dimethylphenyl)-pyrrolidin-2-yl-methanol catalyzed Michael reactions: In an ordinary glass vial equipped with a magnetic stirring bar, to 0.1 mmol of the Hagemann's esters **88a** & **88b** was added 1.0 mL of toluene solvent and then the catalyst (*R*)-bis-(3,5-dimethylphenyl)-pyrrolidin-2-yl-methanol **2r** (0.03 mmol, 30 mol%), thiourea **132b** (0.03 mmol, 30 mol%) was added and then 0.15 mmol of β-nitrostyrene **35a** were added in one-portion and the reaction mixture was stirred at RT for the time indicated in Table 4. The crude reaction mixture was worked up with aqueous NH₄Cl solution and the aqueous layer was extracted with dichloromethane (2 x 20 mL). The combined organic layers were dried (Na₂SO₄), filtered and concentrated. Pure chiral Michael products (+)-**133aa/134aa** and (-)-**133ba/134ba** were obtained by column chromatography (silica gel, mixture of hexane/ethyl acetate).

(S,S)-2-Methyl-3-(2-nitro-1-phenylethyl)-4-oxo-cyclohex-2-enecarboxylic acid ethyl ester (133aa) and (S,R)-2-Methyl-3-(2-nitro-1-phenylethyl)-4-oxo-cyclohex-2-enecarboxylic acid ethyl ester (134aa): Prepared following the procedure 6b and

purified by column chromatography using EtOAc/hexane and isolated as a colorless liquid. The ee was determined by chiral-phase HPLC using a Daicel Chiralcel OD-H column (hexane/*i*-PrOH = 90:10, flow rate 1.0 mL/min, λ = 254 nm), $t_{\rm R}$ = 14.3 min (minor), $t_{\rm R}$ = 17.2 min (major), $t_{\rm R}$ = 20.8 min (major), $t_{\rm R}$ = 25.5 min (minor); $[\alpha]^{25}_{\rm D}$ = +9.5 (c

0.3, CHCl₃, 70% ee, –7% ee); IR (neat): v_{max} 2965, 1727 (O-C=O), 1666 (C=O), 1551, 1375, 1160, 857, 700 and 625 cm⁻¹; ¹H NMR (CDCl₃, 1.3:1 ratio of diastereomers) δ 7.32-7.28 (4H, m), 7.25-7.21 (6H, m), 5.30 (1H, dd, J = 12.4, 9.2 Hz), 5.18 (2H, br d, J = 7.6 Hz), 5.08 (1H, dd, J = 12.4, 6.0 Hz), 4.85 (1H, t, J = 7.6 Hz), 4.79 (1H, dd, J = 8.8, 6.0 Hz), 4.22 (2H, q, J = 7.2 Hz, OC H_2 CH₃), 4.19 (2H, q, J = 7.2 Hz, OC H_2 CH₃), 3.37 (1H, t, J = 4.4 Hz), 3.34 (1H, t, J = 4.4 Hz), 2.62-2.50 (1H, m), 2.47 (1H, dd, J = 12.0, 5.6 Hz), 2.42 (1H, t, J = 4.8 Hz), 2.37 (1H, t, J = 4.0 Hz), 2.34-2.20 (4H, m), 2.16

(3H, s, olefinic-C H_3), 2.13 (3H, s, olefinic-C H_3), 1.30 (3H, t, J = 7.2 Hz, OCH₂C H_3), 1.25 (3H, t, J = 7.2 Hz, OCH₂C H_3); ¹³C NMR (CDCl₃, DEPT-135, 1.3:1 ratio of diastereomers) δ 197.4 (C, C=O), 197.1 (C, C=O), 171.4 (C, O-C=O), 171.3 (C, O-C=O), 155.1 (C), 154.8 (C), 138.5 (C), 138.2 (C), 135.3 (C), 135.0 (C), 128.7 (2 x CH), 128.6 (2 x CH), 127.5 (2 x CH), 127.1 (CH), 127.0 (3 x CH), 77.60 (CH₂), 77.57 (CH₂), 61.6 (CH₂, OCH₂CH₃), 61.4 (CH₂, OCH₂CH₃), 48.4 (CH), 48.3 (CH), 42.0 (CH), 41.9 (CH), 35.3 (CH₂), 35.2 (CH₂), 25.2 (CH₂), 25.1 (CH₂), 21.2 (CH₃, olefinic-CH₃), 21.1 (CH₃, olefinic-CH₃), 14.04 (CH₃, OCH₂CH₃), 14.02 (CH₃, OCH₂CH₃); LRMS m/z 332.00 (M + H⁺), calcd C₁₈H₂₁NO₅ 331.1420; Anal. calcd for C₁₈H₂₁NO₅ (331.14): C, 65.24; H, 6.39; N, 4.23. Found: C, 65.31; H, 6.35; N, 4.26%.

(R,S)-2-Methyl-3-(2-nitro-1-phenylethyl)-4-oxo-cyclohex-2-enecarboxylic acid ethyl ester (133aa) and (R,R)-2-Methyl-3-(2-nitro-1-phenylethyl)-4-oxo-cyclohex-2-enecarboxylic acid ethyl ester (134aa): Prepared following the procedure 6c and

purified by column chromatography using EtOAc/hexane and isolated as a colorless liquid. The ee was determined by chiral-phase HPLC using a Daicel Chiralcel OD-H column (hexane/*i*-PrOH = 90:10, flow rate 1.0 mL/min, λ = 254 nm), t_R = 14.3 min (minor), t_R = 17.2 min (+)-133aa/134aa (major), t_R = 21.2 min (minor), t_R = 25.4 min (major); $[\alpha]^{25}_D$ = +3.5 (c 0.3, CHCl₃, -70% ee, -14% ee); IR (neat): v_{max} 2965, 1727 (O-C=O), 1666 (C=O), 1551, 1375, 1160, 857, 700 and 625 cm⁻¹.

(S,S)-2-Methyl-3-(2-nitro-1-phenylethyl)-4-oxo-cyclohex-2-enecarboxylic acid methyl ester (133sa) and (S,R)-2-Methyl-3-(2-nitro-1-phenylethyl)-4-oxo-cyclohex-2-enecarboxylic acid methyl ester (134sa): Prepared following the procedure 6b and

purified by column chromatography using EtOAc/hexane and isolated as a yellow liquid. The ee was determined by chiral-phase HPLC using a Daicel Chiralcel OD-H column (hexane/i-PrOH = 90:10, flow rate 0.5 mL/min, λ = 254 nm), t_R = 38.0 min (major), t_R = 52.2 min (minor), t_R 133sa/134sa

dr = 1:1.6

= 59.8 min (major), t_R = 64.8 min (minor); $[\alpha]^{25}_D$ = -6.7 (c 0.3, CHCl₃, 61% ee, 11%) ee); IR (neat): v_{max} 3028, 2955, 1732 (O-C=O), 1668 (C=O), 1553, 1497, 1435, 1377, 1204, 1161 and 700 cm $^{-1}$; 1 H NMR (CDCl $_{3}$, 1:1.6 ratio of diastereomers) δ 7.32-7.28 (4H, m), 7.25-7.21 (6H, m), 5.29 (1H, dd, J = 12.4, 9.2 Hz), 5.24-5.12 (2H, m), 5.07 (1H, dd, J = 12.4, 6.0 Hz), 4.85-4.77 (2H, m), 3.77 (3H, s, OCH₃), 3.74 (3H, s, OCH₃),3.40 (1H, t, J = 4.4 Hz), 3.37 (1H, t, J = 4.8 Hz), 2.61-2.52 (1H, m), 2.50-2.41 (2H, m),2.39-2.31 (1H, m), 2.30-2.19 (4H, m), 2.16 (3H, s, olefinic-CH₃), 2.12 (3H, s, olefinic- CH_3); ¹³C NMR (CDCl₃, DEPT-135, 1.6:1 ratio of diastereomers) δ 197.4 (C, C=O), 197.1 (C, C=O), 172.0 (C, O-C=O), 171.8 (C, O-C=O), 155.0 (C), 154.6 (C), 138.5 (C), 138.1 (C), 135.6 (C), 135.1 (C), 128.7 (4 x CH), 127.5 (CH), 127.2 (CH), 127.1 (4 x CH), 77.7 (CH₂), 77.6 (CH₂), 52.6 (CH₃, OCH₃), 52.5 (CH₃, OCH₃), 48.3 (CH), 48.2 (CH), 42.2 (CH), 42.0 (CH), 35.4 (CH₂), 35.3 (CH₂), 25.2 (CH₂), 25.1 (CH₂), 21.3 (CH₃, olefinic-CH₃), 21.1 (CH₃, olefinic-CH₃); LRMS m/z 318.00 (M + H⁺), calcd C₁₇H₁₉NO₅ 317.1263; Anal. calcd for C₁₇H₁₉NO₅ (317.12): C, 64.34; H, 6.03; N, 4.41. Found: C, 64.41; H, 6.08; N, 4.45%.

(S,S)-2-Methyl-3-(2-nitro-1-phenylethyl)-4-oxo-cyclohex-2-enecarboxylic acid tertbutyl ester (133ba) and (S,R)-2-Methyl-3-(2-nitro-1-phenylethyl)-4-oxo-cyclohex-2enecarboxylic acid tert-butyl ester (134ba): Prepared following the procedure 6b and

ĆO₂^tBu 133ba/134ba

purified by column chromatography using EtOAc/hexane and isolated as a yellow liquid. The ee was determined by chiral-phase HPLC using a Daicel Chiralcel OD-H column (hexane/i-PrOH = 95:5, flow rate 0.5 mL/min, $\lambda = 254$ nm), $t_R = 26.2$ min (minor), $t_R = 28.3$ min (major), t_R = 37.0 min (major), t_R = 52.7 min (minor); $[\alpha]^{25}_D$ = +31.0 (c 0.3, CHCl₃, 84% ee, 40% ee); IR (neat): v_{max} 2978, 1724 (O-C=O), 1668 (C=O), 1553, 1372, 1254, 1148 and 698 cm⁻¹; ¹H NMR (CDCl₂, 2.1:1 ratio of diastereomers) δ 7.28 (4H, t, J = 7.2 Hz), 7.24-7.21 (6H, m), 5.30-5.21 (2H, m), 5.15-5.06 (2H, m), 4.84 (1H, m)t, J = 7.6 Hz), 4.76 (1H, dd, J = 8.8, 6.0 Hz), 3.27 (1H, t, J = 4.4 Hz), 3.24 (1H, t, J = 4.4 Hz), 3.24 (1H, t, J = 4.4 Hz), 3.25 (1H, t, J = 4.4 Hz), 3.26 (1H, t, J = 4.4 Hz), 3.27 (1H, t, J = 4.4 Hz), 3.27 (1H, t, J = 4.4 Hz), 3.28 (1H, t, J = 4.4 Hz), 3.29 (1H, t, J = 4.4 Hz), 3.20 (1H, t, J = 4.4 Hz), 3 4.8 Hz), 2.62-2.45 (2H, m), 2.42-2.33 (2H, m), 2.31-2.20 (4H, m), 2.17 (3H, s, olefinicCH₃), 2.12 (3H, s, olefinic-CH₃), 1.48 (9H, s, 3 x CH₃, O-C(CH₃)₃), 1.41 (9H, s, 3 x CH₃, O-C(CH₃)₃); ¹³C NMR (CDCl₃, DEPT-135, 2.1:1 ratio of diastereomers) δ 197.6 (C, C=O), 197.3 (C, C=O), 170.44 (C, O-C=O), 170.37 (C, O-C=O), 155.7 (C), 155.4 (C), 138.6 (C), 138.3 (C), 134.8 (C), 134.5 (C), 128.64 (2 x CH), 128.57 (2 x CH), 127.5 (2 x CH), 127.1 (3 x CH), 127.0 (CH), 82.3 (C, O-C(CH₃)₃), 82.1 (C, O-C(CH₃)₃), 77.6 (CH₂), 77.4 (CH₂), 49.51 (CH), 49.46 (CH), 42.01 (CH), 41.97 (CH), 35.3 (CH₂), 35.2 (CH₂), 27.8 (6 x CH₃, O-C(CH₃)₃), 25.3 (CH₂), 25.1 (CH₂), 21.3 (CH₃, olefinic-CH₃), 21.2 (CH₃, olefinic-CH₃); LRMS m/z 360.00 (M + H⁺), calcd C₂₀H₂₅NO₅ 359.1733; Anal. calcd for C₂₀H₂₅NO₅ (359.17): C, 66.83; H, 7.01; N, 3.90. Found: C, 66.75; H, 7.08; N, 3.98%.

(R,S)-2-Methyl-3-(2-nitro-1-phenylethyl)-4-oxo-cyclohex-2-enecarboxylic acid tert-butyl ester (133ba) and (R,R)-2-Methyl-3-(2-nitro-1-phenylethyl)-4-oxo-cyclohex-2-enecarboxylic acid tert-butyl ester (134ba): Prepared following the procedure 6c and

purified by column chromatography using EtOAc/hexane and isolated as a yellow liquid. The ee was determined by chiral-phase HPLC using a Daicel Chiralcel OD-H column (hexane/i-PrOH = 95:5, flow rate 0.5 mL/min, λ = 254 nm), $t_{\rm R}$ = 28.2 min (major), $t_{\rm R}$ = 31.3 min (minor), $t_{\rm R}$ = 43.4 min (minor), $t_{\rm R}$ = 63.3 min (major); $[\alpha]^{25}_{\rm D}$ = -32.4 (c 0.3, CHCl₃, -85% ee, -8% ee); IR (neat): $v_{\rm max}$ 2978, 1724 (O-C=O), 1668 (C=O), 1553, 1372, 1254, 1148 and 698 cm⁻¹.

(S,S)-3-[1-(2-Chlorophenyl)-2-nitro-ethyl]-2-methyl-4-oxo-cyclohex-2-enecarboxylic acid *tert*-butyl ester (133bc) and (S,R)-3-[1-(2-Chlorophenyl)-2-nitro-ethyl]-2-methyl-4-oxo-cyclohex-2-enecarboxylic acid *tert*-butyl ester (134bc):

Prepared following the procedure **6b** and purified by column chromatography using EtOAc/hexane and isolated as a yellow liquid. The ee was determined by chiral-phase HPLC using a Daicel Chiralcel OD-H column (hexane/i-PrOH = 90:10, flow rate 1.0 mL/min,
$$\lambda$$
 = 254

dr = 1.5:1

nm), $t_R = 7.6 \text{ min (minor)}$, $t_R = 8.6 \text{ min (major)}$, $t_R = 12.2 \text{ min (major)}$, $t_R = 14.4 \text{ min}$ (minor); $[\alpha]^{25}_{D} = +21.4$ (c 0.15, CHCl₃, 83% ee, 33% ee); IR (neat): v_{max} 2976, 2932, 1726 (O-C=O), 1668 (C=O), 1555, 1372, 1258, 1150, 845 and 758 cm⁻¹; ¹H NMR (CDCl₃, 1.5:1 ratio of diastereomers) δ 7.48 (1H, dd, J = 7.6, 1.6 Hz), 7.38-7.33 (3H, m), 7.24-7.18 (4H, m), 5.41 (1H, dd, J = 12.8, 9.6 Hz), 5.27 (1H, dd, J = 12.4, 8.4 Hz), 5.10 (1H, dd, J = 8.4, 6.4 Hz), 5.05-4.98 (2H, m), 4.83 (1H, dd, J = 13.2, 5.6 Hz), 3.26-3.22 (2H, m), 2.64-2.52 (2H, m), 2.43 (1H, td, J = 17.2, 4.8 Hz), 2.35-2.20 (4H, m), 2.17 (3H, s, olefinic- CH_3), 2.02 (3H, s, olefinic- CH_3), 1.48 (9H, s, 3 x CH_3 , O- $C(CH_3)_3$, 1.38 (9H, s, 3 x CH₃, O-C(CH₃)₃); ¹³C NMR (CDCl₃, DEPT-135, 1.5:1 ratio of diastereomers) δ 198.4 (C, C=O), 198.1 (C, C=O), 170.4 (C, O-C=O), 170.3 (C, O-C=O) C=O), 157.0 (C), 156.9 (C), 135.34 (C), 135.30 (C), 133.6 (C), 133.4 (C), 133.2 (C), 132.8 (C), 130.3 (CH), 130.1 (CH), 129.7 (CH), 129.6 (CH), 128.7 (CH), 128.6 (CH), 127.2 (CH), 127.1 (CH), 82.4 (C, O-C(CH₃)₃), 82.0 (C, O-C(CH₃)₃), 76.0 (CH₂), 75.7 (CH₂), 49.6 (2 x CH), 40.3 (CH), 39.9 (CH), 35.34 (CH₂), 35.26 (CH₂), 27.9 (3 x CH₃, $O-C(CH_3)_3$, 27.8 (3 x CH₃, $O-C(CH_3)_3$), 25.2 (CH₂), 25.0 (CH₂), 21.6 (CH₃, olefinic- CH_3), 21.2 (CH₃, olefinic- CH_3); LRMS m/z 394.00 (M + H⁺), calcd C₂₀H₂₄ClNO₅ 393.1343; Anal. calcd for C₂₀H₂₄ClNO₅ (393.13): C, 60.99; H, 6.14; N, 3.56. Found: C, 61.05; H, 6.10; N, 3.61%.

(S,S)-2-Methyl-3-(2-nitro-1-thiophen-2-yl-ethyl)-4-oxo-cyclohex-2-enecarboxylic acid *tert*-butyl ester (133bd) and (S,R)-2-Methyl-3-(2-nitro-1-thiophen-2-yl-ethyl)-4-oxo-cyclohex-2-enecarboxylic acid *tert*-butyl ester (134bd): Prepared following the

NO₂
Me
CO₂^tBu

procedure **6b** and purified by column chromatography using EtOAc/hexane and isolated as a yellow liquid. The ee was determined by chiral-phase HPLC using a Daicel Chiralpak AS-H column (hexane/i-PrOH = 90:10, flow rate 0.5 mL/min, λ = 254 nm), t_R = 14.8 min (major), t_R = 17.8 min (minor), t_R = 19.3 min (major), t_R = 22.3

133bd/134bd min (major), $t_R = 17.8$ min (minor), $t_R = 19.3$ min (major), $t_R = 22.3$ min (minor); $[\alpha]^{25}_D = +43.1$ (c 0.2, CHCl₃, 69% ee, 11% ee); IR (neat): v_{max} 2978, 2930, 1722 (O-C=O), 1667 (C=O), 1555, 1370, 1260, 1150 and 845 cm⁻¹; ¹H NMR

(CDCl₃, 5.2:1 ratio of diastereomers, major isomer) δ 7.16 (1H, t, J = 3.2 Hz), 6.92 (2H, d, J = 3.2 Hz), 5.26-5.20 (1H, m), 5.11-5.05 (2H, m), 3.25 (1H, t, J = 4.8 Hz), 2.64-2.55 (1H, m), 2.37 (1H, td, J = 17.2, 4.8 Hz), 2.29-2.22 (1H, m), 2.18 (3H, s, olefinic-CH₃), 2.18-2.11 (1H, m), 1.48 (9H, s, 3 x CH₃, O-C(CH₃)₃); ¹³C NMR (CDCl₃, DEPT-135, 5.2:1 ratio of diastereomers, major isomer) δ 197.3 (C, C=O), 170.3 (C, O-C=O), 155.7 (C), 141.0 (C), 134.3 (C), 126.8 (CH), 125.4 (CH), 124.8 (CH), 82.4 (C, O-C(CH₃)₃), 78.0 (CH₂), 49.6 (CH), 38.3 (CH), 35.3 (CH₂), 27.9 (3 x CH₃, O-C(CH₃)₃), 25.2 (CH₂), 21.1 (CH₃, olefinic-CH₃); LRMS m/z 364.00 (M – H⁺), calcd C₁₈H₂₃NO₅S 365.1297; Anal. calcd for C₁₈H₂₃NO₅S (365.12): C, 59.16; H, 6.34; N, 3.83. Found: C, 59.32; H, 6.25; N, 3.92%.

(S,S)-3-(2-Nitro-1-phenylethyl)-4-oxo-2-propyl-cyclohex-2-enecarboxylic acid ethyl ester (133ha) and (S,R)-3-(2-Nitro-1-phenylethyl)-4-oxo-2-propyl-cyclohex-2-enecarboxylic acid ethyl ester (134ha): Prepared following the procedure 6b and

O E NO₂
Pr
CO₂Et

133ha/134ha *dr* = 2.0:1

purified by column chromatography using EtOAc/hexane and isolated as a yellow liquid. The ee was determined by chiral-phase HPLC using a Daicel Chiralcel AD-H column (hexane/*i*-PrOH = 95:5, flow rate 0.5 mL/min, $\lambda = 254$ nm), $t_R = 20.6$ min (minor), $t_R = 23.5$ min (major), $t_R = 28.9$ min (major), $t_R = 49.2$ min (minor); $[\alpha]^{25}_D = +7.3$ (*c* 0.25, CHCl₃, 77% ee, 3% ee); IR (neat): $v_{max} = 2964$, 1728 (O-C=O), 1669

(C=O), 1551, 1375, 1188 and 1158 cm⁻¹; ¹H NMR (CDCl₃, 2:1 ratio of diastereomers) δ 7.32-7.28 (4H, m), 7.25-7.20 (6H, m), 5.30-5.05 (4H, m), 4.86 (1H, t, J = 7.2 Hz), 4.79-4.72 (1H, m), 4.21 (2H, q, J = 7.2 Hz, OC H_2 CH₃), 4.18 (2H, q, J = 7.2 Hz, OC H_2 CH₃), 3.48-3.43 (2H, m), 2.66-2.44 (4H, m), 2.42-2.25 (3H, m), 2.24-2.08 (4H, m), 1.55-1.35 (5H, m), 1.29 (3H, t, J = 7.2 Hz, OCH₂CH₃), 1.24 (3H, t, J = 7.2 Hz, OCH₂CH₃), 0.98 (6H, t, J = 7.2 Hz, 2 x CH₂CH₂CH₃); ¹³C NMR (CDCl₃, DEPT-135, 2:1 ratio of diastereomers) δ 197.8 (C, C=O), 197.6 (C, C=O), 171.5 (C, O-C=O), 171.4 (C, O-C=O), 159.0 (C), 158.8 (C), 138.8 (C), 138.5 (C), 135.1 (C), 134.8 (C), 128.6 (4 x CH),

133ae/134aedr = 1.2:1

213

127.3 (2 x CH), 127.0 (4 x CH), 78.1 (2 x CH₂), 61.6 (CH₂, OCH₂CH₃), 61.5 (CH₂, OCH₂CH₃), 46.2 (CH), 45.8 (CH), 41.9 (2 x CH), 36.7 (CH₂), 36.5 (CH₂), 35.0 (2 x CH₂), 25.5 (CH₂), 25.4 (CH₂), 21.3 (2 x CH₂), 14.4 (2 x CH₃, OCH₂CH₃), 14.10 (CH₃, CH₂CH₂CH₃), 14.07 (CH₃, CH₂CH₂CH₃); LRMS m/z 360.00 (M + H⁺), calcd C₂₀H₂₅NO₅ 359.1733; Anal. calcd for C₂₀H₂₅NO₅ (359.17): C, 66.83; H, 7.01; N, 3.90. Found: C, 66.75; H, 7.05; N, 3.96%.

(S,S)-3-[1-(2-Methoxyphenyl)-2-nitroethyl]-2-methyl-4-oxo-cyclohex-2-

enecarboxylic acid ethyl ester (133ae) and (S,R)-3-[1-(2-Methoxyphenyl)-2nitroethyl]-2-methyl-4-oxo-cyclohex-2-enecarboxylic acid ethyl ester (134ae): Prepared following the procedure 6b and purified by column chromatography using EtOAc/hexane and isolated as a yellow liquid. The ee was determined by chiral-phase HPLC using a Daicel Chiralcel OD-H column (hexane/i-PrOH = 95:5, flow rate 1.0 mL/min, $\lambda = 254$ nm), $t_R = 17.5$ min (major), $t_R = 19.0$ min (minor), $t_R = 23.8$ min (minor), $t_R = 26.6$ min (major); $[\alpha]_D^{25} = +26.0$ (c 0.3, CHCl₃, 78% ee, 17% ee); IR (neat): v_{max} 2979, 1728 (O-C=O), 1665 (C=O), 1550, 1492, 1375, 1246, 1160 and 1028 cm⁻¹; ¹H NMR (CDCl₃, 1.2:1 ratio of diastereomers) δ 7.31 (1H, d, J = 7.6 Hz), 7.23-7.18 (3H, m), 6.91-6.84 (4H, m), 5.34-5.22 (2H, m), 5.16 (1H, dd, J = 8.4, 6.0 Hz), 5.05-5.00 (2H, m), 4.90 (1H, dd, J = 12.8, 5.2 Hz), 4.20 (2H, q, J = 7.2 Hz, OCH_2CH_3), 4.15 (2H, q, J = 7.2 Hz, OCH_2CH_3), 3.83 (3H, s, OCH_3), 3.82 (3H, s, OCH_3), 3.35 (1H, t, J = 4.8 Hz), 3.32 (1H, t, J = 4.4 Hz), 2.60-2.51 (2H, m), 2.43-2.28 (2H, m), 2.26-2.20 (3H, m), 2.15 (3H, s, olefinic-CH₃), 2.12-2.07 (1H, m), 2.03 (3H, s, s)olefinic- CH_3), 1.29 (3H, t, J = 7.2 Hz, OCH_2CH_3), 1.22 (3H, t, J = 7.2 Hz, OCH_2CH_3); ¹³C NMR (CDCl₃, DEPT-135, 1.2:1 ratio of diastereomers) δ 197.74 (C, C=O), 197.66 (C, C=O), 171.4 (C, O-C=O), 171.3 (C, O-C=O), 156.54 (C), 156.47 (C), 155.32 (C), 155.26 (C), 134.1 (C), 133.9 (C), 129.0 (CH), 128.9 (CH), 128.23 (CH), 128.17 (CH), 126.0 (C), 125.7 (C), 120.4 (2 x CH), 110.2 (CH), 110.0 (CH), 76.8 (CH₂), 76.6 (CH₂), 61.3 (CH₂, OCH₂CH₃), 61.1 (CH₂, OCH₂CH₃), 55.2 (CH₃, OCH₃), 55.1 (CH₃, OCH₃), 48.4 (CH), 48.2 (CH), 36.8 (CH), 36.2 (CH), 35.3 (CH₂), 35.2 (CH₂), 25.1

(CH₂), 24.9 (CH₂), 21.1 (CH₃, olefinic-CH₃), 20.7 (CH₃, olefinic-CH₃), 14.0 (CH₃, OCH₂CH₃), 13.9 (CH₃, OCH₂CH₃); LRMS m/z 360.25 (M – H⁺), calcd C₁₉H₂₃NO₆ 361.1525; Anal. calcd for C₁₉H₂₃NO₆ (361.15): C, 63.15; H, 6.41; N, 3.88. Found: C, 63.09; H, 6.45; N, 3.95%.

O H NO₂

Me
CO₂Et

133af/134af

dr = 1.2:1

(*S*,*S*)-3-[1-(2-Hydroxyphenyl)-2-nitroethyl]-2-methyl-4-oxocyclohex-2-enecarboxylic acid ethyl ester (133af) and (*S*,*R*)-3-[1-(2-Hydroxyphenyl)-2-nitroethyl]-2-methyl-4-oxo-cyclohex-2-enecarboxylic acid ethyl ester (134af): Prepared following the procedure 6b and purified by column chromatography using EtOAc/hexane and isolated as a dark yellow liquid. The ee was

determined by chiral-phase HPLC using a Daicel Chiralpak AS-H column (hexane/i-PrOH = 90:10, flow rate 1.0 mL/min, $\lambda = 254$ nm), $t_R = 16.0$ min (major), $t_R = 17.6$ min (minor), $t_R = 34.4$ min (minor), $t_R = 44.9$ min (major); $[\alpha]^{25}_D = -10.7$ (c 0.3, CHCl₃, **26% ee, 21% ee**); IR (neat): v_{max} 3360 (O-*H*), 2928, 1730 (O-C=O), 1661 (C=O), 1553, 1456, 1375 and 756 cm⁻¹; ¹H NMR (CDCl₃, 1.2:1 ratio of diastereomers) δ 7.24 (1H, d, J = 7.6 Hz), 7.17-7.13 (2H, m), 7.09 (3H, t, J = 7.6 Hz), 6.86-6.81 (2H, m), 6.78 (2H, m), 5.34 (1H, dd, J = 12.0, 9.2 Hz), 5.27 (1H, dd, J = 12.4, 8.8 Hz), 5.12 (1H, dd, J = 12.4), 5.34 (1H, dd, J = 12.4), 5.35 (1H, dd, J = 12.4), 5.35 (1H, dd, J = 12.4), 5.36 (1H, dd, J = 12.4), 5.37 (1H, dd, J = 12.4), 5.38 (1H, dd, J = 12.4), 5.38 (1H, dd, J = 12.4), 5.39 (1H, dd, J = 12.4), 5.39 (1H, dd, J = 12.4), 5.31 (1H, dd, J = 12.4), 5.31 (1H, dd, J = 12.4), 5.31 (1H, dd, J = 12.4), 5.32 (1H, dd, J = 12.4), 6.32 (1H, dd, J = 12.4), 8.8, 6.0 Hz), 5.07-5.02 (2H, m), 4.98 (1H, dd, J = 16.4, 4.4 Hz), 4.21 (2H, q, J = 7.2 Hz, OCH_2CH_3), 4.15 (2H, q, J = 7.2 Hz, OCH_2CH_3), 3.38 (1H, t, J = 4.4 Hz), 3.34 (1H, t, J = 4.4 Hz) = 4.4 Hz), 2.65-2.54 (2H, m), 2.46 (1H, td, J = 17.2, 4.8 Hz), 2.38 (1H, td, J = 17.2, 4.8 Hz), 2.26-2.23 (3H, m), 2.20 (3H, s, olefinic- CH_3), 2.18-2.13 (1H, m), 2.10 (3H, s, olefinic- CH_3), 1.29 (3H, t, J = 7.2 Hz, OCH_2CH_3), 1.22 (3H, t, J = 7.2 Hz, OCH_2CH_3); ¹³C NMR (CDCl₃, DEPT-135, 1.2:1 ratio of diastereomers) δ 199.3 (C, C=O), 199.2 (C, C=O), 171.6 (C, O-C=O), 171.5 (C, O-C=O), 156.9 (C), 156.8 (C), 153.8 (C), 153.7 (C), 134.3 (C), 134.1 (C), 129.4 (2 x CH), 128.44 (CH), 128.38 (CH), 124.1 (C), 123.9 (C), 120.4 (CH), 120.3 (CH), 115.9 (2 x CH), 76.6 (CH₂), 76.3 (CH₂), 61.6 (CH₂, OCH₂CH₃), 61.5 (CH₂, OCH₂CH₃), 48.5 (CH), 48.4 (CH), 37.6 (CH), 37.0 (CH), 35.14 (CH₂), 35.07 (CH₂), 25.0 (CH₂), 24.8 (CH₂), 21.4 (CH₃, olefinic-CH₃), 21.1

(CH₃, olefinic-CH₃), 13.9 (2 x CH₃, OCH₂CH₃); LRMS m/z 346.20 (M – H⁺), calcd C₁₈H₂₁NO₆ 347.1369; Anal. calcd for C₁₈H₂₁NO₆ (347.13): C, 62.24; H, 6.09; N, 4.03. Found: C, 62.41; H, 6.15; N, 4.10%.

 (S,S)-2-Methyl-3-[2-nitro-1-(2-trifluoromethylphenyl)ethyl]-4-oxo-cyclohex-2-enecarboxylic acid ethyl ester (133ag) and (S,R)-2-Methyl-3-[2-nitro-1-(2-trifluoromethylphenyl)ethyl]-4-oxo-cyclohex-2-enecarboxylic acid ethyl ester (134ag): Prepared following the procedure 6b and purified by column chromatography

dr = 1:1.4using EtOAc/hexane and isolated as a yellow liquid. The ee was determined by chiralphase HPLC using a Daicel Chiralpak AD-H column (hexane/i-PrOH = 90:10, flow rate 1.0 mL/min, $\lambda = 254$ nm), $t_R = 6.0$ min (major), $t_R = 6.7$ min (minor), $t_R = 7.2$ min (major), $t_R = 12.6 \text{ min (minor)}$; $[\alpha]^{25}_D = +5.7 (c 0.3, CHCl_3, 44\% ee, 15\% ee)$; IR (neat): v_{max} 1729 (O-C=O), 1667 (C=O), 1553, 1376, 1309, 1160, 1118, 1038 and 771 cm⁻¹; ¹H NMR (CDCl₃, 1:1.4 ratio of diastereomers) δ 7.72 (1H, d, J = 8.0 Hz), 7.67 (2H, t, J = 8.4 Hz), 7.58 (1H, d, J = 8.0 Hz), 7.51 (2H, m), 7.38 (2H, m), 5.61 (1H, dd, J)= 13.2, 10.8 Hz), 5.34 (1H, dd, J = 12.8, 8.4 Hz), 5.19 (1H, t, J = 7.2 Hz), 5.04-4.97 (2H, m), 4.71 (1H, dd, J = 13.6, 4.8 Hz), 4.21 (2H, q, J = 7.2 Hz, OCH₂CH₃), 4.13 (2H, q)q, J = 7.2 Hz, OCH_2CH_3), 3.37 (1H, t, J = 4.4 Hz), 3.33 (1H, t, J = 4.8 Hz), 2.65-2.45 (3H, m), 2.34 (1H, td, J = 17.2, 3.6 Hz), 2.27-2.24 (3H, m), 2.17-2.07 (1H, m), 2.11 (3H, s, olefinic- CH_3), 1.95 (3H, s, olefinic- CH_3), 1.30 (3H, t, J = 7.2 Hz, OCH_2CH_3), 1.19 (3H, t, J = 7.2 Hz, OCH₂CH₃); ¹³C NMR (CDCl₃, DEPT-135, 1:1.4 ratio of diastereomers) δ 198.8 (C, C=O), 198.3 (C, C=O), 171.2 (C, O-C=O), 171.1 (C, O-C=O) C=O), 156.41 (C), 156.38 (C), 136.4 (C), 136.1 (C), 134.3 (C), 134.2 (C), 132.5 (CH), 132.1 (CH), 131.3 (CH), 131.0 (CH), 128.3 (CF₃, q, J = 30.0 Hz), 128.1 (CF₃, q, J = 30.0 Hz) 30.0 Hz), 127.7 (CH), 127.5 (CH), 126.54 (CH, q, J = 6.0 Hz), 126.51 (CH, q, J = 6.0Hz), 125.8 (C, d, J = 4.0 Hz), 123.1 (C, d, J = 3.0 Hz), 76.3 (CH₂), 76.0 (CH₂), 61.6 (CH₂, OCH₂CH₃), 61.4 (CH₂, OCH₂CH₃), 48.7 (CH), 48.6 (CH), 39.8 (CH), 39.1 (CH), 35.4 (CH₂), 35.3 (CH₂), 24.9 (CH₂), 24.6 (CH₂), 21.3 (CH₃, olefinic-CH₃), 20.9

(CH₃, olefinic-CH₃), 14.01 (CH₃, OCH₂CH₃), 13.98 (CH₃, OCH₂CH₃); LRMS m/z 400.00 (M + H⁺), calcd C₁₉H₂₀F₃NO₅ 399.1294; Anal. calcd for C₁₉H₂₀F₃NO₅ (399.12): C, 57.14; H, 5.05; N, 3.51. Found: C, 57.08; H, 5.09; N, 3.58%.

(S,S)-3-[1-(2-Chlorophenyl)-2-nitroethyl]-2-methyl-4-oxo-cyclohex-2-enecarboxylic acid ethyl ester (133ac) and (S,R)-3-[1-(2-Chlorophenyl)-2-nitroethyl]-2-methyl-4-oxo-cyclohex-2-

enecarboxylic acid ethyl ester (134ac): Prepared following the 133ac/134ac procedure 6b and purified by column chromatography using dr = 1.1:1EtOAc/hexane and isolated as a yellow liquid. The ee was determined by chiral-phase HPLC using a Daicel Chiralcel OD-H column (hexane/i-PrOH = 90:10, flow rate 1.0 mL/min, $\lambda = 254$ nm), $t_R = 11.2$ min (minor), $t_R = 12.9$ min (major), $t_R = 14.9$ min (major), $t_R = 19.0 \text{ min (minor)}$; $[\alpha]^{25}_D = +5.3 \text{ (c 0.3, CHCl}_3, 60\% \text{ ee, } 4\% \text{ ee})$; IR (neat): v_{max} 1728 (O-C=O), 1667 (C=O), 1552, 1442, 1375, 1158 and 1018 cm⁻¹; ¹H NMR (CDCl₃, 1.1:1 ratio of diastereomers) δ 7.49 (1H, dd, J = 8.0, 1.6 Hz), 7.38-7.33 (3H, m), 7.25-7.17 (4H, m), 5.40 (1H, dd, J = 12.8, 3.2 Hz), 5.24 (1H, dd, J = 12.0, 4.0)Hz), 5.11 (1H, t, J = 6.4 Hz), 5.07-5.02 (2H, m), 4.85 (1H, dd, J = 12.8, 5.2 Hz), 4.21 $(2H, q, J = 7.2 \text{ Hz}, OCH_2CH_3), 4.15 (2H, q, J = 7.2 \text{ Hz}, OCH_2CH_3), 3.36-3.34 (2H, m),$ 2.62-2.51 (2H, m), 2.47-2.41 (1H, m), 2.35-2.21 (4H, m), 2.18 (3H, s, olefinic-CH₃), 2.16-2.08 (1H, m), 2.02 (3H, s, olefinic-CH₃), 1.29 (3H, t, J = 7.2 Hz, OCH₂CH₃), 1.21(3H, t, J = 7.2 Hz, OCH₂CH₃); ¹³C NMR (CDCl₃, DEPT-135, 1.1:1 ratio of diastereomers) δ 198.1 (C, C=O), 197.7 (C, C=O), 171.2 (C, O-C=O), 171.1 (C, O-C=O) C=O), 156.3 (C), 156.2 (C), 135.2 (C), 135.1 (C), 133.6 (C), 133.5 (C), 133.4 (C), 133.1 (C), 130.2 (CH), 130.0 (CH), 129.7 (CH), 129.6 (CH), 128.7 (CH), 128.6 (CH), 127.2 (CH), 127.0 (CH), 76.1 (CH₂), 75.7 (CH₂), 61.6 (CH₂, OCH₂CH₃), 61.4 (CH₂, OCH₂CH₃), 48.42 (CH), 48.39 (CH), 40.2 (CH), 39.9 (CH), 35.3 (CH₂), 35.2 (CH₂), 25.1 (CH₂), 24.9 (CH₂), 21.6 (CH₃, olefinic-CH₃), 21.2 (CH₃, olefinic-CH₃), 14.04 $(CH_3, OCH_2CH_3), 14.02 (CH_3, OCH_2CH_3); LRMS m/z 365.00 (M⁺), calcd$

C₁₈H₂₀ClNO₅ 365.1030; Anal. calcd for C₁₈H₂₀ClNO₅ (365.10): C, 59.10; H, 5.51; N, 3.83. Found: C, 59.15; H, 5.54; N, 3.88%.

(S,S)-3-[1-(4-Methoxyphenyl)-2-nitroethyl]-2-methyl-4-oxo-cyclohex-2-

enecarboxylic acid ethyl ester (133ah) and (S,R)-3-[1-(4-Methoxyphenyl)-2-nitroethyl]-2-methyl-4-oxo-cyclohex-2-enecarboxylic acid ethyl ester (134ah): Prepared following the procedure 6b and purified by column chromatography using EtOAc/hexane and isolated as a yellow liquid. The ee was determined by chiral-phase HPLC using a Daicel Chiralpak AD-H column (hexane/i-PrOH = 90:10, flow rate 1.0 mL/min, λ = 254 nm), t_R = 10.3

min (minor), $t_R = 11.1$ min (major), $t_R = 17.1$ min (major), $t_R = 20.8$ min (minor); $[\alpha]^{25}_{D} = +8.3$ (c 0.3, CHCl₃, 29% ee, 27% ee); IR (neat): v_{max} 2926, 2856, 1727 (O-C=O), 1666 (C=O), 1550, 1513, 1376, 1250, 1183 and 1034 cm⁻¹; ¹H NMR (CDCl₃, 1:1.7 ratio of diastereomers) δ 7.18 (2H, d, J = 8.8 Hz), 7.15 (2H, d, J = 8.8 Hz), 6.82 (4H, d, J = 8.4 Hz), 5.25 (1H, dd, J = 12.4, 9.2 Hz), 5.14 (2H, d, J = 7.2 Hz), 5.03 (1H, dd, J = 12.4, 9.2 Hz), 5.14 (2H, d, J = 12.4, 9.2 Hz), 5.14 (2dd, J = 12.8, 6.4 Hz), 4.77 (1H, t, J = 7.6 Hz), 4.71 (1H, dd, J = 8.8, 6.4 Hz), 4.21 (2H, q, J = 7.2 Hz, OCH_2CH_3), $4.18 (2H, q, J = 7.2 Hz, <math>OCH_2CH_3$), $3.76 (6H, s, 2 \times OCH_3)$, 3.36 (1H, t, J = 4.4 Hz), 3.33 (1H, t, J = 4.4 Hz), 2.61-2.51 (1H, m), 2.46 (1H, dd, J =12.0, 5.6 Hz), 2.41 (1H, t, J = 4.8 Hz), 2.35 (1H, t, J = 4.0 Hz), 2.32-2.17 (4H, m), 2.15 (3H, s, olefinic- CH_3), 2.11 (3H, s, olefinic- CH_3), 1.29 (3H, t, J = 7.2 Hz, OCH_2CH_3), 1.24 (3H, t, J = 7.2 Hz, OCH₂CH₃); ¹³C NMR (CDCl₃, DEPT-135, 1:1.7 ratio of diastereomers) δ 197.5 (C, C=O), 197.2 (C, C=O), 171.4 (C, O-C=O), 171.3 (C, O-C=O) C=O), 158.5 (C), 158.4 (C), 154.8 (C), 154.4 (C), 135.4 (C), 135.0 (C), 130.4 (C), 130.1 (C), 128.6 (2 x CH), 128.2 (2 x CH), 113.9 (4 x CH), 77.8 (2 x CH₂), 61.5 (CH₂, OCH₂CH₃), 61.3 (CH₂, OCH₂CH₃), 55.1 (2 x CH₃, OCH₃), 48.3 (CH), 48.2 (CH), 41.5 (CH), 41.3 (CH), 35.3 (CH₂), 35.2 (CH₂), 25.2 (CH₂), 25.0 (CH₂), 21.1 (CH₃, olefinic-CH₃), 21.0 (CH₃, olefinic-CH₃), 14.0 (2 x CH₃, OCH₂CH₃); LRMS m/z $360.20 \text{ (M} - \text{H}^+\text{)}$, calcd $C_{19}H_{23}NO_6$ 361.1525; Anal. calcd for $C_{19}H_{23}NO_6$ (361.15): C, 63.15; H, 6.41; N, 3.88. Found: C, 63.21; H, 6.45; N, 3.84%.

(S,S)-2-Methyl-3-[2-nitro-1-(4-nitrophenyl)ethyl]-4-oxo-cyclohex-2-enecarboxylic

NO₂
NO₂
NO₂
NO₂
Me
CO₂Et

133ai/134ai
dr = 1:1.7

acid ethyl ester (133ai) and (S,R)-2-Methyl-3-[2-nitro-1-(4-nitrophenyl)ethyl]-4-oxo-cyclohex-2-enecarboxylic acid ethyl ester (134ai): Prepared following the procedure 6b and purified by column chromatography using EtOAc/hexane and isolated as a yellow liquid. The ee was determined by chiral-phase HPLC using a Daicel Chiralpak AD-H column (hexane/i-PrOH = 90:10, flow rate 1.0 mL/min, λ = 254

nm), $t_R = 19.2 \text{ min (major)}$, $t_R = 25.2 \text{ min (major)}$, $t_R = 41.1 \text{ min (minor)}$, $t_R = 66.9 \text{ min}$ (minor); $[\alpha]^{25}_{D} = -48.7$ (c 0.3, CHCl₃, 42% ee, 27% ee); IR (neat): v_{max} 2958, 1727 (O-C=O), 1667 (C=O), 1552, 1520, 1346, 1160, 853 and 670 cm⁻¹; ¹H NMR (CDCl₃, 1:1.7 ratio of diastereomers) δ 8.15 (2H, d, J = 8.4 Hz), 8.14 (2H, d, J = 8.4 Hz), 7.44 (2H, d, J = 8.8 Hz), 7.42 (2H, d, J = 8.4 Hz), 5.30 (1H, dd, J = 13.2, 5.6 Hz), 5.20 (2H, d, J = 8.8 Hz), 7.42 (2H, d, J = 8.4 Hz), 5.30 (1H, dd, J = 13.2, 5.6 Hz), 5.20 (2H, d, J = 8.8 Hz), 7.42 (2H, d, J = 8.4 Hz), 5.30 (1H, dd, J = 13.2, 5.6 Hz), 5.20 (2H, d, J = 8.8 Hz), 7.42 (2H, d, J = 8.4 Hz), 5.30 (1H, dd, J = 13.2, 5.6 Hz), 5.20 (2H, d, J = 8.8 Hz), 7.42 (2H, d, J = 8.4 Hz), 7.42 (2H, d, J = 8.4 Hz), 5.30 (1H, dd, J = 13.2, 5.6 Hz), 5.20 (2H, d, J = 8.4 Hz), 7.42 (2H, d, J = 8.4 Hz), 7.42dd, J = 6.8, 3.6 Hz), 5.08 (1H, dd, J = 13.2, 7.2 Hz), 4.96 (1H, t, J = 7.2 Hz), 4.88 (1H, t, J = 7.2 Hz), 4.24 (2H, q, J = 7.2 Hz, OC H_2 CH₃), 4.22 (2H, q, J = 7.2 Hz, OC H_2 CH₃), 3.44 (1H, t, J = 4.4 Hz), 3.41 (1H, t, J = 4.8 Hz), 2.63-2.53 (1H, m), 2.43 (2H, dd, J =11.2, 5.6 Hz), 2.36 (1H, t, J = 4.8 Hz), 2.32-2.26 (3H, m), 2.22 (3H, s, olefinic-C H_3), 2.19 (3H, s, olefinic- CH_3), 2.17-2.12 (1H, m), 1.31 (3H, t, J = 7.2 Hz, OCH_2CH_3), 1.28 (3H, t, J = 7.2 Hz, OCH₂CH₃); ¹³C NMR (CDCl₃, DEPT-135, 1:1.7 ratio of diastereomers) δ 197.1 (C, C=O), 196.7 (C, C=O), 171.1 (C, O-C=O), 170.9 (C, O-C=O) C=O), 156.5 (C), 156.1 (C), 146.8 (C), 146.7 (C), 146.0 (C), 145.9 (C), 134.6 (C), 134.3 (C), 128.3 (2 x CH), 127.8 (2 x CH), 123.7 (4 x CH), 76.8 (CH₂), 76.6 (CH₂), 61.7 (CH₂, OCH₂CH₃), 61.6 (CH₂, OCH₂CH₃), 48.24 (CH), 48.16 (CH), 41.6 (CH), 41.5 (CH₂), 35.0 (CH₂), 34.9 (CH₂), 25.1 (CH₂), 25.0 (CH₂), 21.4 (CH₃, olefinic-CH₃), 21.3 (CH₃, olefinic-CH₃), 14.1 (CH₃, OCH₂CH₃), 14.0 (CH₃, OCH₂CH₃); LRMS m/z $377.25 \text{ (M} + \text{H}^{+})$, calcd $C_{18}H_{20}N_2O_7$ 376.1271; Anal. calcd for $C_{18}H_{20}N_2O_7$ (376.12): C, 57.44; H, 5.36; N, 7.44. Found: C, 57.36; H, 5.39; N, 7.56%.

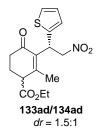
$(S,S) - 3 - (1 - Furan - 2 - yl - 2 - mitroethyl) - 2 - methyl - 4 - oxo - cyclohex - 2 - ene carboxylic \\ acid$

NO₂
Me
CO₂Et
133aj/134aj

ethyl ester (133aj) and (*S*,*R*)-3-(1-Furan-2-yl-2-nitroethyl)-2-methyl-4-oxo-cyclohex-2-enecarboxylic acid ethyl ester (134aj): Prepared following the procedure **6b** and purified by column chromatography using EtOAc/hexane and isolated as a yellow liquid. The ee was determined by chiral-phase HPLC using a Daicel Chiralpak

AS-H column (hexane/i-PrOH = 90:10, flow rate 1.0 mL/min, λ = 254 nm), t_R = 11.4 min (major), $t_R = 12.3$ min (major), $t_R = 14.1$ min (minor), $t_R = 15.5$ min (minor); $[\alpha]^{25}_{D} = +10.7$ (c 0.3, CHCl₃, 47% ee, 6% ee); IR (neat): v_{max} 2983, 1728 (O-C=O), 1670 (C=O), 1553, 1375, 1256, 1189, 1020 and 801 cm⁻¹; ¹H NMR (CDCl₂, 1.1:1 ratio of diastereomers) δ 7.29 (2H, m), 6.29 (2H, m), 6.08 (2H, m), 5.19-5.11 (3H, m), 5.02-4.98 (2H, m), 4.92 (1H, t, J = 7.2 Hz), 4.22 (4H, q, J = 7.2 Hz, 2 x OC H_2 CH₃), 3.38 $(2H, t, J = 4.8 Hz, 2 \times CH), 2.60-2.49 (2H, m), 2.44-2.34 (2H, m), 2.33-2.19 (4H, m),$ 2.16 (3H, s, olefinic- CH_3), 2.12 (3H, s, olefinic- CH_3), 1.29 (3H, t, J = 7.2 Hz, OCH_2CH_3), 1.28 (3H, t, J = 7.2 Hz, OCH_2CH_3); ¹³C NMR (CDCl₃, DEPT-135, 1.1:1 ratio of diastereomers) δ 196.5 (C, C=O), 196.3 (C, C=O), 171.3 (C, O-C=O), 171.2 (C, O-C=O), 155.9 (C), 155.6 (C), 151.4 (C), 151.1 (C), 141.5 (CH), 141.4 (CH), 132.9 (C), 132.7 (C), 110.6 (2 x CH), 106.38 (CH), 106.37 (CH), 75.9 (CH₂), 75.8 (CH₂), 61.6 (CH₂, OCH₂CH₃), 61.5 (CH₂, OCH₂CH₃), 48.5 (CH), 48.3 (CH), 36.6 (CH), 36.3 (CH), 35.1 (CH₂), 34.9 (CH₂), 25.2 (2 x CH₂), 21.1 (CH₃, olefinic-CH₃), 20.8 (CH₃, olefinic-CH₃), 14.1 (CH₃, OCH₂CH₃), 14.0 (CH₃, OCH₂CH₃); LRMS m/z 322.15 (M $+ H^{+}$), calcd $C_{16}H_{19}NO_{6}$ 321.1212; Anal. calcd for $C_{16}H_{19}NO_{6}$ (321.12): C, 59.81; H, 5.96; N, 4.36. Found: C, 59.88; H, 5.93; N, 4.42%.

(S,S)-2-Methyl-3-(2-nitro-1-thiophen-2-vl-ethyl)-4-oxo-cyclohex-2-enecarboxylic



acid ethyl ester (133ad) and (*S*,*R*)-2-Methyl-3-(2-nitro-1-thiophen-2-yl-ethyl)-4-oxo-cyclohex-2-enecarboxylic acid ethyl ester (134ad): Prepared following the procedure **6b** and purified by column chromatography using EtOAc/hexane and isolated as a yellow liquid.

The ee was determined by chiral-phase HPLC using a Daicel Chiralpak AD-H column (hexane/i-PrOH = 98:2, flow rate 1.0 mL/min, λ = 254 nm), t_R = 22.8 min (major), t_R = 24.9 min (major), $t_R = 26.7$ min (minor), $t_R = 43.0$ min (minor); $[\alpha]^{25}_D = +8.7$ (c 0.3, **CHCl₃, 41% ee, 12% ee)**; IR (neat): v_{max} 2979, 1727 (O-C=O), 1667 (C=O), 1551, 1375, 1299, 1196, 1161 and 1039 cm⁻¹; ¹H NMR (CDCl₃, 1.5:1 ratio of diastereomers) δ 7.16 (1H, t, J = 1.6 Hz), 7.15 (1H, t, J = 2.0 Hz), 6.92 (2H, t, J = 2.0 Hz), 6.91-6.89 (2H, m), 5.28 (1H, dd, J = 12.4, 8.4 Hz), 5.16-5.13 (2H, m), 5.08 (2H, dd, J = 8.0, 2.0)Hz), 5.01 (1H, dd, J = 14.0, 4.8 Hz), 4.22 (2H, q, J = 6.8 Hz, OC H_2 CH₃), 4.19 (2H, q, J $= 7.2 \text{ Hz}, \text{ OC}H_2\text{CH}_3), 3.39-3.34 \text{ (2H, m)}, 2.63-2.49 \text{ (2H, m)}, 2.44-2.35 \text{ (2H, m)}, 2.32-$ 2.20 (4H, m), 2.18 (3H, s, olefinic- CH_3), 2.15 (3H, s, olefinic- CH_3), 1.29 (3H, t, J = 7.2Hz, OCH_2CH_3), 1.24 (3H, t, J = 7.2 Hz, OCH_2CH_3); ¹³C NMR (CDCl₃, DEPT-135, 1.5:1 ratio of diastereomers) δ 197.0 (C, C=O), 196.7 (C, C=O), 171.2 (2 x C, O-C=O), 155.3 (C), 155.0 (C), 140.9 (C), 140.7 (C), 134.8 (C), 134.5 (C), 126.8 (2 x CH), 125.4 (CH), 125.0 (CH), 124.7 (CH), 124.5 (CH), 78.1 (CH₂), 77.9 (CH₂), 61.6 (CH₂, OCH₂CH₃), 61.5 (CH₂, OCH₂CH₃), 48.4 (CH), 48.2 (CH), 38.3 (CH), 38.2 (CH), 35.2 (CH₂), 35.0 (CH₂), 25.08 (CH₂), 25.06 (CH₂), 21.0 (CH₃, olefinic-CH₃), 20.8 (CH₃, olefinic-CH₃), 14.03 (CH₃, OCH₂CH₃), 14.01 (CH₃, OCH₂CH₃); LRMS m/z 338.00 $(M + H^{+})$, calcd $C_{16}H_{19}NO_{5}S$ 337.0984; Anal. calcd for $C_{16}H_{19}NO_{5}S$ (337.09): C, 56.96; H, 5.68; N, 4.15. Found: C, 56.88; H, 5.65; N, 4.19%.

7: Decarboxylation procedures:

7a: Synthesis of racemic 3-methyl-2-(2-nitro-1-phenylethyl)-cyclohex-2-enone (\pm) -142aa:

First step: In a 10 mL RB equipped with a magnetic stirring bar, to 0.3 mmol of 2-methyl-3-(2-nitro-1-phenylethyl)-4-oxo-cyclohex-2-enecarboxylic acid ethyl ester (\pm)-133aa/134aa was added mixture of MeOH (0.5 mL), H₂O (0.5 mL), THF (2 mL) as solvent and then LiOH.H₂O (1.8 mmol, 6 equiv.) was added and the reaction mixture was stirred for 3 hours at RT as indicated in Scheme 22. The solvent was removed and washed with DCM (2 x 10 mL). The pH of the reaction mixture was adjusted to 2 with

2 mL of 1N aq. HCl furnished the ketoacid (\pm)-140aa/141aa which extracted with dichloromethane (3 x 20 mL). The combined organic layers were dried with Na₂SO₄ and crude product proceeded for next step.

Second step: Ketoacid (\pm)-**140aa/141aa** (0.1 mmol) was dissolved in a mixture of concentrated HCl (0.1 mL) in THF (1.25 mL) and heated for 3h at 90 °C in a sealed glass tube. The reaction mixture was concentrated and residue was partitioned between water and CH₂Cl₂. The combined organic layers were dried with Na₂SO₄ and concentrated. The residue was chromatographed to yield 3-methyl-2-(2-nitro-1-phenylethyl)-cyclohex-2-enone (\pm)-**142aa** as yellow oil.

7b: Synthesis of 3-methyl-2-(2-nitro-1-phenylethyl)-cyclohex-2-enone (R)-142aa:

First step: In a 10 mL RB equipped with a magnetic stirring bar, to 0.3 mmol of 2-methyl-3-(2-nitro-1-phenylethyl)-4-oxo-cyclohex-2-enecarboxylic acid ethyl ester (+)-**133aa/134aa** was added mixture of MeOH (0.5 mL), H₂O (0.5 mL), THF (2 mL) as a solvent and then LiOH.H₂O (1.8 mmol, 6 equiv.) was added and the reaction mixture was stirred for 0.75 h at RT as indicated in Scheme 22. The solvent was removed and washed with DCM (2 x 10 mL). The pH of the reaction mixture was adjusted to 2 with 2 mL of 1N aq. HCl furnished the ketoacid (+)-**140aa/141aa** which extracted with dichloromethane (3 x 20 mL). The combined organic layers were dried with Na₂SO₄ and crude product proceeded for next step.

Second step: Ketoacid (+)-**140aa/141aa** (0.2 mmol) was dissolved in a mixture of concentrated HCl (0.2 mL) in THF (2.5 mL) and heated for 3 h at 90 °C in a sealed glass tube. The mixture was concentrated and residue was partitioned between water and CH₂Cl₂. The combined organic layers were dried (Na₂SO₄) and concentrated. The residue was chromatographed to yield 3-methyl-2-(2-nitro-1-phenylethyl)-cyclohex-2-enone (*R*)-**142aa** as yellow oil.

7c: Synthesis of 3-methyl-2-(2-nitro-1-phenylethyl)-cyclohex-2-enone (S)-142aa: First step: In a 10 mL RB equipped with a magnetic stirring bar, to 0.33 mmol of 2methyl-3-(2-nitro-1-phenylethyl)-4-oxo-cyclohex-2-enecarboxylic acid tert-butyl ester 133ba/134ba was added 5 mL dry DCM as a solvent and then CH₃SO₃H (0.4 mL) was added dropwisely for 10 minits and the reaction mixture was stirred for 0.5 h at RT as indicated in Scheme 22. The mixture was partitioned between water and CH₂Cl₂. The combined organic layers were dried (Na₂SO₄) and concentrated and the crude product was proceded for next step.

Second step: Ketoacid 140aa/141aa (0.16 mmol) was dissolved in a mixture of concentrated HCl (0.5 mL) in THF (2 mL) and heated for 3 h at 90 °C in a sealed glass tube. The mixture was concentrated and residue was partitioned between water and CH₂Cl₂. The combined organic layers were dried (Na₂SO₄) and concentrated. The residue was chromatographed to yield 3-methyl-2-(2-nitro-1-phenylethyl)-cyclohex-2enone (S)-142aa as yellow oil.

(1S)-3-Methyl-2-(2-nitro-1-phenylethyl)-cyclohex-2-enone (142aa): Prepared

(S)-142aa

using EtOAc/hexane and isolated as a yellow liquid. The ee was determined by chiral-phase HPLC using a Daicel Chiralcel OD-H column (hexane/i-PrOH = 90:10, flow rate 1.0 mL/min, λ = 254 nm), $t_{\rm R} = 12.2 \text{ min (major)}, t_{\rm R} = 15.5 \text{ min (minor)}; [\alpha]^{25}_{\rm D} = -15.5 (c 0.2, \text{CHCl}_3, 44\% \text{ ee});$ IR (neat): v_{max} 2926, 2857, 1663 (C=O), 1614, 1551, 1452, 1377, 1173 and 1055 cm⁻¹; ¹H NMR (CDCl₃) δ 7.31-7.22 (5H, m, Ar-H), 5.28 (1H, dd, J = 12.4, 8.8 Hz), 5.09 (1H, dd, J = 12.4, 6.4 Hz), 4.76 (1H, t, J = 6.4 Hz), 2.46-2.30 (4H, m), 2.11 (3H, s, olefinic- CH_3), 1.92 (2H, t, J = 6.0 Hz); ¹³C NMR (CDCl₃, DEPT-135) δ 198.7 (C, C=O), 160.3 (C), 139.0 (C), 133.4 (C), 128.7 (2 x CH), 127.6 (2 x CH), 127.1 (CH), 77.9 (CH₂), 42.5 (CH), 38.7 (CH₂), 33.8 (CH₂), 21.8 (CH₂), 21.7 (CH₃, olefinic-CH₃); LRMS m/z 260.10 (M + H $^{+}$), calcd C₁₅H₁₇NO₃ 259.1208; Anal. calcd for C₁₅H₁₇NO₃ (259.12): C,

following the procedure 7c and purified by column chromatography

69.48; H, 6.61; N, 5.40. Found: C, 69.35; H, 6.39; N, 5.52%.

(1R)-3-Methyl-2-(2-nitro-1-phenylethyl)-cyclohex-2-enone (142aa): Prepared

NO₂

following the procedure **7b** and purified by column chromatography using EtOAc/hexane and isolated as a yellow liquid. The ee was determined by chiral-phase HPLC using a Daicel Chiralcel OD-H column (hexane/i-PrOH = 90:10, flow rate 1.0 mL/min, λ = 254 nm),

(R)-142aa column (hexane/*i*-PrOH = 90:10, flow rate 1.0 mL/min, λ = 254 nm), $t_{\rm R}$ = 11.6 min (minor), $t_{\rm R}$ = 14.2 min (major); $[\alpha]^{25}_{\rm D}$ = +10.7 (c 0.3, CHCl₃, 33% ee); IR (neat): $v_{\rm max}$ 2926, 2857, 1663 (C=O), 1614, 1551, 1452, 1377, 1173 and 1055 cm⁻¹.